

## SUPPLEMENTARY MATERIAL

# **$\alpha$ -Amido Trifluoromethyl Xanthates: a New Class of RAFT/MADIX Agents**

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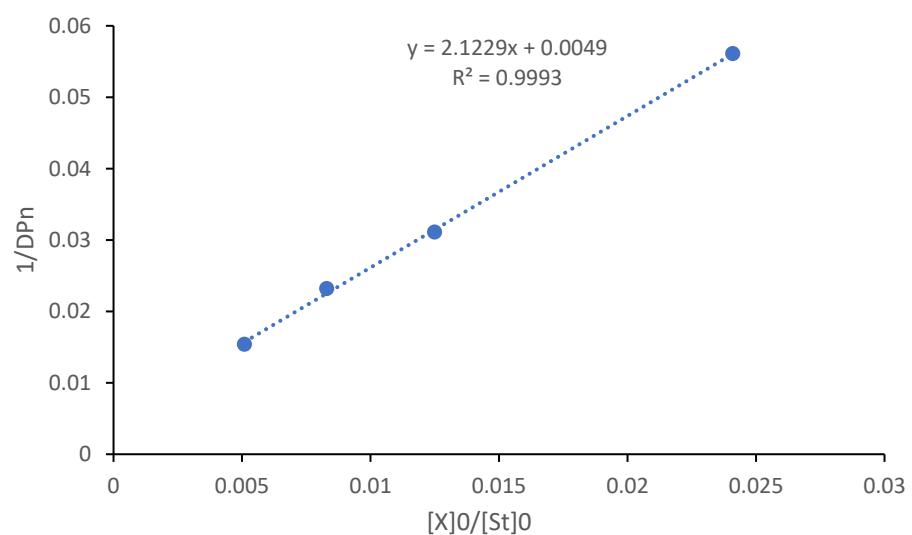
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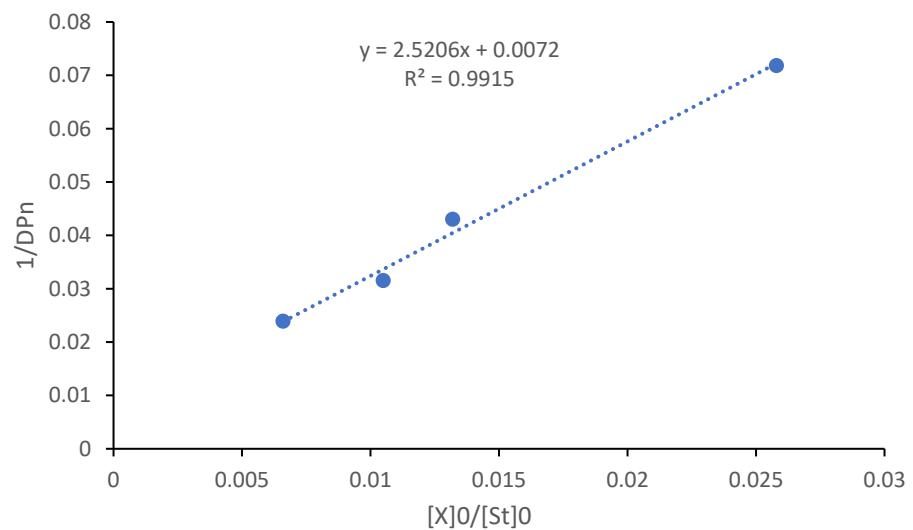
**Table S1.** Data used to determine transfer constants to xanthates  $C_{tr}$  (X) in RAFT/MADIX polymerization of styrene in bulk at 110°C (self-initiation).

Entry	Xanthate	[St] <sub>0</sub> /[X] <sub>0</sub>	t (min)	Conv (%)	$M_n$ (g/mol)	$D$
1		41	90	11.9	2060	2.10
2	<b>9</b>	80	90	10.1	3550	2.17
3		120	90	13.0	4680	2.26
4		197	90	12.1	6960	2.35
5		39	45	12.8	1700	1.85
6	<b>10</b>	76	60	8.4	2680	2.08
7		96	45	4.3	3560	2.24
8		151	45	7.9	4610	2.29
9		38	45	9.7	1980	1.85
10	<b>11</b>	74	60	11.1	3420	2.11
11		131	45	4.8	5420	2.21
12		160	45	8.1	6360	2.17
13		38	30	1.0	980	1.43
14	<b>13</b>	82	30	1.2	1730	1.68
15		120	30	1.3	2440	1.88
16		178	30	0.9	3800	2.10

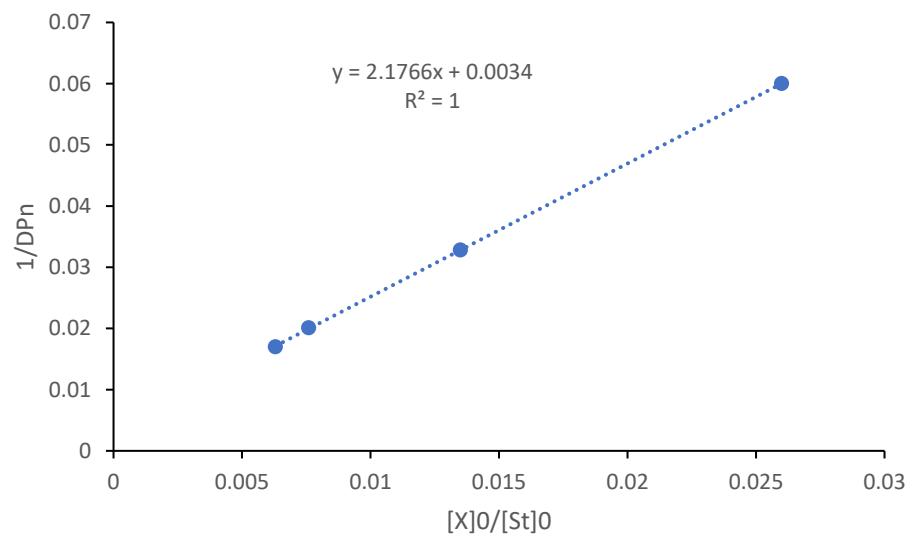
**Figure S1.** Mayo plot for the determination of chain transfer constant to xanthate **9** in styrene polymerization. Conditions of Table S1 (entries 1-4).



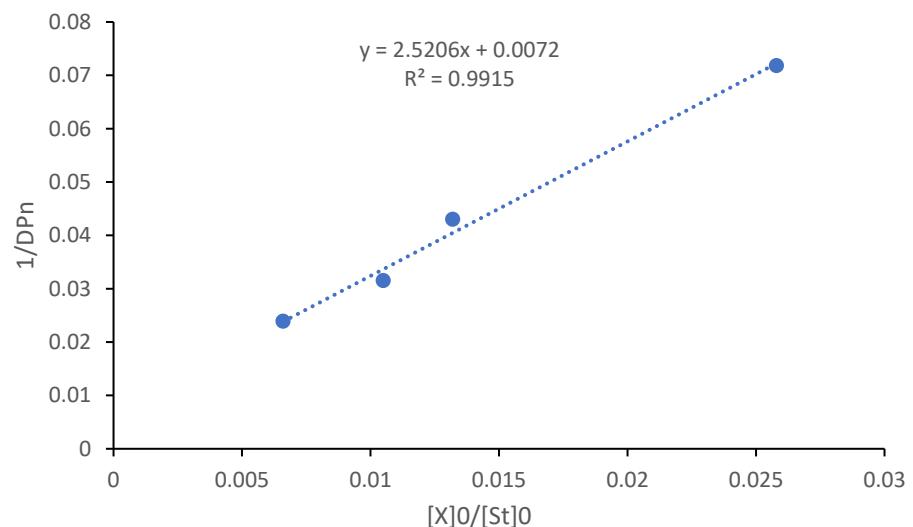
**Figure S2.** Mayo plot for the determination of chain transfer constant to xanthate **10** in styrene polymerization. Conditions of Table S1 (entries 5-8).



**Figure S3.** Mayo plot for the determination of chain transfer constant to xanthate **11** in styrene polymerization. Conditions of Table S1 (entries 9-12).



**Figure S4.** Mayo plot for the determination of chain transfer constant to xanthate **13** in styrene polymerization. Conditions of Table S1 (entries 13-16).



**Table S2.** **13**- and **14**-mediated RAFT/MADIX polymerization of styrene at 110°C (self-initiation).  $[St]_0/[X]_0=80$ .

Entry	Xanthate	t (h)	Conv (%)	$M_n$ (g/mol)	$M_{n\ th}$ (g/mol)	$\bar{D}$
1		0.5	0.8	2060	387	1.69
2		1	3.4	3550	601	1.75
3	<b>13</b>	2	9.6	4680	1119	1.87
4		5	28.3	6960	2690	2.29
5		24	96.0	8220	8361	2.39
6		1	4.12	1190	658	1.69
7	<b>14</b>	2	12.1	1470	1323	1.80
8		5	25.8	2570	2461	1.92
9		24	85.0	6700	7393	1.49

**Figure S5.** An example of SEC chromatograms during St polymerization in the presence of PSt-**14**, for a given  $[BPO]_0$  at 80°C. Plotting  $\ln(S/S_0)$  as a function of time gives access to  $k_{act}$ .

