

Geochemical Assessment of the Evolution of Groundwater under the Impact of Seawater Intrusion in the Mannar District of Sri Lanka

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NO. OF FIGURES: 4

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NO. OF PAGES: 9

Text 1. Calculation of water parameters

The categorization of water hardness is based on CaCO_3 equivalents, where soft water registers below 75 mg/L as CaCO_3 . Slightly hard water falls within the range of 75 to 150 mg/L as CaCO_3 , while moderately hard water encompasses concentrations between 150 and 300 mg/L as CaCO_3 . Water is classified as very hard when the values surpass 300 mg/L as CaCO_3 , according to the classification by Sawyer and McCarty in 1967 [1].

The calculation of hardness is based on the following equation in equivalent of CaCO_3 mg/L.

$$\text{Hardness [mg/L]} = 2.497 [\text{Ca, mg/L}] + 4.118 [\text{Mg, mg/L}]$$

Alkalinity is the acid-neutralizing capacity due to the presence of HCO_3^- , CO_3^{2-} and hydroxide OH^- ions of the water. The total alkalinity was determined by H_2SO_4 (0.01N) titration with methyl orange as the indicator. $[\text{HCO}_3^-]$ is calculated using the following equation.

$$[\text{HCO}_3^-] = 1.22 \times (\text{Total Alkalinity/mg/L})$$

The concentrations in terms of milliequivalents per liter (meq/L) is used to measure the chemical activity of ions in a solution, providing a standardized approach to convey the quantity of substances involved in various chemical reactions.

$$\text{meq} = \frac{\text{mg} \times \text{Valence}}{\text{Molecular weight}}$$

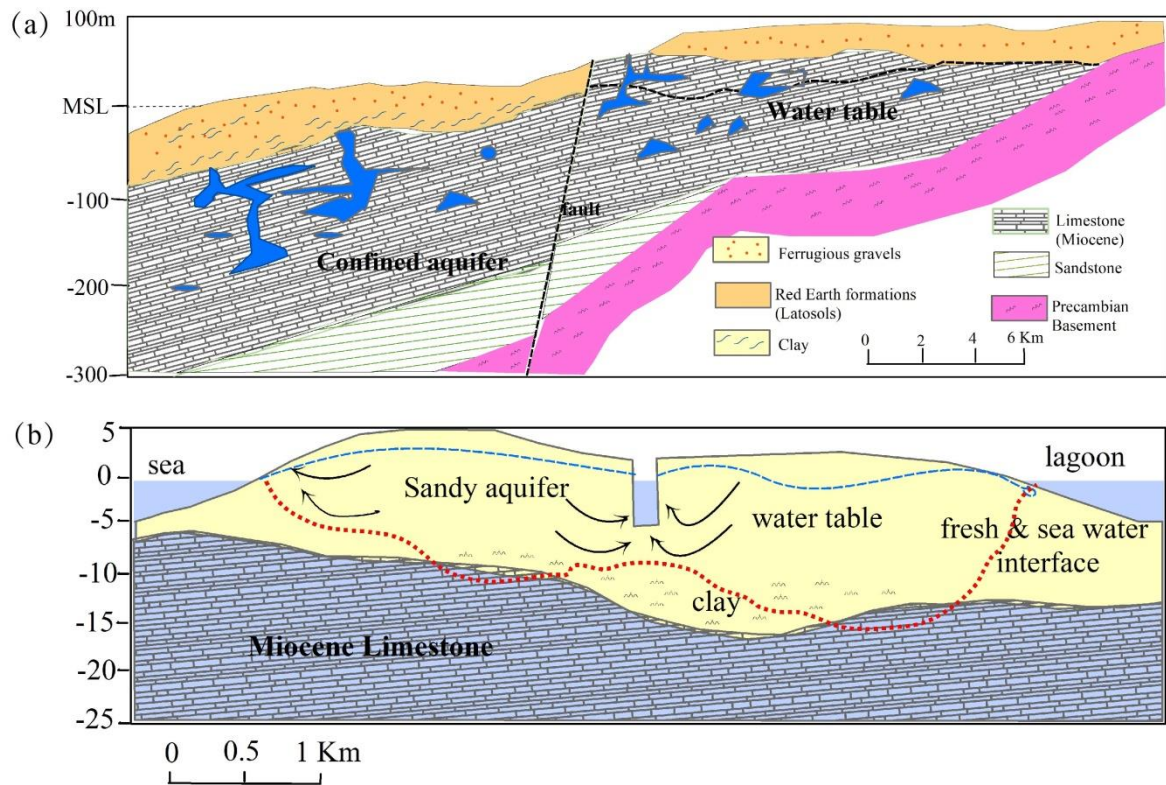


Figure S1. East-West geological schematic cross sections of (a) deep confined aquifer basin system of Mannar mainland [2] and (b) sandy aquifer systems of Mannar Island [3]

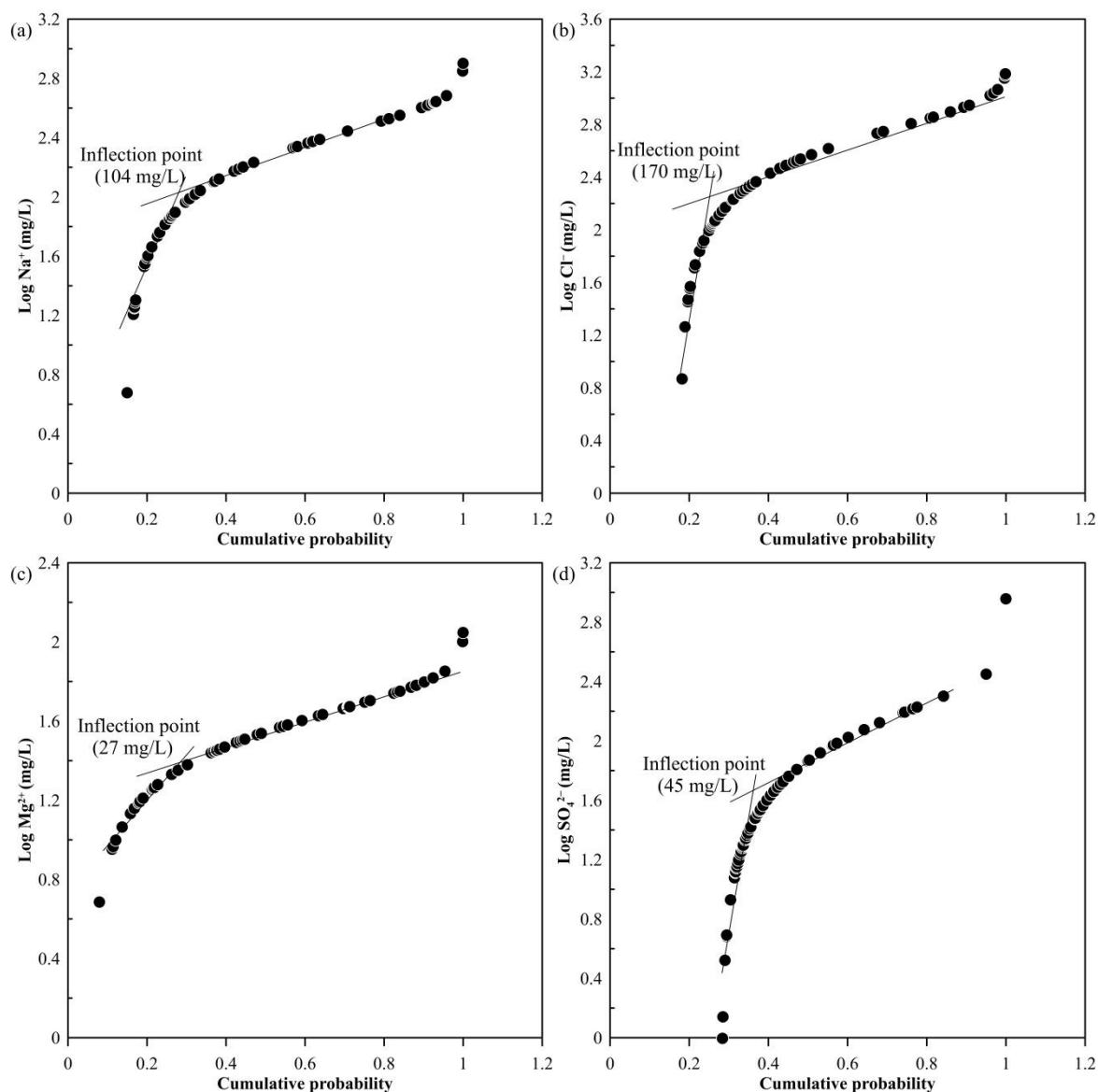


Figure S2. Cumulative probability (%) for calculating the threshold value of critical parameters (Na^+ , Mg^{2+} , Cl^- , SO_4^{2-})

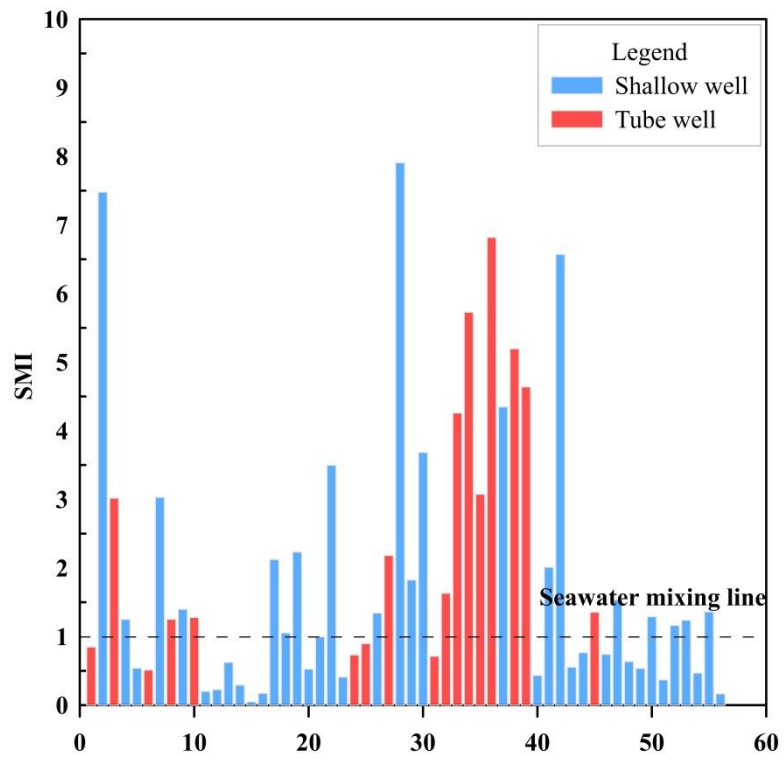


Figure S3. SMI values of groundwater in Mannar district.

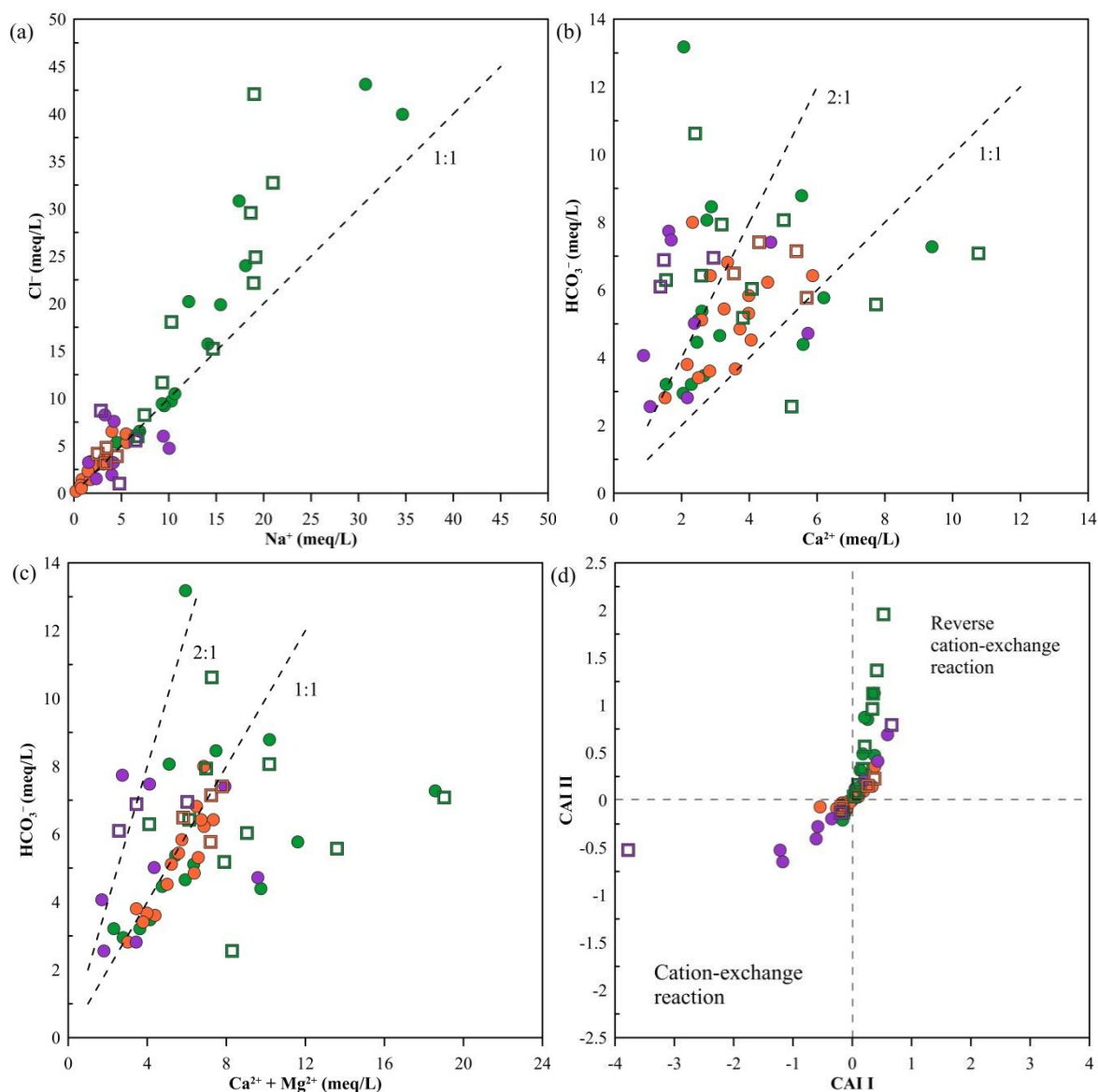


Figure S4: Hydrochemical relationships between ions. Orange, green, and purple colors represent HCO₃-Ca type, Cl-Na type, and Mixed type, respectively. Solid and hollow points correspond to shallow well and tube well groundwater, respectively. (a) Cl^- vs Na^+ . (b) HCO_3^- vs Ca^{2+} . (c) HCO_3^- vs $(\text{Ca}^{2+} + \text{Mg}^{2+})$. (d) chloro-alkaline index diagram CAI II vs CAI I.

Table S1. The statistic of groundwater geochemical compositions.

Indices	Unit	Minimum	Maximum	Average	SLS 614- 2013	WHO 2017
pH		5.89	7.97	6.82	6.5 - 8.5	6.5 - 8.5
Temperature	°C	27.3	28.2	27.6	-	-
EC	µS/cm	384	5910	1787	750	750
TDS	mg/L	192	2950	894	500	500
Na ⁺	mg/L	4.77	797.00	184.03	250	200
K ⁺	mg/L	0.29	68.01	15.24	-	-
Mg ²⁺	mg/L	4.85	111.55	35.07	-	-
Ca ²⁺	mg/L	17.54	215.23	70.72	-	-
HCO ₃ ⁻	mg/L	156	804	354	200	-
F ⁻	mg/L	0.14	2.48	0.69	1.0	1.5
Cl ⁻	mg/L	7.38	1530.69	369.54	250	250
Br ⁻	mg/L	<0.01	5.5	1.2	-	2
NO ₃ ⁻	mg/L	<0.01	85.6	8.1	50	50
SO ₄ ²⁻	mg/L	0.99	906.70	73.19	200	200
Total Hardness	mg/L	84.8	950.8	321.0	250	-
DOC	mg/L	0.59	15.16	5.82	-	-

Table S2. Relative weight of chemical parameters with threshold values.

Parameter	Threshold values (S _i)	Weight (w _i)	Relative Weight (W _i)
pH	6.5–8.5	4	0.105263
Dissolved oxygen	5	5	0.131579
Total Dissolved solids	500	4	0.105263
Alkalinity	200	2	0.052632
Electrical conductivity	750	5	0.131579
Na ⁺	200	1	0.026316
Ca ²⁺	75	2	0.052632
Mg ²⁺	50	2	0.052632
F ⁻	1.5	2	0.052632
Cl ⁻	250	3	0.078947
SO ₄ ²⁻	200	4	0.105263
NO ₃ ⁻	50	4	0.105263
		$\sum w_i=38$	$\sum W_i=1.0000$

Table S3. Sub-stages (in %) in freshening and intrusion phases

	HF _s	Sub-stage	Cation exchange evaluation	Percentages (%)	Percentages (%)
Shallow wells	NaCl	f ₁		15.4	
	NaMixCl	f ₂		5.1	
	NaMixHCO ₃ , MixNaMixHCO ₃	f ₃	freshening	5.1	74.36
	NaHCO ₃ , CaHCO ₃ , MixNaHCO ₃ , MixCaHCO ₃	f ₄ + FW		48.8	
	MixNaMixCl, MixCaMixCl	i ₂		5.1	
	MixNaCl, MixCaCl	i ₃	intrusion	5.1	25.64
	NaCl	i ₄ + SW		15.4	
Tube wells	NaCl	f ₁		23.4	
	NaMixHCO ₃	f ₃	freshening	6.0	64.7
	NaHCO ₃ , CaHCO ₃ , MixNaHCO ₃ , MixCaHCO ₃	f ₄ + FW		35.3	
	NaCl	i ₄ + SW	intrusion	35.3	35.3

References

1. Sawyer CN. *Chemistry for Sanitary Engineers*. McGraw-Hill; 1967.
<https://books.google.lk/books?id=hj9RAAAAMAAJ>
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