

SUPPORTING INFORMATION

Article

Mass Spectrometric Analysis of Purine Intermediary Metabolism Indicates Cyanide Induces Purine Catabolism in Rabbits

Running title: In vivo effects of cyanide on purine metabolism

Jordan Morningstar ¹, Jangwoen Lee ², Sari Mahon ², Matthew Brenner ^{2,3} and Anjali K. Nath ^{1,4,*}

¹ Division of Cardiovascular Medicine, Beth Israel Deaconess Medical Center, Boston, MA 02215, USA; morningj@msc.edu

² Beckman Laser Institute, University of California, Irvine, CA 92697, USA; jangwl@hs.uci.edu (J.L.); mahonsb@hs.uci.edu (S.M.); mbrenner@uci.edu (M.B.)

³ Division of Pulmonary and Critical Care Medicine, Department of Medicine, University of California, Irvine, CA 92697, USA; mbrenner@uci.edu (M.B.)

⁴ Harvard Medical School, Boston, MA 02215, USA

* Correspondence: anath1@bidmc.harvard.edu

Keywords: mass spectrometry (MS); purine; nucleoside/nucleotide metabolism; metabolic regulation; cytochrome c oxidase (Complex IV); animal model; allopurinol

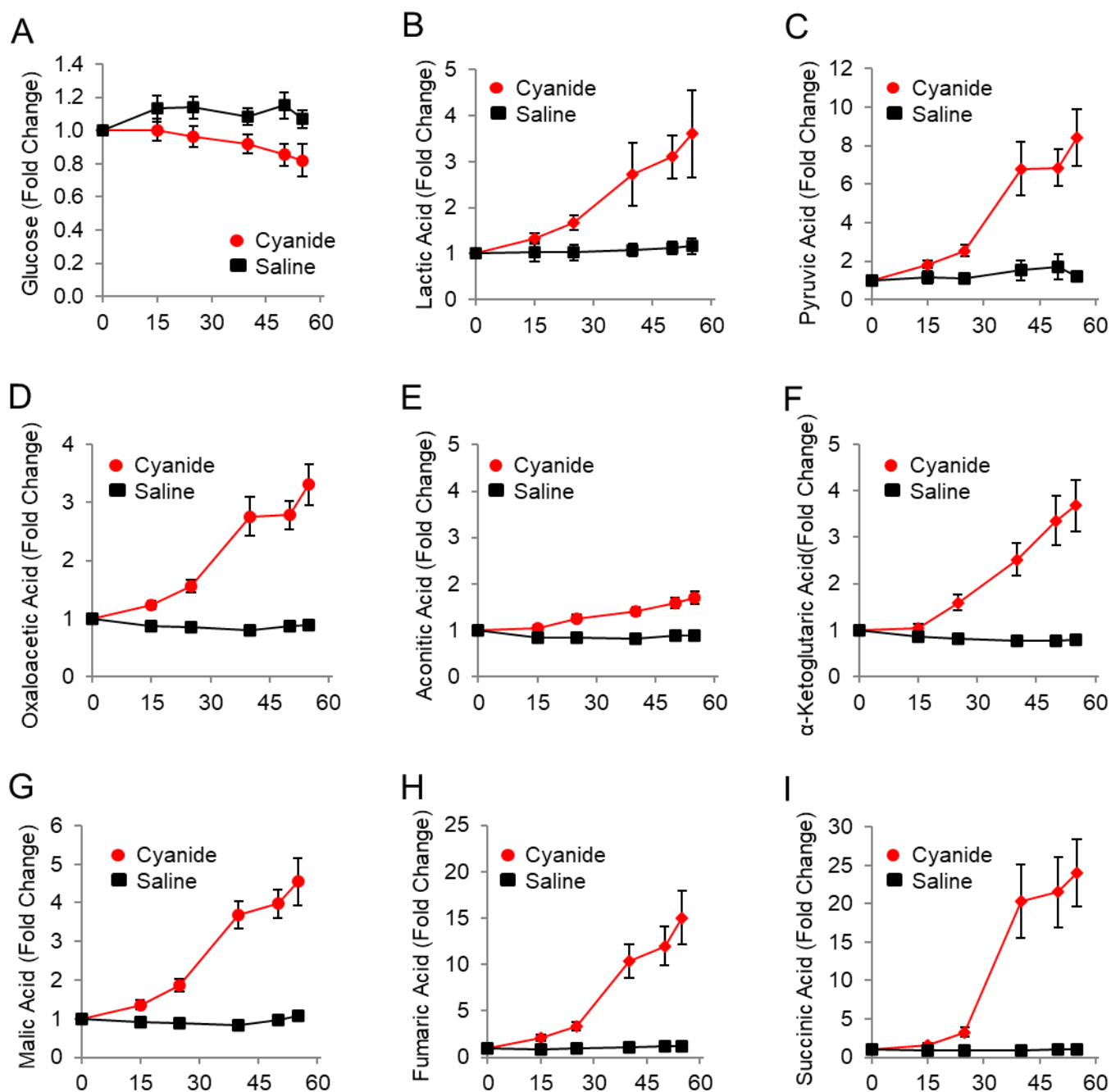


Figure S1. Plasma levels of glycolytic and TCA cycle metabolites increase in rabbits exposed to a lethal dose of cyanide. Metabolites were measured in the plasma of cyanide-treated animals ($n = 15$) and sham control animals ($n = 5$): (A) glucose, (B) lactic acid, (C) pyruvic acid, (D) oxaloacetic acid, (E) aconitic acid, (F) α -ketoglutaric acid, (G) malic acid, (H) fumaric acid, and (I) succinic acid. See Table 1 in the manuscript for P values and q values.

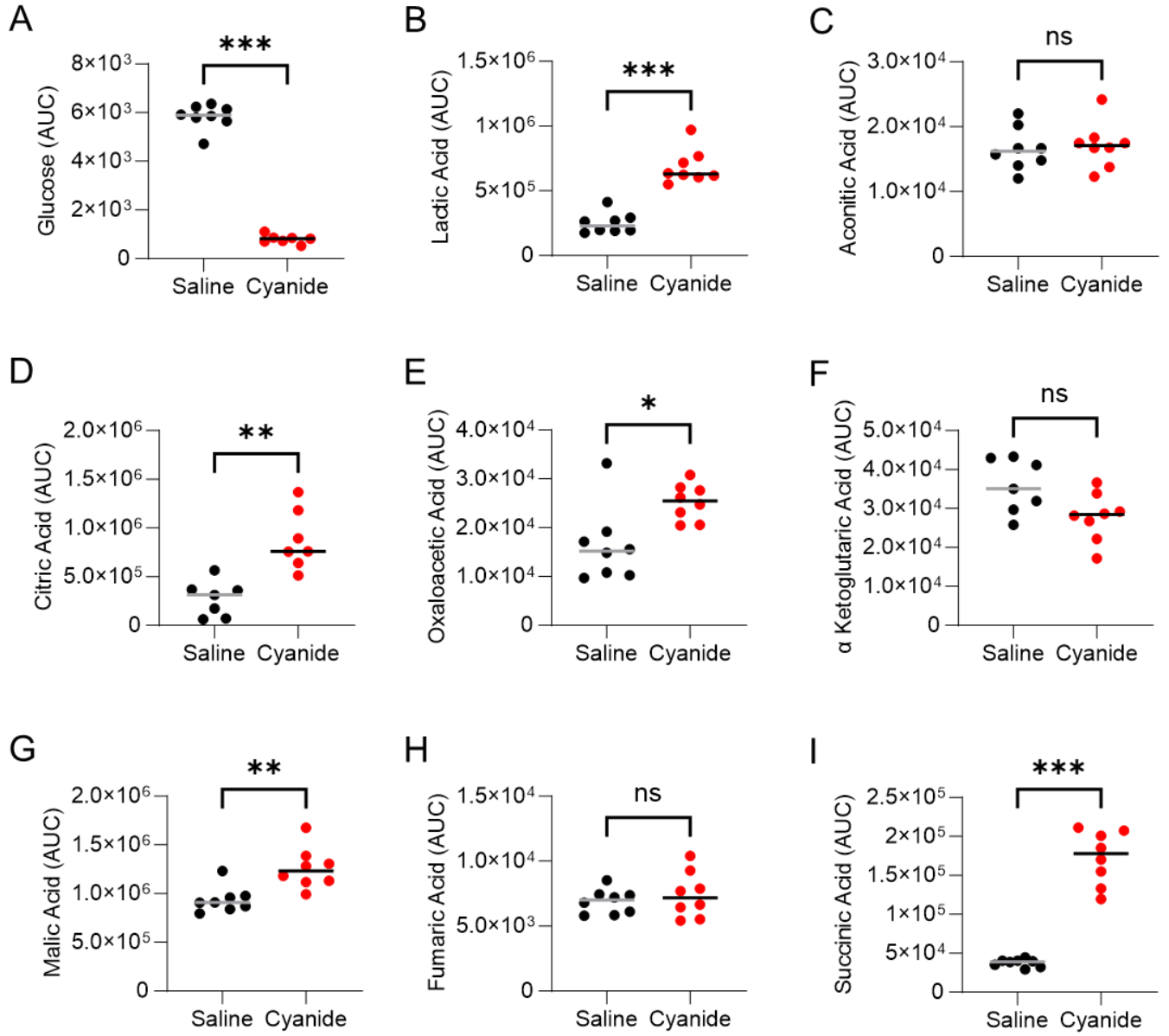


Figure S2. Glycolytic and TCA cycle metabolites increase in zebrafish larvae exposed to a lethal dose of cyanide. Metabolites were measured in lysates ($n = 8$) from whole zebrafish larvae (6 d.p.f.) exposed to saline or 25 μ M potassium cyanide for 2 hours. Each data point represents a group of 10 larvae. (A) Glucose ($P = 0.00031$; $q = 0.00032$), (B) lactic acid ($P = 0.00015$; $q = 0.00024$), (C) aconitic acid (n.s.), (D) citric acid ($P = 0.00116$; $q = 0.00091$), (E) oxaloacetic acid ($P = 0.01041$; $q = 0.00546$), (F) α -ketoglutaric acid (n.s.), (G) malic acid ($P = 0.00186$; $q = 0.00117$), (H) fumaric acid (n.s.), and (I) succinic acid ($P = 0.00015$; $q = 0.00024$). ns = $P > 0.05$; * = $P \leq 0.05$; ** = $P \leq 0.01$, *** = $P \leq 0.001$.

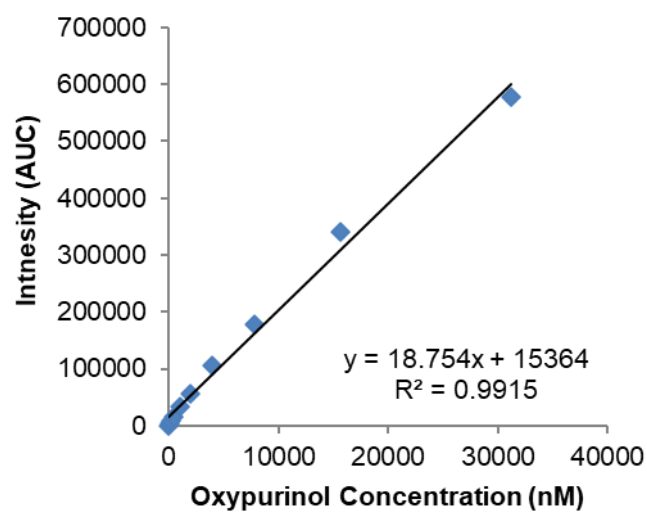


Figure S3. Oxypurinol quantification by mass spectrometry. To generate the oxypurinol standard curve, oxypurinol was spiked into pooled reference plasma, serially diluted, and analyzed by LC-MS/MS. The standard curve was linear from 122 nM to 31.25 μ M. The absolute concentration of oxypurinol in treated animals was calculated using this standard curve. See **Table S1** for multiple reaction monitoring (MRM) transitions.

Table S1. Multiple reaction monitoring transitions of compounds and their coefficient of variation in pooled plasma.

Ionization Mode	Q1 Precursor (m/z)	Q3 Product (m/z)	Collision Energy (eV)	Metabolite Name	Coefficient of Variation*
Negative	181.22	138.2	-6	Citrulline-d ₇	7.7
Negative	271.2	138.9	-22	Inosine- ¹⁵ N ₄	8.5
Negative	129.11	42.1	-14	Thymine-d ₄	7.2
Negative	172.19	154.1	-6	Phenylalanine-d ₈	6.5
Negative	161.1	113.1	-2	Glucose	11.6
Negative	87.05	43	-14	Pyruvic acid	8.1
Negative	89.1	43.1	-16	Lactic acid	5.9
Negative	131.1	87.1	-14	Oxaloacetate	11.3
Negative	191.1	111.1	-15	Citric acid-Isocitric acid	15.0
Negative	173.05	85	-17	Aconitic acid	19.6
Negative	145.1	101.1	-13	α-Ketoglutaric acid	9.1
Negative	117.1	73	-12	Succinic acid	9.1
Negative	115.06	71.01	-13	Fumaric acid	36.8
Negative	133.08	115	-14	Malic acid	18.5
Negative	151.02	42.1	-16	Oxypurinol	7.9
Negative	267.2	135	-27	Inosine	9.9
Negative	346.2	78.8	-50	AMP	23.8
Negative	135.1	92.1	-18	Hypoxanthine	7.6
Negative	157.05	114	-17	Allantoin	16.5
Negative	167.001	124	-17	Uric acid	6.9
Negative	151.1	108	-23	Xanthine	6.4
Negative	283.2	150.9	-24	Xanthosine	21.9

m/z: mass/charge; eV: electron volt. *Calculated from measurements in pooled plasma samples, except for oxypurinol which was calculated from oxypurinol spiked plasma samples.