

Article

Two New Phenols from Scutellaria barbata

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Abstract: Two new phenols, 2(S)-2',7-dihydroxy-5,8-dimethoxyflavanone (1) and (S)-2-(4-hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one (2), were isolated from the ethanol extract of *Scutellaria barbata*. Their structures were elucidated on the basis of spectroscopic analysis.

Keywords: *Scutellaria barbata*; 2(*S*)-2',7-dihydroxy-5,8-dimethoxyflavanone; (*S*)-2-(4-hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one; Labiatae

1. Introduction

Scutellaria barbata (family Labiatae) and some other species from the same genus have been widely used in China, India, Nepal and other Asian countries for a long time as a traditional Chinese medicine or a folk remedy for the treatment of different diseases [1,2], and are also recognized as a source of *neo*-clerodane diterpenoids, as more than 150 *neo*-clerodane diterpenoids have been isolated from this genus [3-15]. BZL101 (Bezielle) is an aqueous extract from *Scutellaria barbata*. BZL101 is currently in phase II clinical trial in patients with advanced breast cancer. The phase I trial showed favorable toxicity profile and promising efficacy [16,17]. In previous investigation we reported four new *neo*-clerodane diterpenoids and structure revision of a series of 13-spiro *neo*-clerodanes [1]. In current study the flavonoid constituents of this plant have been examined. As described in the

experimental section, two new phenols: 2(S)-2',7-dihydroxy-5,8-dimethoxyflavanone (1) and (S)-2-(4-hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one (2), were isolated, together with seven known compounds, from the ethanol extract of whole plant of this plant, which was collected in Yunnan province, in southwest China.

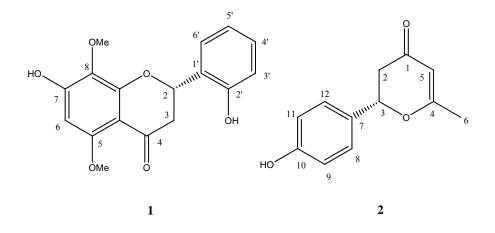
2. Results and Discussion

Compound 1, obtained as an amorphous powder, had a molecular formula of $C_{17}H_{16}O_6$ based on the positive high resolution electrospray ionization (ESI-MS), showing a quasi-molecular ion peak at m/z317.1024 (calcd. for C₁₇H₁₇O₆, 317.1025). The IR spectrum gave absorption bands corresponding to hydroxyl and conjugated carbonyl groups and aromatic rings. The UV spectrum was characteristic of the flavanone series. The flavanone nucleus was also confirmed by ¹H-NMR spectrum, in which the signals due to the C-3 and C-2 protons were observed an ABX system at 2.58 (1H, dd, 2.8 and 16.4 Hz), 2.92 (1H, dd, J = 12.8 and 16.4 Hz) and 5.61 (1H, dd, J = 2.8 and 12.8 Hz) [2]. The ¹H-NMR spectrum further showed the presence of two methoxyls (3.62, 3.69 ppm), two hydroxyls (9.91, 10.45 ppm, no chelated hydroxyl). In the aromatic region of the spectrum, the remaining five protons occurred as a singlet (6.13 ppm, 1H), for the A-ring proton and two doublets (7.43 ppm, 1H, J = 7.0 Hz; 6.86 ppm, 1H, J = 7.7 Hz) and two multiplets (centered at 6.87 ppm, 1H; 7.18 ppm, 1H) for the B-ring protons [2]. The HMBC correlations between H-6 (6.13 ppm), H-2 (5.61 ppm), H-3a (2.92 ppm), and H-3b (2.58 ppm) and carbonyl (188.5 ppm), 2'-OH (9.91 ppm) and C-2' (154.4 ppm), C-5 OMe (3.69 ppm) and C-5 (157.3 ppm), C-8 OMe (3.62 ppm) and C-8 (129.2 ppm), 7-OH (10.45) and C-6 (93.3 ppm), C-7 (157.1 ppm) and C-8 (129.2 ppm), 2'-OH (9.91 ppm) and C-1' (125.4), C-2' (154.4 ppm) and C-3'(115.5 ppm) were observed. It allowed us to position the two methoxyls in the A-ring at the C-5 and C-8. The chemical shifts and splitting patterns of the B-ring protons suggested that the B-ring is substituted at the 2'-position by a hydroxyl. The arrangement of the substituent in the B-ring was also supported by ¹³C-NMR spectrum. It is known that flavanones having 2(S)-configuration exhibit a positive Cotton effect due to $n-\pi^*$ transition (~ 330 nm) and a negative Cotton effect due to π - π * transition (270-290 nm) in the circular dichroism (CD) spectra [18]. The CD curve of 1 exhibited positive and negative maxima at 330 and 285 nm, respectively, which established the 2-(S)-configuration. From these results, the structure of 1 was determined to be 2(S)-2',7-dihydroxy-5,8-dimethoxyflavanone (Figure 1).

Compound **2** was obtained as an amorphous powder. It had a molecular formula of $C_{12}H_{12}O_3$ based on the positive high resolution electrospray ionization (ESI-MS), showing a quasi-molecular ion peak at m/z 227.0683 (calcd. for $C_{12}H_{12}O_3Na$, 227.0684). The ¹³C-NMR spectrum revealed 12 carbon resonances, including one carbonyl carbon at δ_C 191.8, one trisubstituted double bond at δ_C 174.0 (quaternary carbon) and 104.4 (CH), aromatic carbons at δ_C 157.7 (quaternary carbon), 128.7 (quaternary carbon), 128.3 (2×CH), 115.2 (2×CH), one oxygen-bearing carbon at δ_C 80.2 (CH), one methylene carbon at δ_C 41.4 (CH₂), and one methyl at δ_C 20.6 (CH₃). The ¹H-NMR spectrum further showed the presence of one methoxyl (1.99 ppm), one hydroxyl (9.61 ppm), one methylene (2.36 ppm, dd, J = 3.4, 16.6 Hz; 2.83 ppm, dd, J = 14, 16.6 Hz), one proton for the double bond (5.34 ppm, s, 1H), one methine (5.36 ppm, m, 1H). In the aromatic region of the spectrum, the four protons occurred as two doublets (7.27 ppm, 2H, J = 7.5 Hz; 6.76 ppm, 2H, J = 7.5 Hz). The HMBC correlations between H-5 (5.34 ppm), H-2 (2.83, 2.36 ppm) and carbonyl (191.8 ppm), 10-OH (9.61 ppm), H-9, 11 (6.76 ppm), H-8, 12 (7.27 ppm) and C-10 (157.7 ppm), H-9, 11 (6.76 ppm), H-8, 12 (7.27 ppm), H-3 (5.36 ppm) and C-7 (128.7 ppm), 10-OH (9.61 ppm), H-9, 11 (6.76 ppm), H-8, 12 (7.27 ppm) and C-9, 11 (115.2 ppm), H-8, 12 (7.27 ppm), H-2a (2.83 ppm) and C-3 (80.2 ppm), H-5 (5.34 ppm) and C-2 (41.4 ppm) and C-6 (20.6 ppm) were observed. The optical rotation of compound **2** is -21.7, which established the (*S*)-configuration by data comparison with literature [19]. Thus the structure of **2** was established as (*S*)-2-(4-hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one (Figure 1). It was isolated as a new natural product in present study, and synthesized in a previous report [20].

Compounds **3-9** were known compounds and were identified as 2',4'-dihydroxy-2,3',6'-trimethoxychalcone [2], apigenin 5-O-beta-D-glucopyranoside [21], 4',5-dihydroxy-3',5',6,7-tetramethoxyflavone [22], 4'-hydroxywogonin [23], 6-methoxynaringenin [24], 2',5,7-trihydroxy-8-methoxyflavanone [25], 7-hydroxy-2',5, 8-trimethoxyflavanone [2], respectively, by comparison of their spectral data with literature data.

Figure 1. Structures of compounds 1 and 2.



3. Experimental

3.1. General

Optical rotations were measured on a Jasco P-1020 (Jasco International Co., Ltd., Tokyo, Japan) automatic digital polarimeter. UV spectra were obtained using a Shimadzu UV-2401A spectrophotometer. IR spectra were recorded using a Bruker Tensor 27 FT-IR (Bruker Optics GmbH, Ettlingen, Germany) spectrophotometer with KBr pellets. NMR spectra were carried out on either a Bruker DRX-500 or AM-400 (Bruker BioSpin GmbH, Rheinstetten, Germany) spectrometers with the deuterated solvent as an internal standard. ESI-MS (including HR-ESI-MS) were performed on an API-Qstar-Pulsar i (MDS Sciex, Concord, ON, Canada) mass spectrometer. Column chromatography was performed on Silica gel (200-300 mesh, Qingdao Marine Chemical Inc., Qingdao, China) and RP-18 (20–45 μ m, Fuji Silysia Chemical Ltd., Kasugai, Achi, Japan). Fractions were monitored by TLC (GF 254, Qingdao Marine Chemical Inc., Qingdao, China), and spots were visualized by heating silica gel plates sprayed with 10 % H₂SO₄ in EtOH.

3.2. Plant material

The whole plants of *S. barbata* were collected in Xinping County of Yunnan Province, China in March 2008, and identified by Mr. Yu Chen of Kunming Institute of Botany, CAS. A voucher specimen (No. BBP2010010SB) was deposited at BioBioPha.

3.3. Extraction and isolation

Dried and powdered *S. barbata* whole plants (9,500 g) were extracted three times with 95% EtOH (60 L) at room temperature for two days each time. The extract was concentrated to give a residue (1,000 g), which was fractionalized by silica gel column chromatography eluted with a solvent system of petroleum ether (PE)/acetone (20:1, 10:1, 6:1, 3:1, 1:1, 0:1) and then pure methanol to yield fractions A-G. Fraction D eluted by 33% acetone was separated on silica gel using a solvent system of CHCl₃/MeOH (30:1 \rightarrow 8:1) to obtain subfraction, it was further isolated and purified by silica gel, Sephadex LH-20 (CHCl₃/MeOH, 1:1) and MCI (50% \rightarrow 100% MeOH in water) columns to afford compounds **5** (46 mg), and **9** (235 mg). Fraction E eluted by 50% acetone was separated on silica gel using a solvent system of CHCl₃/MeOH (30:1 \rightarrow 8:1) to obtain subfraction E eluted by 50% acetone was separated on silica gel using a solvent system of CHCl₃/MeOH (30:1 \rightarrow 8:1) to obtain subfraction E eluted by 50% acetone was separated on silica gel using a solvent system of CHCl₃/MeOH (30:1 \rightarrow 8:1) to obtain subfraction E eluted by 50% acetone was separated on silica gel using a solvent system of CHCl₃/MeOH (30:1 \rightarrow 8:1) to obtain subfractions I and II. Subfraction I was further isolated and purified by silica gel, Sephadex LH-20 (CHCl₃/MeOH, 1:1) and MCI (50% \rightarrow 100% MeOH in water) columns to afford compounds **1** (67 mg), **2** (70 mg), **3** (94 mg), **4** (543 mg), **6** (86 mg), **7** (128 mg), **8** (3 mg).

3.4. Characterization of compound 1

2(*S*)-2',7-Dihydroxy-5,8-dimethoxyflavanone (**1**), powder; $[\alpha]_{D}^{20}$ –19.1 (*c* 0.021, MeOH). UV (MeOH): λ_{max} (log ε) 207 (3.76), 285 (3.02) nm. IR (KBr): ν_{max} 3266, 1614, 1583, 1510, 1458, 1416, 1367, 1270, 1099, 992, 748 cm⁻¹. ¹H-NMR (DMSO) and ¹³C NMR (DMSO) data, see Table 1. HRMS [(+)ESI]: *m/z* 317.1024 (calcd. 317.1025 for C₁₇H₁₇O₆, [M + H]⁺).

3.5. Characterization of compound 2

(*S*)-2-(4-Hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one (**2**), powder; $[\alpha]_{D}^{20}$ –21.7 (*c* 0.030, MeOH). ¹H NMR (DMSO) and ¹³C NMR (DMSO) data, see Table 1. HRMS [(+)ESI]: *m/z* 227.0683 (calcd. 227.0684 for C₁₂H₁₂O₃Na [M + Na]⁺).

Position	1		2	
	$\delta_{ m H}$	$\delta_{ m C}$	$\delta_{ m H}$	$\delta_{ m C}$
1				191.8 (C)
2	5.61 (dd, 2.8, 12.8)	74.1 (CH)	2.36 (dd, 3.4, 16.6)	41.4 (CH ₂)
			2.83 (dd, 14, 16.6)	
3a	2.58 (dd, 2.8, 16.4)	43.8 (CH ₂)	5.36 (m)	80.2 (CH)
3b	2.92 (dd, 12.8,16.4)			
4		188.5 (C)		174.0 (C)
4a		104.7 (C)		

Table 1. ¹ H	H and ¹³ C NMR	data of 1	and 2 in DMS	O (δ in ppm	, J in Hz).
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	1	able I. Cont.		
5		157.3 (C)	5.34 (s)	104.4 (CH)
6	6.13 (s)	93.3 (CH)	1.99 (s)	20.6 (CH ₃)
7		157.1 (C)		128.7 (C)
8		129.2 (C)	7.27 (d, 7.5)	128.3 (CH)
8a		156.9 (C)		
9			6.76 (d, 7.5)	115.2 (CH)
10				157.7 (C)
11			6.76 (d, 7.5)	115.2 (CH)
12			7.27 (d, 7.5)	128.3 (CH)
1'		125.4 (C)		
2'		154.4 (C)		
3'	6.86 (d, 7.7)	115.5 (CH)		
4'	7.18 (m)	129.4 (CH)		
5'	6.87 (m)	119.4 (CH)		
6'	7.43 (d, 7.0)	126.8 (CH)		
5-OCH ₃	3.69 (s)	55.8 (CH ₃)		
8-OCH ₃	3.62 (s)	60.6 (CH ₃)		
7-OH	10.45 (s)			
2'-ОН	9.91 (s)			
10-OH			9.61(s)	

Table 1. Cont.

4. Conclusions

In conclusion, two new phenols, 2(S)-2',7-dihydroxy-5,8-dimethoxyflavanone (1) and (*S*)-2-(4-hydroxyphenyl)-6-methyl-2,3-dihydro-4H-pyran-4-one (2), were isolated from the ethanol extract of *Scutellaria barbata*. The discovery of compounds 1-2 is a further addition to the diverse plant phenolic compounds. Their biological activities are evaluating in progress.

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Sample Availability: Samples of the compounds **1** and **2** from *Scutellaria barbata* are available from the authors.

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