

Article

Unexpected Behavior of Enaminones: Interesting New Routes to 1,6-Naphthyridines, 2-Oxopyrrolidines and Pyrano[4,3,2-de][1,6]naphthyridines

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Abstract: Reaction of enaminones **1a**–**d** with 2-aminoprop-1-ene-1,1,3-tricarbonitrile (**2**) in the presence of AcOH/NH₄OAc afforded 7-amino-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitrile derivatives **9a**–**d**. On the other hand, 2-aminopyrano[4,3,2-*de*] [1,6]naphthyridine-3-carbonitriles **20a**–**c**,**e** were the only obtained products from the reactions of **1a**–**d** with **2** in the presence of AcOH/NaOAc, while **1d** afforded [3,5-bis-(4-chloro-benzoyl)-phenyl]-(4-chloro-phenyl)-methanone **21** under the same condition. The reaction of **2** with diethyl acetylenedicarboxylate in the presence of AcOH/NH₄OAc afforded (4-cyano-5-dicyanomethylene-2-oxo-2,5-dihydro-1*H*-pyrrol-3-yl)-acetic acid ethyl ester **15B**.

Keywords: enaminones; 3-amino-2-cyanopent-2-enedinitrile; 7-amino-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitrile; 2-aminoprop-1-ene-1,1,3-tricarbonitrile

1. Introduction

During the last decade we have been involved in a program aimed at exploring the synthetic potentials of enaminones [1] as building blocks for polyfunctionally substituted aromatics and heteroaromatics [1–3]. We have in the past successfully developed syntheses of polysubstituted benzenes [4,5] and polysubstituted pyridines [6,7] utilizing enaminones **1a**–**d** as starting materials. In the present article we report our further results in this area where a novel one pot synthesis of 7-amino-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitriles **9a**–**d**, 2-aminopyrano[4,3,2-*de*][1,6]naphthyridine-3-carbonitrile derivatives **20a**–**c**,**e** and (4-cyano-5-dicyanomethylene-2-oxo-2,5-dihydro-1*H*-pyrrol-3-yl)-acetic acid ethyl ester **15B** could be achieved. To our knowledge only one derivative of the 2-aminopyrano[4,3,2-*de*][1,6]naphthyridine-3-carbonitrile system has been reported, prepared via a multistep route [8]. The newly synthesized 2-aminopyrano[4,3,2-*de*][1,6]naphthyridine-3-carbonitrile derivatives **20a**–**c**,**e** seem interesting for biological activity investigations as investigations on such polynuclear aromatics are rare [8].

2. Results and Discussion

Although the reaction of enaminone 1c and 3-amino-2-cyanopent-2-enedinitrile (2) in refluxing acetic acid in presence of ammonium acetate (NH₄OAc) has been reported earlier [1,9] to afford 2,4-diamino-5-benzoylisophthalonitrile, with molecular formula $C_{15}H_{10}N_4O$ and molecular mass $M^+ = 262$, we found, however, that enaminone 1c and compound 2 react in acetic acid in the presence of NH₄OAc to yield a completely different product with the same molecular formula and molecular mass. Both starting compounds the enaminone 1 and 3-amino-2-cyanopent-2-enedinitrile (2) are bifunctional reactants. The carbonyl group of 1 can form an imine $(1 + 2 \rightarrow 3)$ or participate in a Knoevenagel reaction with the methylene group of 2 $(1+2 \rightarrow 6)$. Michael-type additions $1+2 \rightarrow 4$ or $1+2 \rightarrow 5$ are other alternatives. The subsequent ring closure reactions can lead to the pyridine derivatives 7 or 8. Finally, the 1,6-naphthyridine systems 9–12 can also be generated. The yields obtained from 1a–d are between 75 and 90%. We have depicted all possible end products that could be obtained from reacting 1a-d and 2 in Scheme 1 as secondary carbamides, but their tautomers having a hydrogen atom bound to N-1 or to the oxygen atom have to be considered as well. The structure determination of the reaction products of 1a-d and 3-amino-2-cyanopent-2-enedinitrile (2) proved to be very difficult, because all twelve possible 1,6-naphthyridine derivatives should have similar ¹H- and ¹³C-NMR spectra. Therefore, we performed a series of 2D NMR measurements: (1H,1H)COSY, (1H,13C)HSQC, (1H,13C)HMBC, (1H,15N)HSQC, (1H,15N)HMBC, and INADEQUATE. The final decision was made in favor of the structures **9a–d**. Figures 1 and 2 showed the ¹H, ¹³C and ¹⁵N chemical shifts and the results of the 2D-INADEQUATE and the two HMBC measurements which represent the basis for the assignment of the chemical shifts (Scheme 1).

Scheme 1. Formation of compounds 9a-d.

Figure 1. Upper part: (13 C, 1 H)- and (15 N, 1 H) couplings nJ (n = 2–4) according to the crosspeaks observed in the HMBC measurements of **9c**. Lower part: Assignment of all 1 H, 13 C and 15 N signals of **9c**. The δ values obtained in CD₃SOCD₃ at room temperature are related to TMS and NH₃ (liquid). The measurements were performed at 14.1 T (600 MHz for 1 H).

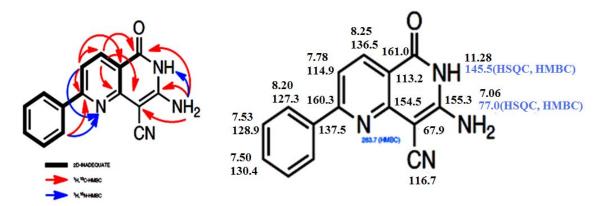
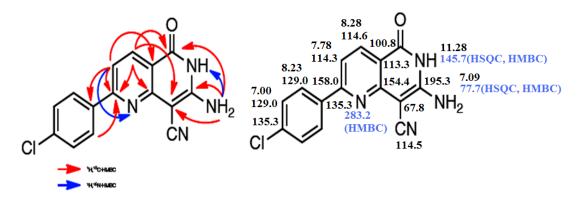
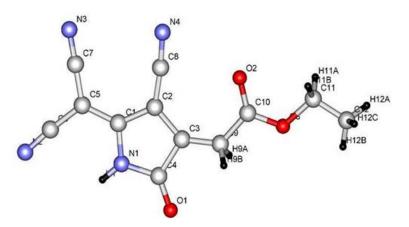


Figure 2. Upper part: (13 C, 1 H)- and (15 N, 1 H) couplings nJ (n = 2–4) according to the crosspeaks observed in the HMBC measurements of **9d**. Lower part: Assignment of all 1 H, 13 C and 15 N signals of **9d**. The δ values obtained in CD₃SOCD₃ at room temperature are related to TMS and NH₃ (liquid). The measurements were performed at 14.1 T (600 MHz for 1 H).



In order to extend this finding further we reacted 3-amino-2-cyanopent-2-enedinitrile (2) with diethyl acetylenedicarboxylate (13) in the presence of AcOH/NH₄OAc. In this case, however, ethyl (4-cyano-5-dicyanomethylene-2-oxo-pyrrolidin-3-ylidene) acetate 15B was obtained as indicated by an X-ray crystal structure determination (Figure 3) [10]. It is believed that compound 2 reacts with 13 to initially afford adduct 14 that cyclizes preferably to the pyrrolidine 15 rather than the alternative pyridine derivative 16 (Scheme 2). Although 15 has been shown to exist as a solid, in DMSO solution only form 15B exists, as indicated by the 1 H-NMR data that showed a singlet at $\delta = 5.50$ ppm for the methylene proton and a broad signal at $\delta = 2.48$ ppm for proton of the dicyanomyl moiety.

Figure 3. X-ray crystal structure of compound **15A**.



We conducted the same reactions of enaminones $1\mathbf{a}-\mathbf{c},\mathbf{e}$ with 3-amino-2-cyanopent-2-enedinitrile (2) in the presence of AcOH/NaOAc. This reaction afforded in this case a different product with molecular formula $C_{20}H_{10}N_4O_2S_2$ and molecular mass $\mathbf{M}^+=402$. It is believed that compound $1\mathbf{a}$ reacted with 3-amino-2-cyanopent-2-enedinitrile (2) to form the highly unsaturated intermediates $17\mathbf{a}-\mathbf{c},\mathbf{e}$ and their anions $18\mathbf{a}-\mathbf{c},\mathbf{e}$, respectively (Scheme 3).

Scheme 2. Formation of compound 15B.

The intermediate 18 can undergo a cyclic π -electron shift ($6\pi \to 3\sigma + 3\pi$, valence isomerization) to 19a-c,e. Protonation of 19a-c,e, followed by 1,5-H-shift and dehydrogenation lead finally to the 2-aminopyrano[4,3,2-de][1,6]naphthyridine-3-carbonitrile 20a-c,e. Yields of 20a-c,e amounted to 85–92% when 2:1-mixtures of 1 and 2 were refluxed in AcOH/NaOAc. The reaction of 1d with 2 under the same conditions afforded [3,5-bis-(4-chloro-benzoyl)-phenyl]-(4-chloro-phenyl)-methanone 21 previously obtain by upon refluxing 1d in AcOH. It has been previously observed that 1d readily trimerise on attempted condensation with nucleophils [12].

Scheme 3. Formation of compounds 20a-c,e.

The structure determination of **20** was based on 2D NMR measurements (COSY, HSQC, HMBC and (¹H, ¹⁵N) HMBC) of **20a** and on a crystal structure analysis of **20c** (Figures 4 and 5) [11].



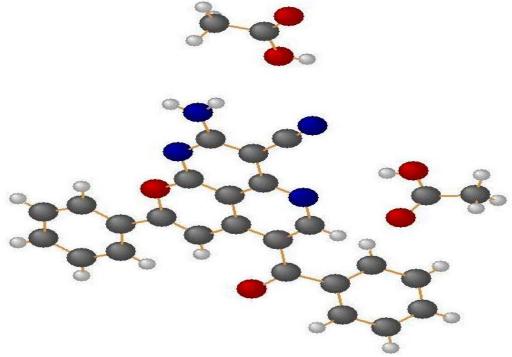


Figure 5. Left part: (13 C, 1 H) couplings ^{n}J (n = 2–4) according to the crosspeaks observed in HMBC measurement of **20a**. Right part: Assignment of all 1 H and 13 C signals and one 15 N signal of **20a**. The δ values obtained in CD₃SOCD₃ at room temperature are related to TMS and NH₃ (liquid). The measurements were performed at 14.1 T (600 MHz for 1 H).

3. Experimental

3.1. General

Melting points are reported uncorrected and were determined with a Sanyo (Gallaenkamp) instrument. Infrared spectra were recorded using KBr pellets and a Perkin-Elmer 2000 FT–IR instrument. ¹H- and ¹³C-NMR spectra were determined by using a Bruker DPX instrument at 600 MHz for ¹H-NMR and 150 MHz for ¹³C-NMR and either CDCl₃ or DMSO-d₆ solutions with TMS as internal standards. Chemical shifts are reported in δ (ppm). Mass spectra were measured using VG Autospec Q MS 30 and MS 9 (AEI) spectrometer, with the EI (70 EV) mode. Elemental analyses were carried out by using a LEOCHNS-932 Elemental Analyzer. X-ray crystal structure determined by using a Single Crystal X-ray Crystallography-Rigaku Rapid II & Bruker X8 Prospector system.

3.2. General Procedure for the Synthesis of 9a-d

A mixture of enaminone **1a–d** (0.01 mol) and 3-amino-2-cyanopent-2-enedinitrile (**2**, 1.32 g, 0.01 mol) in AcOH (25 mL)/NH₄OAc (1 g) was heated under reflux for 2 h (followed until completion by TLC using 1:1 ethyl acetate–petroleum ether as eluent). The mixture was then cooled and poured onto ice-water. The solid, so formed, was collected by filtration and recrystallized from AcOH to give yellow crystals.

7-Amino-5-oxo-2-(thienyl)-5,6-dihydro-1,6-naphthyridine-8-carbonitrile (**9a**). Yield 75%; mp. 291–292 °C. Anal. Calcd. for $C_{13}H_8N_4OS$ (268.29): C, 58.20; H, 3.01; N, 20.88; S, 11.95%. Found: C, 58.17; H, 3.13; N, 20.65; S, 11.92%; IR (KBr, cm⁻¹): 3471 (NH), 3363, 3295 (NH₂) 2252 (CN), 1652 (CO); ¹H-NMR (DMSO-d₆): δ = 7.05 (br, 2H, NH₂, D₂O exchangeable), 7.21 (t, 1H, J = 4.0, thienyle-H), 7.69 (d, 1H, J = 8.0, CH), 7.77 (d, 1H, J = 4.0, thienyl-H), 7.94 (d, 1H, J = 4.0, thienyl-H), 8.22 (d, 1H, J = 8.0, CH), 11.22 (br, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆): δ = 160.6, 156.1, 155.2, 154.4, 143.6, 136.2, 130.9, 128.6, 127.7, 116.2, 113.5, 112.56, 67.4. MS: m/z (%) 268 (M⁺, 40), 256 (35), 241 (15), 213 (25), 185 (25), 169 (20), 129 (55), 97 (40), 73 (100).

7-Amino-2-(furyl)-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitrile (**9b**). Yield 80%; mp. 287–289 °C. Anal. Calcd. for $C_{13}H_8N_4O_2$ (252.23): C, 61.90; H, 3.20; N, 22.21%. Found: C, 61.84; H, 3.31; N, 22.32%; IR (KBr, cm⁻¹): 3424 (NH), 3343, 3240 (NH₂), 2214 (CN), 1662 (CO); ¹H-NMR (DMSO-d₆): $\delta = 6.72$ (t, 1H, J = 4.0, furyl-H), 7.05 (br, 2H, NH₂, D₂O exchangeable), 7.25 (d, 1H, J = 4.0, furyl-H), 7.51 (d, 1H, J = 8.0, CH), 7.94 (d, 1H, J = 4.0, furyl-H), 8.25 (d, 1H, J = 8.0, CH), 11.24 (br, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆): $\delta = 160.1$, 155.3, 154.5, 152.4, 152.3, 145.7, 136.3, 116.4, 113.3, 112.7, 112.6, 111.9, 67.5. MS: m/z (%) 252 (M⁺, 100), 224 (20), 195 (10), 73 (90).

7-Amino-5-oxo-2-phenyl-5,6-dihydro-1,6-naphthyridine-8-carbonitrile (**9c**). Yield 90%; mp. 253–255 °C. Anal. Calcd. for $C_{15}H_{10}N_4O$ (262.27): C, 68.69; H, 3.84; N, 21.36%. Found: C, 68.65; H, 3.75; N, 21.40%; IR (KBr, cm⁻¹): 3325 (NH), 3251, 3209 (NH₂), 2209 (CN), 1673 (CO); ¹H-NMR (DMSO-d6): $\delta = 7.07$ (br, 2H, NH₂, D₂O exchangeable), 7.52–7.53 (m, 3H, Ph-H), 7.75 (d, 1H, J = 8.0, CH), 8.20–8.21 (m, 2H, Ph-H), 8.28 (d, 1H, J = 8.0, CH), 11.27 (br, 1H, NH, D₂O exchangeable); 13 C-NMR (DMSO-d₆): $\delta = 160.8$, 160.2, 155.2, 154.3, 137.5, 136.4, 130.3, 128.8 (2C), 127.2 (2C),

116.5, 114.8, 113.1, 67.8. MS: m/z (%) 262 (M⁺, 100), 234 (15), 217 (10), 192 (5), 164 (10), 129 (10), 83 (10), 73 (25).

7-Amino-2-(4-chlorophenyl)-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitrile (**9d**). Yield 90%; mp. 275–276 °C. Anal. Calcd. for C₁₅H₉ClN₄O (296.72): C, 60.72; H, 3.06; N, 18.88%. Found: C, 60.70; H, 3.11; N, 18.87%; IR (KBr, cm⁻¹): 3436 (NH), 3324, 3216 (NH₂), 2211 (CN), 1678 (CO); ¹H-NMR (DMSO-d₆): δ = 7.06 (s, 2H, NH₂, D₂O exchangeable), 7.54 (d, 2H, Ph-H), 7.69 (d, 1H, J = 8.0, CH), 8.17 (d, 2H, Ph-H), 8.23 (d, 1H, J = 8.0, CH), 11.25 (br, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆): δ = 160.7, 158.8, 155.2, 154.2, 136.4, 136.2, 135.2, 129.0 (2C), 128.8 (2C), 116.4, 114.6, 113.2, 67.8. MS: m/z (%) 296 (M⁺, 100), 268 (20), 216 (15), 189 (15), 164 (10), 130 (10), 88 (15), 73 (20).

Synthesis of (4-cyano-5-dicyanomethylene-2-oxo-2,5-dihydro-1H-pyrrol-3-yl)-acetic acid ethyl ester (15B). A mixture of 3-amino-2-cyanopent-2-enedinitrile (2, 1.32 g, 0.01 mol) and diethyl acetylenedicarboxylate (13, 1.70 g, 0.01 mol) in AcOH (25 mL)/NH₄OAc (1 g) was refluxed for 2 h (followed until completion by TLC using 1:1 ethyl acetate–petroleum ether as eluent). The mixture was cooled and then was poured onto ice-water. The solid, so formed, was collected by filtration and recrystallized from EtOH to give orange crystals, yield 70%; mp. 200–202 °C. Anal. Calcd. for C₁₂H₈N₄O₃ (256.22): C, 56.25; H, 3.15; N, 21.87%. Found: C, 56.22; H, 3.27; N, 22.00%; IR (KBr, cm⁻¹): 3363 (NH), 2223 (CN), 2211 (2CN), 1695 (CO), 1653 (CO); ¹H-NMR (DMSO-d₆): δ = 1.22 (t, 3H, J = 8.0, CH₃), 2.48 (br, 1H, J = 8.0, CH), 4.10 (q, 2H, CH₂), 5.50 (s, 1H, CH), 10.84 (br, 1H, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆): δ = 169.0, 165.8, 161.3, 138.5, 117.2, 116.0, 115.6, 100.1 (2C), 68.2, 59.0, 14.2. MS: m/z (%) 256 (M⁺, 50), 211 (15), 184 (75), 156 (10), 112 (100), 97 (10), 84 (25), 70 (25), 55 (90).

3.3. General Procedure to Syntheses of 20a-c,e

A mixture of enaminone **1a–e** (0.01 mol) and 2-aminoprop-1-ene-1,1,3-tricarbonitrile (**2**, 1.32 g, 0.01 mol) in AcOH/NaOAc (1 g) was refluxed for 2 h (followed until completion by TLC using 1:1 ethyl acetate–petroleum ether as eluent). The mixture was cooled and then was poured onto ice-water. The solid, so formed, was collected by filtration and recrystallized from AcOH to give yellow crystals.

2-Amino-8-(thiophen-2-yl)-6-(thiophene-2-carbonyl)pyrano-[4,3,2-de][1,6]naphthyridine-3-carbonitrile (20a). Yield 92%; mp. 369–370 °C. Anal. Calcd. for C₂₀H₁₀N₄O₂S₂ (402.45): C, 59.69; H, 2.50; N, 13.92; S, 15.93%. Found: C, 59.72; H, 2.33; N, 13.97; S, 15.86%; IR (KBr, cm⁻¹): 33363, 3295 (NH₂), 2213 (CN), 1642 (CO); ¹H-NMR (DMSO-d₆): δ = 7.27 (t, 1H, J = 4.0, thienyl-H), 7.31 (t, 1H, J = 4.0, thienyl-H), 7.47 (s, 1H, CH), 7.78 (d, 1H, J = 4.0, thienyl-H), 7.86 (br, 3H, thienyl-H, NH₂, D₂O exchangeable), 7.94 (d, 1H, J = 4.0, thienyl-H), 8.16 (d, 1H, J = 4.0, thienyl-CH), 8.98 (s, 1H, CH); ¹³C-NMR (DMSO-d₆): δ = 184.8, 162.1, 160.4, 157.2, 156.2, 154.0, 143.7, 136.1, 135.8, 134.1 (2C), 131.6, 129.3, 129.00, 128.9, 118.1, 115.9, 104.0, 98.3, 77.1 . MS: m/z (%) 402 (M⁺, 100), 373 (15), 319 (25), 263 (5), 236 (5), 187 (10), 111 (30), 83 (5).

2-Amino-6-(furan-2-carbonyl)-8-(furan-2-yl)pyrano[4,3,2-de]-[1,6]naphthyridine-3-carbonitrile (**20b**). Yield 90%; mp. 375–377 °C. Anal. Calcd. for $C_{20}H_{10}N_4O_4$ (370.32): C, 64.87; H, 2.72; N, 15.13%. Found: C, 64.88; H, 2.68; N, 15.22%; IR (KBr, cm⁻¹): 3471, 3373(NH₂), 2210 (CN), 1653 (CO); ¹H-NMR (DMSO-d₆): δ = 6.78 (t, 1H, J = 4.0, furyl-H), 6.82 (t, 1H, J = 4.0, furyl-H), 7.19 (d, 1H, J = 4.0, furyl-H), 7.44 (s, 1H, CH), 7.46 (d, 1H, J = 4.0, furyl-H), 7.83 (br, 2H, NH₂, D₂O exchangeable), 8.04 (d, 1H, J = 4.0, furyl-H), 8.24 (d, 1H, J = 4.0, furyl-H), 8.98 (s, 1H, CH); ¹³C-NMR (DMSO-d₆): δ = 179.20, 162.29, 161.98, 160.13, 157.31, 156.08, 151.64, 150.06, 148.72, 147.23, 145.30, 140.47, 121.34, 117.48, 115.76, 113.36, 112.97, 103.98, 97.56, 77.23. MS: m/z (%) 370 (M⁺, 90), 264 (15), 224 (25), 195 (15), 169 (10), 129 (10), 83 (30), 73 (35).

2-Amino-6-benzoyl-8-phenylpyrano[4,3,2-de][1,6]naphthpyrid-ine-3-carbonitrile (**20c**). Yield 88%; mp. 318–319 °C. Anal. Calcd. for $C_{24}H_{14}N_4O_2$ (390.11): C, 73.84; H, 3.61; N, 14.35%. Found: C, 73.92; H, 3.6; N, 14.28%; IR (KBr, cm⁻¹): 3445, 3341 (NH₂), 2216 (CN), 1646 (CO); ¹H-NMR (DMSO-d₆): δ = 7.56–7.61 (m, 5H, Ph-H), 7.65 (s, 1H, CH), 7.69–7.86 (m, 7H, Ph-H, NH₂, D₂O exchangeable), 8.72 (s, 1H, CH); ¹³C-NMR (DMSO-d₆): δ = 193.50, 162.10, 160.86, 158.22, 158.31, 156.20, 141.80, 141.10, 137.70, 135.20, 132.90, 131.60, 130.70, 129.60 (2C), 129.30, 128.80 (2C), 125.70, 117.75, 115.77, 104.21, 100.10, 77.10. MS: m/z (%) 390 (M⁺, 100), 373 (15), 313 (65), 257 (10), 230 (10), 188 (5), 181 (5), 105 (20), 77 (25).

2-Amino-6-(4-methoxybenzoyl)-8-(4-methoxyphenyl)pyrano-[4,3,2-de][1,6]naphthyridine-3-carbonitrile (20e). Yield 85%; mp. 325–327 °C. Anal. Calcd. for $C_{26}H_{18}N_4O_4$ (450.45): C, 69.33; H, 4.03; N, 12.44%. Found: C, 69.40; H, 4.12; N, 12.42%; IR (KBr, cm⁻¹): 3424, 3343 (NH₂), 2214 (CN), 1644 (CO); ¹H-NMR (DMSO-d₆): δ = 3.84 (s, 3H, CH₃), 3.88 (s, 3H, CH₃), 7.07-7.11 (m, 4H, Ph-H), 7.42 (s, 1H, CH), 7.75–7.80 (m, 6H, Ph-H, NH₂, D₂O exchangeable), 8.69 (s, 1H, CH); ¹³C-NMR (DMSO-d₆): δ = 191.90, 163.30, 162.90, 160.82, 158.21, 157.65, 157.50, 156.20, 140.81, 132.11 (2C), 130.16 (2C), 127.50, 123.10, 118.62, 115.53, 114.27 (2C), 114.10 (2C), 103.90, 98.50, 77.15, 55.36, 55.41. MS: m/z (%) 450 (M⁺, 100), 419 (20), 407 (5), 343 (30), 300 (10), 211 (15), 135 (25), 107 (5), 77 (20).

3.4. Synthesis of [3,5-bis-(4-chlorobenzoyl)phenyl]-(4-chlorophenyl)methanone (21)

A mixture of enaminone **1d** (2.09 g, 0.01 mol) and 2-aminoprop-1-ene-1,1,3-tricarbonitrile (**2**, 1.32 g, 0.01 mol) in AcOH/NaOAc (1 gm) was refluxed for 2 h (followed until completion by TLC using 1:1 ethyl acetate-petroleum ether as eluent). The mixture was cooled and then was poured onto ice-water. The solid, so formed, was collected by filtration and recrystallized from AcOH to give faint yellow crystals, yield 80%. This product was also prepared *via* refluxing **1d** in AcOH as described earlier by Elnagdi *et al.* [12].

4. Conclusions

New simple and efficient routes for the synthesis of 7-amino-5-oxo-5,6-dihydro-1,6-naphthyridine-8-carbonitrile derivatives **9a**–**d**, (4-cyano-5-dicyanomethylene-2-oxo-pyrrolidin-3-ylidene)-acetic acid ethyl ester **15B** and 2-aminopyrano[4,3,2-*de*][1,6]naphthyridine-3-carbonitrile derivatives **20a**–**c**,**e** from the reaction of enaminones with 2-aminoprop-1-ene-1,1,3-tricarbonitrile (**2**) have been described.

These products look interesting for potential biological evaluation. Moreover, all the products have latent functional moieties that seem interesting precursors to other derivatives of the described ring systems.

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- 10. CCDC 861196 contains the supplementary crystallographic data for compound **15A**. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk.

11. CCDC 838314 contains the supplementary crystallographic data for compound **20c**. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk.

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Sample Availability: Samples of the all compounds are available from the authors.

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