

pH-responsive Doxorubicin-loaded Fe₃O₄@CaCO₃ Nanocomposites for Cancer Treatment

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- Figure S1.** DLS size distribution of Fe₃O₄@CaCO₃ (0.45 mg/mL of Fe₃O₄ synthesis) by number (top), volume (middle) and intensity (bottom) obtained by adding 0.45 mg/mL.2
- Figure S2.** DLS size distribution of Fe₃O₄@CaCO₃ by number (top), volume (middle), intensity (bottom) after 5 months of storage at 7 °C in deionized water. The particle size was determined by DLS (139 ± 5 nm, PDI of 0.33 ± 0.01).3
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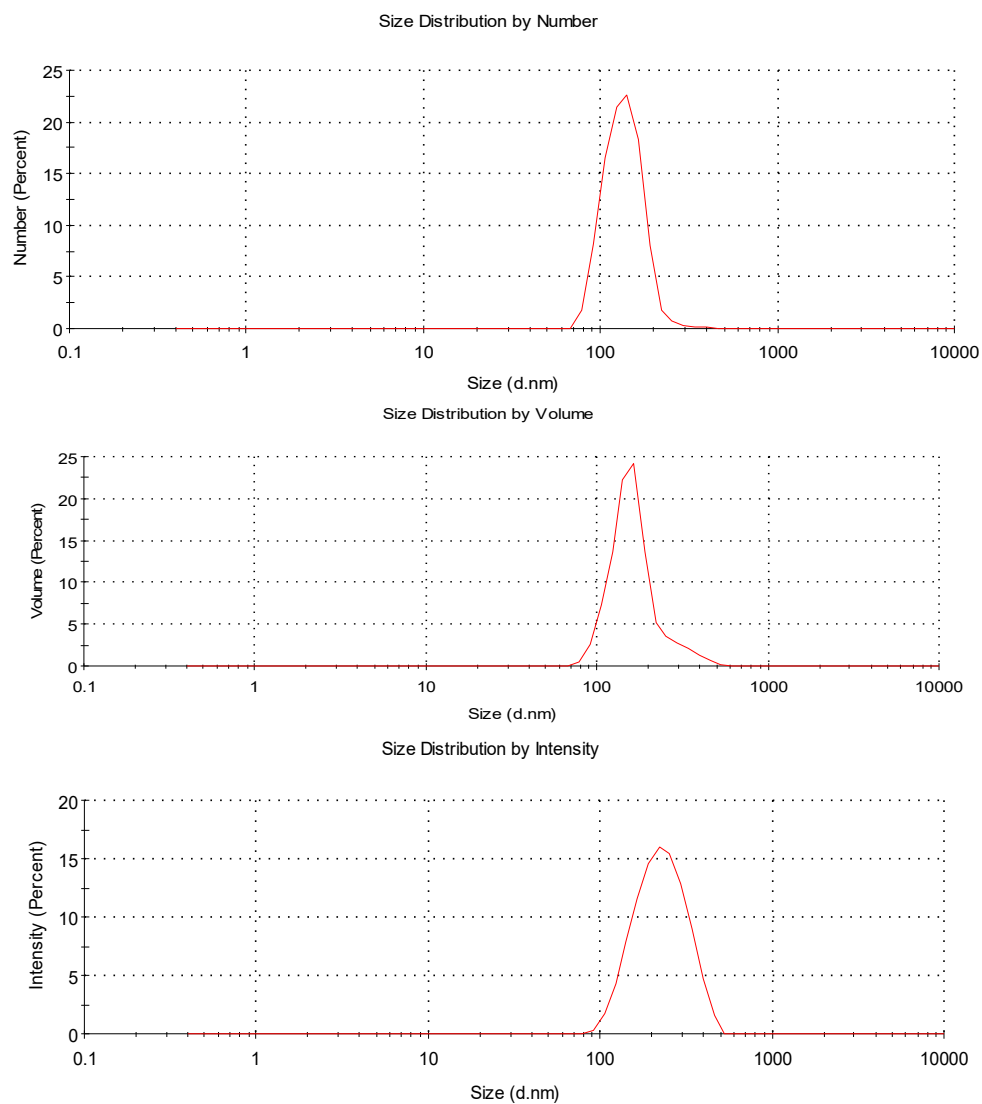


Figure S1. DLS size distribution of $\text{Fe}_3\text{O}_4@\text{CaCO}_3$ (0.45 mg/mL of Fe_3O_4 synthesis) by number (top), volume (middle) and intensity (bottom) obtained by adding 0.45 mg/mL.

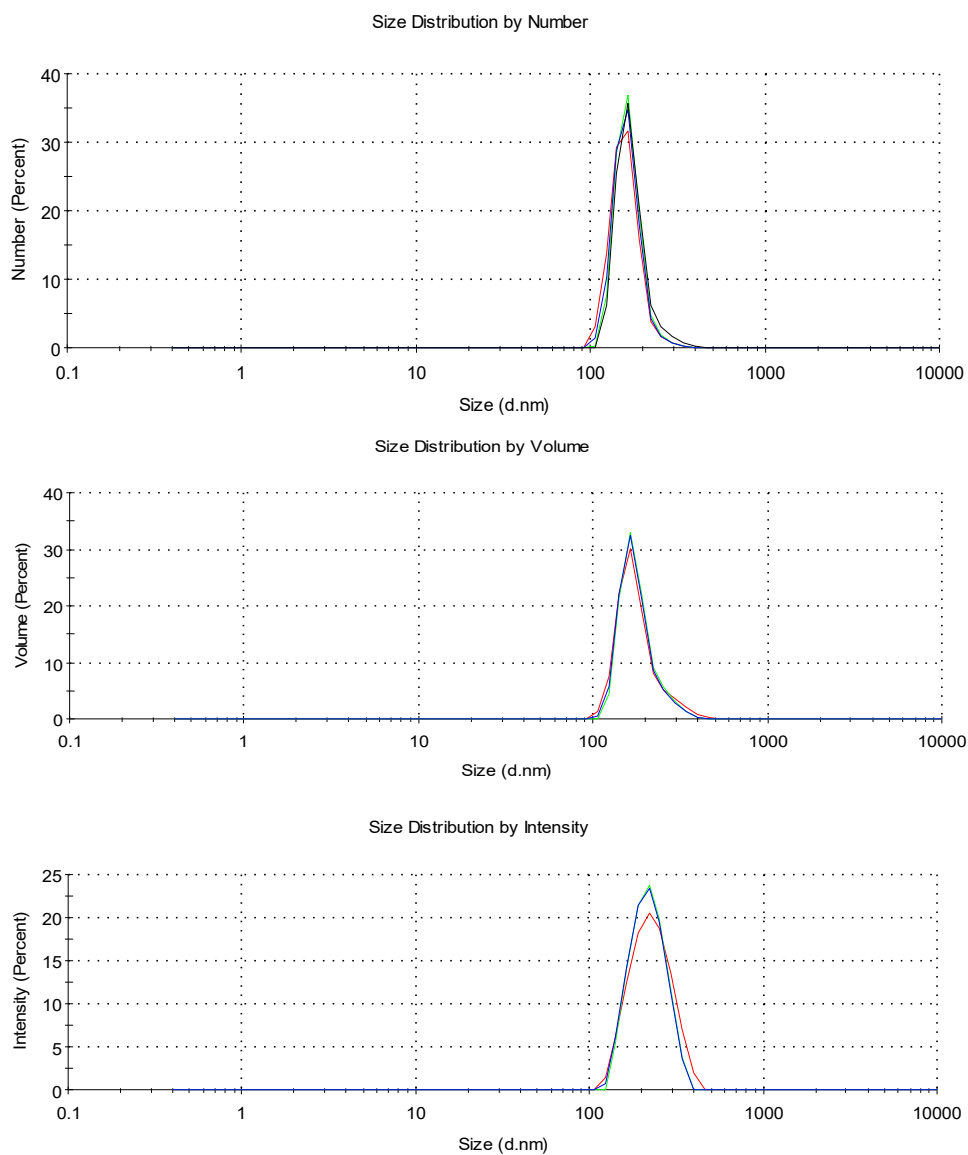


Figure S2. DLS size distribution of $\text{Fe}_3\text{O}_4@ \text{CaCO}_3$ by number (top), volume (middle), intensity (bottom) after 5 months of storage at 7 °C in deionized water. The particle size was determined by DLS (139 ± 5 nm, PDI of 0.33 ± 0.01).

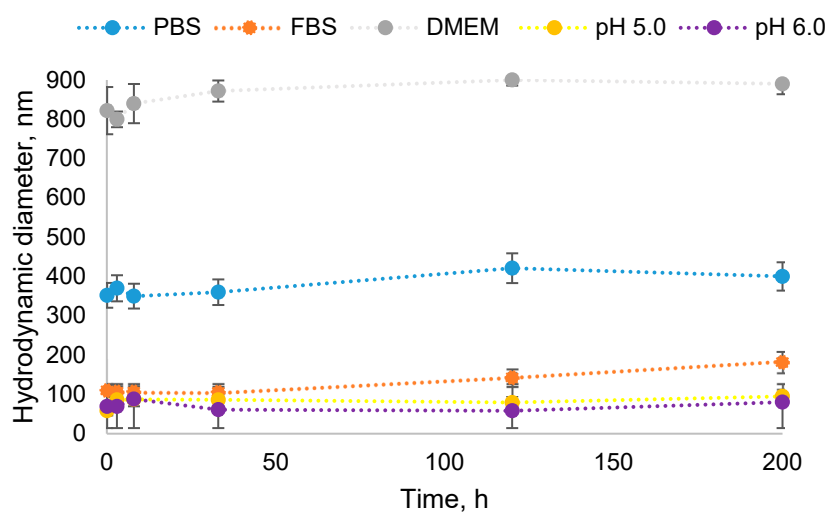


Figure S3. The hydrodynamic diameter by DLS of $\text{Fe}_3\text{O}_4@ \text{CaCO}_3$ in various solutions.

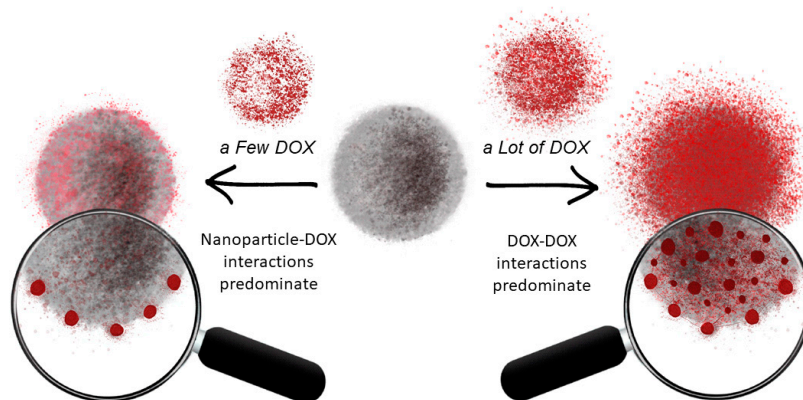
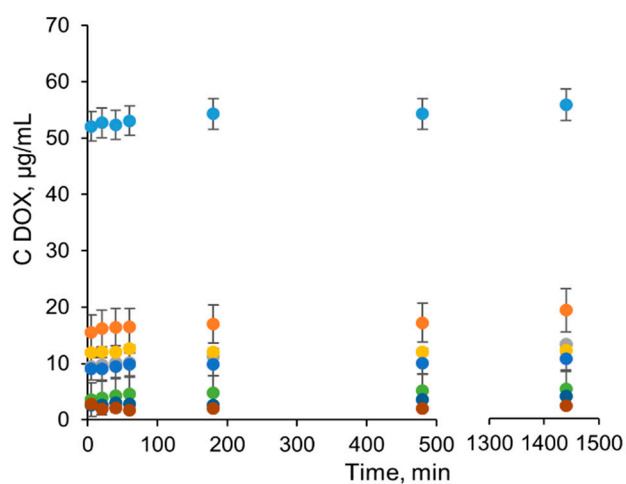


Figure S4. Possible nanoparticle interactions with doxorubicin (DOX).

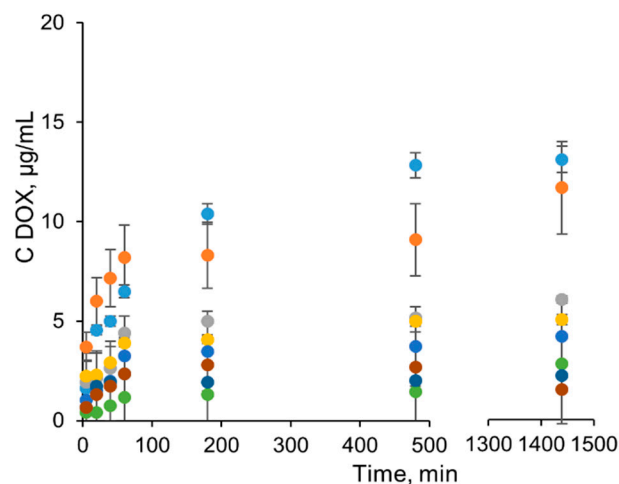
A pH 4.0

- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1900}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1045}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX525}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX295}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX160}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX73}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX45}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX25}$



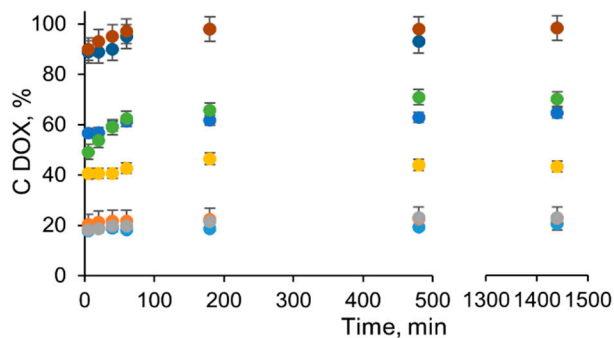
B pH 6.0

- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1900}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1045}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX525}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX295}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX160}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX73}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX45}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX25}$



C pH 4.0

- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1900}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1045}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX525}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX295}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX160}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX73}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX45}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX25}$



D pH 6.0

- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1900}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX1045}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX525}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX295}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX160}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX73}$
- $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX45}$ ● $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX25}$

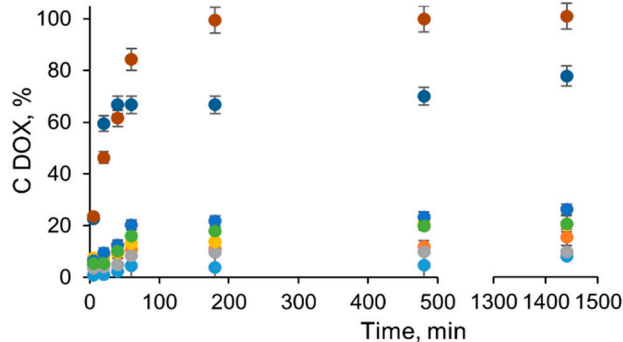


Figure S5. The proportion of DOX release from $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX}$ (capacity, 25–1900 $\mu\text{g}/\text{mg}$) at pH 4.0 (A, C); pH 6.0 (B, D).

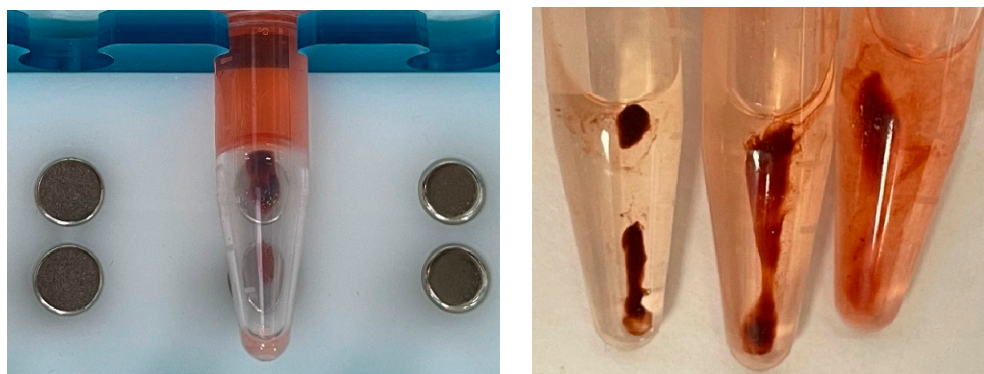


Figure S6. The confirmation of DOX-loading by photography. The left photograph also shows magnetic behavior on the magnetic tube rack.

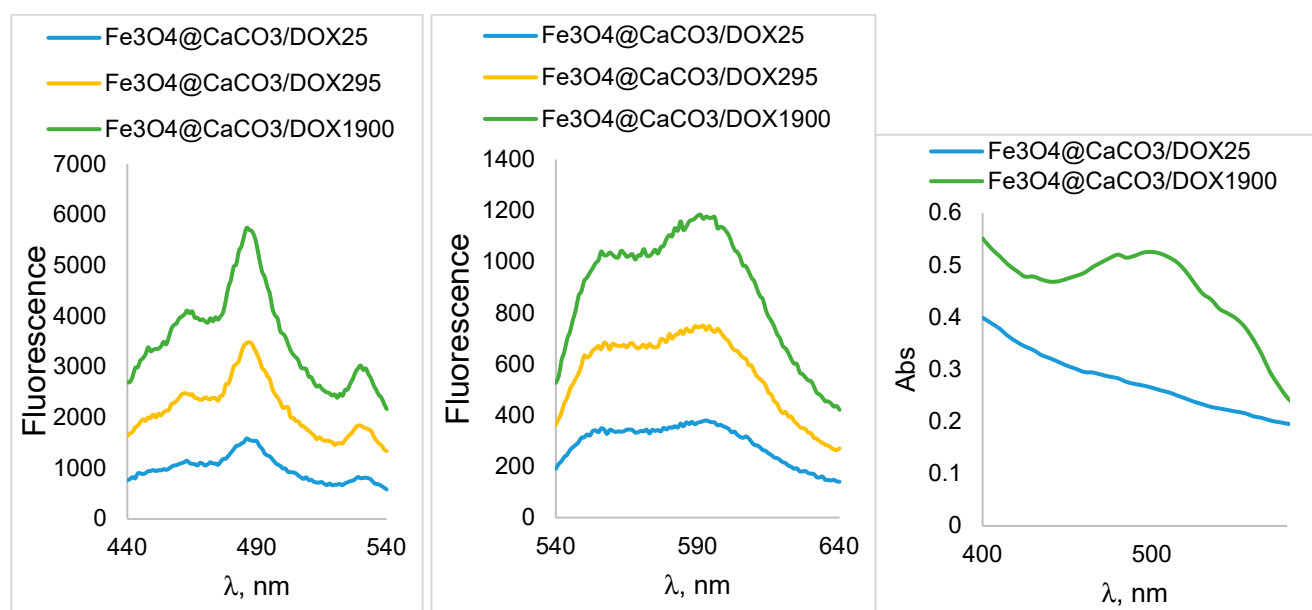


Figure S7. Fluorescence and UV-vis spectra of $\text{Fe}_3\text{O}_4@\text{CaCO}_3/\text{DOX}$ nanocomposites. The measurements were carried out using 100 μl of the solution on the Clariostar (BMG Labtech, Ortenberg, Germany).