

SUPPLEMENTARY MATERIALS

Cori ester as the ligand for monovalent cations

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1. NMR spectra
2. Crystallographic data
3. Cation coordination

NMR spectra of α -D-glucose 1-phosphate complexes

The NMR spectra were recorded using a Bruker Ascend (500 MHz) spectrometer in D₂O as the solvent, with a sample molarity of 0.05 mmol/mL at room temperature, unless specified otherwise. Chemical shifts (δ) are given in ppm relative to residual H₂O (¹H) as a reference. Coupling constants (J) are in Hz. The following abbreviations of signal patterns are as follows: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), br (broad).

NMR data of α -D-Glucose 1-phosphate complexes

Dipotassium α -D-glucose 1-phosphate (Glc-1P2K)

¹H NMR (500 MHz, D₂O) δ ppm 3.28 (dd, J = 10.09, 9.14 Hz, 1 H, H₄), 3.37 (ddd, J = 9.62, 3.47, 1.73 Hz, 1 H, H₂), 3.63 (dd, J = 12.30, 5.04 Hz, 1 H, H_{6a}), 3.68 (t, J = 9.46 Hz, 1 H, H₃), 3.77 (dd, J = 12.14, 2.36 Hz, 1 H, H_{6b}), 3.82 (ddd, J = 10.17, 5.12, 2.36 Hz, 1 H, H₅), 5.34 (dd, J = 7.57, 3.47 Hz, 1 H, H₁);

¹³C NMR (126 MHz, D₂O) δ ppm 60.71 (s), 69.76 (s), 71.91 (s), 72.25 (d, J = 6.36 Hz), 73.16 (s), 93.51 (d, J = 5.45 Hz);

³¹P{¹H} NMR (202 MHz, D₂O) δ ppm 2.39;

³¹P NMR (202 MHz, D₂O) δ ppm 2.37 (d, J = 7.46 Hz).

Disodium α -D-glucose 1-phosphate (Glc-1P2Na)

¹H NMR (500 MHz, D₂O) δ ppm 3.28 (dd, J = 10.25, 9.30 Hz, 1 H, H₄), 3.36 (ddd, J = 9.77, 3.47, 1.89 Hz, 1 H, H₂), 3.63 (dd, J = 12.30, 5.04 Hz, 1 H, H_{6a}), 3.68 (t, J = 9.46 Hz, 1 H, H₃), 3.77 (dd, J = 12.30, 2.52 Hz, 1 H, H_{6b}), 3.82 (ddd, J = 10.01, 5.28, 2.36 Hz, 1 H, H₅), 5.34 (dd, J = 7.57, 3.47 Hz, 1 H, H₁);

¹³C NMR (126 MHz, D₂O) δ ppm 60.70 (s), 69.76 (s), 71.91 (s), 72.25 (d, J = 7.27 Hz), 73.16 (s), 93.51 (d, J = 5.45 Hz);

³¹P{¹H} NMR (202 MHz, D₂O) δ ppm 2.35;

³¹P NMR (202 MHz, D₂O) δ ppm 2.37 (d, J = 7.46 Hz).

Potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa)

¹H NMR (500 MHz, D₂O) δ ppm 3.29 (t, J = 9.62 Hz, 1 H, H₄), 3.37 (ddd, J = 9.77, 3.47, 1.58 Hz, 1 H, H₂), 3.64 (dd, J = 12.30, 5.36 Hz, 1 H, H_{6a}), 3.68 (t, J = 9.50 Hz, 1 H, H₃), 3.77 (dd, J = 12.30, 2.60 Hz, 1 H, H_{6b}), 3.82 (ddd, J = 10.17, 5.12, 2.36 Hz, 1 H, H₅), 5.35 (dd, J = 7.57, 3.47 Hz, 1 H, H₁);

¹³C NMR (126 MHz, D₂O) δ ppm 60.69 (s), 69.74 (s), 71.91 (s), 72.23 (d, J = 6.36 Hz), 73.14 (s), 93.53 (d, J = 4.54 Hz);

³¹P{¹H} NMR (202 MHz, D₂O) δ ppm 2.33;

³¹P NMR (202 MHz, D₂O) δ ppm 2.32 (d, J = 4.98 Hz).

Table S1. Chemical shift and coupling constant for selected atoms.

	C1		H6a		H6b	
	δ [ppm]	J [Hz]	δ [ppm]	J [Hz]	δ [ppm]	J [Hz]
Glc-1P2K	93.51	5.45	3.63	12.30; 5.04	3.77	12.14; 2.36
Glc-1P2Na	93.51	5.45	3.63	12.30; 5.04	3.77	12.14; 2.52
Glc-1PKNa	93.51	4.54	3.64	12.30; 5.36	3.77	12.14; 2.60

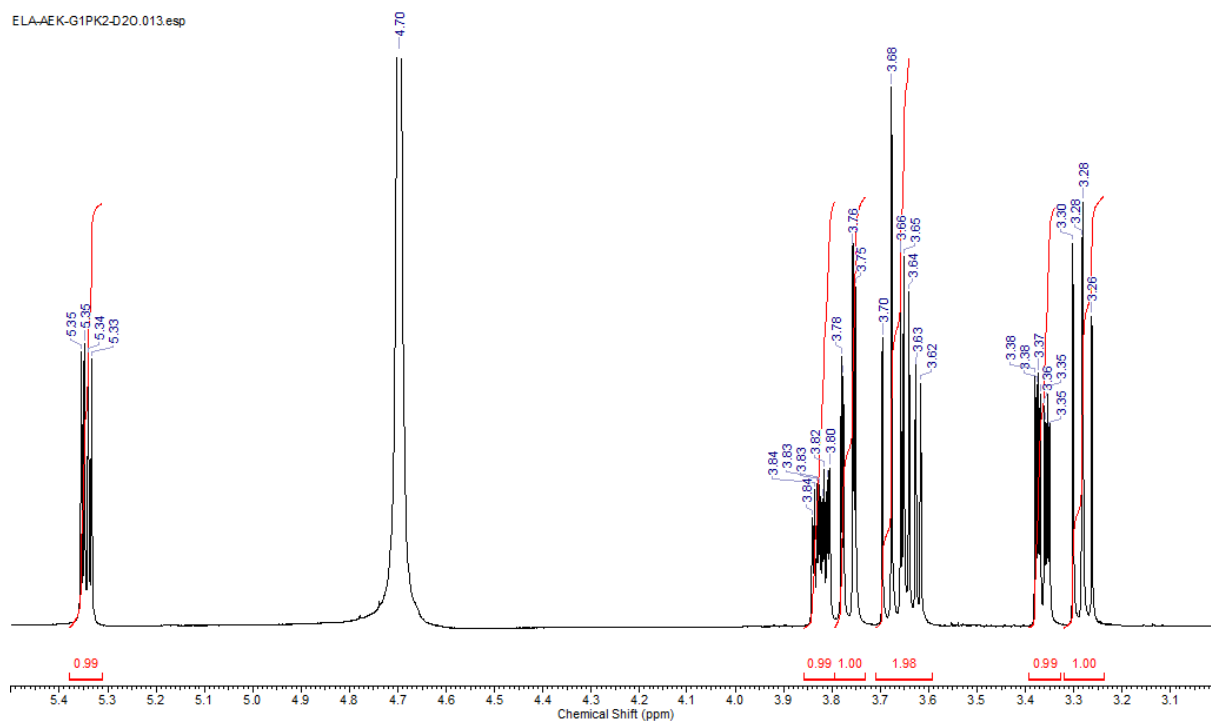


Figure S1. ^1H NMR (D_2O , 500 MHz) spectra of dipotassium α -D-glucose 1-phosphate (Glc-1P2K).

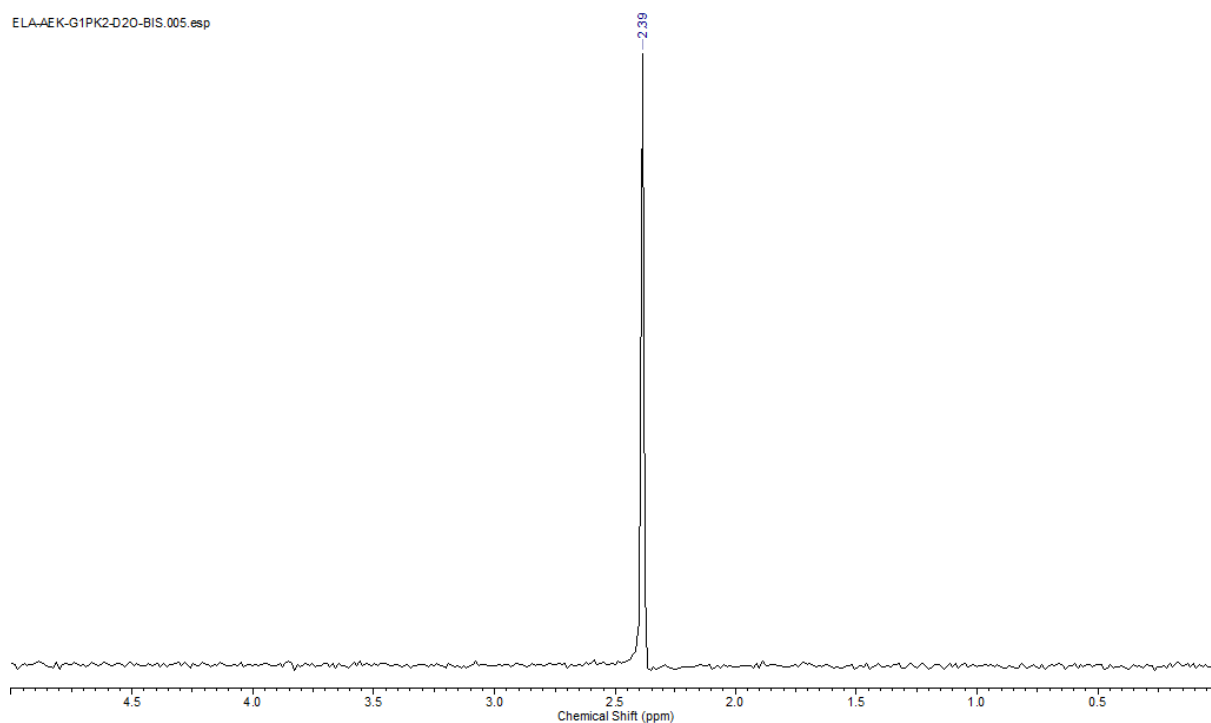


Figure S2. ^{31}P NMR of dipotassium α -D-glucose 1-phosphate (Glc-1P2K) in D_2O , with ^1H decoupling, at 202 MHz.

ELA-AEK-G 1PK2-D2O-BIS.008.esp

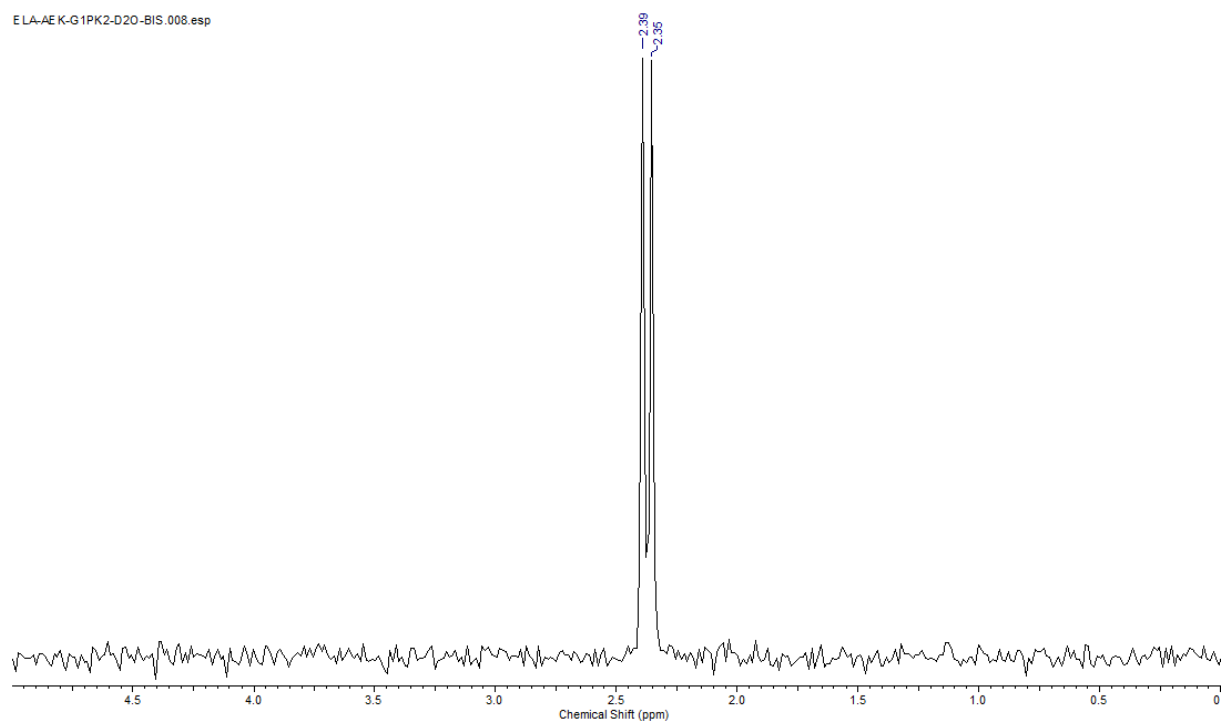


Figure S3. ^{31}P NMR of dipotassium α -D-glucose 1-phosphate (Glc-1P2K) in D_2O , with no decoupling, at 202 MHz.

ELA-AEK-G 1PN A2-D2O.005.esp

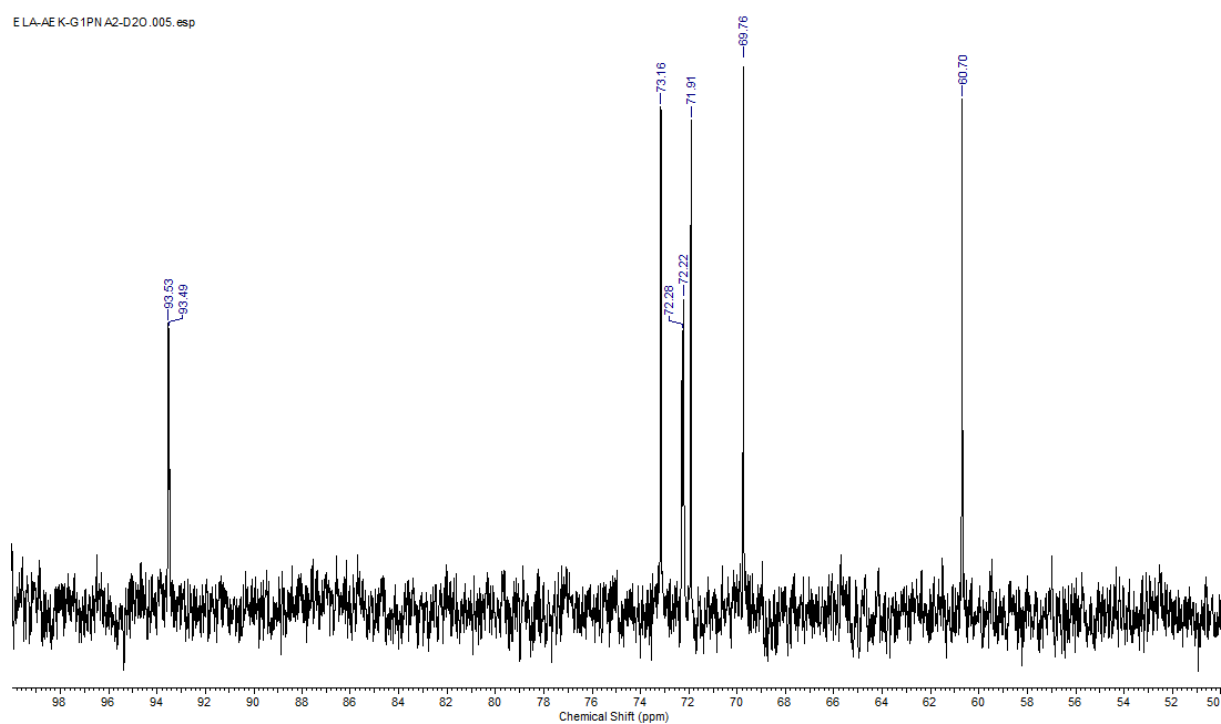


Figure S4. ^{13}C NMR (D_2O , 126 MHz) spectra of dipotassium α -D-glucose 1-phosphate (Glc-1P2K)

ELA-AE-K-G 1PN A2-D2O.006.esp

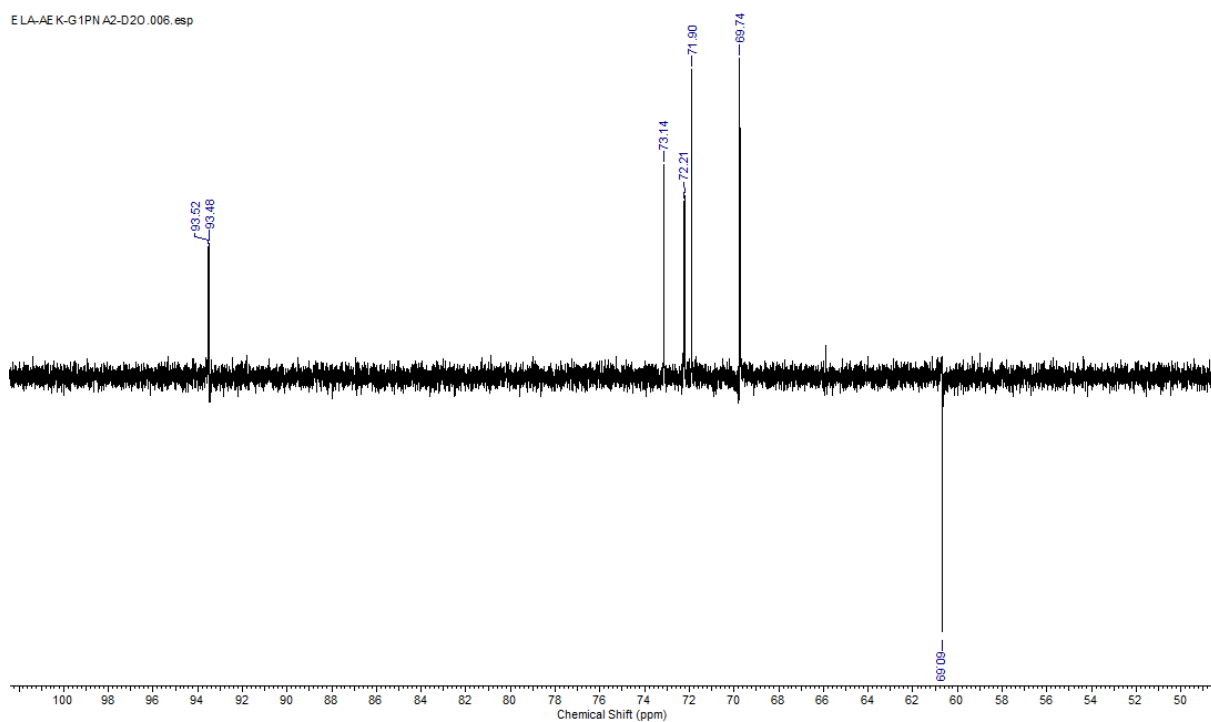


Figure S5. DEPT 135 spectra of dipotassium α -D-glucose 1-phosphate (Glc-1PK₂ Glc-1P2K) (D₂O, 126 MHz).

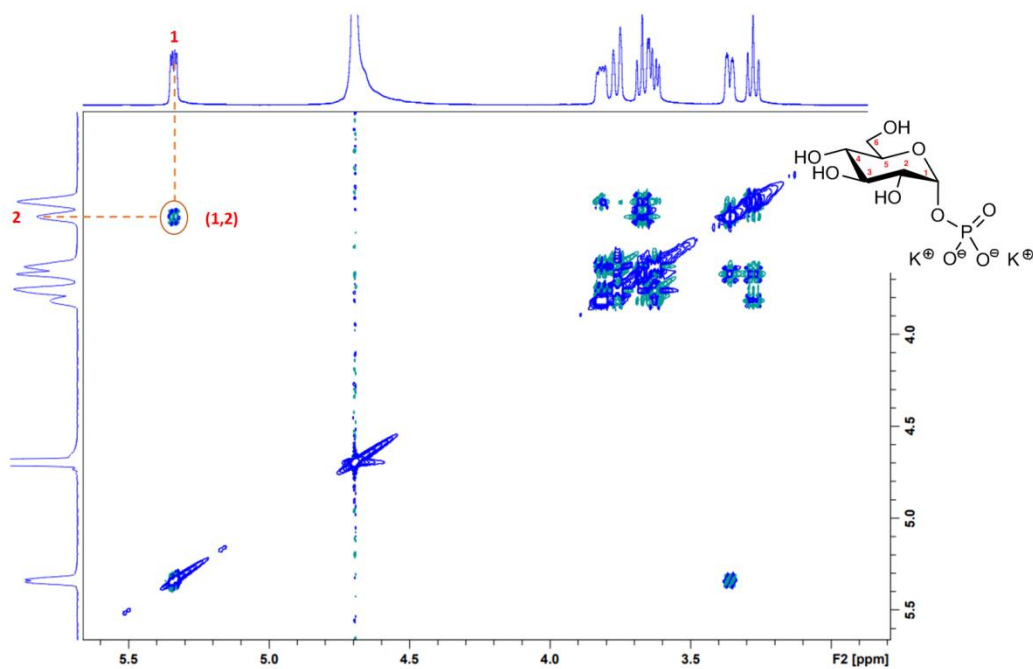


Figure S6. 500 MHz ¹H 2D NOESY spectrum of dipotassium α -D-glucose 1-phosphate (Glc-1P2K).

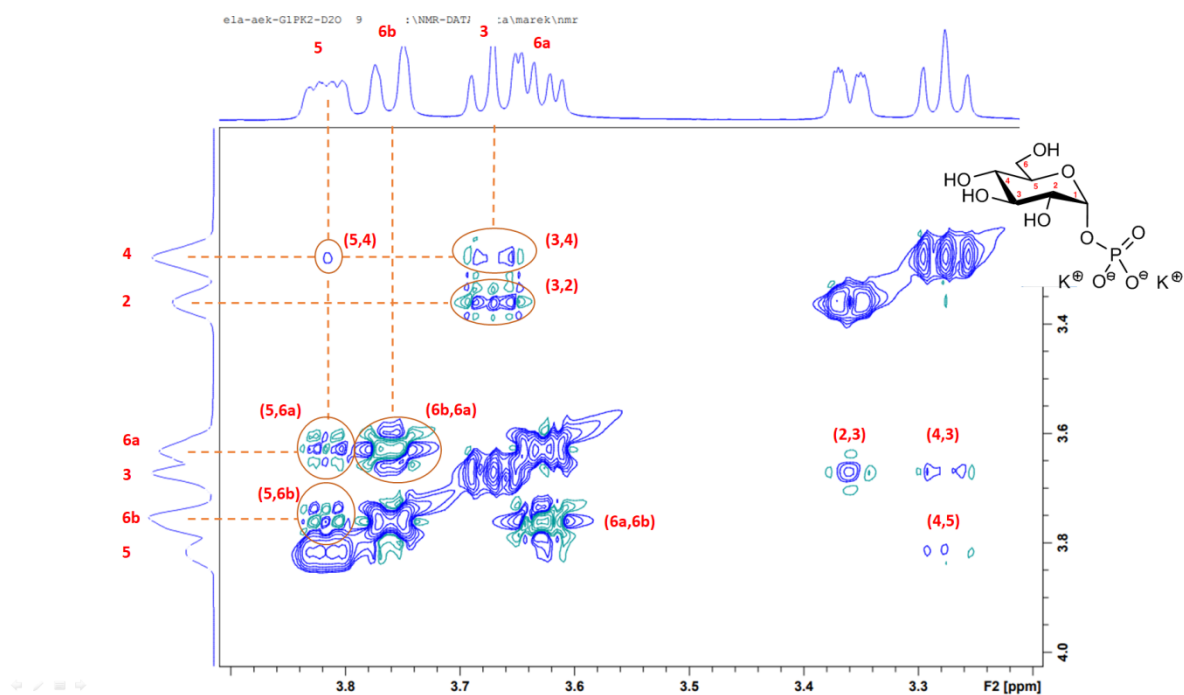


Figure S7. Excerpts of the 500 MHz ^1H 2D NOESY spectrum of dipotassium α -D-glucose 1-phosphate (Glc-1P2K).

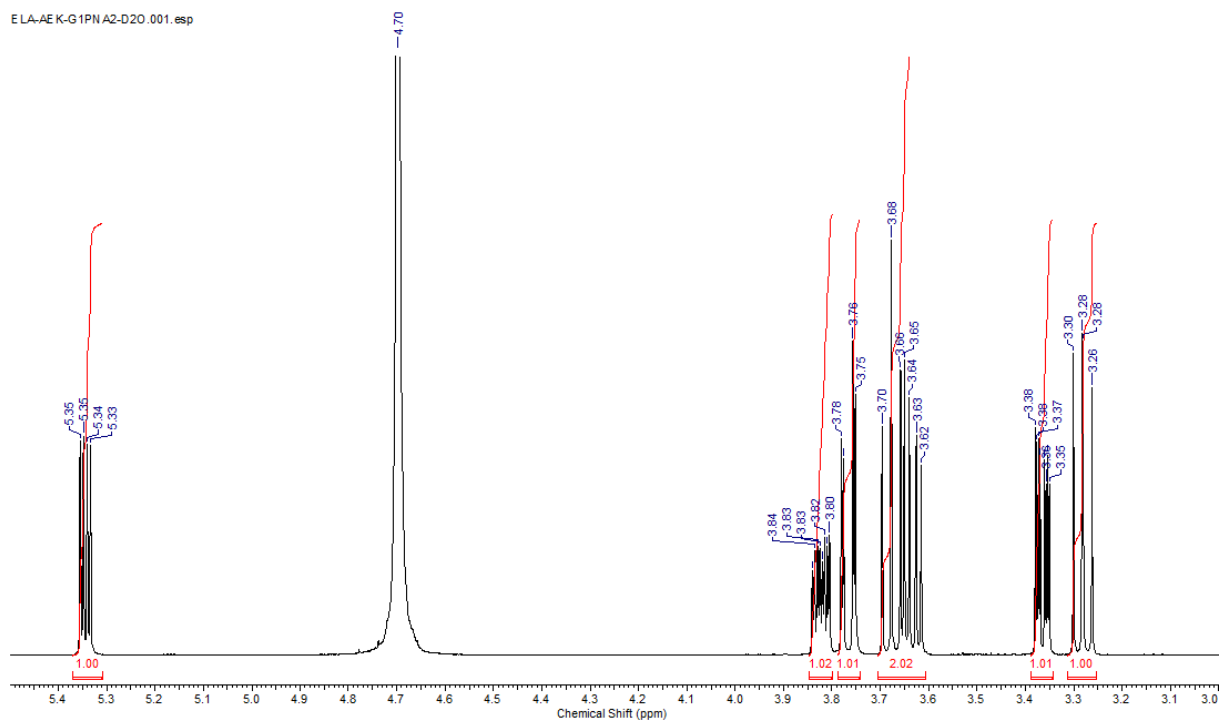


Figure S8. ^1H NMR spectra of disodium α -D-glucose 1-phosphate (Glc-1P2Na) (D_2O , 500 MHz)

NMR-AEK.010.esp

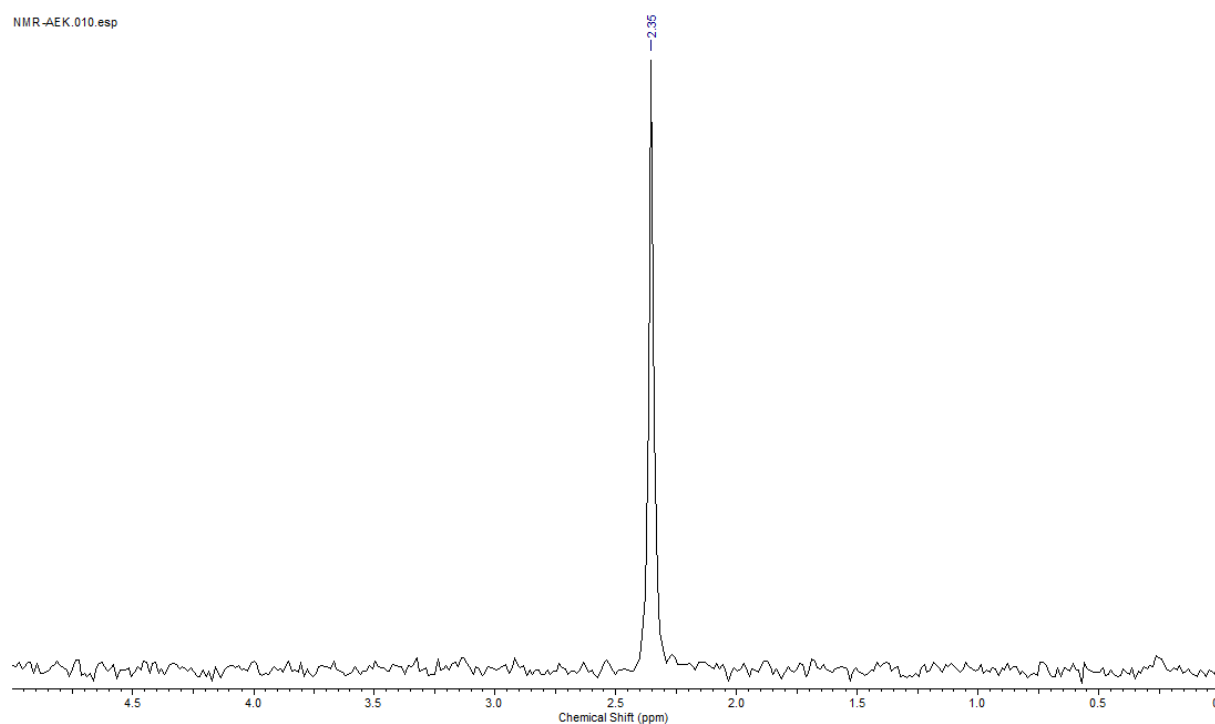


Figure S9. ^{31}P NMR of disodium α -D-glucose 1-phosphate (Glc-1P2Na) in D_2O , with ^1H decoupling, at 202 MHz.

ELA-AEK-G1PN A2-D2O.012.esp

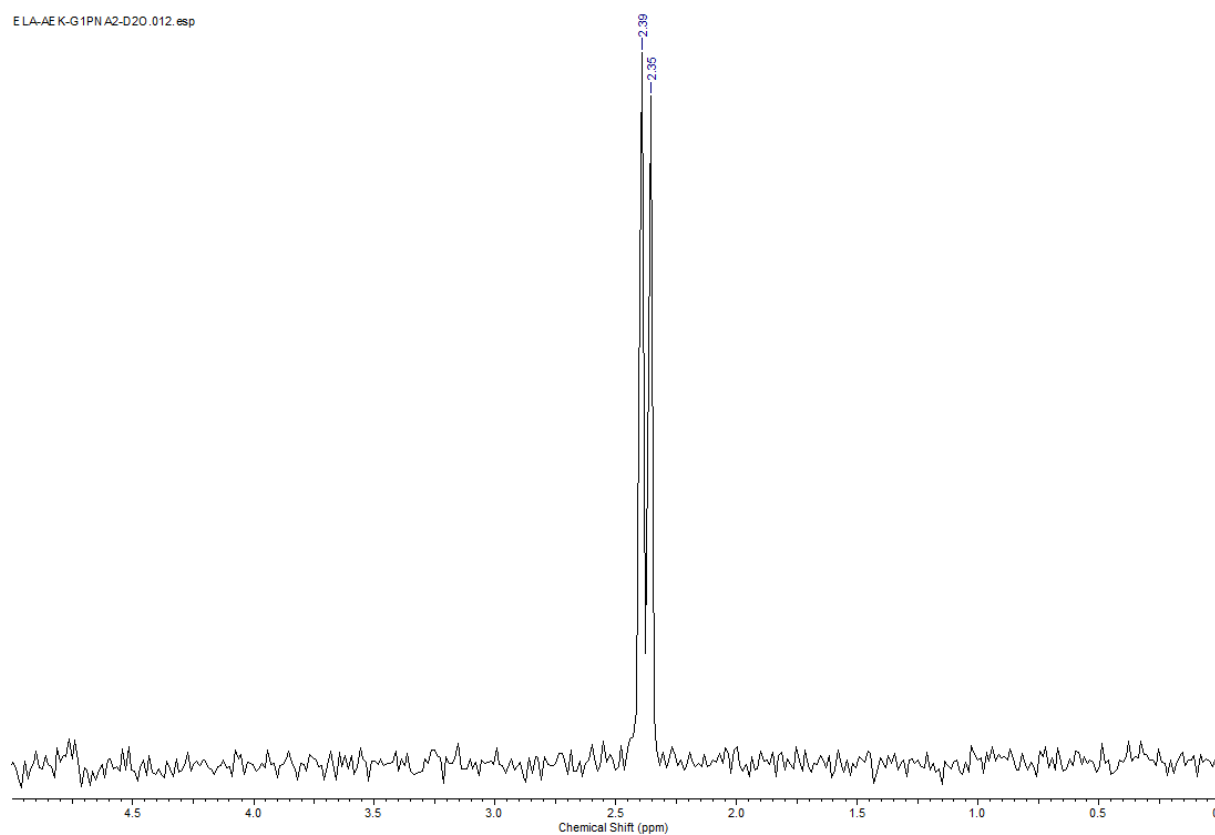


Figure S10. ^{31}P NMR of disodium α -D-glucose 1-phosphate (Glc-1P2Na) in D_2O , with no decoupling, at 202 MHz.

ELA-AE K-G 1PK2-D2O.003.esp

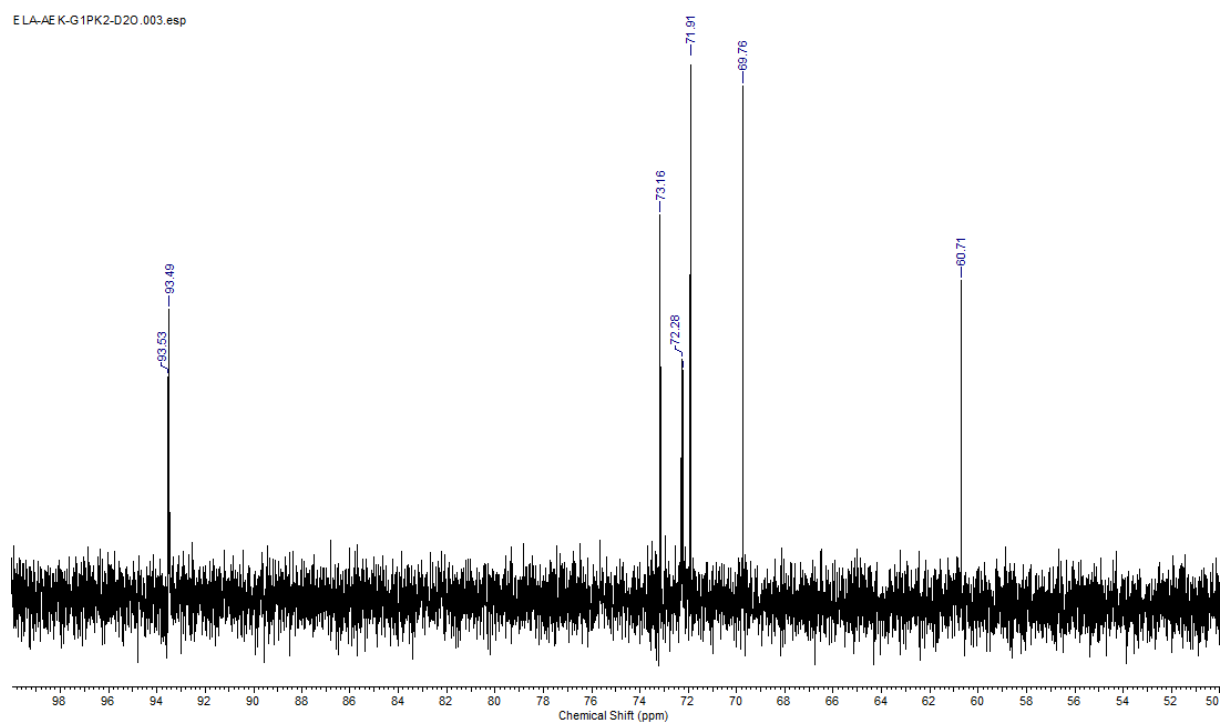


Figure S11. ¹³C NMR spectra of disodium α -D-glucose 1-phosphate (Glc-1P2Na) (D₂O, 126 MHz)

ELA-AE K-G 1PK2-D2O.004.esp

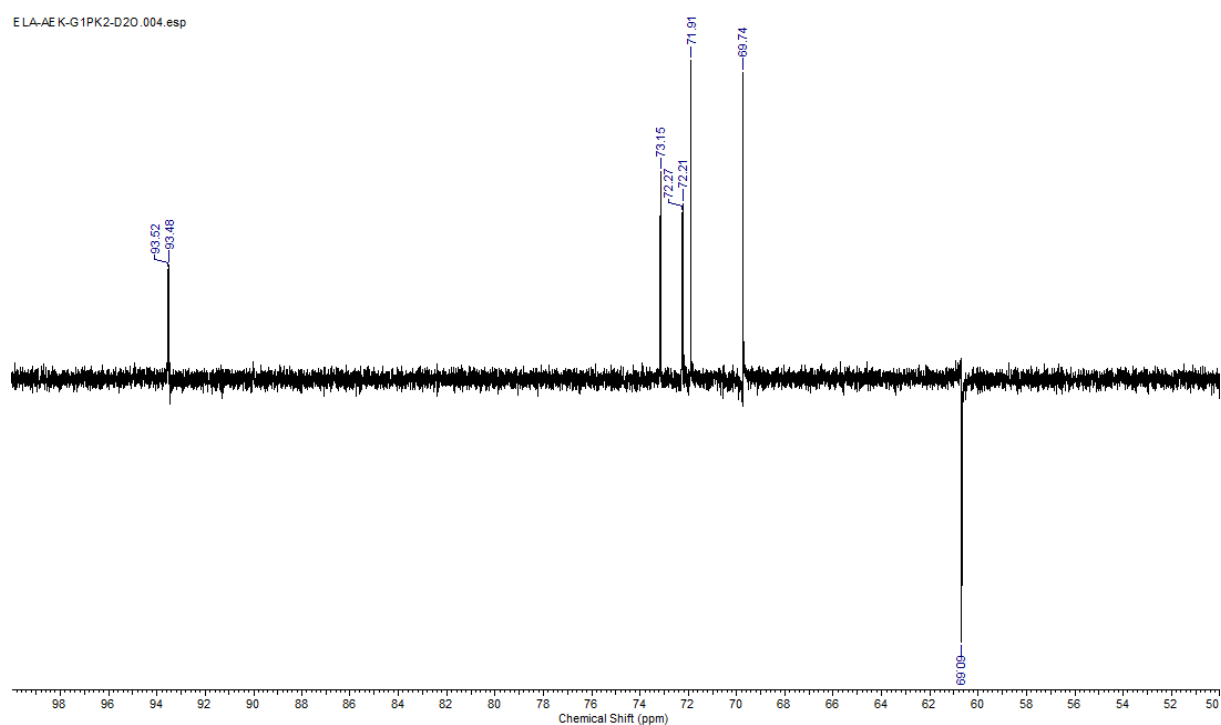


Figure S12. DEPT 135 spectra of disodium α -D-glucose 1-phosphate (Glc-1P2Na) (D₂O, 126 MHz).

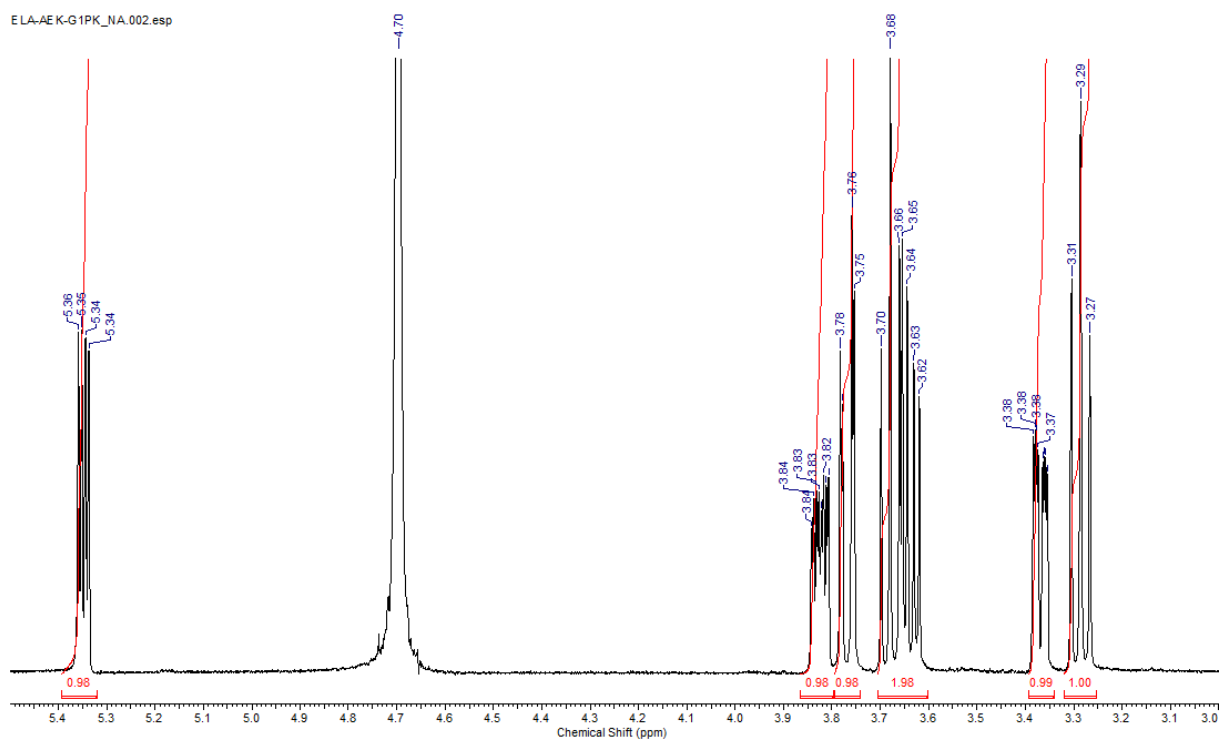


Figure S13. ^1H NMR spectra of potassium sodium α -D-glucose 1-phosphate α -D-Glucose (Glc-1PKNa) (D_2O , 500 MHz)

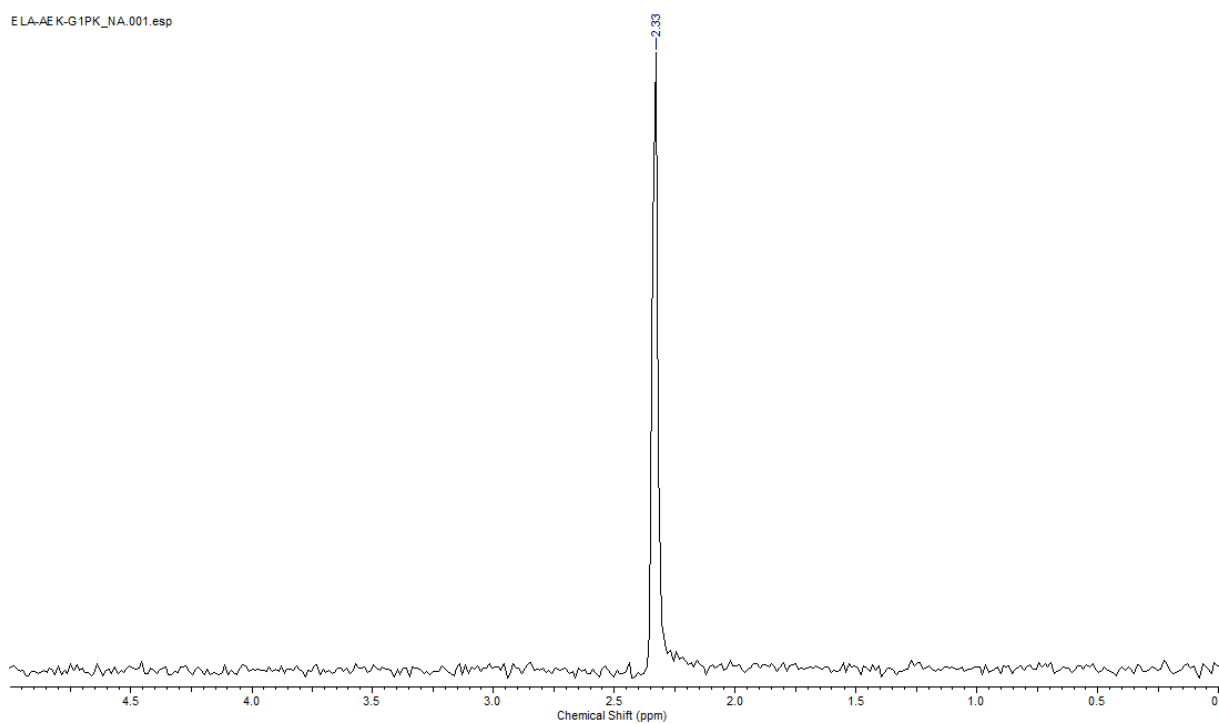


Figure S14. ^{31}P NMR of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa) in D_2O , with ^1H decoupling, at 202 MHz.

ELA-AEK-G1PK_NA.004.esp

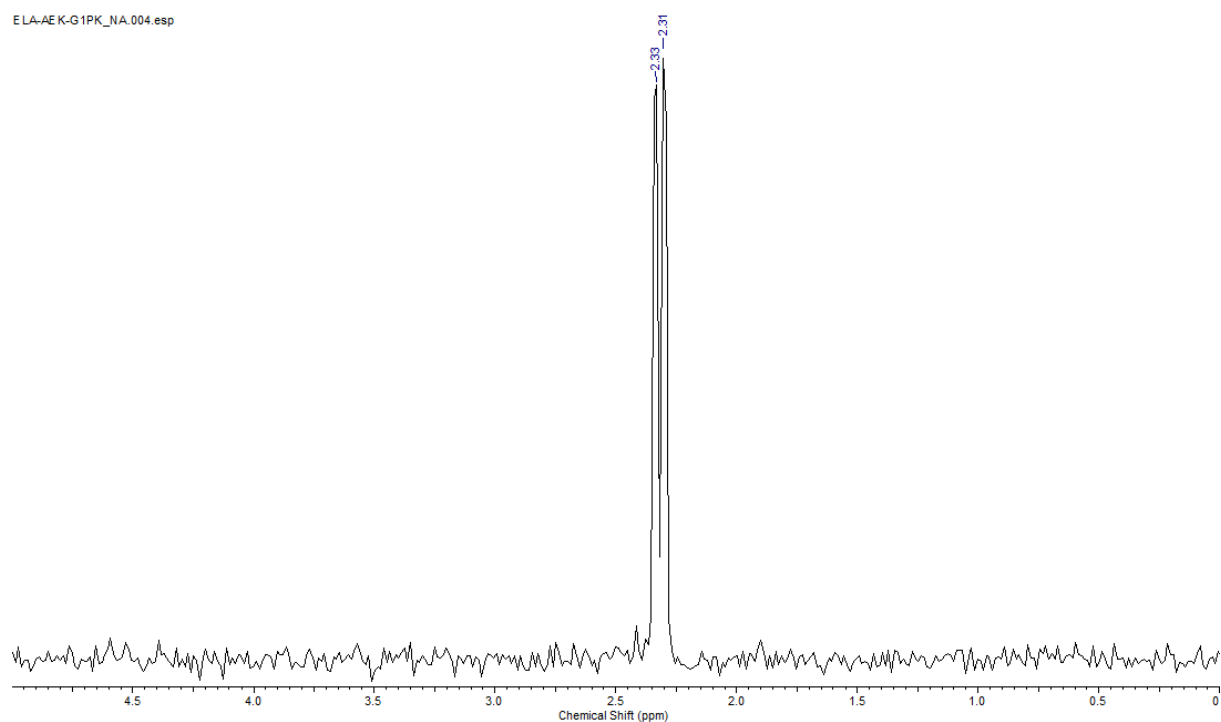


Figure S15. ^{31}P NMR of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa) in D_2O , with no decoupling, at 202 MHz.

ELA-AEK-G1PK_NA.003.esp

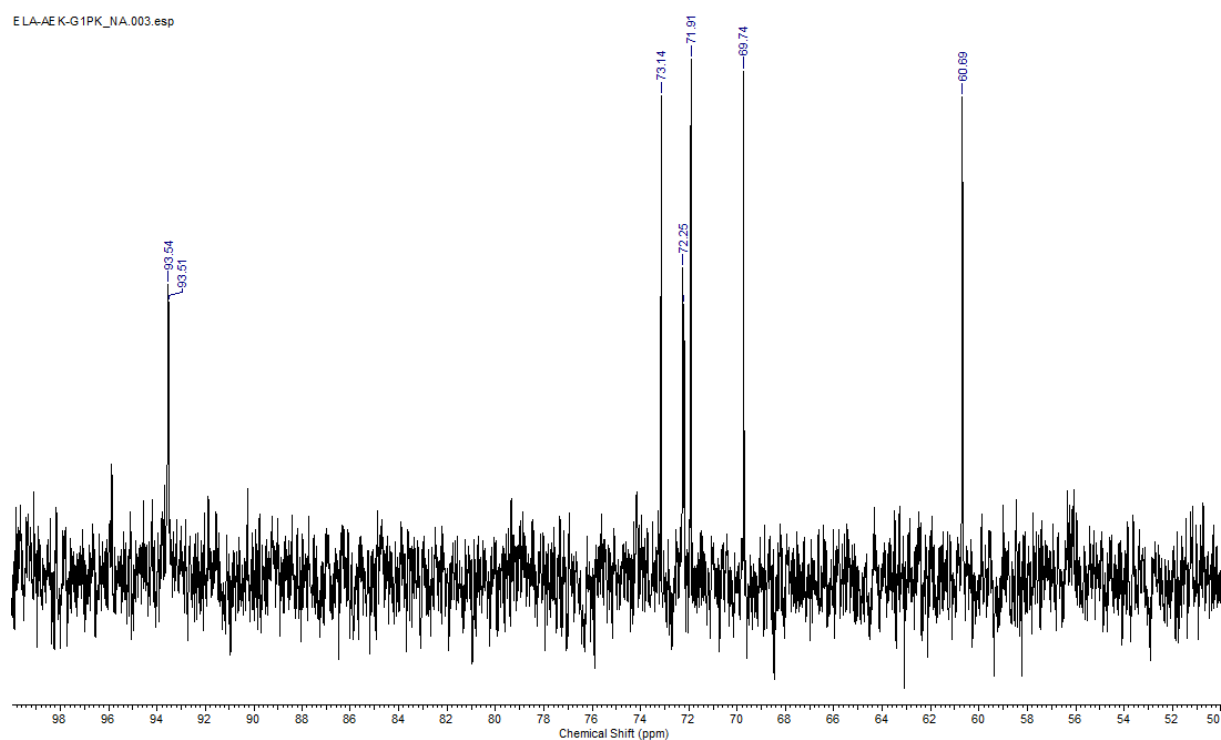


Figure S16. ^{13}C NMR spectra of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa) (D_2O , 126 MHz)

ELA-AEK-G1PK_NA.013.esp

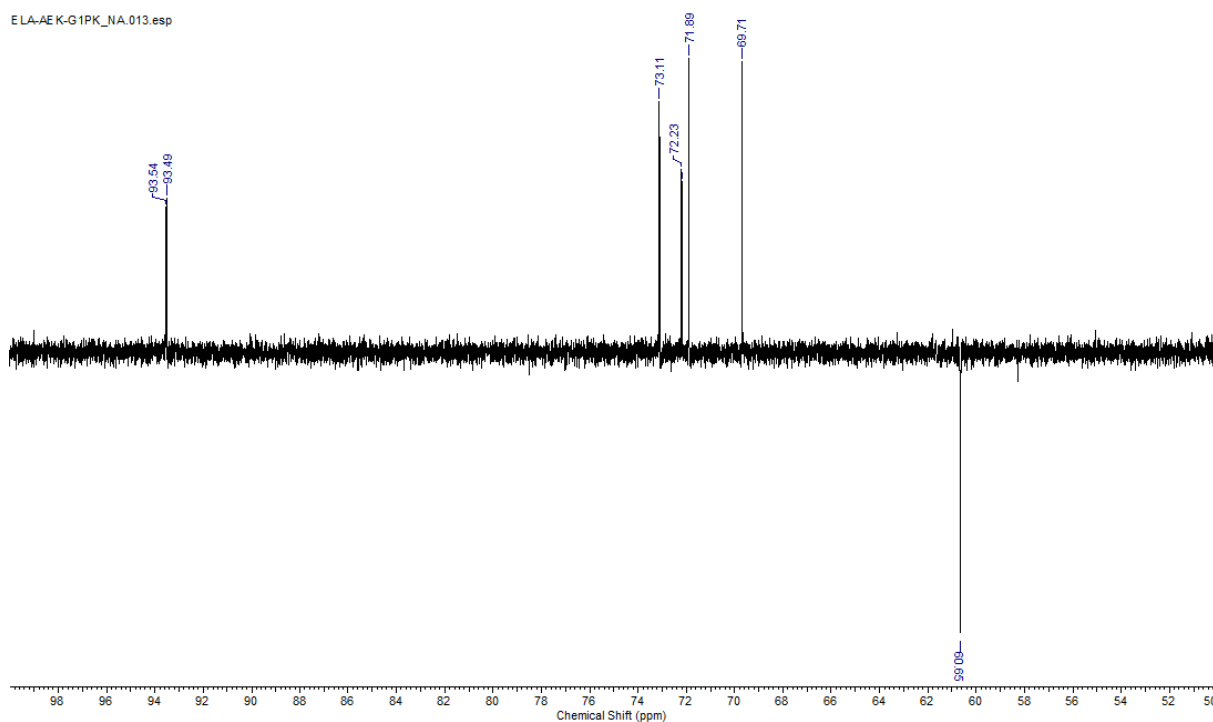


Figure S17. DEPT 135 spectra of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa) (D_2O , 126 MHz).

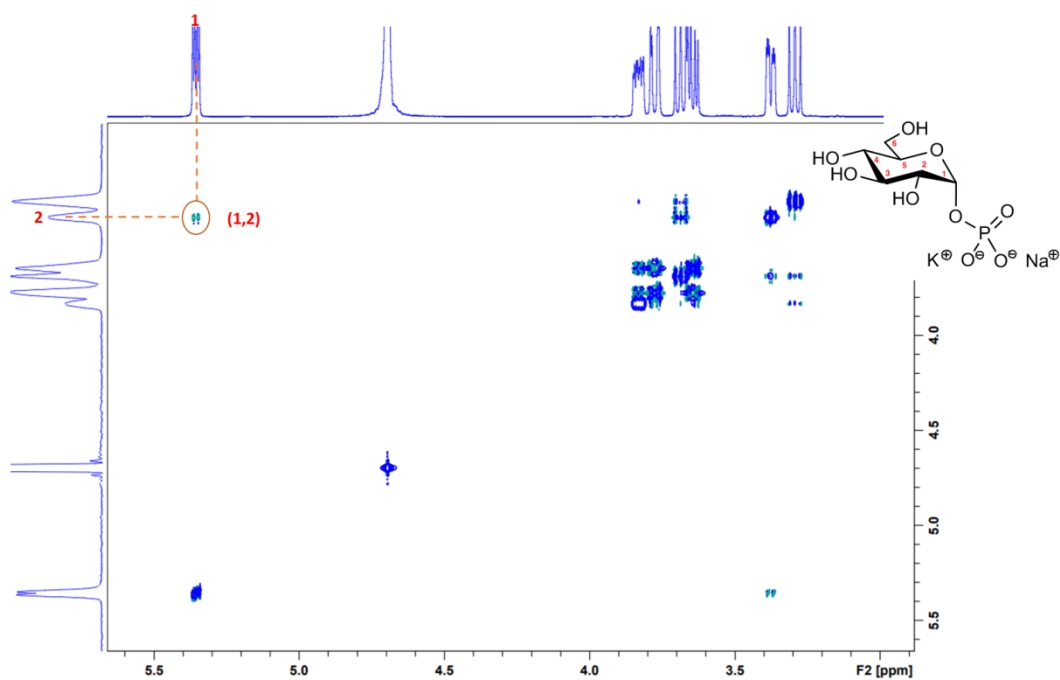


Figure S18. 500 MHz 1H 2D NOESY spectrum of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa)

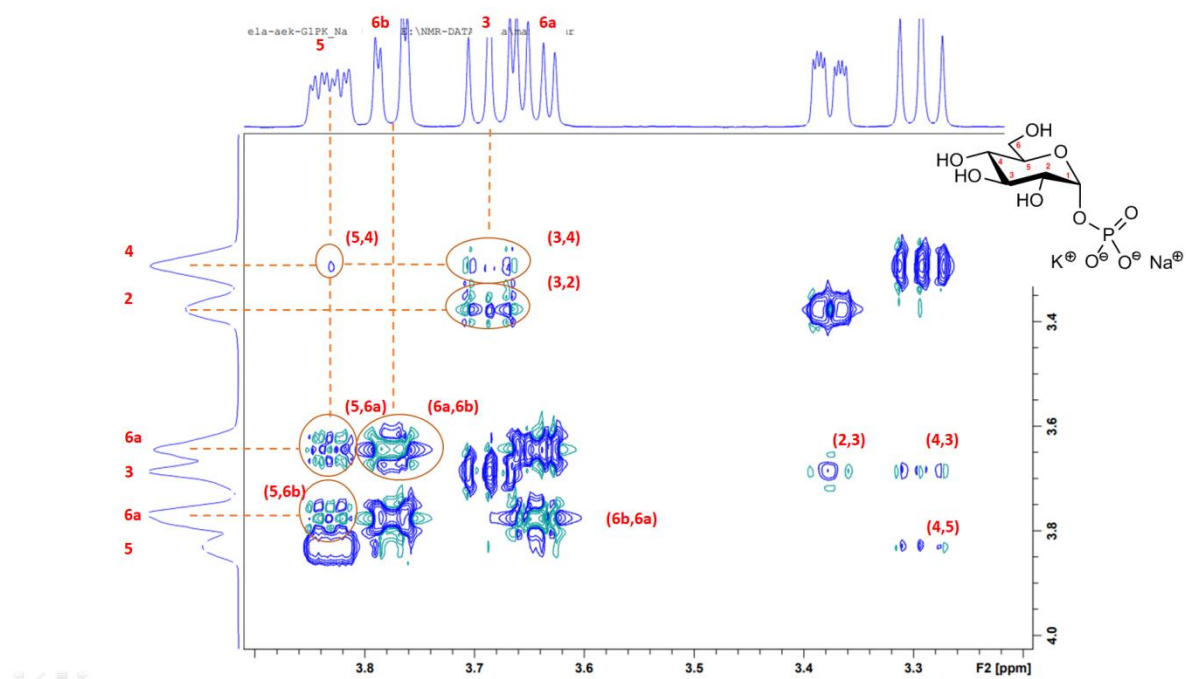


Figure S19. Excerpts of the 500 MHz ^1H 2D NOESY spectrum of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa)

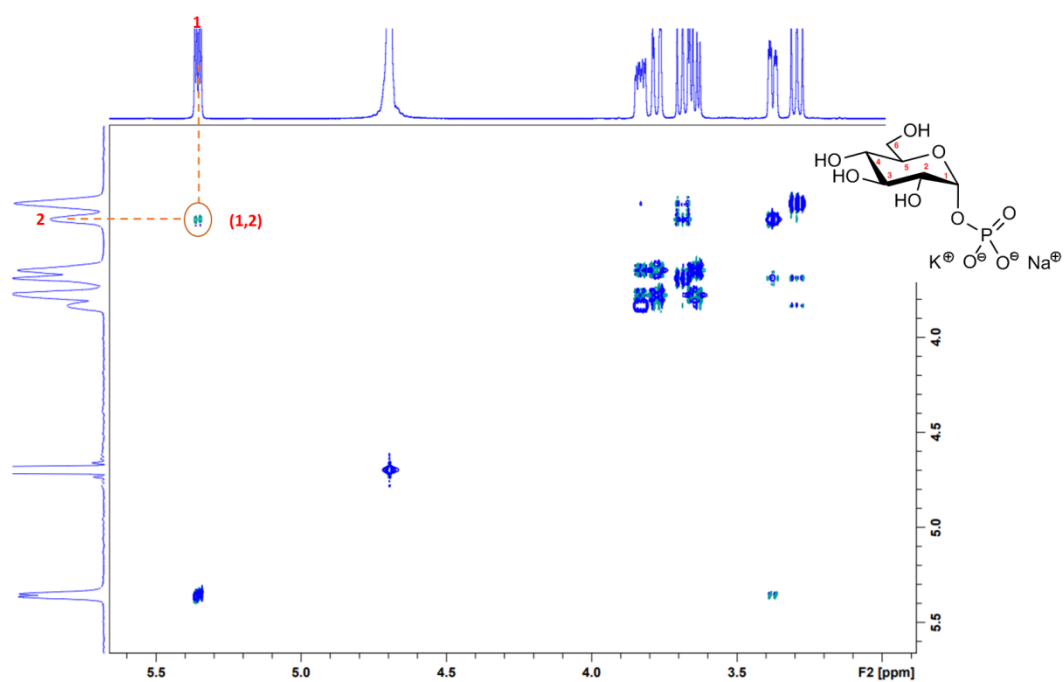


Figure S20. 500 MHz ^1H 2D NOESY spectrum of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa)

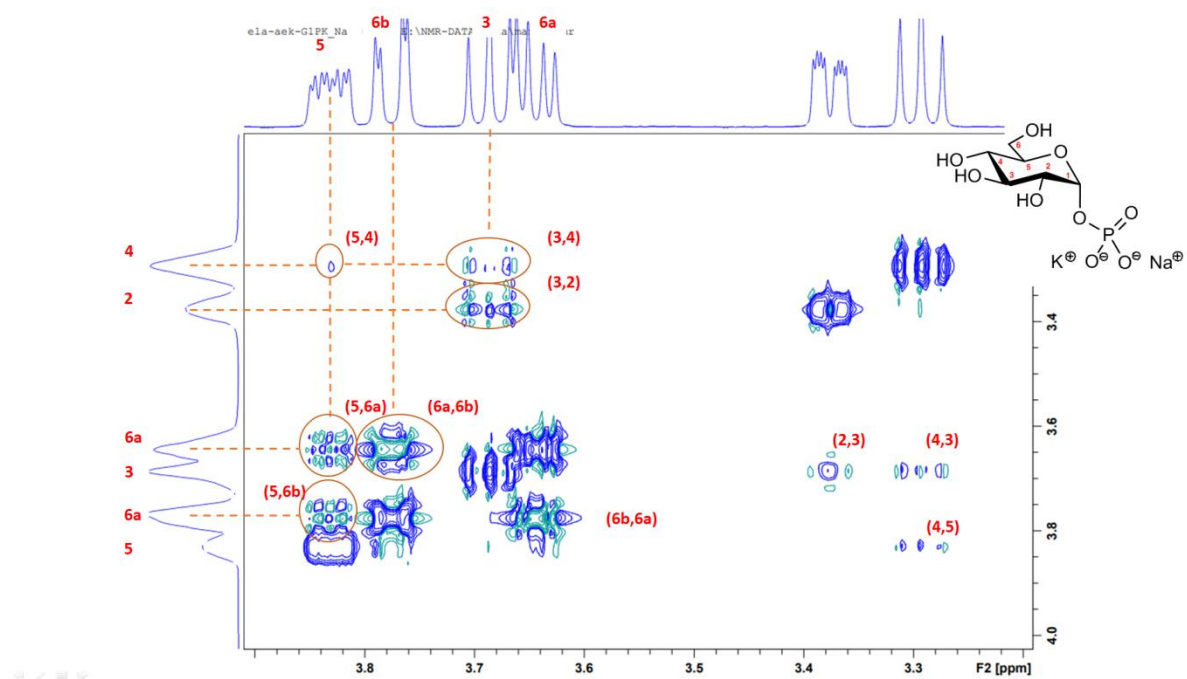


Figure S21. Excerpts of the 500 MHz ^1H 2D NOESY spectrum of potassium sodium α -D-glucose 1-phosphate (Glc-1PKNa)

CRYSTALLOGRAPHIC DATA

Table S2. Potassium sodium α -D-glucose 1-phosphate tetrahydrate

Identification code	G1P-KNa
CCDC No	2345577
Formula	C ₆ H ₁₉ K _{1.50} Na _{0.50} O ₁₃ P
Formula weight	400.33
Wavelength	1.54184 Å
Crystal system	Orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2
Unit cell dimensions	a = 16.8360(4) Å
	b = 13.4089(3) Å
	c = 6.6729(1) Å
Volume	1506.42(5) Å ³
Z	4
Density (calculated)	1.765 g/cm ³
Absorption coefficient	6.121 mm ⁻¹
F(000)	832
Theta range for data collection	4.215 to 76.361°.
Index ranges	-21 ≤ h ≤ 19, -16 ≤ k ≤ 16, -8 ≤ l ≤ 6
Reflections collected	10925
Independent reflections	3123 [R(int) = 0.0314]
Completeness to theta = 67.684°	99.9 %
Data / restraints / parameters	3123 / 0 / 213
Goodness-of-fit on F ²	1.062
Final R indices [I > 2σ(I)]	R1 = 0.0321, wR2 = 0.0852
R indices (all data)	R1 = 0.0343, wR2 = 0.0873
Absolute structure parameter	-0.004(6)
Largest diff. peak and hole	0.46 and -0.45 e.Å ⁻³

Table S3. Disodium α -D-glucose 1-phosphate pentahydrate

Identification code	Glc-1P2Na-5H₂O
CCDC No	2345578
Empirical formula	C ₆ H ₂₁ Na ₂ O ₁₄ P
Formula weight	394.18
Wavelength	1.54184 Å
Crystal system	Orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2 ₁
Unit cell dimensions	a = 6.6347(5) Å
	b = 8.7585(9) Å
β = 99.147(2)°.	c = 27.129(2) Å
Volume	1576.5(2) Å ³
Z	4
Density (calculated)	1.661 g/cm ³
Absorption coefficient	2.783 mm ⁻¹
F(000)	824
Theta range for data collection	5.307 to 76.416°
Index ranges	-8 < = h < = 8, -10 < = k < = 10, -33 < = l < = 24
Reflections collected	10188
Independent reflections	3208 [R(int) = 0.0381]
Completeness to theta = 67.684°	99.7 %
Data / restraints / parameters	3208 / 0 / 253
Goodness-of-fit on F ²	1.161
Final R indices [I > 2sigma(I)]	R1 = 0.0277, wR2 = 0.0613
R indices (all data)	R1 = 0.0367, wR2 = 0.0814
Absolute structure parameter	0.035(16)
Extinction coefficient	0.0016(3)
Largest diff. peak and hole	0.27 and -0.26 e.Å ⁻³

Table S4. Disodium α -D-glucose 1-phosphate trihemihydrate (redetermination – CIMDUX01)

Identification code	G1P2NA-3.5H₂O
CCDC No	23455775
Empirical formula	C ₆ H ₁₈ Na ₂ O _{12.50} P
Formula weight	367.15
Wavelength	1.54184 Å
Crystal system	Monoclinic
Space group	C 2
Unit cell dimensions	a = 8.4325(2) Å
	b = 10.1983(2) Å
	c = 16.5857(4) Å
	β = 99.147(2)°
Volume	1408.19(6) Å ³
Z	4
Density (calculated)	1.732 g/cm ³
Absorption coefficient	2.986 mm ⁻¹
F(000)	764
Theta range for data collection	5.403 to 76.054°.
Index ranges	-10 ≤ h ≤ 10, -12 ≤ k ≤ 12, -19 ≤ l ≤ 20
Reflections collected	2612
Independent reflections	2017 [R(int) = 0.0242]
Completeness to theta = 67.684°	99.6 %
Data / restraints / parameters	2017 / 1 / 265
Goodness-of-fit on F ²	1.043
Final R indices [I > 2σ(I)]	R1 = 0.0243, wR2 = 0.0635
R indices (all data)	R1 = 0.0245, wR2 = 0.0639
Absolute structure parameter	-0.06(2)
Extinction coefficient	0.0081(4)
Largest diff. peak and hole	0.27 and -0.23 e.Å ⁻³

Table S5. Dipotassium α -D-glucose 1-phosphate dihydrate (redetermination – KGLUCP02)

Identification code	G1P2K2H2O
CCDC No	2345574
Empirical formula	C ₆ H ₁₅ K ₂ O ₁₁ P
Formula weight	372.35
Wavelength	1.54184 Å
Crystal system	Monoclinic
Space group	<i>P</i> 2 ₁
Unit cell dimensions	<i>a</i> = 7.5234(3) Å
β = 110.576(5)°	<i>b</i> = 9.0643(3) Å
	<i>c</i> = 10.4588(4) Å
Volume	667.73(5) Å ³
<i>Z</i>	2
Density (calculated)	1.852 g/cm ³
Absorption coefficient	7.959 mm ⁻¹
<i>F</i> (000)	384
Theta range for data collection	4.516 to 76.141°.
Index ranges	-9 ≤ <i>h</i> ≤ 8, -8 ≤ <i>k</i> ≤ 11, -11 ≤ <i>l</i> ≤ 13
Reflections collected	4319
Independent reflections	2178 [<i>R</i> (int) = 0.0272]
Completeness to theta = 67.684°	99.8 %
Data / restraints / parameters	2178 / 1 / 217
Goodness-of-fit on <i>F</i> ²	1.075
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0294, <i>wR</i> 2 = 0.0774
<i>R</i> indices (all data)	<i>R</i> 1 = 0.0298, <i>wR</i> 2 = 0.0779
Absolute structure parameter	0.063(10)
Largest diff. peak and hole	0.25 and -0.50 e.Å ⁻³

Table S6. Diammonium α -D-glucose 1-phosphate trihydrate

Identification code	G1P2NH4
CCDC No	2345576
Empirical formula	C ₆ H ₂₅ N ₂ O ₁₂ P
Formula weight	348.25
Wavelength	0.71069 Å
Crystal system	Orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2
Unit cell dimensions	a = 13.756(7) Å
	b = 16.929(7) Å
	c = 6.502(3) Å
Volume	1514.2(12) Å ³
<i>Z</i>	4
Density (calculated)	1.528 g/cm ³
Absorption coefficient	0.244 mm ⁻¹
F(000)	744
Theta range for data collection	2.825 to 32.568°
Index ranges	-20 ≤ h ≤ 20, -22 ≤ k ≤ 25, -9 ≤ l ≤ 9
Reflections collected	3297
Independent reflections	3297 [R(int) = 0.028]
Completeness to theta = 25.240°	84.8 %
Data / restraints / parameters	3297 / 0 / 204
Goodness-of-fit on F ²	1.083
Final R indices [I > 2σ(I)]	R1 = 0.0440, wR2 = 0.1193
R indices (all data)	R1 = 0.0440, wR2 = 0.1193
Absolute structure parameter	-0.02(8)
Largest diff. peak and hole	0.57 and -0.35 e. Å ⁻³

Table S7. Ammonium sodium α -D-glucose 1-phosphate tetrahydrate

Identification code	Glc-1P(NH₄)Na
CCDC No	2345579
Empirical formula	C ₆ H ₂₅ N _{1.50} Na _{0.50} O ₁₃ P
Formula weight	368.74
Wavelength	1.54184 Å
Crystal system	Orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2
Unit cell dimensions	a = 16.9272(3) Å
	b = 13.3171(3) Å
	c = 6.6436(1) Å
Volume	1497.61(5) Å ³
<i>Z</i>	4
Density (calculated)	1.635 g/cm ³
Absorption coefficient	2.462 mm ⁻¹
F(000)	784
Theta range for data collection	5.226 to 76.184°
Index ranges	-21 ≤ <i>h</i> ≤ 18, -16 ≤ <i>k</i> ≤ 16, -8 ≤ <i>l</i> ≤ 7
Reflections collected	10118
Independent reflections	3069 [R(int) = 0.0359]
Completeness to theta = 67.684°	99.7 %
Data / restraints / parameters	3069 / 0 / 200
Goodness-of-fit on F ²	1.066
Final R indices [I > 2σ(I)]	R ₁ = 0.0579, wR ₂ = 0.1565
R indices (all data)	R ₁ = 0.0647, wR ₂ = 0.1609
Absolute structure parameter	0.008(14)
Largest diff. peak and hole	0.64 and -0.62 e.Å ⁻³

Table S8. Ammonium potassium α -D-glucose 1-hydrogenphosphate

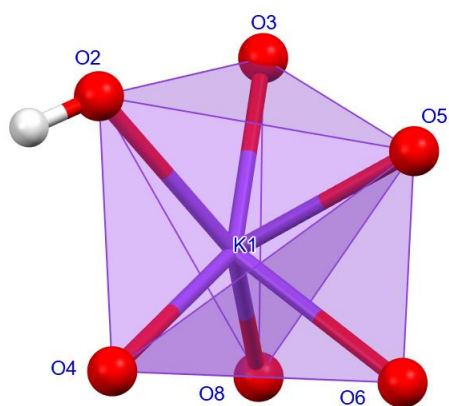
Identification code	Glc-1PH(NH₄)K
CCDC No	2345573
Empirical formula	C ₆ H _{14.75} K _{0.31} N _{0.69} O ₉ P
Formula weight	283.74
Wavelength	1.5418 Å
Crystal system	Orthorhombic
Space group	<i>P</i> 2 ₁ 2 ₁ 2 ₁
Unit cell dimensions	a = 7.147(4) Å
	b = 12.036(5) Å
	c = 12.524(5) Å
Volume	1077.3(9) Å ³
<i>Z</i>	4
Density (calculated)	1.749 g/cm ³
Absorption coefficient	3.793 mm ⁻¹
F(000)	594
Theta range for data collection	5.096 to 74.984°.
Index ranges	-8 ≤ h ≤ 0, -15 ≤ k ≤ 14, -15 ≤ l ≤ 15
Reflections collected	4183
Independent reflections	2211 [R(int) = 0.0427]
Completeness to theta = 67.680°	99.9 %
Data / restraints / parameters	2211 / 0 / 164
Goodness-of-fit on F ²	1.048
Final R indices [I > 2σ(I)]	R1 = 0.0338, wR2 = 0.0851
R indices (all data)	R1 = 0.0382, wR2 = 0.0873
Absolute structure parameter	0.033(12)
Largest diff. peak and hole	0.56 and -0.42 e.Å ⁻³

Table S9. Potassium α -D-glucose 1-hydrogenphosphate [18]

Identification code	Glc-1PHK (JUGTAG)
Formula	C ₆ H ₁₂ O ₉ P K
Formula weight	298.23
Crystal system	Orthorhombic
Space group	$P 2_1 2_1 2_1$
Unit cell dimensions	$a = 7.35(4) \text{ \AA}$
	$b = 9.666(6) \text{ \AA}$
	$c = 15.230(7) \text{ \AA}$
Volume	$1082.16(9) \text{ \AA}^3$
Density	1.83
R1	R1=0.0316

Potassium coordination polyhedron for JUGTAG – Glc-1PHK

LK = 6 TRIGONAL PRISM



CATION COORDINATION

CSD search for complexes of chemical composition C+H+O+cation (Na/K); R1 < 10%

Cation coordination number	Na CN = 5	Na CN = 6	K CN = 6	K CN = 8
Ionic radii (Å) [Shannon]	1.14	1.16	1.52	1.65
Na/K....O distance range (Å)	2.13 – 2.62	2.15 – 2.74	2.51 – 3.27	2.59 – 3.17
<i>No of spheres</i>	18	48	7	35
<i>No of bonds</i>	18x5	48x6	7x6	35x8

Table S10. Cation coordination in the Glc-1P2K 2H₂O crystal

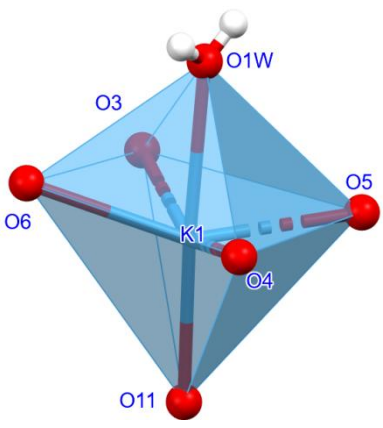
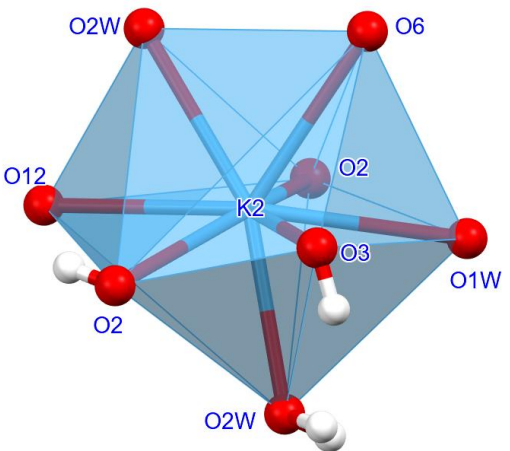
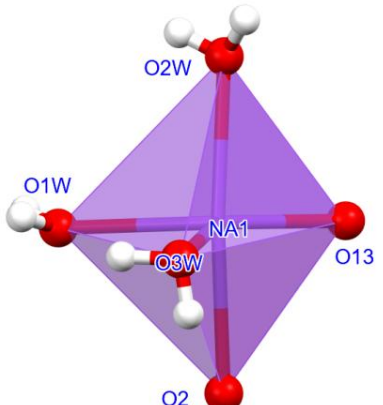
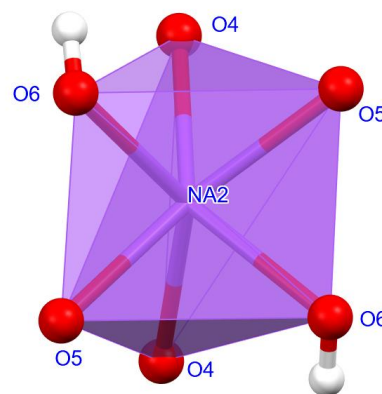
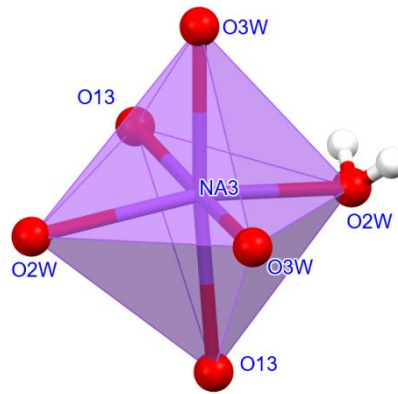
<p>LK = 6</p> <p>K(1)-O(6) 2.716(3)</p> <p>K(1)-O(11) 2.712(3)</p> <p>K(1)-O(4)#1 2.717(3)</p> <p>K(1)-O(5)#2 2.736(2)</p> <p>K(1)-O(3)#3 2.760(3)</p> <p>K(1)-O(1W) 2.881(3)</p> <p>#1 x-1,y,z</p> <p>#2 -x,y-1/2,-z+1</p> <p>#3 -x+1,y-1/2,-z+1</p>	 <p>OCTAHEDRON (OC-6)</p>
<p>LK = 8</p> <p>K(2)-O(2) 2.831(3)</p> <p>K(2)-O(2)#8 3.085(3)</p> <p>K(2)-O(3)#8 3.072(3)</p> <p>K(2)-O(6)#5 2.962(3)</p> <p>K(2)-O(12)#7 3.013(3)</p> <p>K(2)-O(1W)#5 2.939(3)</p> <p>K(2)-O(2W) 2.797(4)</p> <p>K(2)-O(2W)#6 2.943(4)</p> <p>#5 x,y,z-1 #6 -x+1,y+1/2,-z</p> <p>#7 x+1,y,z #8 -x+1,y-1/2,-z</p>	 <p>DODECAHEDRON (DD-8)</p>

Table S11. Cation coordination in the Glc-1P2Na 3.5 H₂O crystal

<p>LK = 5</p> <p>Na(1)-O(13) 2.314 (2)</p> <p>Na(1)-O(2)#1 2.424(2)</p> <p>Na(1)-O(1W) 2.360(2)</p> <p>Na(1)-O(2W) 2.347(2)</p> <p>Na(1)-O(3W) 2.431(2)</p>	 <p>TRIGONAL BIPYRAMID (TBPY-5)</p>
<p>LK =6 (2-fold axis)</p> <p>Na(2)-O(5) 2.440(2)</p> <p>Na(2)-O(6) 2.394(2)</p> <p>Na(2)-O(5)#2 2.440(2)</p> <p>Na(2)-O(6)#2 2.394(2)</p> <p>Na(2)-O(4)#3 2.405(2)</p> <p>Na(2)-O(4)#4 2.405(2)</p>	 <p>TRIGONAL PRISM (TPR-6)</p>
<p>LK =6 (2-fold axis)</p> <p>Na(3)-O(13) 2.526(2)</p> <p>Na(3)-O(13)#5 2.526(2)</p> <p>Na(3)-O(2W) 2.397(2)</p> <p>Na(3)-O(2W)#5 2.397(2)</p> <p>Na(3)-O(3W)#6 2.417(2)</p> <p>Na(3)-O(3W)#7 2.417(2)</p>	 <p>OCTAHEDRON (OC-6)</p>

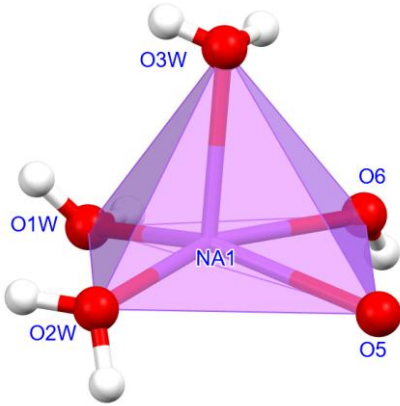
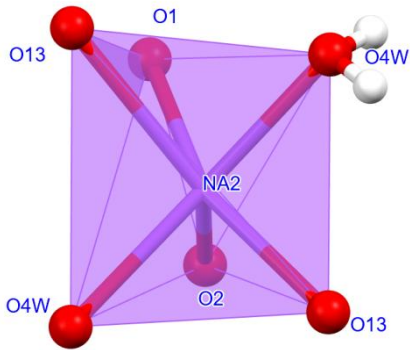
Symmetry transformations used to generate equivalent atoms:

#1 $x-1/2, y-1/2, z$ #2 $-x+1, y, -z$ #3 $-x+1/2, y-1/2, -z$

#4 $x+1/2, y-1/2, z$ #5 $-x+1, y, -z+1$ #6 $-x+1/2, y+1/2, -z+1$

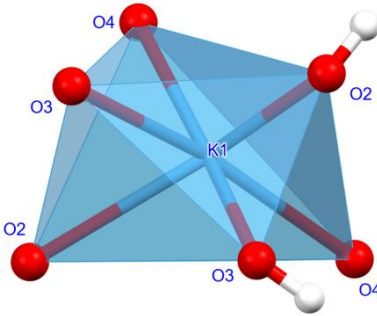
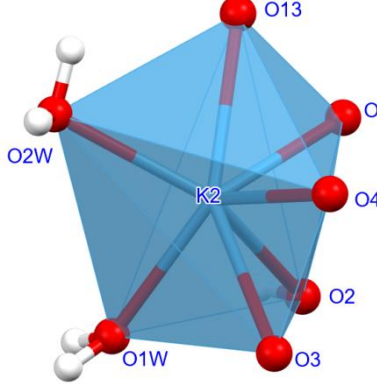
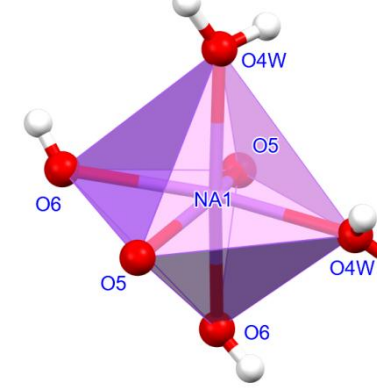
#7 $x+1/2, y+1/2, z$ #8 $x-1/2, y+1/2, z$

Table S12. Cation coordination in the Glc-1P2Na 5 H₂O crystal

<p>LK = 5</p> <p>Na(1)-O(5) 2.471(3)</p> <p>Na(1)-O(6) 2.317(3)</p> <p>Na(1)-O(1W) 2.405(4)</p> <p>Na(1)-O(2W) 2.356(4)</p> <p>Na(1)-O(3W) 2.331(4)</p>	 <p>TETRAGONAL PYRAMID (SPY-5)</p>
<p>LK = 6</p> <p>Na(2)-O(1)#2 2.525(3)</p> <p>Na(2)-O(2)#2 2.492(3)</p> <p>Na(2)-O(13) 2.344(3)</p> <p>Na(2)-O(13)#2 2.610(3)</p> <p>Na(2)-O(4W) 2.416(3)</p> <p>Na(2)-O(4W)#1 2.397(4)</p>	 <p>TRIGONAL PRISM (TPR-6)</p>

#1 $x+1/2, -y+3/2, -z$ #2 $x-1/2, -y+3/2, -z$

Table S13. Cation coordination in the Glc-1PKNa crystal

<p>LK =6 (2-fold axis)</p> <p>K(1)-O(2) 2.980(3)</p> <p>K(1)-O(3) 2.735(3)</p> <p>K(1)-O(2)#1 2.980(3)</p> <p>K(1)-O(3)#1 2.735(3)</p> <p>K(1)-O(4)#2 2.913(3)</p> <p>K(1)-O(4)#3 2.913(3)</p>	 <p>TRIGONAL ANTIPRISM (OC-6)</p>
<p>LK = 7</p> <p>K(2)-O(1W) 2.709(4)</p> <p>K(2)-O(2W) 2.750(3)</p> <p>K(2)-O(1)#8 2.820(2)</p> <p>K(2)-O(2)#8 2.944(3)</p> <p>K(2)-O(13)#8 2.859(3)</p> <p>K(2)-O(3)#9 2.837(3)</p> <p>K(2)-O(4)#9 3.280(3)</p>	 <p>CAPPED OCTAHEDRON (OCF-7)</p>
<p>LK = 6 (on 2-fold axis)</p> <p>Na(1)-O(5) 2.578(2)</p> <p>Na(1)-O(6) 2.344(3)</p> <p>Na(1)-O(4W) 2.273(4)</p> <p>Na(1)-O(5)#10 2.578(2)</p> <p>Na(1)-O(6)#10 2.344(3)</p> <p>Na(1)-O(4W)#10 2.273(4)</p>	 <p>OCTAHEDRON (OC-6)</p>

Symmetry transformations used to generate equivalent atoms:

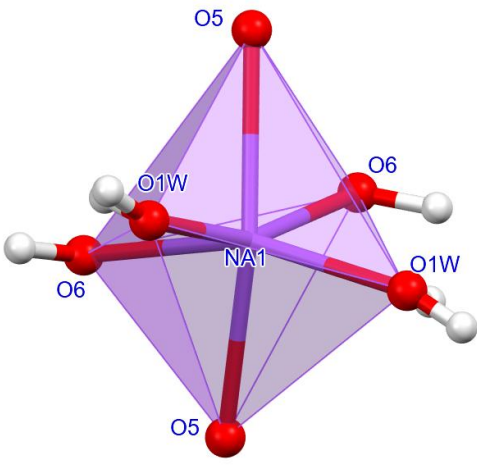
#1 -x+1,-y+1,z #2 -x+1,-y+1,z+1 #3 x,y,z+1

#4 -x+1/2,y+1/2,-z #5 x+1/2,-y+1/2,-z #6 x+1/2,-y+1/2,-z+1

#7 -x+1/2,y+1/2,-z+1 #8 -x+1/2,y-1/2,-z #9 x-1/2,-y+1/2,-z

#10 -x+1,-y,z #11 -x+1/2,y-1/2,-z+1 #12 x,y,z-1

Table S14. Cation coordination and hydrogen bonds in the Glc-1P(NH₄)Na crystal

LK = 6			
Na(1)-O(5)	2.535(3) Å		
Na(1)-O(6)	2.365(6)		
Na(1)-O(1W)	2.317(8)		
Na(1)-O(5)#1	2.535(3)		
Na(1)-O(6)#1	2.365(6)		
Na(1)-O(1W)#1	2.317(8)		
Symmetry transformations used to generate equivalent atoms: #1 -x+1,-y,z		OCTAHEDRON	

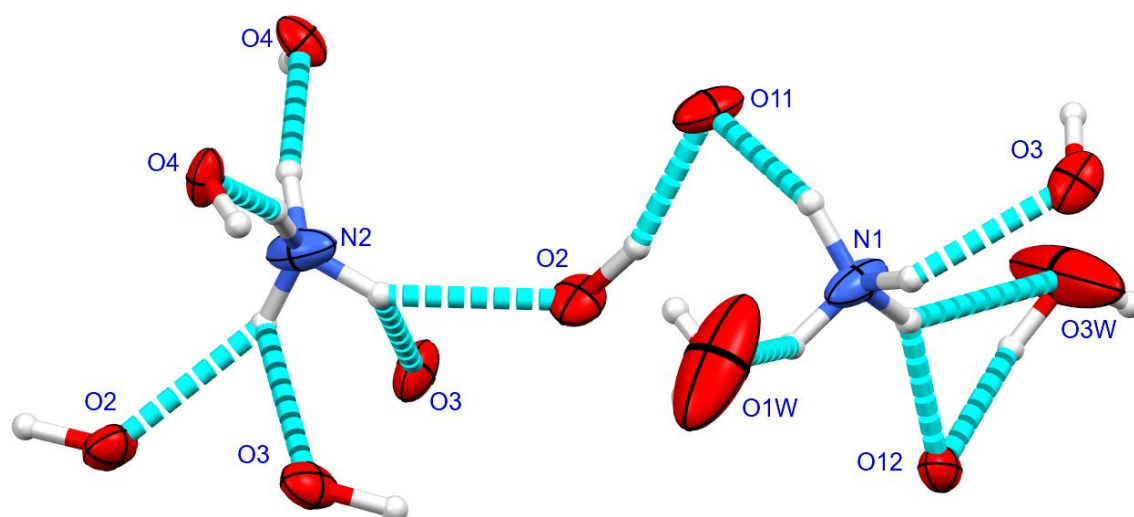
D-H...A	d(D-H) (Å)	d(H...A) (Å)	d(D...A) (Å)	<(DHA) (°)
N(1)-H(1N1)...O(12)	1.02	1.90	2.782(6)	143
N(1)-H(2N1)...O(11)#2	1.01	2.02	2.817(6)	134
N(1)-H(3N1)...O(3)#3	1.01	2.10	2.782(6)	123
N(2)-H(1N2)...O(2)	1.00	2.18	3.091(5)	151
N(2)-H(2N2)...O(4)#4	1.02	1.90	2.893(5)	166
O(2)-H(2O)...O(11)#2	0.99	1.82	2.780(6)	161
O(3)-H(3O)...O(4W)	0.98	2.13	3.007(8)	148
O(4)-H(4O)...O(12)#5	0.99	1.63	2.611(5)	169
O(6)-H(6O)...O(13)#5	0.99	1.77	2.675(5)	151
O(1W)-H(2W1)...O(2W)#2	0.81	1.91	2.434(16)	122
O(1W)-H(1W1)...O(3W)#6	0.77	2.48	3.069(15)	135
O(2W)-H(1W2)...O(6)#1	0.99	1.80	2.753(12)	161
O(2W)-H(2W2)...O(5)	0.99	2.65	3.205(12)	116
O(2W)-H(2W2)...O(11)	0.99	1.79	2.473(13)	123
O(3W)-H(1W3)...O(13)	0.95	1.76	2.690(7)	167
O(3W)-H(2W3)...O(12)#7	1.01	1.82	2.761(7)	154
O(4W)-H(1W4)...O(3W)#4	1.04	1.86	2.823(9)	152
O(4W)-H(2W4)...O(13)#8	0.86	2.06	2.883(7)	160

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,z #2 x,y,z+1 #3 x-1/2,-y+1/2,-z+2

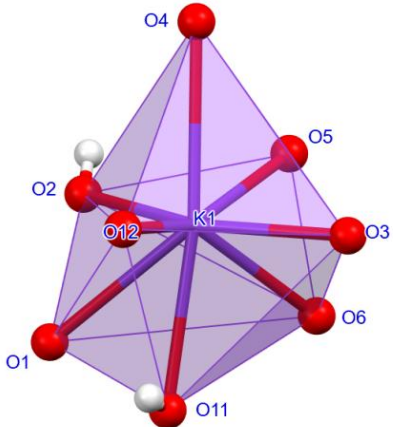
#4 -x+1,-y+1,z+1 #5 x+1/2,-y+1/2,-z+1 #6 -x+1/2,y-1/2,-z+1

#7 -x+1/2,y+1/2,-z+1 #8 -x+1,-y+1,z



Ammonium cation..... water/hydroxyl interactions

Table S15. Cation coordination in the Glc-1PH(NH₄)K crystal. The K cation occupancy is 31%

LK = 8		
K(1)-O(3)#1	2.725(10)	
K(1)-O(2)	2.725(10)	
K(1)-O(6)#2	2.857(9)	
K(1)-O(5)#2	2.908(9)	
K(1)-O(12)#3	3.017(9)	
K(1)-O(11)	3.026(10)	
K(1)-O(1)	3.125(10)	
K(1)-O(4)#4	3.274(10)	BICAPPED TETRAGONAL PRISM

Symmetry transformations used to generate equivalent atoms:

#1 $-x+1, y+1/2, -z+1/2$ #2 $-x+3/2, -y+1, z-1/2$

#3 $x-1/2, -y+3/2, -z+1$ #4 $-x+1/2, -y+1, z-1/2$

Hydrogen bonds O-H...O

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(2)-H(2O)...O(12)#2	0.82	1.90	2.698(3)	166
O(3)-H(3O)...O(6)#9	0.82	2.42	2.927(3)	121
O(4)-H(4O)...O(13)#10	0.82	1.94	2.751(3)	169
O(6)-H(6O)...O(12)#11	0.82	1.95	2.764(3)	171
O(11)-H(11)...O(13)#3	0.82	1.77	2.564(3)	162

Symmetry transformations used to generate equivalent atoms:

#2 $-x+3/2, -y+1, z-1/2$ #3 $x-1/2, -y+3/2, -z+1$

#9 $x-1/2, -y+1/2, -z+1$ #10 $x-1, y, z$

#11 $-x+1, y-1/2, -z+3/2$ #12 $x+1, y, z$

Table S16. Hydrogen bonds in the Glc-1P2(NH₄) crystalNH₄⁺.....O distances

N(1)...O(1)	2.973(4)	N(2) – on 2-fold axis	
N(1)...O(2)	3.016(4)	N(2)...O(2)	3.013(3)
N(1)...O(13)	2.906(4)	N(2)...O(4)#5	2.937(4)
N(1)...O(3)#1	2.949(4)	N(3) - disordered	
N(1)...O(4)#1	3.327(4)	N(3)...O(5)	2.806(12)
N(1)...O(2W)#3	3.093(6)	N(3)...O(5)#2	3.099(11)
N(1)...O(3WA)#4	2.850(10)	N(3)...O(6)#2	2.759(19)
N(1)...O(3WB)#4	2.691(10)	N(3)...O(2W)#5	2.940(12)

Geometry of hydrogen bonds

Donor acceptor	N-H (Å)	H....O (Å)	N...O (Å)	N-H...O (°)
N(1)-H(1N1)...O(2W)#3	0.96	2.23	3.093(6)	149
N(1)-H(2N1)...O(1)	0.96	2.29	2.973(4)	128
N(1)-H(2N1)...O(13)	0.96	2.16	2.906(4)	134
N(1)-H(3N1)...O(3WA)#4	0.95	2.18	2.850(10)	126
N(1)-H(3N1)...O(3WB)#4	0.95	1.82	2.691(10)	151
N(1)-H(3N1)...O(2)	0.95	2.52	3.016(4)	112
N(1)-H(4N1)...O(3)#1	0.94	2.40	2.949(4)	117
N(1)-H(4N1)...O(4)#1	0.94	2.46	3.327(4)	152
N(2)-H(1N2)...O(4)#5	0.95	2.00	2.937(4)	167
N(2)-H(2N2)...O(2)	0.95	2.07	3.013(3)	172
N(3)-H(1N3)...O(5)	1.02	1.85	2.806(12)	154
N(3)-H(2N3)...O(2W)#5	0.98	2.09	2.940(12)	144
N(3)-H(4N3)...O(5)#2	0.90	2.27	3.099(11)	153
N(3)-H(4N3)...O(6)#2	0.90	2.13	2.759(19)	126

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,z #2 -x+1,-y+1,z #3 x-1/2,-y+1/2,-z+1

#4 x-1/2,-y+1/2,-z+2 #5 x,y,z+1 #6 x+1/2,-y+1/2,-z+1

#7 -x+1/2,y+1/2,-z+2 #8 -x+1/2,y+1/2,-z+1