
Supplementary Materials for

*Enhancement of Peroxydisulfate Activation for Complete
Degradation of Refractory Tetracycline by 3D
Self-Supported MoS₂/MXene Nanocomplex*

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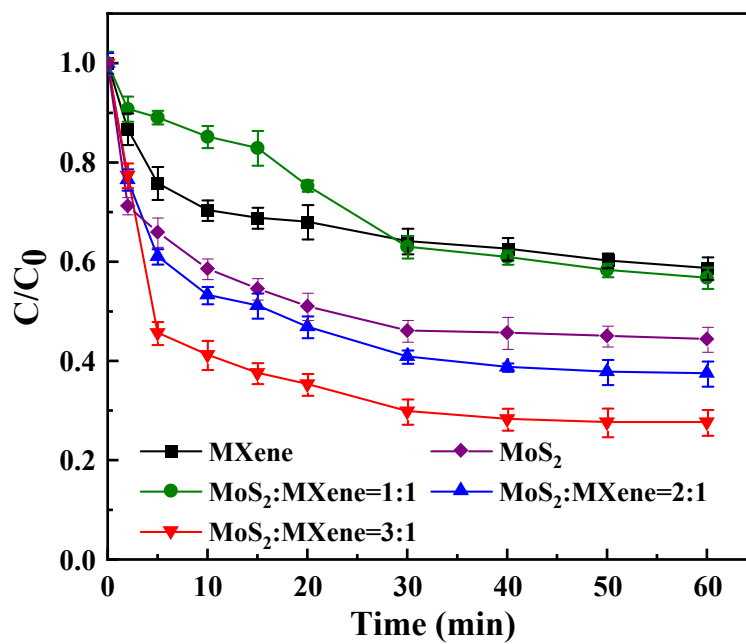


Figure S1. Effect of different materials on the degradation efficiency of TC.

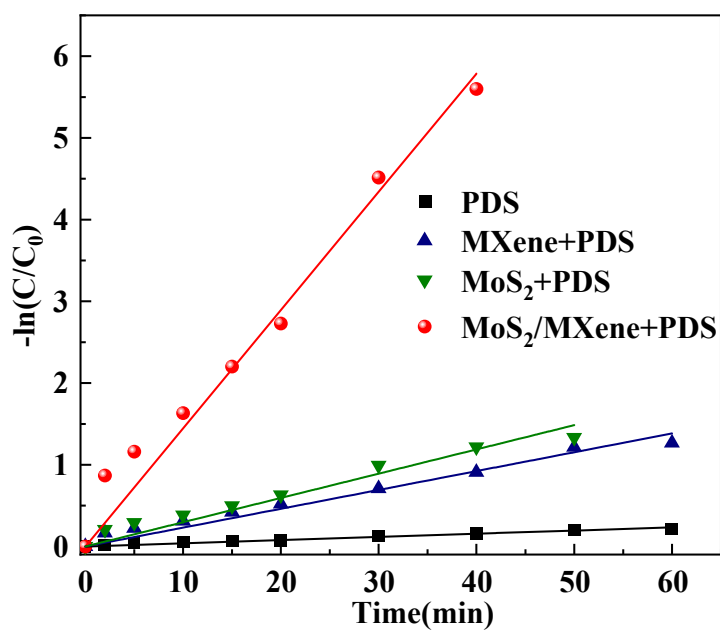


Figure S2. Kinetic fitting models for different materials/systems.

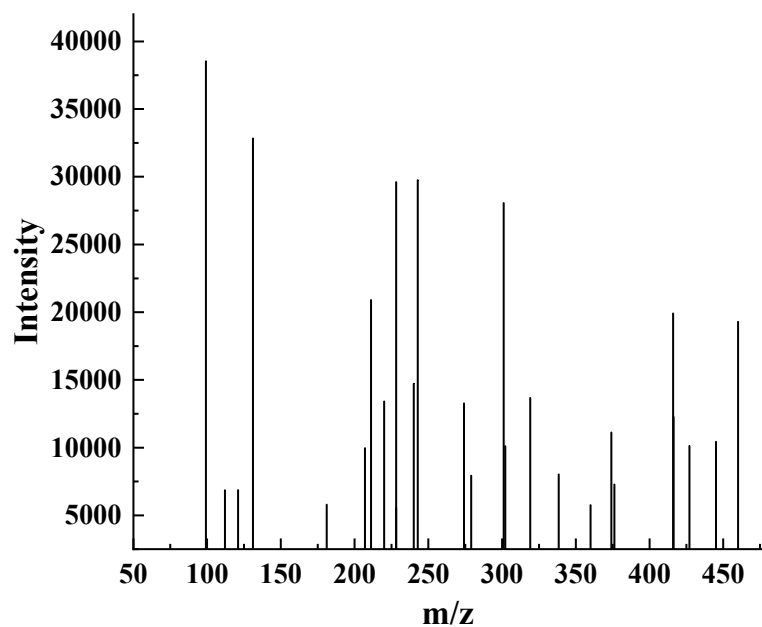


Figure S3. MS full scan spectra of intermediates in TC degradation in the MoS₂/MXene/PDS system by LC-MS.

Table S1. The elemental weight percentages and atomic percentages of MoS₂/MXene were calculated by EDX spectroscopy.

Element	Weight %	Atomic %
C	8.42	26.14
O	5.68	13.23
Mo	47.79	18.58
S	32.18	37.43
Ti	5.94	4.62

Table S2. The fitting parameters of kinetic models of TC degradation by MoS₂/MXene/PDS.

Materials/Systems	Equations	$k(\text{min}^{-1})$	R^2
PDS	$y=0.0039x$	0.0039	0.992
MXene/PDS	$y=0.0231x$	0.0231	0.988
MoS ₂ /PDS	$y=0.0297x$	0.0297	0.983
MoS ₂ /MXene/PDS	$y=0.1447x$	0.1447	0.991

Table S3. The reaction rate constants of radicals with the corresponding scavenger.

Free Radical Species	Scavenger	Rate Constant (k) ($\text{M}^{-1} \text{s}^{-1}$)
$\text{SO}_4^{\bullet-}/\bullet\text{OH}$	Ethanol (EtOH)	$(1.6\text{-}7.7) \times 10^7 / (1.2\text{-}2.8) \times 10^9$
	Methanol (MeOH)	$(1.6\text{-}7.7) \times 10^7 / 9.7 \times 10^8$
	Tert-butanol (TBA)	$(4.0\text{-}9.1) \times 10^5 / (3.8\text{-}7.6) \times 10^8$
$\text{O}_2^{\bullet-}$	Benzoquinone(p-BQ)	3.0×10^{10}

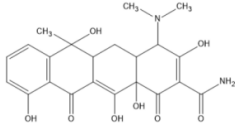
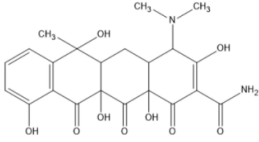
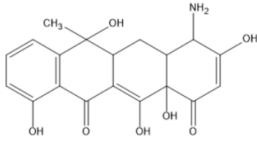
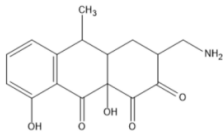
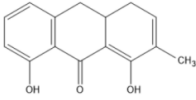
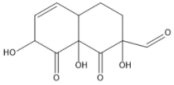
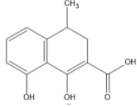
Table S4. The Elemental peak positions and contents of fresh and used MoS₂/MXene

Elements	Fresh		Used	
	Peak Positions	Contents (%)	Peak Positions	Contents (%)
C 1s	285.1eV	66.62	284.8 eV	56.32
O 1s	532.2 eV	15.77	530.9 eV	23.20
Mo 3d	228.4 eV	5.31	228.4 eV	6.85
S 2p	161.8 eV	9.29	161.9 eV	9.13
Ti 2p	456.1 eV	3.01	458.7 eV	4.49

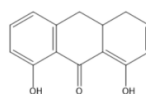
Table S5. The group percentages of fresh and used MoS₂/MXene.

Samples	C (%)			O (%)			Mo (%)		S (%)	
	C-C	C-O	Ti-C	C-O	Ti-O	C=O	Mo ⁴⁺	Mo ⁶⁺	S ²⁻	S ₂ ²⁻
Fresh	32.57	5.45	1.59	16.52	5.61	5.93	5.93	0.07	15.37	0
Used	41.41	11.21	0	9.88	8.57	3.99	4.54	0.86	12.63	0.67

Table S6. Liquid-phase mass spectrometry (LC-MS) analysis of intermediates and possible structural formulae.

Number	m/z	Supposed Structure
TC	446	
P11	460	
P12	374	
P13	302	
P14	242	
P15	240	
P16	220	

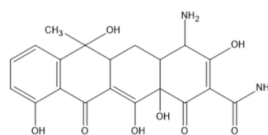
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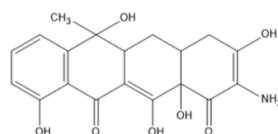
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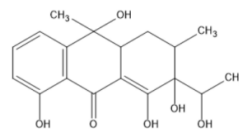
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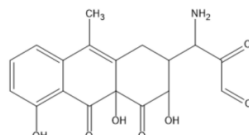
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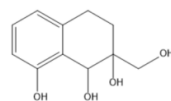
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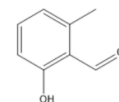
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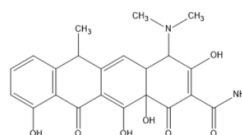
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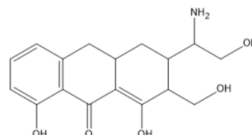
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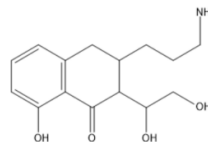
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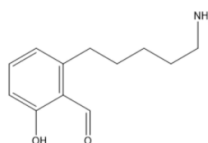
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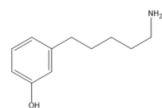
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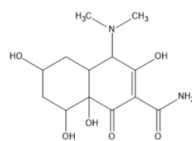
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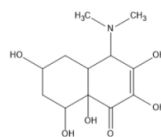
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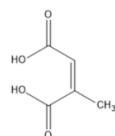
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P37 274



P38 131



P39 112

