



Abstract Recovery of Waste Industrial Waters Containing Red Congo by Multifunctionalized Mesoporous Silica Nanomaterials[†]

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- + Presented at the 17th International Symposium "Priorities of Chemistry for a Sustainable Development" PRIOCHEM, Bucharest, Romania, 27–29 October 2021.

Keywords: platinum nanomaterials; functionalized mesoporous silica; congo red; porphyrin derivatives



Citation: Fratilescu, I.; Fagadar-Cosma, E. Recovery of Waste Industrial Waters Containing Red Congo by Multifunctionalized Mesoporous Silica Nanomaterials. *Chem. Proc.* 2022, 7, 19. https://doi.org/10.3390/ chemproc2022007019

Academic Editors: Mihaela Doni, Florin Oancea, Zina Vuluga and Radu Claudiu Fierăscu

Published: 2 March 2022

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Introduction: Colored wastewaters, containing toxic dyes, such as Congo red, methylene blue, malachite green, fuchsine B, bromothymol blue, must first focus on the removal of dyes, before any other recovery treatment. Comparing the efficiency of adsorbent materials used to remove dyes, functionalized silica derivatives, tailored to have specific surface areas exceeding $600 \text{ m}^2/\text{g}$, are among the best performing [1]. Particularly, for the removal of Congo red dye, silica and chitosan materials have also been shown in the literature to be effective adsorbents [2,3]. Materials and Methods: A set of mesoporous silica nanomaterials, functionalized either with Pt-metalloporphyrin (PtTAOPP-silica hybrid), or with PtNPs (PtNPs-silica hybrid) or bis-functionalized with both the same porphyrin-base and PtNPs (TAOPP-PtNPs)-silica hybrid, were successfully used as adsorbent materials for methylene blue, malachite green, fuchsine B, bromothymol blue and Congo red [1,4]. Results: A novel example of monitoring (by UV-vis spectrometry) the capacity of the three impregnated-silica hybrid materials, PtTAOPP-silica hybrid, PtNPs-silica hybrid and (TAOPP-PtNPs)-silica hybrid, in comparison with silica control, for Congo red removal from wastewaters, is given in Figure 1. From Figure 1, it is clear that the absorption intensity of the Congo red solution is significantly decreased, meaning that most of the dye was absorbed by the silica materials. After the Congo red was adsorbed, very interesting haystack aggregates, randomly distributed, were deposited on silica matrices, proving the absorption of the dye. In Figure 2, the atomic force microscopy (AFM) images of PtNPs-silica hybrid material are shown, after Congo red adsorption from waters containing an initial concentration of 5×10^{-5} M Congo red. The adsorption capacity of Congo red (for initial concentration of the dye: 5×10^{-5} M) realized by the tested silica materials, at a loading of 3.33 g/L, are: (a) *silica control*: $q_e = 8.25 \text{ mg/g}$; $\eta = 78.13\%$; (b) *PtTAOPP-silica hybrid*: $q_e = 8.43 \text{ mg/g}$; $\eta = 79.84\%$; (c) (*TAOPP-PtNPs*)-silica hybrid: $q_e = 8.73 \text{ mg/g}$; $\eta = 82.75\%$; (d) *PtNPs-silica hybrid*: $q_e = 8.02 \text{ mg/g}$; $\eta = 76.03\%$. Conclusions: As demonstrated in this study, all the functionalized silica materials have a good performance in removing red Congo dye, with the best result attaining a yield of 82%, registered for the bis-functionalized (TAOPP-PtNPs)-silica hybrid.

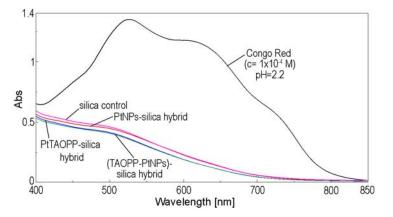


Figure 1. Overlapped UV-vis spectra of the remaining Red Congo concentration (initial conconcentration concentration 5×10^{-5} M Congo red), after 160 min of contact between the Congo red solution and the functionalized silica hybrid materials. The solutions were centrifugated and filtrated.

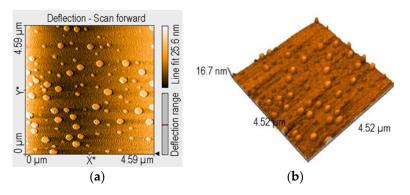


Figure 2. 2D (**a**) and 3D (**b**) AFM images of PtNPs-silica hybrid material after red Congo adsorption from waters containing 5×10^{-5} M Congo red.

Funding: This research was funded by UEFISCDI-Romania, grant number 76-PCCDI/2018, belonging to PNIII-Future and Emerging Technologies and partially to Programme 3-2021, from the Institute of Chemistry "Coriolan Dragulescu".

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not Applicable.

Data Availability Statement: All the data are available from the authors at ICT laboratories.

Conflicts of Interest: The authors declare no conflict of interest.

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