Counter Anion Effect on the Photophysical Properties of Emissive Indolizine-Cyanine Dyes in Solution and Solid State

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Atom Number	Atom	Charge	2 O
1	Ν	-0.052098	1 <mark>1 1</mark>
2	Ο	-0.315967	
3	0	-0.315967	
4	0	-0.315967	5 4

Table S1. Mulliken charges on NO3⁻ atoms.

Table S2. Mulliken charges on ClO₄⁻ atoms.

Atom	Atom	Charge	20
Number			
1	Cl	1.351688	
2	Ο	-0.587922	
3	Ο	-0.587922	5 7
4	Ο	-0.587922	5 0 ⁻
5	Ο	-0.587922	

Table S3. Mulliken charges on PF_6^- atoms.

Atom Number	Atom	Charge	2 _F
1	Р	3.403124	7 F 1 > F3
2	F	-0.733854	P<
3	F	-0.733854	6F \ \ F4
4	F	-0.733854	БË
5	F	-0.733854	5.
6	F	-0.733854	
7	F	-0.733854	

Atom	Atom	Charge
Number		_
1	S	1.176786
2	С	0.742771
3	F	-0.309487
4	F	-0.294206
5	F	-0.266768
6	Ν	-0.878432
7	S	1.117281
8	С	0.828531
9	F	-0.315792
10	F	-0.282991
11	F	-0.274752
12	Ο	-0.556669
13	Ο	-0.568497
14	Ο	-0.551996
15	0	-0.565780



 Table S4. Mulliken charges on TFSI atoms.

Atom	Atom	Charge
Number	Atom	Charge
1	P	0 158270
1	Б	-0.138270
2	C	0.481455
3	C C	-0.481433
4	C	-0.416219
5	C	-0.330027
0	C	-0.416219
7	U	-0.461433
0	п	0.178000
9	П	0.143291
10	п	0.145492
11	П	0.143291
12	Г	1 129740
13	C	1.128749
14	C	-0.461433
15	C	-0.416219
10	C	-0.330027
17	C	-0.416219
10	с u	-0.461455
19	П	0.178000
20	п	0.143291
21	П	0.145492
22	п	0.143291
23	Г	1 129740
24	C	0.491455
25	C	-0.461433
20	C	-0.416219
27	C C	-0.330027
20	C	-0.416219
29	U U	-0.461455
30	п	0.178000
31	П Ц	0.143291
32	П Ц	0.143492
33	П Ц	0.143291
25	II C	1 128740
35	C C	0.481455
30	C	-0.481433
37	C C	-0.416219
30	C	-0.550027
39	C	-0.418219
40	U U	-0.481433
41	п	0.1/8000
42	П Ц	0.145291
43	П	0.145492
44	н	0.145291
45	Н	0.1/8060





Atom	Atom	Charge
Number		
1	В	0.246940
2	С	-0.467115
3	С	0.030678
4	С	-1.081474
5	С	-0.353031
б	С	-0.525904
7	С	-0.424496
8	Н	0.215602
9	С	1.885359
10	F	-0.322164
11	F	-0.302350
12	F	-0.273043
13	Н	0.220926
14	С	1.771129
15	F	-0.296042
16	F	-0.274135
17	F	-0.332978
18	Н	0.220021
19	С	-0.337231
20	С	0.010444
21	С	-1.091433
22	С	-0.315577
23	С	-0.594870
24	С	-0.373494
25	Н	0.215116
26	С	1.752906
27	F	-0.304481
28	F	-0.268874
29	F	-0.327559
30	Н	0.220899
31	С	1.838879
32	F	-0.272759
33	F	-0.301823
34	F	-0.329219
35	Н	0.221551
36	С	-0.409728
37	С	-0.022202
38	С	-1.101163
39	С	-0.314471
40	С	-0.573120

 Table S6. Mulliken charges on BARF atoms.

Atom	Atom	Charge
Number		0
41	С	-0.341871
42	Н	0.215555
43	С	1.737881
44	F	-0.269514
45	F	-0.304927
46	F	-0.328041
47	Н	0.221053
48	С	1.828482
49	F	-0.271783
50	F	-0.300031
51	F	-0.329847
52	Н	0.220930
53	С	-0.391629
54	С	-0.316589
55	С	-0.740104
56	С	-0.283551
57	С	-1.233593
58	С	0.015076
59	Н	0.219623
60	С	1.870671
61	F	-0.332687
62	F	-0.271987
63	F	-0.297009
64	Н	0.220620
65	С	1.887648
66	F	-0.265933
67	F	-0.303000
68	F	-0.330092
69	Н	0.214931
F ₃ C		, ^{CF} 3
F ₃ C		¹⁴ CF ₃
	7	6/
		—
	,,B <u>1</u> {	> 5
<u>~</u>		
130		9 ⁰⁻³
/	\searrow \checkmark	
F₃C	\sim	CF ₃











Figure S5. ¹H NMR (DMSO) of C5-TFSI





Figure S7. ¹H NMR (DMSO) of C5-TPB

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Figure S8. ¹H NMR (DMSO) of C5-BARF

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Σ Đ 0 F3C F₃C F₃C - 50 CF3 CF3 -61.6012 CF3 CF3 0 - 100 - 150 - 200

Figure S9. ¹⁹F NMR (DMSO) of C5-BARF



Figure S10. Full spectrum (300-1000 nm) of molar absorptivity and emission plot in MeCN.



Figure S11. Full spectrum (300-1000 nm) of molar absorptivity and emission plot in DCM.



Figure S12. Concentration versus absorbance plot for C5-Cl in DCM.



Figure S13. Concentration versus absorbance plot for C5-NO3 in DCM.



Figure S14. Concentration versus absorbance plot for C5-PF₆ in DCM.



Figure S15. Concentration versus absorbance plot for C5-TFSI in DCM.



Figure S16. Concentration versus absorbance plot for C5-TPB in DCM.



Figure S17. Concentration versus absorbance plot for C5-BARF in DCM.



Figure S18. Concentration versus absorbance plot for C5-Cl in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S19. Concentration versus absorbance plot for C5-NO₃ in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S20. Concentration versus absorbance plot for C5-PF₆ in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S21. Concentration versus absorbance plot for **C5-TFSI** in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S22. Concentration versus absorbance plot for **C5-TPB** in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S23. Concentration versus absorbance plot for C5-BARF in MeCN. A positive deviation from the Beer-Lambert Law is observed.



Figure S24. Absorption spectrum of varying concentrations of **C5-Cl** in MeCN showing a change in features as concentration changes.



Figure S25. Raw film absorption spectrum not normalized. The blank glass absorption spectrum with no dye is shown for comparison.



Figure S26. Film absorption with C5-Cl on glass prepared with varying concentrations of the dye.



Figure S27. Film absorption with C5-TPB on glass prepared with varying concentrations of the dye.



Figure S28. Film absorption with C5-TPB and C5-Cl on fluorine doped tin oxide (FTO).