



Abstract Biocatalytic Preparation of Perillic Derivatives as an Alternative to Limonene Valorization ⁺

Stefania-Alexandra Voicea and Madalina Tudorache *

Faculty of Chemistry, University of Bucharest, 4-12 Regina Elisabeta Av., 030018 Bucharest, Romania; stefania.voicea1@gmail.com

* Correspondence: madalina.sandulescu@g.unibuc.ro

+ Presented at the 17th International Symposium "Priorities of Chemistry for a Sustainable Development" PRIOCHEM, Bucharest, Romania, 27–29 October 2021.

Keywords: biocatalysis; perillic derivatives; ALDH

Perillic-type derivatives of limonene from the class of the oxygenated derivatives have not been effectively utilized to date, despite the large number of applications that they can fulfill based on their phytochemical properties [1,2]. Perillyl derivatives (e.g., alcohol, aldehyde, and acid) are plant compounds designed as monoterpenes with low toxicity and prominent biological action, called phytochemicals, which are considered valuable intermediates for functional foods and novel therapies. They are increasingly important for their flavors, antimicrobial properties, and anticancer properties [3,4]. The aim of this study was the development of an alternative for the biocatalytic preparation of perillic acid from perillic aldehyde. Aldehyde dehydrogenase will catalyze the biocatalytic transformation of perillic aldehyde.

The tests were performed in 2 mL Eppendorf tubes. The sample contained 1mM perillic aldehyde, 1 mM NAD⁺, biocatalyst (aldehyde dehydrogenase such as F-ALDH, ALD-S1, and ALD-S2), and buffer, until a volume of 1000 μ L was reached in each reaction vessel. The mixture was vortexed for 10 minutes and then incubated in the thermoshaker for 24h, 1000 rpm, at a temperature of 25 °C. After reaction, the sample content was monitored using an HPLC-DAD system. Before analysis, the sample was mixed with the mobile phase (1:1, v/v). An analysis was performed in an isocratic regime using 20:20:60 acetonitrile/sulfuric acid/water as a composition of the mobile phase, 1 mL/min of flow rate, a reaction time of 30 min, and 10 μ L of injected volume, at a temperature of 60 °C of the detector.

The oxidation reaction of perillic aldehyde to perillic acid was studied. The reaction was catalyzed by aldehyde dehydrogenase in the presence of a NAD⁺ cofactor. The screening of the enzyme biocatalyst was performed initially in the presence of three different enzymes, aldehyde dehydrogenase type (F-ALDH, ALD-S1, ALD-S2). Moreover, the experimental parameters of the biocatalytic system have been optimized in order to increase the process efficiency. The performance of the system was evaluated by calculating the conversion of perillic aldehyde and selectivity to perillic acid. The biocatalysts ALD-S1 and ALD-S2 showed a similar behavior. A substrate conversion of 80% has been achieved, with a total selectivity in perillic acid. We developed a biocatalytic approach for the efficient conversion of perillic aldehyde into acid derivatives, which is a valuable alternative to limonene valorization from the biomass residues.

Author Contributions: Conceptualization, M.T.; methodology, M.T.; formal analysis, S.-A.V.; investigation, M.T.; resources, M.T.; data curation, M.T.; writing—original draft preparation, S.-A.V.; writing—review and editing, M.T.; supervision, M.T.; project administration, M.T. All authors have read and agreed to the published version of the manuscript.



Citation: Voicea, S.-A.; Tudorache, M. Biocatalytic Preparation of Perillic Derivatives as an Alternative to Limonene Valorization. *Chem. Proc.* 2022, 7, 52. https://doi.org/ 10.3390/chemproc2022007052

Published: 23 March 2022

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). **Funding:** The work of this paper was financially supported by PNCDI III PED project (contract no. 376PED/2020) from UEFISCDI, Romania.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

References

- 1. Kris-Etherton, P.M.; Hecker, K.D.; Bonanome, A.; Coval, S.M.; Binkoski, A.E.; Hilpert, K.F.; Griel, A.E.; Etherton, T.D. Bioactive compounds in foods: their role in the prevention of cardiovascular disease and cancer. *Am. J. Med.* **2002**, *113*, 71–88. [CrossRef]
- 2. Surh, Y.-J. Cancer chemoprevention with dietary phytochemicals. *Nat. Rev. Cancer* 2003, *3*, 768. [CrossRef] [PubMed]
- 3. Shojaei, S.; Kiumarsi, A.; Moghadam, A.R.; Alizadeh, J.; Marzban, H.; Ghavami, S. *Chapter Two—Perillyl Alcohol (Monoterpene Alcohol), Limonene*; Bathaie, S.Z., Tamanoi, F., Eds.; Academic Press: Cambridge, MA, USA, 2014; Volume 36, pp. 7–32.
- 4. Berman, B.; Amini, S.; Valins, W.; Block, S. Pharmacotherapy of actinic keratosis. Expert opinion on pharmacotherapy. *Expert Opin. Pharmacother.* **2009**, *10*, 3015. [CrossRef] [PubMed]