

Article

Diterpenoids with Anti-Inflammatory Activity from the Wood of *Cunninghamia konishii*

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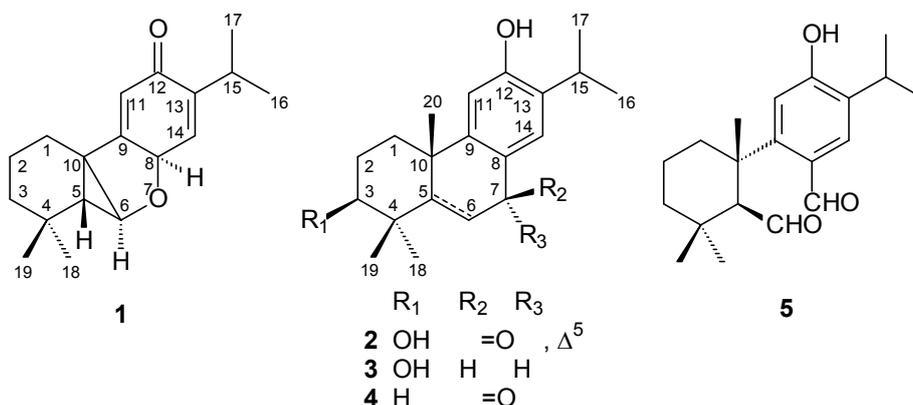
Abstract: Two new diterpenoids, konishone (**1**) and 3 β -hydroxy-5,6-dehydrosugiol (**2**), along with three known diterpenoids—hinokiol (**3**), sugiol (**4**), and 12-hydroxy-6,7-secoabieta-8,11,13-triene-6,7-dial (**5**)—were isolated from the wood of *Cunninghamia konishii*. Compound **1** is a novel skeleton of the 7,20-dinorabietane-type diterpene. In addition, when RAW264.7 macrophages were treated with different concentrations of compounds **1**, **3**, and **5** together with LPS, a significant concentration-dependent inhibition of NO production was detected. The IC₅₀ values for inhibition of nitrite production of compounds **1**, **3**, and **5** were about 9.8 \pm 0.7, 7.9 \pm 0.9, and 9.3 \pm 1.3 μ g/mL, respectively. This study presents the potential utilization of compounds **1**, **3**, and **5**, as lead compounds for the development of anti-inflammatory drugs.

Keywords: *Cunninghamia konishii*; Taxodiaceae; diterpenoid; 7,20-dinorabietane; konishone; 3 β -hydroxy-5,6-dehydrosugiol; anti-inflammatory

1. Introduction

Cunninghamia konishii Hayata (Taxodiaceae), one of the *Cunninghamia* which has two species in eastern Asia, is an endemic coniferous tree distributed in the northern and central part of Taiwan at altitudes of 1,300–2,700 m [1]. The wood of this tree is one of the best building materials available in Taiwan. The chemical components of wood [2–9], bark [10], leaf [8], and whole plant [11] of *C. konishii* have been reported. Some isolates of this plant exhibit antifungal activity [8,9] and cytotoxicity [11]. In connection with our interest in the components of this plant, a further chemical study on the same EtOAc-soluble fraction led to the isolation of two new diterpenoids, konishone (**1**) and 3 β -hydroxy-5,6-dehydrosugiol (**2**), along with three known diterpenoids, namely hinokiol (**3**), sugiol (**4**), and 12-hydroxy-6,7-secoabieta-8,11,13-triene-6,7-dial (**5**) (Figure 1). The isolation and structural elucidation of these new compounds and the anti-inflammatory activity of the isolates are described herein.

Figure 1. The chemical structures of compounds **1–5**.



2. Results and Discussion

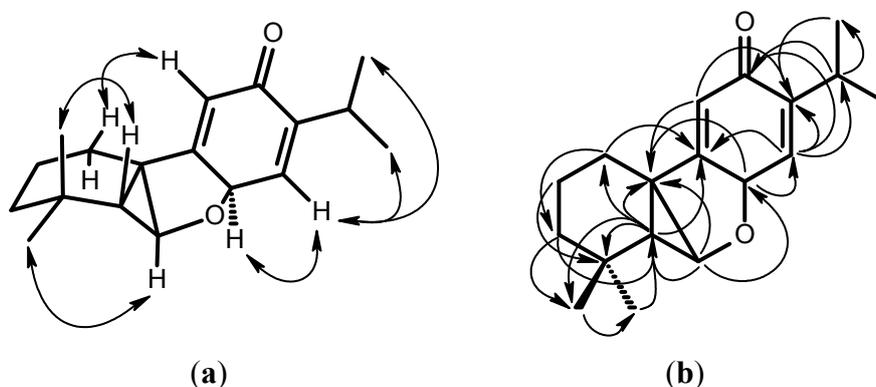
Konishone (**1**) was obtained as yellowish oil. Its molecular formula was established as C₁₈H₂₄O₂ by the HRMS peak at m/z 272 [M⁺]. The presence of a cross conjugated dienone was revealed by the UV (237, 260 nm), IR spectrum data (3078, 1653, 1626, 1602 cm⁻¹), ¹H-NMR (Table 1) [δ 6.06 (s, 1H), 6.31 (s, 1H)], and ¹³C-NMR (Table 1) [δ 187.1 (C=O), 161.9, 149.6, 138.5, 126.5]. The ¹H-NMR spectrum (Table 1) showed an isopropyl group [δ 1.05 (d, J = 6.0 Hz, H-16), 1.07 (d, J = 6.0 Hz, H-17), 2.96 (sept, J = 6.0 Hz, H-15)] attached on a double bond. Two methyl group signals at δ 1.00 (H-19) and 1.21 (H-18) showing HMBC correlations with δ 34.5 (C-4) suggested the presence of a geminal dimethyl group. Two oxygenated carbon signals at δ 79.7 (C-6) and 67.5 (C-8) have corresponding proton signals at δ 4.61 (d, J = 10.3 Hz, H-6) and 4.11 (s, H-8), respectively. H-8 showed a HMBC correlation with C-6, C-10, C-11 and C-14, and H-6 showed HMBC correlations with C-5, C-8 and C-10. The result suggested that C-6 was linked to C-8 by an ether linkage. H-5 expressed HMBC correlations with C-1, C-3, C-4, C-6, C-10, C-18, and C-19. The molecular formula C₁₈H₂₄O₂ indicated index of hydrogen deficiency (IHD) of seven. According to the above evidence and the HMBC correlations, konishone was proposed to be a 7,20-dinorabietane diterpene. By using AM1 theoretical calculations, the most stable conformation has the angles of H-C₅-C₆-H and H-C₈-C₁₄-H being 136.5°

and 87.6°, respectively. The result agreed to the coupling constant of H-5 and H-6 ($J = 10.3$ Hz); H-8 and H-14 (singlet). According to the above data, the structure of konishone was elucidated as **1**, which was further confirmed by COSY, NOESY (Figure 2), ^{13}C -NMR (Table 1), DEPT, HMQC and HMBC (Figure 2) experiments. This is a novel 7,20-dinorabietane-type diterpene skeleton.

Table 1. ^1H - (400 MHz) and ^{13}C -NMR (100 MHz) data (CDCl_3) of compounds **1** and **2**. Chemical shifts δ in ppm relative to TMS, J in Hz.

Position	1		2	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1	1.42 (m, 1H), 1.72 (m, 1H)	38.1	1.63 (ddd, 13.6, 3.6, 3.6, 1H) 2.13 (ddd, 13.6, 3.6, 3.6, 1H)	34.5
2	1.58 (m, 1H), 1.72 (m, 1H)	18.1	1.95 (m, 1H), 2.01 (m, 1H)	27.2
3	1.20 (m, 1H), 1.41 (m, 1H)	42.9	3.39 (dd, 11.6, 4.8, 1H)	76.5
4	–	34.5	–	43.2
5	1.33 (d, 10.3, 1H)	52.5	–	171.0
6	4.61 (d, 10.3, 1H)	79.7	6.51 (s, 1H)	125.7
7	–	–	–	185.1
8	4.11 (s, 1H)	67.5	–	129.3
9	–	161.9	–	153.3
10	–	56.0	–	40.6
11	6.31 (s, 1H)	126.5	6.82 (s, 1H)	111.1
12	–	187.1	–	157.5
13	–	149.6	–	133.7
14	6.06 (s, 1H)	138.5	7.98 (s, 1H)	125.0
15	2.96 (sept, 6.0, 1H)	26.3	3.16 (sept, 6.8, 1H)	26.9
16	1.05 (d, 6.0, 3H)	21.6	1.25 (d, 6.8, 3H)	22.3
17	1.07 (d, 6.0, 3H)	22.4	1.28 (d, 6.8, 3H)	22.5
18	1.21 (s, 3H)	21.3	1.33 (s, 3H)	22.4
19	1.00 (s, 3H)	34.8	1.28 (s, 3H)	27.4
20	–	–	1.49 (s, 3H)	32.3

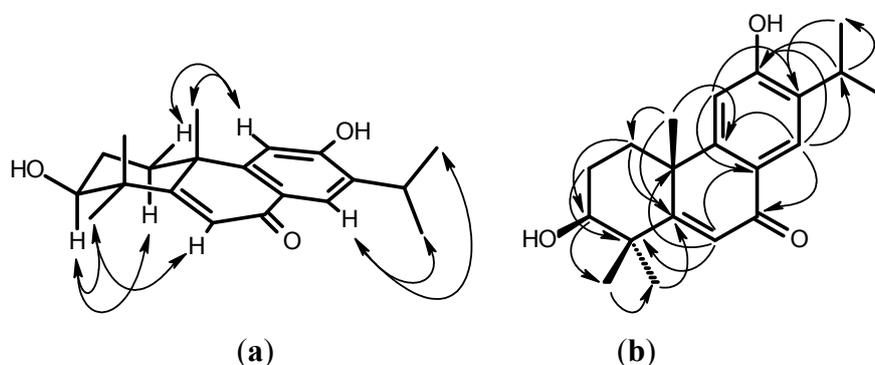
Figure 2. Key NOESY contacts (a) and HMBC connectivities (b) of compound **1**.



3 β -Hydroxy-5,6-dehydrosugiol (**2**) was obtained as a yellowish amorphous solid. Its molecular formula was established as $\text{C}_{20}\text{H}_{26}\text{O}_3$ by the HRMS peaks at m/z 314 [M^+]. Absorptions in the IR

spectrum were attributable to a hydroxy (3363 cm^{-1}), a conjugated carbonyl (1639 cm^{-1}), and a benzene ring ($1613, 1502\text{ cm}^{-1}$). The ^1H - and ^{13}C -NMR spectra (Table 1) showed three singlet methyl groups [δ 1.33 (H-18), 1.28 (H-19), 1.49 (H-20)], an isopropyl group [δ 1.25 (d, $J = 6.8\text{ Hz}$, H-16), 1.28 (d, $J = 6.8\text{ Hz}$, H-17), 3.16 (sept, $J = 6.8\text{ Hz}$, H-15)] attached on a phenyl group [δ 6.82 (s, H-11), 7.98 (s, H-14), 129.3 (C-8), 153.3 (C-9), 111.1 (C-11), 157.5 (C-12), 133.7 (C-13), 125.0 (C-14)], and a conjugated enone [δ 6.51 (s, H-6), 185.1 (C-7), 171.0 (C-5), 125.7 (C-6)]. The ^1H -NMR spectrum (Table 1) of **2** was similar to that of 5,6-dehydrosugiol [12], except for a hydroxy group at C-3. The proton resonating at δ 3.39 (dd, $J = 11.6, 4.8\text{ Hz}$) was attributable to H-3 with an α -axial orientation and was geminal to the hydroxy group. According to the above data, the structure of 3 β -hydroxy-5,6-dehydrosugiol was elucidated as **2**, which was further confirmed by COSY, NOESY (Figure 3), ^{13}C -NMR (Table 1), DEPT, HMQC and HMBC (Figure 3) experiments.

Figure 3. Key NOESY contacts (a) and HMBC connectivities (b) of compound **2**.



The three known compounds, hinokiol (**3**) [13], sugiol (**4**) [14], and 12-hydroxy-6,7-secoabieta-8,11,13-triene-6,7-dial (**5**) [15], were readily identified by comparison of physical and spectroscopic data (UV, IR, ^1H -NMR, $[\alpha]_D$, and mass spectrometry data) with values found in the literature.

RAW264.7 is a mouse macrophage cell line used to model macrophage-mediated inflammatory events *in vitro*; thus, RAW264.7 cells were used to assess the effect of the isolated compounds on NO synthesis. The effect of compounds **1**, **3**, **4**, and **5** on RAW264.7 cell viability was determined by a MTT assay. Cells cultured with compounds **1**, **3**, and **4** at the concentrations (0, 2.5, 5, and 10 $\mu\text{g/mL}$) or compound **5** (0, 2.5, 5, 10, and 20 $\mu\text{g/mL}$) used in the presence of 100 ng/mL LPS for 24 h did not change cell viability (Table 2).

Compounds **1**, **3**, **4**, and **5** did not interfere with the reaction between nitrite and Griess reagents at 10 or 20 $\mu\text{g/mL}$ (data not shown). Unstimulated macrophages, after 24 h of incubation in culture medium produced background levels of nitrite. When RAW264.7 macrophages were treated with different concentrations of compounds **1**, **3**, and **5** together with LPS (100 ng/mL) for 24 h, a significant concentration-dependent inhibition of nitrite production was detected. The IC_{50} values for inhibition of nitrite production of compounds **1**, **3**, and **5** were about 9.8 ± 0.7 , 7.9 ± 0.9 , and $9.3 \pm 1.3\ \mu\text{g/mL}$. There was either a significant decrease in the nitrite production of group treated with **1**, **3**, and **5** (10 $\mu\text{g/mL}$) when compared with the LPS-alone group ($p < 0.01$).

Table 2. Cell viability and effect of compounds 1, 3–5 on LPS-induced NO production in macrophages ^a.

Compound	Dose (µg/mL)	Cell viability (% of control)	NO level	NO inhibition (% of control)	IC ₅₀ (µg/mL)
Control	(–)	93.0 ± 4.9	–0.5 ± 0.1	(–)	
LPS	(+)	98.7 ± 8.0	45.4 ± 2.7 ^{###}	(–)	
1	2.5	103.7 ± 2.0	39.0 ± 1.2	14.2 ± 2.7	
	5	101.1 ± 3.1	34.6 ± 1.1 *	23.9 ± 2.5	9.8 ± 0.7
	10	98.9 ± 6.7	22.2 ± 1.1 **	51.0 ± 2.4	
	20	34.6 ± 9.6	(–)	(–)	
	2.5	96.2 ± 1.5	39.3 ± 3.7	13.5 ± 8.2	
3	5	87.1 ± 1.5	32.5 ± 0.6 *	28.5 ± 1.4	7.9 ± 0.9
	10	82.6 ± 1.5	16.0 ± 0.7 **	64.8 ± 1.5	
	20	75.8 ± 2.5	(–)	(–)	
4	2.5	90.4 ± 1.7	44.9 ± 1.9	1.2 ± 4.2	
	5	83.1 ± 1.8	40.5 ± 1.3	10.8 ± 2.9	>20
	10	81.8 ± 1.0	28.7 ± 2.1 *	36.9 ± 4.7	
	20	53.8 ± 1.9	(–)	(–)	
5	2.5	98.0 ± 1.2	38.0 ± 1.1	16.2 ± 2.4	
	5	100.4 ± 4.5	33.0 ± 1.3	27.4 ± 2.8	9.3 ± 1.3
	10	91.1 ± 2.0	21.1 ± 2.3 **	53.5 ± 5.1	
	20	82.2 ± 1.2	11.1 ± 4.4 ***	75.6 ± 9.7	

^a The data were presented as mean ± S.D. for three different experiments performed in triplicate; ^{###} compared with sample of control group; * $p < 0.05$, ** $p < 0.01$, and *** $p < 0.001$ were compared with LPS-alone group.

3. Experimental

3.1. General

Silica gel 60 F_{254} precoated plates (*Merck*). Column chromatography (CC): Silica gel (*Merk* 70–230 mesh). High performance liquid chromatography (HPLC): *LDC Analytical-III* system; column: *LiChrosorb Si 60*, 7 µm, 250 × 10 mm. M.p.: *Yanaco-MP-S3* micro-melting-point apparatus; uncorrected. Optical rotation: *Jasco-DIP-180* polarimeter; in CHCl_3 . UV Spectra: *Hitachi-S-3200* spectrophotometer; λ_{max} (log ϵ) in nm; in MeOH. IR Spectra: *Perkin-Elmer-983G* FT-IR spectrophotometer; ν in cm^{-1} . ¹H-, ¹³C-, and 2D-NMR Spectra: *Varian-Unity-Plus-400* spectrometers; δ in ppm rel. to SiMe_4 , J in Hz. EI-MS and HR-EI-MS: *Jeol-JMS-HX300* mass spectrometer; m/z (rel. %).

3.2. Plant Material

The wood of *C. konishii* was collected at Luantashan, Nantau County, Taiwan, in December 1996 and was identified by Prof. Shao-Shun Ying (Department of Forestry, National Taiwan University). A voucher specimen (013492) has been deposited at the Herbarium of the Department of Botany, National Taiwan University, Taipei, Taiwan.

3.3. Extraction and Isolation

Dried wood (6.5 kg) of *C. konishii* was crushed into pieces and extracted with MeOH (60 L) three times (7 days each time) at room temperature. After removal of the solvent under vacuum, the extract (60.2 g) was suspended in water (500 mL) and successively partitioned into *n*-hexane (500 mL × 3), EtOAc (500 mL × 4), and *n*-BuOH (500 mL × 3). The EtOAc fraction (15.6 g) was chromatographed by silica gel column chromatography (using hexane–EtOAc and EtOAc–MeOH mixtures as solvent systems). Elution with *n*-hexane–EtOAc (7:3), (2:3), and (3:7) gave crude **4**, **3**, and **2**, respectively, while crude compounds **1** and **5** were eluted with *n*-hexane–EtOAc (9:1). Further purification by HPLC gave **1** (2.5 mg), **2** (2.5 mg), **3** (14.0 mg), **4** (7.3 mg), and **5** (5.0 mg) using *n*-hexane–CH₂Cl₂–EtOAc (10:5:1), *n*-hexane–EtOAc–*i*-PrOH (15:5:1), CH₂Cl₂–acetone–*i*-PrOH (30:5:1), *n*-hexane–CH₂Cl₂–EtOAc–*i*-PrOH (50:10:5:1), and *n*-hexane–CH₂Cl₂–EtOAc (10:5:1), respectively.

3.4. Konishone (**1**)

Yellowish oil; $[\alpha]_D^{27} +27.9^\circ$ (*c* 0.23, CHCl₃); UV (MeOH) λ_{\max} (log ϵ): 237 (4.06), 260 (4.13) nm; IR (neat) ν_{\max} : 3078 (C–H, vinyl), 1653 (C=O, conjugated), 1626, 1602 (C=C), 1374, 1268 cm⁻¹; EIMS: *m/z* (rel. int.): 272 (M⁺, 23), 271 (53), 252 (19), 243 (30), 217 (52), 230 (100), 189 (42), 179 (48); HREIMS: 272.1776 (C₁₈H₂₄O₂⁺, calc. 272.1772); ¹H- and ¹³C-NMR: see Table 1.

3.5. 3β-Hydroxy-5,6-dehydrosugiol (**2**)

Yellowish amorphous solid; $[\alpha]_D^{26} +26.3^\circ$ (*c* 0.23, CHCl₃); UV (MeOH) λ_{\max} (log ϵ): 219 (4.20), 245 (4.30), 317 (4.11) nm; IR (KBr) ν_{\max} : 3363 (OH), 1639 (C=O, conjugated), 1613, 1587, 1567, 1502, 1461, 1328, 1036 cm⁻¹; EIMS: *m/z* (rel. int.): 314 (M⁺, 100), 299 (19), 281 (32), 271 (9), 258 (54), 257 (75); HREIMS: 314.1879 (C₂₀H₂₆O₃⁺, calc. 314.1883); ¹H- and ¹³C-NMR: see Table 1.

3.6. Chemicals

LPS (endotoxin from *Escherichia coli*, serotype 0127:B8), Carr (type IV), indomethacin, MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) and other chemicals were purchased from Sigma Chemical Co. (St. Louis, MO, USA).

3.7. Cell Culture

A murine macrophage cell line RAW264.7 (BCRC No. 60001) was purchased from the Bioresources Collection and Research Center (BCRC, Hsinchu, Taiwan) of the Food Industry Research and Development Institute (Hsinchu, Taiwan). Cells were cultured in plastic dishes containing Dulbecco's Modified Eagle Medium (DMEM, Sigma, St. Louis, MO, USA) supplemented with 10% fetal bovine serum (FBS, Sigma) in a CO₂ incubator (5% CO₂ in air) at 37 °C and subcultured every 3 days at a dilution of 1:5 using 0.05% trypsin-0.02% EDTA in Ca²⁺-, Mg²⁺-free phosphate-buffered saline (DPBS).

3.8. Cell Viability

Cells (2×10^5) were cultured in 96-well plate containing DMEM supplemented with 10% FBS for 1 day to become nearly confluent. Then cells were cultured with compounds **1–5** in the presence of 100 ng/mL LPS (lipopolysaccharide) for 24 h. After that, the cells were washed twice with DPBS and incubated with 100 μ L of 0.5 mg/mL MTT for 2 h at 37 °C testing for cell viability. The medium was then discarded and 100 μ L dimethyl sulfoxide (DMSO) was added. After 30-min incubation, absorbance at 570 nm was read using a microplate reader (Molecular Devices, Sunnyvale, CA, USA).

3.9. Measurement of Nitric Oxide/Nitrite

NO production was indirectly assessed by measuring the nitrite levels in the cultured media and serum determined by a colorimetric method based on the Griess reaction. The cells were incubated with different concentration of samples in the presence of LPS (100 ng/mL) at 37 °C for 24 h. Then, cells were dispensed into 96-well plates, and 100 μ L of each supernatant was mixed with the same volume of Griess reagent (1% sulfanilamide, 0.1% naphthylethylenediamine dihydrochloride and 5% phosphoric acid) and incubated at room temperature for 10 min, the absorbance was measured at 540 nm with a Micro-Reader (Molecular Devices).

3.10. Statistical Analysis

IC₅₀ values were estimated using a non-linear regression algorithm (Sigma Plot 8.0; SPSS Inc. Chicago, IL, USA). Statistical evaluation was carried out by one-way analysis of variance (ANOVA followed by Scheffe's multiple range tests).

4. Conclusions

Konishone (**1**) and 3 β -hydroxy-5,6-dehydrosugirol (**2**) are new compounds from the wood of *C. konishii*. Compound **1** has a novel 7,20-dinorabietane skeleton. Compounds **1**, **3**, and **5** displayed a significant concentration-dependent inhibition of NO production. The compounds **1**, **3**, and **5** may lead to the development of novel non-steroidal anti-inflammatory drugs.

Acknowledgments

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Sample Availability: Samples of the compounds **1–5** are available from the authors.

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