OPEN ACCESS **MOLECULES** ISSN 1420-3049 www.mdpi.com/journal/molecules

Article

Synthesis of Novel 1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3aryl-1*H*-pyrazole-4-carbaldehydes

Huanan Hu^{1,*}, Changhua Ge¹, Lisheng Ding² and Anjiang Zhang^{3,*}

- ¹ College of Pharmaceutical and Chemical Engineering, Taizhou University, Linhai 317000, China; E-Mail: gechhua@163.com
- ² Chengdu Institute of Biology, Chinese Academy of Sciences, Chengdu 610041, China;
 E-Mail: lsding@cib.ac.cn
- ³ Ningbo Institute of Material Technology and Engineering, Chinese Academy of Sciences, Ningbo 315211, China
- * Author to whom correspondence should be addressed; E-Mails: hhn12192000@yahoo.com.cn (H.H.); anjiangzhang@gmail.com (A.Z.).

Received: 7 September 2010; in revised form: 29 September 2010 / Accepted: 2 October 2010 / Published: 25 October 2010

Abstract: A series of novel 1-[(2,6-dichloro-4-trifluoromethyl)phenyl]-3-aryl-1*H*-pyrazole-4-carbaldehydes **6a-i** were synthesized using the Vilsmeier-Haack reagent. The structures of all the title compounds have been confirmed by elemental analysis, ¹H-NMR and ¹³C-NMR and in addition, the structure of intermediate **5b** was investigated by X-ray crystallography.

Keywords: Vilsmeier-Haack reagent; phenylhydrazine; phenylhydrazone; 1*H*-pyrazole-4-carbaldehyde

1. Introduction

The application of the Vilsmeier-Haack (VH) reagent (POCl₃/DMF) for formylation of a variety of both aromatic and heteroaromatic substrates is well documented [1]. Besides this, the reagent has also been extensively used for effecting various chemical transformations with other classes of compounds. Many of these reactions have led to novel and convenient routes for the synthesis of various heterocyclic compounds [2]. A notable example that finds significant application in heterocyclic

chemistry is the synthesis of 4-formylpyrazoles from the double formylation of hydrazones with the VH reagent [3,4]. These observations, coupled with the recent developments in the simple synthesis of pyrazole derivatives [5], especially 4-functionalized 1,3-diphenylpyrazoles as antibacterial [6], antiinflammatory [7], antiparasitic [8], and antidiabetic drugs [9], prompted us to undertake the synthesis of pyrazole-4-carbaldehyde derivatives using the VH reagent.

It has been known for some time that fluorine atoms can lead to unexpected biological activity results due to the special properties of the fluorine atom, such as the high electronegativity of fluorine and the high carbon-fluorine bond energy [10]. As a consequence, trifluoromethyl-containing molecules have been found considerable utilization in the agrochemical and pharmaceutical industries [10–12], for example, pyrazole derivatives bearing trifluoromethyl groups, such as fipronil and analogs [13,14] are widely used phenylpyrazole insecticides applied in granular or bait form for residential and commercial control of turf grass pests, and celecoxib [15], widely prescribed to treat acute or chronic inflammation by providing symptomatic pain relief, are examples of such compounds. In this paper, we show a simple and efficient synthetic method using the VH reagent [4,16,17] that affords a series of novel 1-[(2,6-dichloro-4-trifluoromethyl)phenyl]-3-aryl-1H-pyrazole-4-carbaldehydes.

2. Results and Discussion

Phenylhydrazine **3** was synthesized from 2,6-dichloro-4-trifluoromethylphenylamine (**1**) through diazotization, reduction and hydration (Scheme 1). The reducing reagent plays an important role in the reaction. When the intermediate **2** was reduced by $SnCl_2$, an almost quantitative yield of product was obtained. Use of $Na_2S_2O_3$ or Zn as reducing reagents led to lower yields.





Phenylhydrazones **5** were next synthesized in almost quantitative yields by the reaction between the phenylhydrazine **3** and 3 or 4-substituted aryl methyl ketones, regardless of whether the ketones contained electron-withdrawing or electron-donating groups (Scheme 2).



X-ray diffraction

To verify the structural assignment compound **5b** was selected as an example for an X-ray diffraction study (Figure 1 and Table 1). The purified product **5b** was dissolved in 50 % ethanol/acetone (1:1 v/v) and kept at room temperature for 5 days and single crystals of **5b** were thus formed. CCDC 685556 contains the supplementary crystallographic data for this comound [18].

Figure 1. ORTEP drawing of the compound 5b showing the atom numbering scheme.



Table 1. Crystal data and summary of data collection and structure refinement.

| Compound | $C_{15} H_{10} Cl_3 F_3 N_2$ | |
|--|------------------------------|--|
| Color | Colorless | |
| Formula weight | 381.60 | |
| Crystal system | Monoclinic | |
| Temperature,° | 25(298K) | |
| Cell constants | | |
| a (Å) | 16.8770(7) | |
| b (Å) | 8.0054(8) | |
| c (Å) | 24.119(2) | |
| α (°) | 90 | |
| β (°) | 99.654(2) | |
| γ (°) | 90 | |
| Volume (Å ³) | 3212.5(4) | |
| Formula units | 8 | |
| Calculated density (g/cm ⁻³) 1.578 | | |

| F(000) | 1536 | |
|--|---|--|
| Absorption coefficient, mµ ⁻³ | 0.599 | |
| Limiting indices | -20<=h<=16, -7<=k<=9, -27<=l<=28 | |
| Reflections collected / unique | 8110 / 2841 [R(int) = 0.0402] | |
| Absorption correction | Semi-empirical from equivalents | |
| Max. and min. transmission | 0.9370 and 0.8314 | |
| Refinement method | Full-matrix least-squares on F ² | |
| Data / restraints / parameters | 2841 / 0 / 208 | |
| Goodness-of-fit on F ² | 1.015 | |
| Final R indices | $R_1 = 0.1641$, $wR_2 = 0.5152$ | |
| Largest diff. peak and hole (e Å ⁻³) | 0.869 and -0.919 | |

 Table 1. Cont.

The Vilsmeier cyclization of hydrazones could provide an efficient route for the preparation of 1H - pyrazole-4-carbaldehydes. The reaction was carried out at 80–90 °C for 4 h using 3 equiv. of the VH reagent, affording a series of novel 1-[(2,6-dichloro-4-trifluoromethyl)phenyl]-3-aryl-1*H*-pyrazole-4-carbaldehydes in good yields (Scheme 3). The results are shown in Table 2. 1*H*-Pyrazole-4-carbaldehydes **6** cannot be obtained if the DMF used is not anhydrous.

Scheme 3. Synthesis of 1*H*-pyrazole-4-carbaldehydes 6.



Table 2. Synthesis of 1*H*-pyrazole-4-carbaldehydes.

| Entry | Ar | Products(6) | $\text{Yield}(\%)^a$ |
|-------|----------------|-------------|----------------------|
| 1 | C_6H_5 | 6a | 89 |
| 2 | $4-ClC_6H_4$ | 6b | 88 |
| 3 | $3-ClC_6H_4$ | 6c | 83 |
| 4 | $4-BrC_6H_4$ | 6d | 86 |
| 5 | $3-BrC_6H_4$ | 6e | 82 |
| 6 | $4-MeOC_6H_4$ | 6f | 83 |
| 7 | $4-CF_3C_6H_4$ | 6g | 86 |
| 8 | $4-NO_2C_6H_4$ | 6h | 85 |
| 9 | | 6i | 81 |

^{*a*} Isolated yields with regard to the quantity of hydrazones **5**

3. Experimental

3.1. General

All melting points were determined on an XT-4A apparatus and are uncorrected. TLC was performed using precoated silica gel GF_{254} (0.25mm), column chromatography was performed using silica gel (200–300 mesh). The ¹H- and ¹³C-NMR spectra were measured at 25 °C at 300 and 75 MHz, respectively, on a Bruker Advance 300 spectrometer, using TMS as internal standard. *J*-values are given in Hz. The IR spectra were taken on a Bruker Vector 55 spectrometer. Elemental analyses were carried out with an EA 1112 elemental analyzer. All the reagents used were AR grade.

3.2. Synthesis of (2,6-Dichloro-4-trifluoromethyl)phenylhydrazine (3)

2,6-Dichloro-4-trifluoromethylphenylamine (**1**, 1.73 g, 7.5 mmol) was dissolved in concentrated HCl (10 mL) and water (10 mL) and cooled to 0 °C, sodium nitrite (0.62 g, 9.0 mmol) was added and the yellow solution was stirred at 0 °C for 2 h to get a solution of diazotizated compound **2**. SnCl₂ (2.85 g, 15 mmol) was dissolved in concentrated HCl (10 mL) and cooled to 7 °C, then the diazotizated solution of **2** was added slowly dropwise. After 1 h of reaction, the precipitate was filtered, washed with water and air-dried. The precipitate was neutralized by 25% NaOH in water to pH 8, the light yellow sediment was collected and air-dried to afford a brown solid **3** (1.53 g, 83%), m.p. 65–67 °C. ¹H-NMR (CDCl₃) δ : 7.52 (s, 2H, Ar-H), 5.85 (s, 1H, NH), 3.86 (br, 2H, NH₂); ¹³C- NMR (CDCl₃) δ : 137.1, 132.5 (q, *J* = 33.8 Hz), 130.3, 123.8, 122.3 (q, *J* = 271.8 Hz); IR (KBr) v: 3517, 3418, 1622, 1583, 1492, 1329, 1281, 1144, 883 cm⁻¹.

3.3. General procedure for synthesis of N-(2,6-Dichloro-4-trifluoromethyl)phenyl-N'-(1-phenyl-ethylidene) hydrazines **5a-i**

To a solution of compound **3** (12 mmol) in absolute C_2H_5OH (30 mL) was added substituted acetophenone **4** (10 mmol). Two drops of concentrated HCl were added to a stirred solution, and the mixture was refluxed for 1 h. The solid obtained on cooling was filtered, dried and recrystallized from C_2H_5OH to give compounds **5a-i**.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-(1-phenylethylidene)hydrazine* (**5a**): White solid, m.p. 90–91 °C; ¹H-NMR (CDCl₃) δ : 7.82 (d, *J* = 3.6 Hz, 2H, N'-Ar-H), 7.67 (s, 1H, NH), 7.58 (s, 2H, N-Ar-H), 7.34–7.42 (m, 3H, N'-Ar-H), 2.35 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 147.2, 140.8, 136.1, 132.5 (q, *J* = 33.8 Hz), 127.3, 126.8, 126.0, 125.8, 123.8, 122.3 (q, *J* = 271.8 Hz), 11.7; IR (KBr) v: 3350, 3072, 2927, 1603, 1538, 1492, 1390, 1351, 1311, 1114, 1009, 887 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(4-chlorophenyl)ethylidene]hydrazine* (**5b**): White solid, m.p. 118–120 °C; ¹H-NMR (CDCl₃) δ : 7.68–7.76 (m, 3H, N'-Ar-H and NH), 7.58 (s, 2H, N-Ar-H), 7.34–7.37 (m, 2H, N'-Ar-H), 2.33 (s, 3H, CH₃); ¹³C NMR (CDCl₃) δ : 146.1, 140.7, 136.3, 134.5 (q, *J* = 33.5 Hz), 128.3, 126.7, 126.2, 126.0, 123.9, 121.8 (q, *J* = 271.8 Hz), 11.8; IR (KBr) v: 3360, 3070, 2924, 1605, 1532, 1490, 1392, 1352, 1303, 1114, 1004, 884 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(3-chlorophenyl)ethylidene]hydrazine* (**5c**): White solid, m.p. 118–120 °C; ¹H-NMR (CDCl₃) δ : 7.83 (s, 1H, N'-Ar-H), 7.71–7.76 (m, 2H, N'-Ar-H and NH), 7.58 (s, 2H, N-Ar-H), 7.36–7.40 (m, 2H, N'-Ar-H), 2.32 (s, 3H, CH3); ¹³C-NMR (CDCl₃) δ : 146.1, 140.7, 136.3, 133.5 (q, *J* = 33.3 Hz), 129.2, 128.3, 127.5, 126.7, 126.1, 125.6, 123.9, 122.1 (q, *J* = 271.8 Hz), 12.1; IR (KBr) v: 3358, 3073, 2926, 1603, 1536, 1489, 1390, 1352, 1313, 1108, 1011, 886 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(4-bromophenyl)-ethylidene]hydrazine* (**5d**): White solid, m.p. 114–115 °C; ¹H-NMR (CDCl₃) δ : 7.65–7.69 (m, 3H, N'-Ar-H and NH), 7.58 (s, 2H, N-Ar-H), 7.48–7.52 (m, 2H, N'-Ar-H), 2.33 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 146.5, 141.7, 134.3, 132.5 (q, *J* = 33.5 Hz), 127.6, 127.1, 126.7, 125.6, 123.9, 121.9 (q, *J* = 271.8 Hz), 11.7; IR (KBr) v: 3362, 3074, 2928, 1606, 1540, 1495, 1393, 1354, 1308, 1106, 1012, 887 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(3-bromophenyl)ethylidene]hydrazine* (**5e**): White solid, m.p. 114–115 °C; ¹H-NMR (CDCl₃) δ : 7.94 (s, 1H, N'-Ar-H), 7.70–7.75 (m, 2H, N'-Ar-H and NH), 7.58 (s, 2H, N-Ar-H), 7.46–7.49 (m, 2H, N'-Ar-H), 2.32 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 147.1, 141.9, 139.3, 132.3 (q, *J* = 33.5 Hz), 128.3, 127.6, 127.1, 126.7, 126.0, 124.6, 123.9, 121.9 (q, *J* = 271.8 Hz), 11.9; IR (KBr) v: 3360, 3072, 2924, 1600, 1538, 1492, 1390, 1356, 1307, 1115, 1019, 889 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(4-methoxyphenyl)ethylidene]-hydrazine* (**5f**): White solid, m.p. 89–90 °C; ¹H-NMR (CDCl₃) δ : 7.81 (s, 1H, N'-Ar-H), 7.79 (s, 1H, N'-Ar-H), 7.67 (s, 1H, NH), 7.58 (m, 2H, N-Ar-H), 6.73–6.89 (m, 2H, N'-Ar-H), 3.83 (s, 3H, OCH₃), 2.32 (s, 3H, CH₃). ¹³C-NMR (CDCl₃) δ : 158.1, 147.5, 136.2, 133.6 (q, *J* = 33.6 Hz), 127.8, 126.2, 125.6, 124.7, 121.9 (q, *J* = 270.2 Hz), 114.2, 55.1, 11.7; IR (KBr) v: 3355, 3072, 2925, 1603, 1530, 1489, 1391, 1351, 1307, 1112, 1016, 887 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(4-trifluoromethylphenyl)-ethylidene]hydrazine* (**5g**): White solid, m.p. 165–166 °C; ¹H-NMR (CDCl₃) δ : 7.91 (d, *J* = 8.4 Hz, 2H, N'-Ar-H), 7.78 (s, 1H, NH), 7.59–7.65 (s, 4H, N' and N-Ar-H), 2.36 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 145.5, 144.7, 133.5 (q, *J* = 32.6 Hz), 130.4 (q, *J* = 32.1 Hz), 126.3, 125.8, 125.2, 124.6, 124.4, 123.2 (q, *J* = 270.2 Hz), 121.8 (q, *J* = 272.6 Hz), 11.9; IR (KBr) v: 3351, 3070, 2926, 1601, 1535, 1490, 1389, 1350, 1313, 1116, 1021, 889 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl)phenyl-N'-[1-(4-nitrophenyl)ethylidene]hydrazine* (**5h**): Yellowish solid, m.p. 146–147 °C; ¹H-NMR (CDCl₃) δ : 8.24 (d, *J* = 6.0 Hz, 2H, N'-Ar-H), 7.88–7.97 (m, 3H, N'-Ar-H and NH), 7.60 (s, 2H, N-Ar-H), 2.38 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 147.5, 144.3, 143.9, 140.4, 132.8 (q, *J* = 33.6 Hz), 126.5, 125.1, 124.7, 123.6, 120.9 (q, *J* = 272.6 Hz), 11.8; IR (KBr) v: 3320, 3073, 2926, 1583, 1513, 1393, 1339, 1307, 1117, 890 cm⁻¹.

N-(2,6-*Dichloro-4-trifluoromethyl*)*phenyl*-*N*'-[*1*-(6-*methoxy-naphthalen-2-yl*)-*ethylidene*]*hydrazine* (**5i**): White solid, m.p. 191–193 °C; ¹H-NMR (CDCl₃) δ : 8.11 (d, *J* = 6.2 Hz, 1H, N'-Ar-H), 8.00 (s, 1H, N'-Ar-H), 7.71–7.78 (m, 3H, N'-Ar-H and NH), 7.58 (s, 2H, N-Ar-H), 7.14-7.17 (m, 2H, N'-Ar-H) H), 3.84 (s, 3H, OCH₃), 2.43 (s, 3H, CH₃); ¹³C-NMR (CDCl₃) δ : 157.9, 147.5, 134.6, 133.2 (q, J = 32.5 Hz), 129.6, 128.2, 126.7, 126.2, 126.1, 125.3, 124.8, 123.8, 123.7, 121.7 (q, J = 271.6 Hz), 118.7, 105.5, 55.0, 11.7; IR (KBr) v: 3353, 3069, 2938, 1602, 1528, 1487, 1390, 1346, 1305, 1127, 1026, 889 cm⁻¹.

3.4. General procedure for synthesis of 1-[(2,6-dichloro-4-trifluoromethyl)phenyl]-3-aryl-1H-pyrazole-4- carbaldehydes **6a-i**

 $POCl_3$ (0.5 g, 3.0 mmol) was added dropwise to an ice-cold stirred solution of hydrazone 5 (1.0 mmol) in dry DMF (4 mL), the reaction mixture was allowed to attain room temperature and then heated at 80 °C for 4 h. The resulting mixture was poured onto crushed ice, neutralized with dilute sodium hydroxide and left standing overnight. The pale yellow precipitate obtained was purified by flash column chromatography with ethyl acetate-petroleum ether mixture to yield the products **6a-i**.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-phenyl-1H-pyrazole-4-carbaldehyde (**6a**): White solid, m.p. 147–149 °C; ¹H-NMR (CDCl₃) δ : 10.11 (s, 1H, CHO), 8.22 (s, 1H, pyrazole H), 7.79–7.83 (m, 4H, 1 and 3-Ar-H), 7.27–7.54 (m, 3H, 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 182.4, 153.8, 137.8, 135.0, 133.9 (q, *J* = 33.8 Hz), 129.0, 126.1, 125.8, 125.3, 124.6, 123.7, 122.6, 121.9 (q, *J* = 271.8 Hz); IR (KBr) v: 3107, 3062, 2843, 1695, 1602, 1531, 1403, 1310, 1131, 850, 719 cm⁻¹; Anal. Calcd. for C₁₇H₉Cl₂F₃N₂O: C, 53.01; H, 2.36; N, 7.27. Found: C, 53.30; H, 2.45; N, 7.33%.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(4-chlorophenyl)-1H-pyrazole-4-carbaldehyde (**6b**): White solid, m.p. 108–110 °C; ¹H-NMR (CDCl₃) δ : 10.08 (s, 1H, CHO), 8.22 (s, 1H, pyrazole H), 7.79–7.83 (m, 4H, 1 and 3-Ar-H), 7.45–7.48 (m, 2H, 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 183.6, 152.8, 138.8, 136.0, 133.8 (q, *J* = 33.8 Hz), 129.1, 128.5, 126.1, 125.8, 125.3, 123.7, 122.6, 121.9 (q, *J* = 271.8 Hz); IR (KBr) v: 3107, 3060, 2845, 1695, 1600, 1529, 1401, 1314, 1130, 858, 719 cm⁻¹; Anal. Calcd. for C₁₇H₈Cl₃F₃N₂O: C, 48.66; H, 1.92; N, 6.68. Found: C, 48.77; H, 2.03; N, 6.60%.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(3-chlorophenyl)-1H-pyrazole-4-carbaldehyde (6c): White solid, m.p. 105–106 °C; ¹H-NMR (CDCl₃) δ : 10.09 (s, 1H, CHO), 8.21 (s, 1H, pyrazole H), 7.83–7.91 (m, 3H, 1 and 3-Ar-H), 7.31–7.62 (m, 3H, 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 183.6, 152.8, 137.8, 136.0, 159.6, 133.8 (q, *J* = 33.8 Hz), 129.1, 128.5, 128.1, 126.1, 125.8, 125.2, 123.6, 122.7, 121.8 (q, *J* = 271.8 Hz); IR (KBr) v: 3106, 3062, 2840, 1696, 1602, 1530, 1403, 1310, 1134, 856, 719 cm⁻¹; Anal. Calcd. for C₁₇H₈Cl₃F₃N₂O: C, 48.66; H, 1.92; N, 6.68. Found: C, 48.81; H, 1.90; N, 6.80%.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(4-bromophenyl)-1H-pyrazole-4-carbaldehyde (6d): White solid, m.p. 116–118 °C; ¹H-NMR (CDCl₃) δ : 10.08 (s, 1H, CHO), 8.22 (s, 1H, pyrazole H), 7.74–7.79 (m, 4H, 1 and 3-Ar-H), 7.61–7.64 (m, 2H, 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 183.6, 152.8, 138.8, 136.0, 134.6, 133.8 (q, *J* = 33.8 Hz), 129.1, 128.5, 125.8, 125.3, 123.5, 122.6, 121.9 (q, *J* = 271.8 Hz); IR (KBr) v: 3108, 3063, 2841, 1697, 1604, 1532, 1407, 1315, 1131, 858, 719 cm⁻¹; Anal. Calcd. for C₁₇H₈BrCl₂F₃N₂O: C, 44.00; H, 1.74; N, 6.04. Found: C, 43.87; H, 1.79; N, 5.96%. *1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(3-bromophenyl)-1H-pyrazole-4-carbaldehyde* (**6e**): White solid, m.p. 110–112 °C; ¹H-NMR (CDCl₃) δ : 10.09 (s, 1H, CHO), 8.22 (s, 1H, pyrazole H), 7.79–7.83 (m, 4H, 1 and 3-Ar-H), 7.59–7.62 (m, 1H, 3-Ar-H), 7.35–7.38 (m, 1H, 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 182.9, 153.8, 139.0, 136.7, 134.2, 133.8 (q, *J* = 33.8 Hz), 129.1, 128.5, 126.1, 125.9, 125.3, 123.7, 123.3, 122.6, 121.8 (q, *J* = 271.8 Hz); IR (KBr) v: 3106, 3062, 2840, 1695, 1602, 1530, 1401, 1313, 1132, 856, 719 cm⁻¹; Anal. Calcd. for C₁₇H₈BrCl₂F₃N₂O: C, 44.00; H, 1.74; N, 6.04. Found: C, 43.70; H, 1.83; N, 6.36%.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(4-methoxyphenyl)-1H-pyrazole-4-carbaldehyde (**6f**): White solid, m.p. 75–77 °C; ¹H-NMR (CDCl₃) δ : 10.08 (s, 1H, CHO), 8.20 (s, 1H, pyrazole H), 7.76–7.79 (m, 4H, 1 and 3-Ar-H), 7.00–7.03 (m, 2H, 3-Ar-H), 3.87 (s, 3H, OCH₃); ¹³C-NMR (CDCl₃) δ : 183.2, 160.4, 154.5, 136.1, 134.9, 133.8 (q, *J* = 33.8 Hz), 130.0, 128.2, 125.8, 125.7, 122.8, 121.9 (q, *J* = 271.6 Hz), 113.9, 55.0; IR (KBr) v: 3110, 3060, 2963, 2850, 1695, 1600, 1530, 1403, 1312, 1130, 858, 719 cm⁻¹; Anal. Calcd. for C₁₈H₁₁C₁₂F₃N₂O₂: C, 52.07; H, 2.67; N, 6.75. Found: C, 52.10; H, 2.65; N, 6.78%.

1-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(4-trifluoromethylphenyl)-1H-pyrazole-4-carbaldehyde (**6g**): White solid, m.p. 88–90 °C; ¹H-NMR (CDCl₃) δ : 10.11(s, 1H, CHO), 8.25 (s, 1H, pyrazole H), 8.01–8.03 (m, 2H, 3-Ar-H), 7.74–7.83 (m, 4H, 1 and 3-Ar-H); ¹³C-NMR (CDCl₃) δ : 183.2, 151.8, 148.3, 139.1, 137.8, 136.9, 135.0, 133.9 (q, *J* = 33.8 Hz), 129.8 (q, *J* = 32.6 Hz), 129.0, 126.1, 123.7 (q, *J* = 270.1 Hz), 122.6, 121.9 (q, *J* = 271.8 Hz); IR (KBr) v: 3110, 3060, 2842, 1695, 1600, 1528, 1400, 1313, 1130, 858, 719 cm⁻¹; Anal. Calcd. for C₁₈H₈C₁₂F₆N₂O: C, 47.71; H, 1.78; N, 6.18. Found: C, 47.84; H, 1.85; N, 6.23%.

 $\begin{aligned} & I-[(2,6-Dichloro-4-trifluoromethyl)phenyl]-3-(4-nitrophenyl)-1H-pyrazole-4-carbaldehyde & (6h): \\ & Yellowish solid, m.p. 206-208 °C; ¹H-NMR (CDCl₃) \delta: 10.12 (s, 1H, CHO), 8.33-8.36 (m, 2H, 3-Ar-H), 8.27(s, 1H, pyrazole H), 8.13-8.15 (m, 2H, 3-Ar-H), 7.82 (s, 2H, 1-Ar-H); ¹³C-NMR (CDCl₃) \delta: 183.2, 151.8, 148.3, 139.1, 137.8, 136.9, 135.0, 133.8 (q,$ *J*= 33.8 Hz), 129.6, 126.1, 123.7, 122.6, 121.9 (q,*J* $= 271.6 Hz); IR (KBr) v: 3106, 3062, 2843, 1695, 1602, 1530, 1403, 1310, 1132, 858, 719 cm⁻¹; Anal. Calcd. for C₁₇H₈C₁₂F₃N₃O₃: C, 47.47; H, 1.87; N, 9.77. Found: C, 47.56; H, 1.93; N, 9.62%. \end{aligned}$

1-[(2,6-Dichloro-4-trifluoromethyl) phenyl]-3-(6-methoxy-naphthalen-2-yl)-1H-pyrazole-4-carb-index and a start of the st

aldehyde (**6i**): White solid, m.p. 144–146 °C; ¹H-NMR (CDCl₃) δ : 10.19(s, 1H, CHO), 8.25 (s, 1H, pyrazole H), 7.80–7.92 (m, 5H, 1 and 3-Ar-H), 7.19–7.23 (m, 3H, 3-Ar-H), 3.94 (s, 3H, OCH₃); ¹³C- NMR (CDCl₃) δ : 183.2, 158.2, 155.0, 136.4, 134.9, 134.7, 133.8 (q, *J* = 33.8 Hz), 129.7, 128.3, 128.2, 127.0, 126.4, 125.8, 125.4, 123.5, 122.1, 121.9 (q, *J* = 271.6 Hz), 119.2, 105.4, 55.0; IR (KBr) v: 3108, 3062, 2963, 2840, 1695, 1600, 1528, 1400, 1313, 1130, 858, 719 cm⁻¹; Anal. Calcd. for C₂₂H₁₃Cl₂F₃N₂O₂: C, 56.79; H, 2.82; N, 6.02. Found: C, 56.65; H, 2.93; N, 6.15%.

4. Conclusions

In conclusion, The Vilsmeier cyclization of hydrazones provided an efficient route for the preparation of 1*H*-pyrazole-4-carbaldehydes, and we have successfully developed a general method

for the synthesis in good yields of a series of 1-[(2,6-dichloro-4-trifluoromethyl)phenyl]-3-aryl-1*H*-pyrazole-4-carbaldehydes **6a-i** using the Vilsmeier-Haack reagent. The structures of all the title compounds have been confirmed by elemental analysis, ¹H-NMR and ¹³C-NMR and in addition, the structure of intermediate **5b** was confirmed by X-ray crystallography.

Acknowledgements

We are grateful for financial support from Taizhou University (Project No. 2010PY21) and the Nature Science Foundation of China (Project No. 20905057).

References and Notes

- 1. Jones, G.; Stanforth, S.P. The Vilsmeier reaction of fully conjugated carbocycles and heterocycles. *Org. React.* **2001**, *49*, 1-315.
- Majo, V.J.; Perumal, P.T. Intramolecular cyclization of azides by iminium species. a novel method for the construction of nitrogen heterocycles under Vilsmeier conditions. *J. Org. Chem.* 1998, 63, 7136-7142.
- 3. Kira, M.A.; Abdel-Rahman, M.O.; Gadalla, K.Z. The Vilsmeier reaction-III, cyclization of hydrazones to pyrazoles. *Tetrahedron Lett.* **1969**, *10*, 109-110.
- 4. Selvi, S.; Perumal, P.T. Facile synthesis of [1]benzopyrano[4,3-c]pyrazoles, 1-aryl-3-(2-formamidophenyl)pyrazoles and 1-aryl-3-phenyl-4-alkylpyrazoles using Vilsmeier reagent. *Indian J. Chem.* **2002**, *41B*, 1887-1893.
- 5. Elguero, J. *Comprehensive Heterocyclic Chemistry*; Katrizky, A., Ed.; Pregamon Press: Oxford, UK, 1984; p. 277.
- 6. El-Emary, T.I.; Bakhite, E.A. Synthesis and biological screening of new 1,3-diphenylpyrazoles with different heterocyclic moieties at position-4. *Pharmazie* **1999**, *54*, 106-111.
- 7. Bratenko, M.K.; Vovk, M.V.; Sydorchuk, I.J. Synthesis and antibacterial and antifungal activity of hydrazones and azines of 1-phenyl-3-R-4-formylpyrazole. *Farm. Zh.* **1999**, 68-71.
- Rathelot, P.; Azas, N.; EL-Kashef, H.; Delmas, F.; Giorgio, C.D.; David, P.T.; Maldonado, V.P. 1,3-Diphenylpyrazoles: Synthesis and antiparasitic activities of azomethine derivatives. *Eur. J. Med. Chem.* 2002, *37*, 671-679.
- 9. Cottineau, B.; Toto, P.; Marot, C.; Pipaud, A.; Chenault, J. Synthesis and hypoglycemic evaluation of substituted pyrazole-4-carboxylic acids. *Biorg. Med. Chem. Lett.* **2002**, *12*, 2105-2108.
- 10. Welch, J.T. Advances in the preparation of biologically active organofluorine compounds. *Tetrahedron* **1987**, *43*, 3123-3197.
- 11. Bertrand, F.; Pevere, V.; Quiclet-Sire, B.; Zard, S.Z. A xanthate transfer radical process for the introduction of the trifluoromethyl group. *Org. Lett.* **2001**, *3*, 1069-1071.
- 12. Song, J.J.; Tan, Z.; Reeves, J.T.; Gallou, F.; Yee, N.K.; Senanayake, C.H. N-heterocyclic carbene catalyzed trifluoromethylation of carbonyl compounds. *Org. Lett.* **2005**, *7*, 2193-2195.
- Caboni, P.; Sammelson, R.E.; Casida, J.E. Phenylpyrazole insecticide photochemistry, metabolism, and GABAergic action: Ethiprole compared with fipronil. J. Agric. Food Chem. 2003, 51, 7055-7061.

- 14. Sammelson, R.E.; Casida, J.E. Synthesis of a tritium-tabeled, fipronil-based, highly potent, photoaffinity probe for the GABA receptor. *J. Org. Chem.* **2003**, *68*, 8075-8079.
- Ranatunge, R.R.; Earl, R.A.; Garvey, D.S.; Janero, D.R.; Gordon Letts, L.; Martino, A.M.; Murty, M.G.; Richardson, S.K.; Schwalb, D.J.; Young, D.V.; Zemtseva, I.S. 3-(2-Methoxytetrahydrofuran-2-yl)pyrazoles: A novel class of potent, selective cyclooxygenase-2 (COX-2) inhibitors. *Bioorg. Med. Chem. Lett.* 2004, 14, 6049-6052.
- Sridhar, R.; Perumal, P.T.; Etti, S.; Shanmugam, G.; Ponnuswamy, M.N.; Prabavathyc, V.R.; Mathivanan, N. Design, synthesis and anti-microbial activity of 1*H*-pyrazole carboxylates. *Bioorg. Med. Chem. Lett.* 2004, 14, 6035-6040.
- Prakash, O.; Pannu, K.; Naithani, R.; Kaur, H. One-pot synthesis of oxime derivatives of 1,3diphenylpyrazole-4-carboxaldehydes from acetophenone phenylhydrazones using Vilsmeier-Haack reagent. *Synth. Commun.* 2006, *36*, 3479-3485.
- 18. CCDC 685556 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge, CB2 1EZ, UK; E-Mail: deposit@ccdc.cam.ac.uk.

Sample Availability: Samples of the compounds 6a-i are available from the authors.

 \bigcirc 2010 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution license (http://creativecommons.org/licenses/by/3.0/).