

Article

Multi-Component One-Pot Synthesis and Antimicrobial Activities of 3-Methyl-1,4-diphenyl-7-thioxo-4,6,8,9-tetrahydro-pyrazolo[5,4-*b*]pyrimidino[5,4-*e*]pyridine-5-one and Related Derivatives

Talaat I. El-Emary * and Shawkat A. Abd El-Mohsen

Department of Chemistry, Faculty of Science, Assiut University, Assiut 71516, Egypt

* Author to whom correspondence should be addressed; E-Mail: emarytalaat@yahoo.com.

Received: 24 October 2012; in revised form: 25 November 2012 / Accepted: 27 November 2012 /

Published: 6 December 2012

Abstract: The synthesis of 3-methyl-1,4-diphenyl-7-thioxo-4,6,8,9-tetrahydropyrazolo[5,4-*b*] pyrimidino[5,4-*e*]pyridine-5-one (6) was achieved by two different one-pot multi-component synthesis (one-pot three-component and one-pot four component synthesis). Mono and dialkylation of 6 under different conditions gave compounds 7–11. The hydrazine 12 produced from reaction of 9 with N₂H₄ was subjected to reactions with some aromatic aldehydes, ethyl acetoacetate, acetyl acetone, ethyl cyanoacetate and triethyl orthoformate to give 13–17, respectively. Compound 12 upon reaction with CS₂, nitrous acid, benzoin, chloroacetone and phenacyl bromide gave 18,20,21,22. Alkylation of 18 with ethyl iodide, ethyl chloroacetate and phenacyl bromide gave 19a–c. The antibacterial and antifungal activities of selected derivatives were evaluated.

Keywords: one-pot reactions; pyrazolo[5,4-*b*]pyrimidino[5,4-*e*]pyridines; synthesis; anti-microbial activity

1. Introduction

Multicomponent domino reactions (MDRs), particularly those performed in aqueous media, have become an increasingly useful tool for the synthesis of chemically and biologically important compounds because of their convergence, atom economy, and other suitable characteristics from the point of view of green chemistry [1–10]. Pyrazoles are excellent precursors for the synthesis of condensed polyfunctionally substituted heterocycles [11–16]. Pyrazolo-annulated heterocycles such as

pyrazolopyridopyrimidines have attracted considerable interest because their derivatives display a wide range of pharmacological activities, e.g., as anticonvulsants [17], antiproliferative agents [18], anti-inflammatories and analgesic agents [19]. In addition these types of compounds are inhibitors of cyclic guanosine-3',5'-monophosphate phosphodiesterase (cGMP PDE), and are thereby agents against erectile dysfunction [20]. They also show miscellaneous biological properties such as virucidal, anticancer, fungicidal, bactericidal and vasodilatory activities [21]. For all the benefits mentioned above and as part of our program investigating syntheses using pyrazole and fused pyrazoles that have biological importance [22–33], we report herein the synthesis of pyrazolo[5,4-*b*]pyrimidino[5,4-*e*]-pyridinethiones in a one-pot four component environmental friendly method in light of recently reported methods [34–37].

2. Results and Discussion

2.1. Chemistry

We first describe a comparison between two methods for the construction of pyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (6), a one-pot four-component domino reaction of phenyl hydrazine (1), 3-aminocrotononitrile (2), benzaldehyde (3) and thiobarbituric acid (4) in water in the presence of one equivalent of p-toluenesulfonic acid (p-TSA) (Method A) and the three-component reaction method involving 5-amino-3-methyl-1-phenylpyrazole (5), benzaldehyde (3) and thiobarbituric acid (4) in presence of p-TSA under solvent-free conditions [35] (Method B).

The one-pot four-component method A showed some crucial advantages, such as short reaction time, excellent yield and high purity, which makes it more efficient and broadly applicable. The percentage yield and the reaction time of the one-pot four-components method in comparison with the one-pot three-component one to produce compound 6 was found to be 91/66% and 1 h/10 h, respectively (Scheme 1).

Scheme 1. Synthetic pathways for compound 6.

The plausible mechanism for the formation of compound 6 is proposed (Scheme 2) and it is in agreement with that proposed in the literature [38]. The domino sequence of reactions is presumably triggered by the formation of 5-amino-3-methyl-1-phenylpyrazole (5) from the acid-catalyzed reaction of phenylhydrazine (1) with 3-aminocrotononitrile (2). The readily formed 5 reacts *in situ* with benzaldehyde to afford intermediate A. The later reacted with thiobarbituric acid under the acidic conditions to presumably furnish the intermediate B, which subsequently undergoes annulations leading to the final intermediate C. This final intermediate gave compound 6 upon losing a molecule of water.

Scheme 2. Mechanistic pathways for the formation of compound **6**.

The structural features of compound 6 were elucidated from its spectral and analytical data. Thus, the IR spectrum revealed absorption bands at 3310, 3290, 3240, 1690, 1345 cm⁻¹characteristic for three NH, C=O and C=S groups, respectively. The 1 H-NMR (DMSO- d_{6}) displayed three NH singlets (exchangeable with D₂O) at 13.8, 12.2 and 9.4 ppm. A characteristic singlet peak for the pyridine CH-4 appeared at 5.57 ppm. Several alkylated derivatives were obtained from the versatile compound 6. Thus, upon treatment of 6 with ethyl iodide, ethyl chloroacetate, phenacyl bromide, chloroacetone, chloroacetamide and chloroacetonitrile in the presence of anhydrous sodium acetate in refluxing ethanol, the S-alkylated derivatives 7a-f were obtained. The IR spectral data of compounds 7a-f displayed no absorption band for C=S and showed bands at 1280-1345 cm⁻¹confirming the S-alkylation, as well as, compounds 7b-e revealed absorption bands for C=O at 1690-1735 cm⁻¹. A characteristic absorption band at 2240 cm⁻¹ has been observed for CN, confirming the structure of 7f. Regioselective dialkylation reactions of compound 6 using two equivalent of ethyl iodide indicate that, in the introduction of the second ethyl group into the pyrimidine ring there is a competition between N1 or N3 and revealed that it depends on the base conditions used in the reactions. Thus, refluxing in ethanolic NaOH for 4 h (Method A) afforded the two isomeric compounds 8 and 9 in 63% and 25% yield, respectively, whereas treatment of 6 with two equivalents of ethyl iodide in DMF in the presence of anhydrous K₂CO₃ at room temperature (Method B) afforded in high yield the isomer 9 and only traces of the isomer 8 (Scheme 2). The reaction was monitored by TLC. The reaction mixture was

chromatographically work up over silica gel using Pet. ether (b.p. 60–80 °C) and ethyl acetate (1:1) as eluent, to afforded two products in pure form. A crystalline solid, m.p. 162–164 °C obtained in the first fraction was characterized as 6-ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4b]pyrimidino[5,4-e] pyridine-5-one (9). The second fraction afforded a solid, m.p. 190–192 °C, which was identified as 8-ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b] pyrimidino-[5,4-e]pyridin-5-one (8). By TLC comparison, data analysis and melting point determination, it was found that one of the products 9 was identical to that obtained from Method B. The chemical shift for the pyrimidinone carbonyl is markedly affected by the nature of the adjacent nitrogen [39–43]. The 13 C-NMR spectral data of compound 9 showed that the δ values of the pyrimidine C=O at 165.82 ppm suggest that N-3 near to C=O is sp³-hybridized (pyrrole type) and different from the C=O adjacent to a sp²-hybridized nitrogen (pyridine type) in compound 8, which appears at 178.14 ppm [42,43]. Moreover, reaction of 6 with highly electron withdrawing aromatic halo compounds such as 2,4-dinitrochlorobenzene in DMF at room temperature yielded the S-aryl derivative 10, whereas, upon heating, benzothiazolo derivative 11 was obtained via ring closure reaction with the N3-pyrimidine ring [44]. As chemical evidence, the formation of compound 11 was achieved through heating of a sample of 10 in DMF. The structures of 10 and 11 were deduced from their satisfactory spectral and analytical data, for example, the mass spectrum of compound 11 showed its correct parent ion peak at m/z 506 (M⁺,100%) (Scheme 3).

Scheme 3. Synthetic pathways for compounds 7a-f, 8, 9, 10 and 11.

The sulfur-free compound **12**, which was identified as 6-ethyl-7-hydrazino-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-*b*]pyrimidino[5,4-*e*]pyridin-5-one, was obtained in excellent yield (85%) *via* nucleophilic displacement of the thioethyl group in compound **9** with hydrazine in boiling ethanol (Scheme 4).

Scheme 4. Synthetic pathways for compounds 12–17.

The hydrazine derivative 12 was used as the key intermediate for the synthesis of some polyheterocyclic compounds. Thus, heating of compound 12 with some aromatic aldehydes namely, benzaldehyde, 4-chlorobenzaldeyde and 4-methoxybenzaldeyde in the presence of a few drops of acetic acid in ethanol resulted in the formation of the corresponding hydrazones 13a-c. The structure of compounds 13a-c was characterized by the disappearance of the NHNH₂ group and revealed in each case two bands at 3425-3250 and 3170-3100 cm⁻¹ assignable to two NH groups. Also, their ¹H-NMR spectra showed the presence of the azomethine and two NH protons at 8.9–9.5 and 10.85–12.75 ppm, respectively. On the other hand, upon heating the hydrazino compound 12 with ethyl acetoacetate, acetylacetone and ethyl cyanoacetate in ethanolic sodium ethoxide solution, the N-pyrazolo derivatives 14–16 were produced. For example, the ¹H-NMR spectrum for compound 15 revealed a singlet at 6.15 ppm due to the 4-H-pyrazole moiety. The ¹³C-NMR spectral data displayed two characteristic singlets at 151.12 and 152.5 ppm for C3 and C5 of the pyrazole nucleus, respectively, which was in agreement with the literature value [45]. The formation of the tetracyclic pyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridine (17) was achieved upon reaction of 12 with triethyl orthoformate in ethanol in presence of a few drops of acetic acid (Scheme 4). The ¹H-NMR spectrum of compound 17 revealed the disappearance of NHNH₂ signals and appearance of a singlet signal at 8.55 ppm due to the triazole CH.

Furthermore, the interaction of the hydrazine 12 with CS_2 in pyridine furnished the angular tetracyclic pyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridin-3(2H)-thione (18) in good yield (Scheme 5).

Scheme 5. Synthetic pathways for compounds 18–22a,b.

The thione **18** was easily converted into a series of S-alkylated derivatives **19a**–**c** upon treatment with ethyl iodide, ethyl chloroacetate and phenacyl bromide in the presence of anhydrous sodium acetate in refluxing ethanol, respectively. As well, the reaction of **11** with sodium nitrite in acetic acid at 0 °C, led to the formation of the pyrazolotetrazolopyrimidinopyridine **20** in 75% yield. This new ring system is in equilibrium with the corresponding 2-azido tautomer **20** [46], which was confirmed by a characteristic absorption peak of the azido group at 2250 cm⁻¹ in the IR. The tetracyclic pyrazolotriazinopyrimidinopyridine compounds **21** and **22a**,**b** were obtained in good to high yield when compound **12** was allowed to react with benzoin in acetic anhydride- pyridine mixture and/or with some α-haloketones (chloroacetone and phenacyl bromide) in DMF/AcOH, respectively (Scheme 5). The structures of compounds **21** and **22a**,**b** were confirmed by spectral data. For example, the ¹H-NMR spectrum of compound **22b** revealed a singlet at 9.3 due to CH-triazine and displayed two broad singlets (D₂O-exchangable) at 10.75 and 11.25 due to NH groups. The mass spectra of compound **22b** showed its correct parent ion peak at *m/z* 513 (M⁺, 100%).

2.2. Biological Results

Using the agar well-diffusion method [47], ten selected derivatives (compounds 6, 7a, 9, 11, 12, 13b, 17, 19c, 20, and 22a) were evaluated for their antibacterial and antifungal activities. Thus, these compounds were screened against *Staphylococcus aureus*, *Bacillus cereus*, *Micrococcus luteus* as a Gram positive bacteria and *Escherichia coil*, *Pseudomonas aeruginosa* and *Serratia marcescens* as Gram negative bacteria using chloramphenicol as control (Table 1). The MIC results indicated that three of the tested compounds (11, 19 and 22a) showed significant activity against *Staphylococcus aureus* (Table 1). Compounds 12 and 19c showed highly significant activity against *Bacillus cereus* and moderate activity against *Micrococcus luteus* (Table 1). Compound 22a revealed moderate activity against *Staphylococcus aureus* and *Pseudomonas aeruginosa* (Table 1). The rest of tested compounds were inactive against all bacterial strains used.

Table 1. Antibacerial activity data [inhibition zone in mm/MICs (in mM).

Compound -	Diameter of the inihibition zone (mm) MIC (mM)							
	S. aureus AUMC B.54	S. cereus AUMC B.52	M. luteus AUMC B.112	E. coli AUMC B.53	p. aeruginosa AUMC B.73	S. marcescens AUMC B.55		
6	-	=	-	-	-	-		
7a	_	-	-	-	-	-		
9	-	-	-	-	-	-		
11	8 (2.5)	-	-	-	-	-		
12	-	10 (1.25)	18 (20)	-	-	-		
13b	-	-	-	-	-	-		
17	-	-	-	-	-	-		
19c	8 (1.25)	10 (0.15)	10 (5)	-	-	-		
20	-	=	-	-	-	-		
22a	8 (5)	-	-	-	11 (20)	-		
CHL	10 (0.08)	12 (1.25)	12 (2.5)	10 (0.08)	12 (0.3)	13 (1.25)		

CHL = chloramphenicol as control.

The same compounds (6, 7a, 9, 11, 12, 13b, 17, 19c, 20 and 22a) were screened for their antifungal activities against six fungal strains: (Candida albicans AUMC No. 418, Trichophyton rubrum AUMC No. 1804, Aspergillus flavus AUMC No. 1276, Fusarium oxysporum AUMC No. 5119, Scopulariopsis brevicaulis AUMC No. 729, Geotrichum candidum AUMC No. 226) using clotrimazole as control. The results are listed in Table 2. The MIC values showed that compounds 22a, 12 and 17 exhibit moderate to low activity against Candida albicans AUMC No. 418. Compounds 19c, 12 and 11 showed moderate to low activity against Geotrichum candidum AUMC No. 226. Compounds 17, 19c, 12 and 10 revealed moderate to low activity against Trichophyton rubrum AUMC No. 1804. Compounds 19c, 12 and 17 showed moderate to poor activity against Aspergillus flavus AUMC No. 1276. Compounds 19c, 12 and 17 showed moderate to poor activity against Fusarium oxysporum AUMC No. 5119. Compounds 19c, 12, 17 and 11 showed moderate to poor activity against Fusarium oxysporum AUMC No. 5119. Compounds 19c, 12, 17 and 11 showed moderate to poor activity against Fusarium oxysporum AUMC No. 5119. The rest of tested compounds were inactive against all fungal strains used.

Compound No.	Diameter of the inihibition zone (mm) MIC (mM)							
	C. albicans AUMC 418	G.candidum AUMC 226	T. rubrum AUMC 1804	A.flavus AUMC 3214	F. oxysporum AUMC 5119	S. brevicaylis AUMC 729		
6	-	-	-	-	-	-		
7a	_	-	-	-	-	-		
9	-	-	-	-	-	-		
11	-	11 (20)	8 (20)	-	-	10 (20)		
12	9 (2.5)	15 (5)	10 (10)	10 (2.5)	16 (20)	20 (5)		
13b	-	-	-	-	-	-		
17	10 (20)	-	24p.i (20)	10 (20)	8 (20)	12 (10)		
19c	9 (2.5)	16 (2.5)	12 (5)	14 (2.5)	10 (10)	28 (5)		
20	-	-	-	-	-	-		
22a	10 (0.6)	-	_	-	_	-		

Table 2. Antifungal activity data [inhibition zone in mm/ MICs (in mM).

CLO = Clotrimazole as control; p.i = Partial inhibition.

15 (0.15)

14 (0.15)

24 p.i (0.3)

35 (0.08)

3. Experimental

CLO

12 (0.08)

14 (0.08)

3.1. General

All melting points were determined on a Kofler melting point apparatus. IR spectra were recorded on a Pye Unicam SP3-100 spectrophotometer using the KBr wafer technique. The ¹H-NMR spectra were recorded on a Bruker ARX 200 spectrometer (200 MHz for ¹H and 50 MHz for ¹³C) at the Faculty of Science, University of King Saoud, Saudi Arabia, Riyadh and on a Jeol LA 400 MHz (400 MHz for ¹H, 100 MHz for the ¹³C) at Assiut University, ¹H and ¹³C NMR chemical shifts (δ) were reported in parts per million (ppm) and were referenced to the solvent peak; CDCl₃ (7.26 ppm for ¹H and 76.90 ppm for ¹³C) and DMSO-d₆ (2.50 ppm for ¹H and 39.70 ppm for ¹³C). Multiplicities are represented by s (singlet), d (doublet), t (triplet), q (quartet) and m (multiplet). Coupling constants (*J*) are reported in Hertz (Hz). Mass spectra were taken on a JEOL JMS600 spectrometer at an ionizing potential of 70 eV (EI) at Assiut University. Elemental analyses were carried out using a Perkin-Elmer 240 C Micro analyzer at Assiut University and they were found to be within ± 0.4% of the theoretical values.

3.2. Synthesis of 3-Methyl-1,4-diphenyl-7-thioxo-4,6,8,9-tetrahydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (6)

Method A: A mixture of thiobarbituric acid (1.44 g, 0.01 mol), 3-methyl-1-phenyl-1*H*-pyrazol-5-amine (1.73 g, 0.01 mol), benzaldehyde (1.1 g, 0.012 mol) and *p*-TSA (1 g, 0.05 mol) was heated at 100 °C for 10 h (monitored by TLC). After cooling, the reaction mixture was washed with water (20 mL) and residue recrystallized from EtOH to afford the pure product **6** as a yellow powder. Yield: 66%. Mp 202–204 °C. IR (KBr) cm⁻¹: 3240, 3290, 3310 (3NH), 1690 (C=O), 1620 (C=N), 1345 (C=S). ¹H-NMR (DMSO-d₆): 2.75 (s, 3H, CH₃), 5.57 (s, 1H, CH pyridine), 7.25–7.98 (m, 10H, Ar-H), 9.45

(s, 1H, NH), 12.20 (br s, 1H, NH), 13.80 (br s, 1H, NH). 13 C-NMR (DMSO-d₆) δ (ppm): 11.93 (CH₃), 33.8 (CH sp³), 102.11–160.12 (17C, sp² carbon atoms), 161.73 (C=O), 174.45 (C=S). Anal. Calcd. For $C_{21}H_{17}N_5OS$ (387.45): C, 65.10; H, 4.42; N, 18.08; S, 8.28. Found: C, 65.15; H, 4.38; N, 18.04; S, 8.23.

Method B: A mixture of phenylhydrazine (1 mmol), 3-aminocrotononitrile (1mmol) and *p*-TSA (0.5 mmol) in water (10 mL) was added to benzaldehyde (1 mmol) and thiobarbituric acid (1 mmol) and the reaction mixture was heated under reflux at 100 °C for 1 h, after completion of the reaction (TLC), the reaction mixture cooled to room temperature, the precipitate filtered off and washed with water and recrystallized from EtOH to afford the pure product 6 as a yellow powder. Yield: 91%. All of spectral and physical data were in agreement with that described in method A.

3.3. General Procedure for the Preparation of 7a-f

A mixture of 6 (3.8 g, 0.01 mol), ethyl iodide and/or α -haloketone compound (0.01 mol) in ethanol (50 mL) was refluxed in the presence of anhydrous sodium acetate (0.9 g, 0.011 mol) for 4 h. The solid product separated from the hot mixture was filtered off, washed with water and recrystallized from the proper solvent.

7-Ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (**7a**). Yellow crystals. Yield: 72%. Mp. 135–137 °C (acetic acid). IR (KBr) cm⁻¹: 3220, 3315 (2NH), 1690 (C=O), 1620 (C=N). 1 H-NMR (DMSO-d₆) δ (ppm): 1.55 (t, J = 7.4 Hz, 3H, CH₃), 2.78 (s, 3H, CH₃), 3.65 (q, J = 7.4 Hz, 2H, CH₂), 5.55 (s, 1H, CH), 7.2–7.8 (m, 10H, Ar-H), 8.9 (s, 1H, NH), 10.2 (s, 1H, NH). 13 C-NMR (DMSO-d₆) δ (ppm): 12.03 (CH₃), 15.34 (CH₃), 24.12 (CH₂), 33.5 (CH sp³), 103.15–160.12 (18C, sp² carbon atoms), 161.73 (C=O), Anal. Calcd. for C₂₃H₂₁N₅OS (415.52): 66.48; H, 5.09; N, 16.85; S, 7.71. Found: C, 66.61; H, 5.13; N, 16.79; S, 7.68.

Ethyl-2-(3-methyl-5-oxo-1,4-diphenyl-4,6,9-trihydro-pyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-7-ylthio)acetate (**7b**). Yellow crystals. Yield: 67%. Mp. 95–97 °C (methanol). IR (KBr) cm⁻¹: 3290, 3310 (2NH), 1735 (C=O), 1690 (C=O), 1620 (C=N). ¹H-NMR (DMSO-d₆) δ (pp): m1.27 (t, J = 7.4 Hz, 3H, CH₃), 2.80 (s, 3H, CH₃), 3.95 (s, 2H, SCH₂), 4.25 (q, J = 7.4 Hz, 2H, CH₂), 5.52 (s, 1H, CH), 7.2–7.9 (m, 10H, Ar-H), 8.7 (s, 1H, NH), 11.1 (s, 1H, NH). ¹³C-NMR (CDCl₃) δ (ppm): 12.03 (CH₃), 15.34 (CH₃), 29.12 (CH₂), 33.55 (CH sp³), 61.11 (CH₂), 103.15–160.12 (18C, sp² carbon atoms), 161.73 (C=O), 170.25 (C=O ester). Anal. Calcd. for: C₂₅H₂₃N₅O₃S (473.54): C, 63.41; H, 4.90; N, 14.79; S, 6.77. Found: C, 63.49; H, 5.01; N, 14.82; S, 6.86.

3-Methyl-7-(2-oxo-2-phenylethylthio)-1,4-diphenyl-4,6,9-trihydropyrazolo [5,4-b]pyrimidino[5,4-e]-pyridin-5-one (**7c**). Light yellow crystals. Yield: 71%. Mp. 170–172 °C (methanol). IR (KBr) cm⁻¹: 3245, 3290 (2NH), 1,685 (C=O), 1690 (C=O), 1620 (C=N); 1 H-NMR (DMSO-d₆) δ (ppm): 2.72 (s, 3H, CH₃), 4.19 (s, 2H, SCH₂CO), 5.57 (s, 1H, CH), 6.9–8.6 (m, 15H, Ar-H), 8.84 (s, 1H, NH), 10.73 (s, 1H, NH); 13 C-NMR (DMSO-d₆) δ (ppm): 12.93 (CH₃), 34.42 (CH sp³), 35.60 (CH₂), 102.75–160.30 (18C, sp² carbon atoms), 162.03 (C=O), 195.30 (C=O) Anal. Calcd. for: C₂₉H₂₃N₅O₂S (505.58): C, 68.89; H, 4.59; N, 13.85; S, 6.34. Found: C, 68.97; H, 4.64; N, 13.91; S, 6.29.

3-Methyl-7-(2-oxopropylthio)-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (7d). Yellow crystals. Yield: 65%. Mp. 223–225 °C (ethanol); IR (KBr) cm⁻¹: 3195, 3290 (2NH), 1685 (C=O), 1700 (C=O); 1 H-NMR (CDCl₃) δ (ppm): 2.75 (s, 3H, CH₃), 3.35 (s, 3H, COCH₃), 4.10 (s, 2H, SCH₂), 5.60 (s, 1H, CH), 7.35–7.95 (m, 10H, Ar-H), 9.15 (s, 1H, NH), 10.2 (s, 1H, NH). 13 C-NMR (CDCl₃) δ (ppm): 11.93 (CH₃), 29.32 (CH₃), 33.82 (CH sp³), 38.22 (CH₂), 103.15–160.12 (18C, sp² carbon atoms), 161.53 (C=O), 202.11 (C=O). Anal. Calcd. for: C₂₄H₂₁N₅O₂S (443.52): C, 64.99; H, 4.77; N, 15.79; S, 7.23. Found: C, 65.13; H, 4.81; N, 15.83; S, 7.19.

2-(3-Methyl-5-oxo-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-7-ylthio)acetamide (7e). Orange crystals. Yield: 69%. Mp. 166–168 °C (DMF). IR (KBr) cm⁻¹: 3420, 3400 (NH₂), 3250, 3310 (2NH), 1685 (C=O), 1665 (C=O), 1690 (C=O). ¹H-NMR (CDCl₃) δ (ppm): 2.65 (s, 3H, CH₃), 3.70 (s, 2H, SCH₂), 5.30 (br.s, 2H, NH₂), 6.15 (s, 1H, CH), 7.35–7.96 (m, 10H, Ar-H), 8.95 (s, 1H, NH), 10.78 (s, 1H, NH); ¹³C-NMR (CDCl₃) δ (ppm): 12.11 (CH₃), 30.34 (CH₂), 34.52 (CH sp³), 104.85–159.90 (18C, sp² carbon atoms), 162.30 (C=O), 171.45 (C=O) Anal. Calcd. for $C_{23}H_{20}N_6O_2S$ (444.50): C, 62.15; H, 4.54; N, 18.91; S, 7.21. Found: C, 62.29; H, 4.61; N, 19.03; S, 7.15.

2-(3-Methyl-5-oxo-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-7-ylthio) ethanenitrile (7f). Orange crystals. Yield: 61%. Mp. 147–149 °C (ethanol). IR (KBr) cm $^{-1}$: 3215, 3200 (2NH), 2,240 (C=N), 1665 (C=O); 1 H-NMR (DMSO-d₆)) δ (ppm): 2.70 (s, 3H, CH₃), 3.90 (s, 2H, SCH₂CO), 6.12 (s, 1H, CH), 7.55–8.22 (m, 10H, Ar-H), 9.20 (s, 1H, NH), 10.6(s, 1H, NH). 13 C-NMR (DMSO-d₆) δ (ppm): 12.31 (CH₃), 15.38 (CH₂), 33.84 (CH sp 3), 104.80–159.95 (18C, sp 2 carbon atoms), 161.90 (C=O). Anal. Calcd. for C₂₃H₁₈N₆OS (426.49): C, 64.77; H, 4.25; N, 19.70; S, 7.52. Found: C, 64.89; H, 4.31; N, 19.61; S, 7.61.

3.4. 8-Ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (**8**) and 6-Ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (**9**)

Method A: A mixture of 6 (3.8 g, 0.01 mol), ethyl iodide (3.1 g, 0.021 mol) and sodium hydroxide (0.8 g, 0.02 mol) in ethanol (50 mL) was refluxed for 4 h. Excess ethanol was distilled off and the residue obtained on cooling was found to be mixture of two products as evidenced by the TLC. The residue was subjected to column chromatography over silica gel using mixtures of pet-ether/ethyl acetate of increasing polarity of eluent to yield compound 9 as an orange solid and compound 8 as a yellow solid in the second fraction.

8-Ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (8). Yield: 63%. Mp. 190–192 °C. IR (KBr) cm⁻¹: 3190 (NH), 1700 (C=O), ¹H-NMR (DMSO-d₆) δ (ppm): 1.00 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 1.35 (t, J = 7.4 Hz, 3H, SCH₂CH₃), 2.68 (s, 3H, CH₃), 3.15 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 3.95 (q, J = 7.4 Hz, 2H, SCH₂CH₃), 5.65 (s, 1H, CH), 7.40–7.80 (m, 10H, Ar-H), 10.45 (s, 1H, NH). ¹³C-NMR (DMSO-d₆) δ (ppm): 12.50 (CH₃), 12.88 (CH₃), 15.09 (CH₃), 26.13 (S-CH₂), 37.04 (CH₂), 39.25 (CH sp³), 105.77–162.15 (18C, sp² carbon atoms), 178.14 (C=O).

6-Ethyl-7-ethylthio-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (9). Yield: 25%. Mp. 162–164 °C. IR (KBr) cm⁻¹: 3235 (NH), 1695 (C=O), ¹H-NMR (DMSO-d₆) δ (ppm): 1.32 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 1.65 (t, J = 7.4 Hz, 3H, SCH₂CH₃), 2.78 (s, 3H, CH₃), 3.45 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 4.05 (q, J = 7.4 Hz, 2H, SCH₂CH₃), 5.75 (s, 1H, CH), 7.40–7.96 (m, 10H, Ar-H), 10.26 (s, 1H, NH). ¹³C-NMR (DMSO-d₆) δ (ppm): 11.81 (CH₃), 12.56 (CH₃), 14.88 (CH₃), 24.28 (S-CH₂), 33.14 (CH₂), 35.22 (CH sp³), 103.77–160.15 (18C, sp² carbon atoms), 163.16 (C=O); Anal. Calcd. for: C₂₅H₂₅N₅OS (443.56) C, 67.69; H, 5.68; N, 15.79; S, 7.23. Found: C, 67.81; H, 5.61; N, 15.81; S, 7.16.

Method B: A mixture of 6 (1.90 g, 0.005 mol) and ethyl iodide (1.55g, 0.01 mol) in DMF (30 mL) in the presence of anhydrous potassium carbonate (0.6 g, 0.002 mol) was stirred at room temperature for 10 h. The reaction mixture was cooled, poured onto ice cold water. The solid product separated was filtered off, washed with water and recrystallized from dioxane to yield compound 9. All of the physical and analytical data were in agreement with those of the product obtained using method A.

3.5. 7-(2,4-Dinitrophenylthio)-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one (10)

A mixture of **6** (1.14 g, 0.003 mol) and 2,4-dinitrochlorobenzene (0.6 g, 0.003 mol) in DMF (40 mL) was stirred at room temperature for 5 h. The reaction mixture was cooled to 0 °C for 2 h, a yellow fine crystals was obtained, it was filtered off dried and recrystallized from dioxane. Yield: 68%. Mp. 251–253 °C. IR (KBr) cm⁻¹: 3325, 3295 (2NH), 1690 (C=O), ¹H-NMR (DMSO-d₆) δ (ppm): 2.75 (s, 3H, CH₃), 5.65 (s, 1H, CH), 7.14–8.92 (m, 13H, Ar-H), 10.26 (s, 1H, NH), 11.35 (s, 1H, NH). ¹³C-NMR (DMSO-d₆) δ (ppm): 11.90 (CH₃), 34.35 (CH sp³), 103.77–161.17 (25C, sp² carbon atoms), 162.99 (C=O); Anal. Calcd. for: C₂₇H₁₉N₇O₅S (553.54) C, 58.58; H, 3.46; N, 17.71; S, 5.79. Found: C, 58.96; H, 3.69; N, 17.97; S, 6.14.

3.6. 3-Methyl-8-nitro-1,4-diphenyl-4,6,13-trihydrobenzothiazolo[3',2'-2,1]pyrimidino[5,4-e] pyrazolo-[5,4-b]pyridin-5-one (11)

Method A: A mixture of **6** (0.38 g, 0.001 mol) and 2,4-dinitrochlorobenzene (0.2 g, 0.001 mol) in DMF (20 mL) was refluxed for 2 h. The reaction mixture was concentrated for its half volume and allowed to cool, yellowish crystals was obtained, filtered off, dried and recrystallized from acetic acid. Yield: 59%. Mp. >300 °C. IR (KBr) cm⁻¹: 3180 (NH), 1685 (C=O), ¹H-NMR (DMSO-d₆) δ (ppm): 2.79 (s, 3H, CH₃), 4.95 (s, 1H, CH), 7.10–7.85 (m, 13H, Ar-H), 9.95 (s, 1H, NH). ¹³C-NMR (DMSO-d₆) δ (ppm): 12.30 (CH₃), 34.15 (CH sp³), 102.57–162.11 (25C, sp² carbon atoms), 163.09 (C=O); Anal. Calcd. for: $C_{27}H_{18}N_6O_3S$ (506.53) C, 64.02; H, 3.58; N, 16.59; S, 6.33. Found: C, 64.42; H, 3.99; N, 16.91; S, 6.64. MS m/z (%) 506.12 (M⁺, 100).

Method B: A sample of **10** (0.5 g, 0.001 mol) was heated under reflux in DMF (30 mL) for 2 h. The separated product **11** was obtained and purified as described in method A. All physical and analytical data of the two final products obtained from both methods A and B are identical.

3.7. 6-Ethyl-7-hydrazino-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]-pyridin-5-one (12)

A mixture of **9** (1.76 g, 0.004 mol) and hydrazine hydrate (15 mL) in absolute ethanol (40 mL) was refluxed for 12 h. The reaction mixture was poured onto ice. The product was isolated and crystallized from acetic acid as white needles. Yield: 85%. Mp. 280–281 °C, IR (KBr) cm⁻¹: 3440, 3335, 3210 (NH, NH₂), 1695 (C=O); ¹H-NMR (DMSO-d₆) δ (ppm): 1.23 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.81 (s, 3H, CH₃), 3.35 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 4.95 (s, 2H, NH₂, D₂O exchangeable), 6.04 (s, 1H, CH), 7.55–8.22 (m, 10H, Ar-H), 9.82 (s, 1H, NH, D₂O exchangeable), 10.55 (s, 1H, NH). ¹³C-NMR (DMSO-d₆) δ (ppm): 12.21 (CH₃), 13.01 (CH₃), 30.58(CH₂), 34.92 (CH sp³), 102.65–162.95 (18C, sp² carbon atoms), 163.56 (C=O). Anal. Calcd. for: C₂₃H₂₃N₇O (413.47): C, 66.81; H, 5.61; N, 23.71; Found: C, 66.95; H, 5.67; N, 23.79.

3.8. General Procedure for the Preparation of 13a-c

A mixture of compound **12** (0.413 g, 0.001 mol) and the appropriate aromatic aldehyde (0.001 mol) was stirred under reflux in ethanol (30 ml) in the presence of a few drops of glacial acetic acid for 5 h. The reaction mixture was allowed to cool to room temperature, poured into water, whereby a solid formed that was filtered off and crystallized from an appropriate solvent to produce **13a–c** in good yields.

6-Ethyl-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one-7-benzaldehyde hydrazone (**13a**). Pale white crystals from acetic acid. Yield: 70%. Mp. 268–269 °C; IR (KBr) cm⁻¹: 3425, 3295 (2NH), 1675 (C=O), 1625 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.20 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.78 (s, 3H, CH₃), 3.30 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 6.10 (s, 1H, CH), 7.55–8.25 (m, 15H, Ar-H), 8.95 (s,1H, azomethine proton), 11.10 (brs, 1H, NH, D₂O exchangeable), 12.75 (brs, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆) δ (ppm): 11.91 (CH₃), 12.99 (CH₃), 30.81 (CH₂), 35.20 (CH sp³), 102.65–162.95 (21C, sp² carbon atoms), 162.66 (C=O), 164.11 (N=C). Anal. Calcd. for: C₃₀H₂₇N₇O (501.58): C, 71.84; H, 5.43; N, 19.55; Found: C, 71.93; H, 5.59; N, 19.61.

6-Ethyl-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one-7-(p-chloro)benzaldehyde hydrazone (**13b**). Pale light yellow crystals, from dioxane. Yield: 72%. Mp. 310–311 °C (dec.); IR (KBr) cm⁻¹: 3320, 3170 (2NH), 1685 (C=O), 1645 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.25 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.88 (s, 3H, CH₃), 3.35 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.98 (s, 1H, CH), 7.25–8.15 (m, 14H, Ar-H), 9.15 (s,1H, azomethine proton), 10.85 (brs, 1H, NH, D₂O exchangeable), 12.75 (brs, 1H, NH, D₂O exchangeable); Anal. Calcd. for C₃₀H₂₆ClN₇O (536.03): C, 67.22; H, 4.89; Cl, 6.61; N, 18.29; Found: C, 67.33; H, 4.94; Cl, 6.59; N, 18.41.

6-Ethyl-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]pyrimidino[5,4-e]pyridin-5-one-7-(p-methoxy)benzaldehyde hydrazone (13c). Pale white crystals from dioxane. Yield: 68%. Mp. 216–218 °C; IR (KBr)cm⁻¹: 3330, 3100 (2NH), 1690 (C=O), 1635 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.15 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.90 (s, 3H, CH₃), 3.30(q, J = 7.4 Hz, 2H, NCH₂CH₃), 3.94 (s, 3H, OCH₃) 5.88 (s, 1H, CH), 7.20–7.95 (m, 14H, Ar-H), 9.50 (s,1H, methylenic proton), 10.77 (brs, 1H, NH, D₂O

exchangeable), 11.45(brs, 1H, NH, D₂O exchangeable). Anal. Calcd. for C₃₁H₂₉N₇O₂ (531.61): C, 70.04; H, 5.50; N, 18.44; Found: C, 70.14; H, 5.59; N, 18.41.

 $3.9.\ 6-Ethyl-3-methyl-7-(3-methyl-5-oxo(2-pyrazolinyl))-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]-pyrimidino[5,4-e]pyridin-5-one (14)$

A solution of compound **12** (0.413 g, 0.001 mol) and ethyl acetoacetate (0.130 g, 0.001 mol) in sodium ethoxide solution [prepared by dissolving sodium metal (0.023 g, 0.001 mol) in absolute ethanol (30 mL)] was heated under reflux with stirring for 4 h. The reaction mixture was allowed to cool and poured into cold water (100 mL) and neutralized by acetic acid, whereby a solid was precipitated, which was filtered off and crystallized from chloroform. Yield: 73%. Mp. 231–233 °C; IR (KBr) cm⁻¹: 3315 (NH), 1698, 1676 (2C=O), 1550 (C=N), 1 H-NMR (CDCl₃) δ (ppm): 1.25 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.33 (s, 3H, CH₃), 2.55 (s, 2H, CH₂), 2.88 (s, 3H, CH₃), 3.50 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.88 (s, 1H, CH), 7.20–7.66 (m, 10H, Ar-H), 10.85(brs, 1H, NH, D₂O exchangeable); 13 C-NMR (CDCl₃) δ (ppm): 11.33 (CH₃), 12.79 (CH₃), 24.37 (CH₃), 30.65 (CH₂), 34.78 (CH sp³), 44.65 (CH₂ pyrazole), 103.65–162.95 (19C, sp² carbon atoms), 163.66 (C=O), 164.815 (C=O pyrazole). Anal. Calcd. for: C₂₇H₂₅N₇O₂ (479.53): C, 67.63; H, 5.25; N, 20.45; Found: C, 67.71; H, 5.29; N, 20.41.

3.10. 7-(3,5-Dimethylpyrazolyl)-6-ethyl-3-methyl-1,4-diphenyl-4,6,9-trihydropyrazolo[5,4-b]-pyrimidine[5,4-e]pyridin-5-one (15)

A mixture of compound **12** (0.413 g, 0.001 mol) and acetylacetone (0.1 g, 0.001 mol) in absolute ethanol (30 mL) was stirred under reflux for 5 h. The reaction mixture was allowed to cool to 0 °C for 3 h. The precipitate was filtered off, dried and crystallized from ethanol as pale light crystals. Yield: 80%. Mp. 222–224 °C. IR (KBr) cm⁻¹: 3225 (NH), 1700 (C=O), 1650 (C=N), ¹H-NMR (CDCl₃) δ (ppm): 1.27 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.41 (s, 3H, CH₃), 2.43 (s, 3H, CH₃), 2.68 (s, 3H, CH₃), 3.63 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.77 (s, 1H, CH), 6.15 (1H, CH pyrazole), 7.20–7.82 (m, 10H, Ar-H), 10.65(s, 1H, NH); ¹³C-NMR (CDCl₃) δ (ppm): 11.33 (CH₃), 12.79 (CH₃),17.62 (CH₃), 24.37 (CH₃), 30.65 (CH₂), 33.66 (CH sp³), 102.25–160.95 (21C, sp² carbon atoms), 162.90 (C=O). Anal. Calcd. for: C₂₈H₂₇N₇O (477.56): C, 70.42; H, 5.70; N, 20.53; Found: C, 70.52; H, 5.79; N, 20.61.

 $3.11.\ 7$ -(3-Amino-5-oxo(2-pyrazolinyl))-6-ethyl-3-methyl-1,4-diphenyl-4,6,9-tri-hydropyrazolo[5,4-b]-pyrimidino[5,4-e]pyridin-5-one (16)

To a warmed ethanolic sodium ethoxide solution [prepared by dissolving sodium metal (0.023 g, 0.001 mol) in absolute ethanol (30 mL)] was added compound **12** (0.413 g, 0.001 mol) and ethyl cyanoacetate (0.113 g, 0.001 mol). The mixture was stirred under reflux for 12 h, the reaction mixture was allowed to cool to room temperature, then poured into cold water (100 mL) and neutralized with acetic acid. The solid product was filtered off, washed with water, ethanol, dried and crystallized from ethanol as pale brown crystals. Yield: 85%. Mp. 310–312 °C. IR (KBr) cm⁻¹: 3455, 3350 (NH₂), 3220 (NH), 1695 (C=O), 1700 (C=O), 1640 (C=N), 1 H-NMR (CDCl₃) δ (ppm): 1.31 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.71 (s, 3H, CH₃), 3.45 (s, 2H, CH₂ pyrazole), 3.65 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 6.01

(s, 1H, CH), 7.15–7.89 (m, 10H, Ar-H), 10.25 (s, 1H, NH), 12.11 (brs, NH₂, D₂O exchangeable). Anal. Calcd. for: C₂6H₂4N₈O₂ (480.52): C, 64.99; H, 5.03; N, 23.32; Found: C, 65.08; H, 5.10; N, 23.41.

3.12. 11-Ethyl-8-methyl-6,9-diphenyl-4,5,9,11-tetrahydropyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]-pyrimidino[5,6-e]pyridin-10-one (17)

A mixture of compound **12** (0.413 g, 0.001 mol) and triethyl orthoformate (0.192 g, 0.0013 mol) in ethanol (30 mL) was refluxed in the presence of few drops of acetic acid for 3 h. The solid product that separated from the hot mixture was filtered off, and recrystallized from acetic acid as yellow crystals. Yield: 56%. Mp. >300 °C. IR (KBr)cm⁻¹: 3315 (NH), 1695 (C=O), 1650 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.24 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.71 (s, 3H, CH₃), 3.45 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 6.15 (s, 1H, CH), 7.15–7.92 (m, 10H, Ar-H), 8.55 (s, 1H, CH triazole), 11.05 (s, 1H, NH). Anal. Calcd. for C₂₄H₂₁N₇O (423.47): C, 68.07; H, 5.01; N, 23.15; Found: C, 68.16; H, 4.96; N, 23.22. MS m/z (%) 423.21 (M⁺, 100).

3.13. 11-Ethyl-8-methyl-6,9-diphenyl-10-oxo-4,5,9,11-tetrahydropyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridin-3(2H)-thione (18)

A mixture of compound **12** (2.06g, 0.005 mol) and carbon disulfide (0.95 g, 0.005 mol) in ethanolic sodium hydroxide (10 mL, 10%) was heated on a water bath for 2 h. The solvent was evaporated under reduced pressure, the residue was diluted with water (30 mL), acidified with HCl. The solid product was filtered off and recrystallized from acetic acid as orange crystals. Yield: 69% Mp.285–287 °C. IR (KBr) cm⁻¹: 3295 (NH), 1705 (C=O), 1620 (C=N), ¹H-NMR (DMSO- d_6) δ (ppm): 0.99 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.45 (s, 3H, CH₃), 3.15 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.65 (s, 1H, CH), 7.30–7.99 (m, 10H, Ar-H), 10.75 (s, 1H, NH). Anal. Calcd. for: C₂₄H₂₁N₇OS (455.53): C, 63.28; H, 4.65; N, 21.52; S, 7.04; Found: C, 63.36; H, 4.72; N, 21.62.; S, 6.98. MS m/z (%) 455.17 (M⁺, 100).

3.14. General Procedure for the Preparation of 19a-c

These compounds were synthesized following a procedure analogous to that for compounds 7a-f using a mixture of 18 (0.44 g, 0.001 mol), ethyl iodide and/or α -haloketone (0.001 mol). The solid product separated from the hot mixture was filtered off, washed with water and recrystallized from ethanol.

3-Ethylthio-11-ethyl-8-methyl-6,9-diphenyl-4,5,9,11-tetrahydropyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridin-10-one (**19a**). Yellow crystals. Yields: 63%. Mp. 245–247 °C. IR (KBr) cm⁻¹: 3305 (NH), 1705 (C=O), 1630 (C=N), ¹H-NMR (DMSO-d₆) ppm δ: 1.30 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 1.58 (t, J = 7.4 Hz, 3H, SCH₂CH₃), 2.69 (s, 3H, CH₃), 3.11 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 3.55(q, J = 7.4 Hz, 2H, SCH₂CH₃), 5.95 (s, 1H, CH), 7.40–7.85 (m, 10H, Ar-H), 10.30 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ (ppm): 11.13 (CH₃), 12.50 (CH₃),14.65 (CH₃), 29.75 (CH₂), 35.60 (CH sp³), 42.80 (CH₂), 103.15–160.95 (19C, sp² carbon atoms), 162.20 (C=O). Anal. Calcd. for: C₂6H₂5N₇OS (483.59): C, 64.58; H, 5.21; N, 20.27; S, 6.63; Found: C, 64.69; H, 5.30; N, 20.32; S, 6.78.

Ethyl-3-(11-ethyl-8-methyl-6,9-dipheny-10-oxo--4,5,9,11-tetrahydro-pyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridine)acetate (**19b**). Yellow crystals. Yield: 79%. Mp. 189–191 °C; IR (KBr) cm⁻¹: 3280 (NH), 1705 (C=O), 1735 (C=O), 1600 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.25 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 1.45 (t, J = 7.4 Hz, 3H, SCH₂CH₃), 2.70 (s, 3H, CH₃), 3.10 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 3.80 (s, 2H, SCH₂), 4.15 (q, J = 7.4 Hz, 2H, SCH₂CH₃), 6.15 (s, 1H, CH), 7.40–7.85 (m, 10H, Ar-H), 11.03 (s, 1H, NH, D₂O exchangeable). Anal. Calcd. for: C₂₈H₂₇N₇O₃S (541.49): C, 62.09; H, 5.02; N, 18.10; S, 5.92; Found: C, 62.23; H, 4.87; N, 18.19; S, 5.99.

11-Ethyl-3-(2-oxo-2-phenylethylthio)--8-methyl-6,9-dipheny-10-oxo-4,5,9,11-tetrahydro-pyrazolo[5,4-b]1,2,4-triazolo[4',3'-2,1]pyrimidino[5,6-e]pyridine-10-one (**19c**). Pale yellow crystals. Yield: 68%. Mp. 272–274 °C. IR (KBr) cm⁻¹: 3290 (NH), 1705 (C=O), 1690 (br, C=O), 1600 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.18 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.77 (s, 3H, CH₃), 3.32 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 4.35 (s, 2H, SCH₂), 5.95 (s, 1H, CH), 7.20–8.25 (m, 15H, Ar-H), 10.80 (s, 1H, NH, D₂O exchangeable). Anal. Calcd. for C₃₂H₂₇N₇O₂S (573.67): C, 67.01; H, 4.74; N, 17.09; S, 5.59; Found: C, 67.21; H, 4.83; N, 17.23.; S, 5.49. MS m/z (%) 573.57 (M⁺, 100).

3.15. 4-Ethyl-7-methyl-6,9-diphenyl-4,6,10,11-tetrahydropyrazolo[5,4-b]1,2,3,4-tetraazolo[1',5'-1,2]-pyrimidino[5,6-e]pyridin-5-one (**20**)

A solution of sodium nitrite (0.07 g, 0.001 mol) in the least amount of water was added dropwise to an ice-cold solution of compound **12** (0.413 g, 0.001 mol) in acetic acid (10 mL) kept in an ice bath at -5 °C. The reaction mixture was allowed to stand overnight at room temperature, then it was poured into water (100 mL). The precipitate that formed was filtered off and crystallized from dioxane. It separated as pale yellow needles. Yield: 75%. Mp. > 300 °C; IR (KBr) cm⁻¹: 3350 (NH), 2250 (N₃), 1695 (C=O), 1620 (C=N). 1 H-NMR (DMSO-d₆) δ (ppm): 1.33 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.60 (s, 3H, CH₃), 3.54 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.85 (s, 1H, CH), 7.20–7.90 (m, 10H, Ar-H), 11.25 (s, 1H, NH, D₂O exchangeable). 13 C-NMR (DMSO-d₆) δ (ppm): 11.03 (CH₃), 12.35 (CH₃), 33.45 (CH sp³), 40.65 (CH₂), 101.75–160.15 (18C, sp² carbon atoms), 163.30 (C=O). Anal. Calcd. for C₂₃H₂₀N₈O (424.46): C, 65.08; H, 4.75; N, 26.40; Found: C, 65.19; H, 4.83; N, 26.55.

3.16. 5-Ethyl-8-methyl-1,2,7,10-tetraphenyl-5,7,11,12-tetrahydro-1H-pyrazolo[5,4-b]1,2,4-triazino[4',3'-2,1]-pyrimidino[5,6-e]pyridin-6-one (**21**)

A mixture of compound **12** (0.413g, 0.001 mol) and benzoin (0.21 g, 0.001 mol) was heated under reflux in a mixture of pyridine and acetic anhydride (20 mL) (1:1) for 5 h. The reaction mixture was allowed to cool, poured onto ice cold water and neutralized with dilute HCl, The solid product was filtered off and recrystallized from acetic as pale gray needles. Yield: 55%. Mp. > 300 °C; IR (KBr) cm⁻¹: 3270 (NH), 1690 (C=O), 1640(C=N). ¹H-NMR (DMSO-d₆) δ (ppm): 1.35 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.40 (s, 3H, CH₃), 3.15 (s, 1H, CH), 3.35 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 6.05 (s, 1H, CH), 7.20–7.90 (m, 20H, Ar-H), 10.40 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ (ppm): 11.03 (CH₃), 12.55 (CH₃), 32.35 (CH sp³), 38.70 (CH₂), 44.55 (CH sp³), 105.80–161.85 (31C, sp² carbon atoms), 162.90 (C=O). Anal. Calcd. for C₃₇H₃₁N₇O (589.66): C, 75.35; H, 5.82; N, 16.19; Found: C, 75.73; H, 6.12; N, 16.53.

3.17. General Procedure for the Preparation of 22a,b

A mixture of compound **12** (0.001 mol) with chloroacetone or phenacyl bromide (0.001 mol) in DMF (30 mL) and drops of glacial acid (0.2 mL) was heated in 90 °C for 10 h. The solid that precipitated upon cooling was filtered off and crystallized from ethanol.

5-Ethyl-1,8-dimethyl-7,10-diphenyl-5,7,11,12-tetrahydro-3H-pyrazolo[5,4-b]1,2,4-triazino[4',3'-2,1]-pyrimidino[5,6-e]pyridin-6-one (**22a**). Pale white crystals. Yield: 51%. Mp. 241–243 °C (dec.). IR (KBr) cm⁻¹: 3310, 3290 (2NH), 1685 (C=O) 1630 (C=N), ¹H-NMR (DMSO-d₆) δ (ppm): 1.20 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.33 (s, 3H, CH₃), 2.65 (s, 3H, CH₃), 3.54 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 5.70 (s, 1H, CH), 7.15–7.95 (m, 10H, Ar-H), 9.25 (s, 1H, triazine), 10.80 (s, 1H, NH, D₂O exchangeable), 10.95 (s, 1H, NH, D₂O exchangeable). ¹³C-NMR (DMSO-d₆) δ (ppm): 11.90 (CH₃), 12.45 (CH₃), 16.55 (CH³), 30.98 (CH₂), 34.95 (CH sp³), 36.15 (CH sp³), 104.90–161.85 (19C, sp² carbon atoms), 162.90 (C=O). Anal. Calcd. for C₂6H₂5N₇O (451.52): C, 69.16; H, 5.58; N, 21.71; Found: C, 69.24; H, 5.73; N, 21.81.

5-Ethyl-8-methyl-1,7,10-triphenyl-5,7,11,12-tetrahydro-3H-pyrazolo[5,4-b]1,2,4-triazino[4',3'-2,1]-pyrimidino[5,6-e]pyridin-6-one (22b). Pale white crystals. Yield: 63%. Mp. >300 °C, IR (KBr) cm⁻¹: 3400, 3390 (2NH), 1685 (C=O) 1630 (C=N), 1 H-NMR (DMSO-d₆) δ (ppm): 1.19 (t, J = 7.4 Hz, 3H, NCH₂CH₃), 2.55 (s, 3H, CH₃), 4.04 (q, J = 7.4 Hz, 2H, NCH₂CH₃), 6.10 (s, 1H, CH), 7.15–8.15 (m, 15H, Ar-H), 9.30 (s, 1H, triazine), 10.75 (s, 1H, NH, D₂O exchangeable), 11.25 (s, 1H, NH, D₂O exchangeable). Anal. Calcd. for: C₃₁H₂₇N₇O (513.59): C, 72.50; H, 5.30; N, 19.09; Found: C, 72.64; H, 5.41; N, 19.16. MS m/z (%) 513.48 (M⁺, 100).

3.18. Antimicrobial Activity

The antimicrobial activity of 10 new chemical compounds was tested in vitro against six bacterial species obtained from contaminated soil, water and food substances (Staphylococcus aureus [AUMC No. B-54], Bacillus cereus [AUMC No. B-52], Micrococcus luteus (+ve) [AUMC NoB-112], Escherichia coli [AUMC No. B-53], Pseudomonas aeruginosa [AUMC No. B-73] and Serratia marcescens [AUMC No. B-55]. They were also tested against six fungal species which are involved in human and animal diseases (Trichophyton rubrum [AUMC No. 1804], Candida albicans [AUMC No. 418], Geotrichum candidum [AUMC No. 226], Scopulariopsis brevicaulis[AUMC No. 729] and Aspergillus flavus[AUMC No. 3214] or plant diseases (Fusarium oxysporum [AUMC No. 5119]. These strains are common contaminants of the environment in Egypt and some of all microbial strains were kindly provided by the Assiut University Mycological Centre (AUMC). To prepare inocula for bioassay, bacterial strains were individually cultured for 48 h in 100 mL conical flasks containing 30 mL nutrient broth medium. Fungi were grown for 7 days in 100 mL conicals containing 30 mL Sabouraud's dextrose broth. Bioassay was done in 10 cm sterile plastic Petri plates in which microbial suspension (1 mL/plate) and 15 mL appropriate agar medium (15 mL/plate) were poured. Nutrient agar and Sabouraud's dextrose agar were respectively used for bacteria and fungi. After solidification of the media, 5 mm diameter cavities were cut in the solidified agar (4 cavities/plate) using sterile cork borer. Chemical compounds dissolved in DMSO at 2%w/v (=20 mg/mL) were pipetted in the cavities

(20 μL/cavity). Cultures were then incubated at 28 °C for 48 h in case of bacteria and up to 7 days in case of fungi. Results were read as the diameter (in mm) of inhibition zone around cavities [29]. To determine the minimum inhibitory concentrations (MICs), chemical compounds giving positive results were diluted with DMSO to prepare a series of descending concentrations down to 0.02 mg/mL. Diluted chemicals were similarly assayed as mentioned before and the least concentration (below which no activity) was recorded as the MIC.

4. Conclusions

In conclusion, we have reported the synthesis of some novel heterocyclic pyrazolo-pyrimidinopyridines and related compounds. Ten of the newly synthesized compounds have been screened for their biological activities against three Gram positive, three Gram negative bacteria, as well as six fungal strains. Most of the tested compounds showed activities against the strains used. Compound **19c** proved to be the most potent compound of all those used.

Supplementary Materials

Supplementary materials can be accessed at: http://www.mdpi.com/1420-3049/17/12/14464/s1.

Acknowledgements

The authors would like to express their gratitude to R. M. Mahfouz, Professor of Inorganic Chemistry at Assiut University for running the ¹³C-NMR analysis at King Saoud University, Saudi Arabia.

References

- 1. Ganem, B. Strategies for innovation in multicomponent reaction design. *Acc. Chem.* **2009**, *42*, 463–472.
- 2. Padwa, A. Domino reactions of rhodium (II) carbenoids for alkaloid synthesis. *Chem. Soc. Rev.* **2009**, *38*, 3072–3081.
- 3. Ali, R.; Amir, T.M.; Morteza, R.; Aram, R. Novel three-component reaction of a secondary amine and a 2-hydroxybenzaldehyde derivative with an isocyanide in the presence of silica gel: an efficient one-pot synthesis of benzo[b] furan derivatives. *Tetrahedron Lett.* **2009**, *50*, 5625–5627.
- 4. Zeinab, Z.; Mehdi, K.; Ali, R.; Alireza, F.; Ali, S.; Katarzyna, Ś.; Tadeusz Lis, A.S.; Synthesis of functionalized furo[3,2-c] coumarins via a one-pot oxidative pseudo three-component reaction in poly(ethylene glycol. *Tetrahedron* **2012**, *68*, 6721–6726.
- 5. Domling, A. Multicomponent reactions. *Chem. Rev.* **2006**, *106*, 17–89.
- 6. D'Souza, D.M.; Muller, T.J. Multi-component syntheses of heterocycles by transition-metal catalysis. *J. Chem. Soc. Rev.* **2007**, *36*, 1095–1108.
- 7. Ali, R.; Ali, S. Iminophosphorane-mediated one-pot synthesis of 1,3,4-oxadiazole Derivatives. *ARKIVOC* **2008**, *xvi*, 235–242.

8. Ali, S.; Ali, R.; Nouri, B.; Richard, W. The reaction of (*N*-isocyanimino)triphenylphosphorane with dialkyl acetylenedicarboxylates in the presence of 1,3-diphenyl-1,3-propanedione: A novel three-component reaction for the stereoselective synthesis of dialkyl (*Z*)-2-(5,7-diphenyl-1,3,4-oxadiazepin-2-yl)-2-butenedioates. *Tetrahedron Lett.* **2007**, *48*, 2617–2620.

- 9. Ali, S.; Ali, R. The reaction of (*N*-isocyanimino)triphenylphosphorane with benzoic acid derivatives: A novel synthesis of 2-aryl-1,3,4-oxadiazole derivatives. *Tetrahedron Lett.* **2007**, *48*, 1549–1551.
- 10. Shore, G.; Yoo, W.J.; Li, C.J.; Organ, M. Propargyl amine synthesis catalysed by gold and copper thin films using microwave assistant, continuous flow organic synthesis. *Chem. Eur. J.* **2010**, *16*, 126–133.
- 11. Hardy, C.R. The chemistry of pyrazolopyridine. Adv. Heterocyl. Chem. 1984, 36, 343–409.
- 12. Elnagdi, M.H.; Elgemeie, G.H.; El-Moghayar, R.M.H. Chemistry of pyrazolopyrimidines. *Adv. Heterocycl. Chem.* **1987**, *41*, 319–376.
- 13. Jiaro, Q.; Jaime, P.; Silva, C.; Rodrigo, A.; Braulio, I.; Manuel, N.; Justo, C.; Mike, H. Solvent free synthesis of fused pyrazolo [1,5-*a*]pyrimidines by reaction of 5-amino-1-*H*-pyrazoles and β-triketones. *Open Org. Chem. J.* **2008**, *2*, 92–99.
- 14. Jiaro, Q.; Jaime, P.;Rodrigo, A.; Manuel, N.; Justo, C.; Mike, H. Regioselective synthesis of novel substituted pyrazolo[1,5-a]pyrimidines under solvent-free conditions. *Tetrahedron Lett.* **2008**, *49*, 6254–6256.
- 15. Jiaro, Q.; Jaime, P.; Hugo, S.; Rodrigo, A.; Braulio, I.; Manuel, N.; Justo, C. Regioselective synthesis of fused benzopyrazolo[3,4-b]quinolines under solvent-free conditions. *Tetrahedron Lett.* **2007**, *48*, 1987–1990.
- 16. Aly, M.F.; El-Naggar, G.M.; El-Emary, T.I.; Girgg, R.; Metwally, S.A.; Sivagnanam, S. X=Y-ZH Compounds as potential 1,3-Dipoles part 41. Azomethine Ylide Formation from the reaction of -Amino acids and esters wth Alloxan (Strecker Degradation) and with 1-phenyl-3-methyl pyrazoline 4,5-dione. *Tetrahedron* **1994**, *50*, 895–906.
- 17. Selleri, S.; Burni, F.; Castanzo, A.; Gueririni, G.; Malmbeg-Aiello, P.; lavarone, G.; Martini, C. Synthesis and preliminary evaluation of pyrazolo[1,5-*a*]pyrido[3,4-*e*]pyrimidin-6-(7*H*)-ones and related comounds as benzodiazepine receptor ligands and anticonvulsant agents. *Eur. J. Med. Chem.* **1992**, *27*, 985–990.
- 18. Poreba, K.; Wietrzyk, J.; Opolski, A. Synthesis and antiproliferative activity *in vitro* of new 2,3 or 4 substituted pyrido[2',3':3,4]pyrazolo[1,5-*a*]pyrimidines. *Acta Pol. Pharm.* **2006**, *63*, 189–194.
- 19. Ismail, M.M.F.; Ammar, Y.A.; El-Zahaby, H.S.A.; Eisa, S.I.; Barakat, S.E. Synthesis of Novel-1-pyrazolylpyridinopyrimidin2-ones as potential Anti-inflammatory and Analgesic Agent. *Arch. Pharm. Chem. Life Sci.* **2007**, *340*, 476–481.
- 20. Bi, Y.; Stoy, P.; He, B.; Adam, L.; Krupinski, J.; Normandin, D.; Pongrac, R.; Seliger, L.; Waston, A.; Macor, J.E. The discovery of novel, potent and selective PDE inhibitors. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 2461–2464.
- 21. Ahluwalia, V.K.; Dahiya, A.; Garg, V. Reaction of 5-amino-4-formyl-3-methyl(or phenyl)-1-phenyl-1*H*-pyrazoles with active methylene compounds: Synthesis of fused heterocyclic rings. *Indian J. Chem.* **1997**, *36B*, 88–90 and references sited therein.

22. Metwally, S.A.; El-Naggar, G.M.; El-Emary, T.I. Reactions of 4-(Dicyanomethylene)-3-methyl-1-phenyl-2-pyrazolin-5-one towards methylene comoponds. *Liebgs Ann. Chem.* **1991**, *62*, 961–962.

- 23. Metwally, S.A.; El-Naggar, G.M.; Younis, M.I.; Elnagdi, M.H.; El-Emary, T.I. Reactions of 4-(Dicyanomethylene)-3-methyl-1-phenyl-2-pyrazolin-5- one towards Amines and Phenols. *Liebgs Ann. Chem.* **1989**, *40*, 1037–1040.
- 24. El-Emary, T.I.; El-Dean, A.M.; El-Kashef, H.S. Facile synthesis of some new pyrazolo [3,4-*b*] pyrazines and their antifungal activity. *II Farmaco* **1998**, *53*, 383–388.
- 25. El-Kashef, H.S.; El-Emary, T.I.; Gasquet, M.; Timon-David, M.J.; Vanele, P. New pyrazolo [3,4-*b*] pyrazines: Synthesis and biological activity. *Pharmazie* **2000**, *55*, 572–577.
- 26. El-Emary, T.I.; El-kashef, H.S.; Hussein A.M. New Polycyclic Azines Derived from Pyrazolo [3,4-b] pyridine. *Pharmazie* **2000**, *55*, 356–358.
- 27. Hussein, A.M.; El-Emary, T.I. Polycyclic Pyrazoles: Routes to New Pyrazoloazines. *J. Chem. Res.* **1998**, 228–236.
- 28. El-Emary, T.I.; Bakhite, E.A. Synthesis and biological screening of new 1,3- diphenylpyrazoles with different moieties at position-4. *Pharmazie* **1999**, *2*, 106–111.
- 29. El-Emary, T.I.; Abdel-Mohsen, Sh.A. Synthesis and antimicrobial activity of some new 1,3-diphenyl pyrazoles bearing pyrimidine, pyrimidinethione, thiazolopyrimidine, triazolopyrimidine, thio and alkyl-thiotriazolpyrimidinone moieties at 4- position. *Phosphorous Sulfur Silicon* **2006**, *181*, 2459–2474.
- 30. El-Emary, T.I.; Khalil, A.; Ali, G.A.; El-Adasy, A.A. A facile synthesis of some new Thiazolo[3,2-a]pyridines containing pyrazolyl moiety and their antimicrobial activity. *Phosphorous Sulfur Silicon* **2005**, *180*, 19–30.
- 31. El-Emary, T.I. Synthesis of newly substituted pyrazoles and substituted pyrazolo[3,4-b]pyridines based on 5-amino-3-methyl-1-phenylpyrazole. *J. Chin. Chem. Soc.* **2007**, *54*, 507–518.
- 32. El-Emary, T.I. Synthesis, reactions and biological activity of some new pyrazolo [3,4-b] pyrazines. *J. Chin. Chem. Soc.* **2006**, *53*, 391–401.
- 33. El-Emary, T.I. Synthesis of some newly condensed and uncondensed pyrazolo[3,4-b]pyridines. *Assiut Univ. J. Chem.* **2006**, *35*, 45–63.
- 34. Jairo, Q.; Diana, M.; Braulio, I.; Ridrigo, A.; Manuel, N.; Adolfo, S.; Justo, C.; John, N. Regioselective synthesis of 4,7,8,9-tetrahydro-2*H*-pyrazolo[3,4-*b*]quinolin-5(6*H*)-ones. Mechanism and structural analysis. *Tetrahedron* **2001**, *57*, 6947–6953.
- 35. Bazgir, A; Khanaposhanti, M.; Sooki, A. One-pot synthesis and antibacterial activities of pyrazolo[4',3':5,6]pyrido[2,3-d]pyrimidine-dione derivatives. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 5800–5803.
- 36. Chebanov, V.A.; Sakhno, Y.I.; Desenk, S.M.; Chernenko, V.N.; Musatov, V.I.; Shishkina, S.V.; Shishkin, O.V.; Kappe, O. Cyclocondensation reactions of 5-aminopyrazoles, pyruvic acids and aldehydes. Multicomponent approaches to pyrazolopyridines and related products. *Tetrahedron* **2007**, 1229–1242.
- 37. Shaabani, A.; Seyyedhamez, M.; Maleki, A.; Behnan, M.; Rezazdeh, F. Synthesis of fully substituted pyrazolo[3,4-b]pyridine-5-carboxamide derivatives via a one-pot four-component reaction. *Tetrahedron Lett.* **2009**, *50*, 2911–2913.

38. Balamurugan, K.; Perumal, S.; Menedez, J.C. New four-component reaction in water: a convergent approach to the metal-free synthesis of spiro[indoline/ acenaphthylene-3,4'-pyrazolo[3,4-b]pyridine derivatives. *Tetrahedron* **2011**, *67*, 3201–3208.

- 39. Shawali, A.S.; Elghandour, A.H.; Sayed, A.R.A novel one-pot synthesis of 3-arylazo[1,2,4]triazolo[4,3-*a*]pyrimidin-5-(1*H*)-ones. *Synth. Commun.* **2001**, *31*, 731–740.
- 40. Bedford, G.R.; Taylor, P.J.; Webb, G.A. ¹⁵N-NMR studies of guanidines. II—The fused-in guanidine unit of some oxoheterocycles: A combined ¹⁵N-NMR, ¹³C-NMR and IR study. *Magn. Res. Chem.* **1995**, *33*, 389–394.
- 41. Elguero, J.; Goya, P.; Martinez, A.; Rozas, I. On the Tautomerism of 2-Phenacyl-4-pyrimidinones and Related Compounds. *Chem. Ber.* **1989**, *122*, 919–924.
- 42. Greenhill, J.V.; Ismail, M.J.; Bedford, G.R.; Edwards, P.N.; Taylor, P.J. Conformational and tautmeric studies of acyl guanidines Part 2. Vibrations and C-13 nuclear magnetic resonance spectroscopy. *J. Chem. Soc. Perkin Trans.* **1985**, *2*, 1265–1274.
- 43. Reiter, J.; Bongo, L.; Dyortsok, P. On triazoles XI. structure elucidation of isomeric 1,2,4-triazolopyrimidinones. *Tetrahedron* **1987**, *43*, 2497–2504.
- 44. Rami, V.J. One-Pot Synthesis of Mono- and Dinitro-1,2,4-triazino[3,2-b]benzothiazoles. *Liebigs Ann. Chem.* **1988**, *11*, 1089–1090.
- 45. Heinisch, G.; Holzer, W. Pyrazoles 3. N-1 Protected 4-Substituted Pyrazoles—Synthesis and Nmr Investigation. *Heterocycles* **1988**, *27*, 2443–2457.
- 46. Khalil, Z.H.; Geies, A.A. Synthesis and reactions of some thieno[2,3-d]pyrimidine derivatives. *Phosphorus Sulfur Silicon* **1991**, *60*, 223–231.
- 47. Kwon-Chung, K.J.; Bennett, J.W. Principles of antifungal and antibacterial therapy. *Med. Mycol. Lea Febiger. Philadel.* **1992**, 81–102.

Sample Availability: Samples of the compounds are available from the authors.

© 2012 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution license (http://creativecommons.org/licenses/by/3.0/).