## Supplementary Information for

Synthesis and X-ray structural characterization of amidine, amide, urea and isocyanate derivatives of the closo-aminododecaborate $\operatorname{anion}\left[\mathrm{B}_{12} \mathrm{H}_{11}\left(\mathbf{N H}_{3}\right)\right]^{-}$

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## I General Information

## Chemicals

If not otherwise specified, reagents and organic solvents were commercially available and used without further purification. Anhydrous solvents were prepared by passage through activated $\mathrm{Al}_{2} \mathrm{O}_{3}$ and stored over $3 \AA$ molecular sieves. $\mathrm{CD}_{3} \mathrm{CN}$ and $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ were purchased from Cambridge Isotope Laboratories and filtered through $\mathrm{Al}_{2} \mathrm{O}_{3}$ prior to use. $\left[\mathrm{B}_{12} \mathrm{H}_{12}\right]^{2-}$ and $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}_{3}\right]^{-}$salts and dodecaborate amides $\mathbf{3 a}-\mathbf{e}$ were prepared according to the literature.[1-3]

## Reaction Conditions

Glassware for air-sensitive reations was dried at $150{ }^{\circ} \mathrm{C}$ and allowed to cool in a vacuum. Reactions carried out in a glovebox were run under a nitrogen atmosphere with $\mathrm{O}_{2}, \mathrm{H}_{2} \mathrm{O}<1 \mathrm{ppm}$.

## Characterization

Thin-layer chromatography (TLC) was carried out using silica gel 60, F254 with a thickness of 0.25 mm . Column chromatography was performed on silica gel 60 (200-30 mesh).

Low-resolution ESI-MS data were recorded on Advion Expression CMS instrument. High-resolution MS data were recorded using IT-TOF detection (Shimadzu, Japan) equipped with an electrospray ionization source (ESI). Accurate mass determination was corrected by calibration using sodium trifluoroacetate clusters as a reference.

Single-crystal X-ray diffraction studies were performed on an Oxford Diffraction Gemini A Ultra diffractometer equipped with an 135mm Atlas CCD detector and using Mo K- $\alpha$ radiation

NMR spectra were recorded on a Bruker AVANCE III 500 spectrometer ( ${ }^{1} \mathrm{H}$ NMR 500.13 MHz, ${ }^{13} \mathrm{C}$ NMR $125.77 \mathrm{MHz},{ }^{11}{ }^{1}$ NMR 160.46 MHz ) or a Bruker AVANCE III 400 spectrometer ( ${ }^{1} \mathrm{H}$ NMR $400.13 \mathrm{MHz},{ }^{13} \mathrm{C}$ NMR $100.62 \mathrm{MHz},{ }^{11} \mathrm{~B}$ NMR $128.38 \mathrm{MHz})$ at the temperature indicated. Data are reported as follows: Chemical shift in ppm, multiplicity $(\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{m}=$ multiplet, $\mathrm{dd}=$ doublet of doublets, etc.), coupling constant $J$ in Hz , integration, and (where applicable) interpretation. Signals were referenced against solvent peaks $\left({ }^{1} \mathrm{H}\right.$ : residual $\mathrm{CHD}_{2} \mathrm{C}(\mathrm{O}) \mathrm{CD}_{3}=2.05 \mathrm{ppm}$, residual $\mathrm{CHD}_{2} \mathrm{CN}=1.94 \mathrm{ppm}$, residual $\mathrm{CHDCl}_{2}$ $=5.32 \mathrm{ppm},{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}: C \mathrm{D}_{3} \mathrm{C}(\mathrm{O}) C \mathrm{D}_{3}=29.84 \mathrm{ppm}, C \mathrm{D}_{3} \mathrm{CN}=1.32 \mathrm{ppm}, \mathrm{CD}_{2} \mathrm{Cl}_{2}=$ $53.32 \mathrm{ppm}) .{ }^{11} \mathrm{~B}$ and ${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectra were calibrated against external $\mathrm{BF}_{3} * \mathrm{Et}_{2} \mathrm{O}$ $=0 \mathrm{ppm}\left(\mathrm{BF}_{3} * \mathrm{Et}_{2} \mathrm{O}\right.$ capillary in $\left.\mathrm{C}_{6} \mathrm{D}_{6}\right)$.

## II Experimental Section



Synthesis of $\left[\mathbf{E t}_{3} \mathbf{N H}\right][\mathbf{3 e - H}]$ : In a glovebox filled with $\mathrm{N}_{2}$, a 20 mL vial was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}_{3}\right](212.4 \mathrm{mg}, 0.817 \mathrm{mmol}, 1$ equiv), $\mathrm{NaH}(138.2 \mathrm{mg}, 5.758$ mmol, 7 equiv) and a stir bar. THF ( 4 mL ) and DMF ( 4 mL ) were added, and the mixture was stirred at room temperature for 10 minutes until there was no $\mathrm{H}_{2}$ evolution anymore. Then pyridine-2-carbonyl chloride hydrochloride $\mathrm{PyCOCl} \cdot \mathrm{HCl}$ ( $220.2 \mathrm{mg}, 1.237 \mathrm{mmol}, 1.5$ equiv) was slowly added. The conversion was complete after stirring for 5 h . The flask was transferred out of the glovebox. $\mathrm{H}_{2} \mathrm{O}(4 \mathrm{~mL})$ was added, and the pH value of the reaction mixture was adjusted to $2-3$ with 1 M aqueous $\mathrm{HCl} .\left[\mathrm{NEt}_{3} \mathrm{H}\right] \mathrm{Cl}(300 \mathrm{mg}, 2.180 \mathrm{mmol}, 2.7$ equiv) was added, and the reaction mixture was extracted with $\mathrm{MeCN} / \operatorname{EtOAc}(1: 2 v / v)$. The organic layers were concentrated on a rotary evaporator. The residue was purified by recrystallization from methanol to afford yellowish crystals of $\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{3 e - H}](150 \mathrm{mg}, 50 \%)$.
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=8.96(\mathrm{~s}, 1 \mathrm{H}$, anionic $\mathrm{N} H), 8.90-8.86(\mathrm{~m}, 1 \mathrm{H}$, Py H), 8.18-8.14 (overlapping m, 2H, Py H), 7.89-7.72 (m, 1H, Py H), 6.63 (t, $1 \mathrm{H}, J_{\mathrm{NH}}$ $=52 \mathrm{~Hz}, \mathrm{~N} H), 3.27(\mathrm{~s}, 1 \mathrm{H}, \mathrm{N} H), \quad 3.20-3.15\left(\mathrm{~m}, 6 \mathrm{H}\right.$, cationic $\left.\mathrm{N}-\mathrm{CH}_{2}\right), 1.47$ (broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}$ ), 1.24 (t, $J=7.4 \mathrm{~Hz}, 9 \mathrm{H}$, cationic $\mathrm{CH}_{3}$ ), 1.20 (broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}$ ), 1.13 (broad signal, 1H, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (101 MHz, $\mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=166.7,149.5,143.9,141.5,129.8$, 124.5 (6 anionic signals), 48.0, 9.2 ( 2 cationic signals).
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-7.6(1 \mathrm{~B}, B-\mathrm{N}),-15.3(5 \mathrm{~B}, B-\mathrm{H}),-15.7$ (overlapping signals, $6 \mathrm{~B}, B-\mathrm{H}$ ).

High-resolution ESI-MS (negative mode, MeOH ): $m / z$ calcd for $\left[\mathrm{C}_{6} \mathrm{H}_{17} \mathrm{~B}_{12} \mathrm{~N}_{2} \mathrm{O}\right]^{-}$ 263.2430. Found: 263.2459.


Transformation of $\left[\mathbf{E t}_{3} \mathbf{N H}\right][3 \mathrm{e}-\mathrm{H}]$ to $\left[\mathbf{E t}_{3} \mathbf{N H}\right]_{2}[\mathbf{3 e}]:$ A 20 mL vial was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right][3 \mathrm{e}-\mathrm{H}](50 \mathrm{mg})$ and a stir bar. $\mathrm{MeCN}(3 \mathrm{~mL})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.5 \mathrm{~mL})$ were added, and the solution was stirred at room temperature for 1 h . Then the stir bar was removed, and the solution was concentrated on a rotary evaporator and dried overnight under vacuum at $80{ }^{\circ} \mathrm{C}$ to afford compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 e}]$ in quantitative yield.

This method can also be applied for the transformation of other compounds 3-H to $\mathbf{3}$ quantitatively. ${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectra of $\mathbf{3 b}, \mathbf{3 b} \mathbf{-} \mathbf{H}, \mathbf{3} \mathbf{e}$ and $\mathbf{3 e} \mathbf{- H}$ are displayed in Figure S1.
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=8.56($ broad signal, 1 H, Py H), 8.09-8.00 (m, 1 H, Py H), 7.99-7.80 (overlapping m, 2H, Py H and amide N-H), 7.50-7.38 (m, 1 H , Py H), 4.63 (broad t, $2 \mathrm{H}, J_{\mathrm{NH}}=52 \mathrm{~Hz}, \mathrm{~N}-\mathrm{H}$ from cation), 3.25-3.01 (m, 12 H , cationic $\mathrm{N}-\mathrm{CH}_{2}$ ), $1.34(\mathrm{~s}, 5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 1.24\left(\mathrm{t}, J=7.4 \mathrm{~Hz}, 9 \mathrm{H}\right.$, cationic $\mathrm{CH}_{3}$ ), 1.03 (broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 0.89$ (broad signal, 1H, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $101 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=166.2,152.9,149.0,138.5,126.4$, 122.2 (6 anionic signals), 47.8, 9.1 (2 cationic signals).
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-5.3$ (1B, B-N), -15.3 (5B, B-H), $-16.4(5 \mathrm{~B}$, B-H), -18.7 (1B, B-H).
High-resolution ESI-MS (negative mode, MeOH ): $m / z$ calcd for $\left[\mathrm{C}_{6} \mathrm{H}_{17} \mathrm{~B}_{12} \mathrm{~N}_{2} \mathrm{O}\right]^{2-}$ 131.1226. Found: 131.1254.

The ${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectra of $\mathbf{3 b}, \mathbf{3 b}-\mathbf{H}, \mathbf{3 e}$ and $\mathbf{3 e} \mathbf{- H}$ are shown in Figure S 1 as representative examples to demonstrate the effect of protonation. For both product pairs $\mathbf{3 b} / \mathbf{3 b}-\mathbf{H}$ and $\mathbf{3 e} / \mathbf{3 e}-\mathbf{H}$, similar effects are observed. Upon protonation, the $\mathrm{B}-\mathrm{N}$ signal is shifted from -5 ppm to -8 ppm . On the other hand, the $\mathrm{B}-\mathrm{H}$ vertices become more deshielded; the B12 signal appears at -19 ppm in the dianionic form and overlaps with the B2-11 resonances in the monoanionic form.


Figure S1. ${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectra of $\mathbf{3 b}, \mathbf{3 b} \mathbf{- H}, \mathbf{3 e}$ and $\mathbf{3 e - H}$ (acetonitrile- $d_{3}, 128 \mathrm{MHz}$, $23^{\circ} \mathrm{C}$ ).


Synthesis of amidine [ $\mathbf{E t}_{\mathbf{3}} \mathbf{N H}$ ][6a]: In a glovebox, a dry 20 mL vial, equipped with a stir bar, was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}_{3}\right](102 \mathrm{mg}, 0.40 \mathrm{mmol}$, 1 equiv). Then anhydrous DMF ( 1 mL ) was added. The vial was transferred to a fumehood, and dry $\mathrm{Et}_{3} \mathrm{~N}\left(1.0 \mathrm{~mL}, 7.20 \mathrm{mmol}, 18\right.$ equiv) was added to the solution under $\mathrm{N}_{2}$ protection. Then 2,4,6-trimethylphenylcarboxylic acid chloride ( $110 \mathrm{mg}, 0.60 \mathrm{mmol}, 1.5$ equiv) was added. The mixture was stirred at $25^{\circ} \mathrm{C}$ for 4 h . The reaction was quenched with an aqueous $\left[\mathrm{Et}_{3} \mathrm{NH}\right] \mathrm{Cl}$ solution $\left(2 \mathrm{~mL} \mathrm{H}_{2} \mathrm{O}+2\right.$ equiv $\left.\left[\mathrm{Et}_{3} \mathrm{NH}\right] \mathrm{Cl}\right)$; the pH value at this point was ca. $7-8$. The mixture was extracted with $\mathrm{DCM} / \mathrm{MeCN}=4: 1(8 \times 10 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{MgSO}_{4}$, and the solution was filtered and concentrated by rotary evaporation. The cloudy residue was purified by silica gel column chromatography (eluent $\mathrm{DCM} / \mathrm{MeCN}=10: 3$, fraction size 20 mL ). The combined eluates were concentrated on a rotary evaporator and dried under vacuum at $60^{\circ} \mathrm{C}$ overnight to afford compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][6 a]$ as a colorless solid $(50.4 \mathrm{mg}, 40 \%)$. ${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $\left.400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta 7.76(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{N}=\mathrm{CH}-\mathrm{N})$, 6.41 (broad signal, $1 \mathrm{H}, \mathrm{N}-\mathrm{H}$ ), 3.13 ( $\mathrm{q}, ~ J=7.2 \mathrm{~Hz}, 6 \mathrm{H}$, cationic N-CH2), $3.08(\mathrm{~s}, 3 \mathrm{H}$, anionic $\left.\mathrm{N}-\mathrm{CH}_{3}\right), 2.83\left(\mathrm{~s}, 3 \mathrm{H}\right.$, anionic $\left.\mathrm{N}-\mathrm{CH}_{3}\right), 1.26($ broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 1.24(\mathrm{t}, J$ $=7.2 \mathrm{~Hz}, 9 \mathrm{H}$, cationic $\left.\mathrm{N}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.03($ broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 0.85($ broad signal, $1 \mathrm{H}, \mathrm{B}-\mathrm{H})$.
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta 157.3(\mathrm{~N}=\mathrm{C}-\mathrm{N}), 48.0$ (cationic $\mathrm{CH}_{2}$ ), 43.1, 35.7 (two N-C signals), 9.2 (cationic $\mathrm{CH}_{3}$ ).
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR $\left(160 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta-4.2(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-14.5$ to $-17.0(10 \mathrm{~B}$, B-H), -19.0 (1B, B-H).

High-resolution ESI-MS (negative mode, MeOH ): $\mathrm{m} / \mathrm{z}$ calcd for $\left[\mathrm{C}_{3} \mathrm{H}_{19} \mathrm{~B}_{12} \mathrm{~N}_{2}\right]^{-}$:
213.2738. Found: 213.2762 .


Synthesis of amidine $\left[\mathbf{E t}_{\mathbf{3}} \mathbf{N H}\right][\mathbf{6 b}]$ : A dry 20 mL vial, equipped with a stir bar, was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCOC}_{6} \mathrm{H}_{5}\right](101 \mathrm{mg}, 0.22 \mathrm{mmol}$, 1 equiv). Then anhydrous $\mathrm{MeCN}(3 \mathrm{~mL})$ was added, and dry $\mathrm{Et}_{3} \mathrm{~N}(0.3 \mathrm{~mL}, 2.16 \mathrm{mmol}, 9.8$ equiv $)$ was added to the solution under $\mathrm{N}_{2}$ protection. Pentafluorophenylcarboxylic acid chloride ( $80.0 \mathrm{mg}, 0.35 \mathrm{mmol}, 1.5$ equiv) was added at $25^{\circ} \mathrm{C}$. The temperature was raised to $50^{\circ} \mathrm{C}$. After 30 min , aniline ( $61 \mathrm{mg}, 0.66 \mathrm{mmol}, 3.0$ equiv) was added. The mixture was stirred for another 4 h and concentrated by rotary evaporation. The cloudy residue was purified by silica gel column chromatography (eluent $\mathrm{DCM} / \mathrm{MeCN}=4: 1$, fraction size 20 mL ). The combined eluates were concentrated on a rotary evaporator and dried under vacuum at $60^{\circ} \mathrm{C}$ overnight to afford compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{6 b}]$ as a yellow solid $(87.7 \mathrm{mg}, 91 \%)$.
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}, 23{ }^{\circ} \mathrm{C}\right): \delta 10.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{N}-\mathrm{H}), 7.53-7.48(\mathrm{~m}, 1 \mathrm{H}$, phenyl H), 7.41-7.35 (overlapping m, 4H, phenyl H), 7.24-7.09 (overlapping m, 3H, phenyl H), 7.03-6.78 (overlapping broad signal and m, 3H, phenyl H and $\mathrm{N}-\mathrm{H}$ ), 6.65 (broad signal, $1 \mathrm{H}, \mathrm{N}-\mathrm{H}) 3.29-3.22\left(\mathrm{~m}, 6 \mathrm{H}\right.$, cationic $\left.\mathrm{N}-\mathrm{CH}_{2}\right), 1.62$ (broad signal, 5 H , B-H), $1.40\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 9 \mathrm{H}\right.$, cationic $\left.\mathrm{N}-\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.22($ broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 1.05$ (broad signal, 1H, B-H).

This spectrum contained small signals at $7.18,6.71$ and 6.67 ppm ascribed to residual aniline
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100 MHz, $\left.\mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta 165.6(\mathrm{~N}=\mathrm{C}-\mathrm{N}), 138.1,133.2,131.3$, $130.4,130.1,129.9,127.5,125.6$ ( 8 aryl signals), 48.3 (cationic $\mathrm{N}-\mathrm{CH}_{2}$ ), 9.4 (cationic $\mathrm{N}-\mathrm{CH}_{3}$ ).

This spectrum showed small signals at $149.1,130.2,118.3$ and 115.6 ppm ascribed to residual aniline.
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (128 MHz, $\left.\mathrm{CD}_{2} \mathrm{Cl}_{2}, 23{ }^{\circ} \mathrm{C}\right): \delta-5.8(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-13.5$ to $-16.5(10 \mathrm{~B}$, B-H), -17.4 (1B, B-H).

High-resolution ESI-MS (negative mode, MeOH ): $m / z$ calcd for $\left[\mathrm{C}_{13} \mathrm{H}_{23} \mathrm{~B}_{12} \mathrm{~N}_{2}\right]^{-}$: 337.3056. Found: 337.2382.


Synthesis of amidine $\left[\mathbf{E t}_{\mathbf{3}} \mathbf{N H}\right][\mathbf{6 c}]$ : A dry 20 mL vial, equipped with a stir bar, was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCOC}_{6} \mathrm{H}_{3} \mathrm{Cl}_{2}\right](177 \mathrm{mg}, 0.33 \mathrm{mmol}, 1$ equiv). Then anhydrous $\mathrm{MeCN}(3 \mathrm{~mL})$ was added, and dry $\mathrm{Et}_{3} \mathrm{~N}(0.45 \mathrm{~mL}, 3.25 \mathrm{mmol}, 9.8$ equiv) was added to the solution under $\mathrm{N}_{2}$ protection. Pentafluorophenylcarboxylic acid chloride ( $128 \mathrm{mg}, 0.55 \mathrm{mmol}, 1.7$ equiv) was added at $25^{\circ} \mathrm{C}$. The temperature was raised to $50^{\circ} \mathrm{C}$. After $30 \mathrm{~min}, N, N$-dimethylethylamine ( $88 \mathrm{mg}, 1.00 \mathrm{mmol}, 3.0$ equiv) was added. The mixture was stirred for another 4 h , and 1 M aqueous $\mathrm{HCl}(5 \mathrm{~mL})$ was added. The suspension was extracted with EtOAc/MeCN 3:1 (5 x 10 mL ). The combined organic layers were dried over $\mathrm{MgSO}_{4}$, and the solution was filtered and concentrated by rotary evaporation. The cloudy residue was purified by silica gel column chromatography (eluent $\mathrm{DCM} / \mathrm{MeCN}=4: 3$, fraction size 20 mL ). The combined eluates were concentrated and dried under vacuum at $60{ }^{\circ} \mathrm{C}$ overnight to afford compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{6 c}]$ as a yellow solid $(132 \mathrm{mg}, 100 \%)$.
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta 8.54$ (broad signal, $\left.1 \mathrm{H}, \mathrm{N}-\mathrm{H}\right), 7.59-7.55$ (overlapping m, 3 H , aryl H), 7.46 (broad signal, $1 \mathrm{H}, \mathrm{N}-\mathrm{H}$ ), 6.98 (very broad signal, $1 \mathrm{H}, \mathrm{N}-\mathrm{H}), 3.43\left(\mathrm{dt}, J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.24\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.77$ $\left(\mathrm{s}, 6 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right), 1.41$ (broad signal, $\left.5 \mathrm{H}, \mathrm{B}-\mathrm{H}\right), 1.12$ (broad signal, $\left.5 \mathrm{H}, \mathrm{B}-\mathrm{H}\right), 1.06$ (broad signal, 1H, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta 161.9(\mathrm{~N}=\mathrm{C}-\mathrm{N}), 134.6,134.2,129.8$, 129.2 (4 aryl signals), 56.9, 44.8, 40.0.
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\} \quad \mathrm{NMR}\left(128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}, 23{ }^{\circ} \mathrm{C}\right): \delta-6.9(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-13.0$ to -18.0 (overlapping signals with peaks at -15.2 and $-16-1 \mathrm{ppm}, 11 \mathrm{~B}, \mathrm{~B}-\mathrm{H}$ ).

High-resolution ESI-MS (negative mode, MeOH ): $\mathrm{m} / \mathrm{z}$ calcd for $\left[\mathrm{C}_{11} \mathrm{H}_{27} \mathrm{~B}_{12} \mathrm{Cl}_{2} \mathrm{~N}_{3}-\mathrm{H}\right]^{-}$: 400.2699. Found: 400.2714.


Synthesis of urea $\left[\mathrm{NBu}_{4}\right]_{2}[7 \mathrm{a}]$ : In a glovebox filled with $\mathrm{N}_{2}$, a 20 mL vial was charged with $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}_{3}\right](260 \mathrm{mg}, 1.00 \mathrm{mmol}, 1$ equiv), $\mathrm{NaH}(53 \mathrm{mg}, 2.2$ mmol, 2.2 equiv) and a stir bar. THF ( 10 mL ) was added, and the mixture was stirred at room temperature for 10 minutes until there was no $\mathrm{H}_{2}$ evolution anymore. Phenyl isocyanate ( $238 \mathrm{mg}, 2.0 \mathrm{mmol}$, 2 equiv) was slowly added. The conversion was complete after stirring for 5 h . The flask was transferred out of the glovebox. The solvent was removed under vacuum, and $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ was added. The aqueous solution was heated to $50^{\circ} \mathrm{C}$, and $\left[\mathrm{NBu}_{4}\right] \mathrm{Br}(677 \mathrm{mg}, 2.1 \mathrm{mmol}, 2.1$ equiv) was added. A white solid precipitated immediately and was collected by filtration. It was dried under vacuum overnight to afford $\left[\mathrm{NBu}_{4}\right]_{2}[7 \mathrm{a}]$ as a colorless microcrystalline product ( $685 \mathrm{mg}, 90 \%$ ).
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=8.52$ (broad $\mathrm{s}, 1 \mathrm{H}$, anionic NH ), $7.41(\mathrm{~d}, 2 \mathrm{H}$, $J=8.2 \mathrm{~Hz}, \mathrm{Ph} \mathrm{H}), 7.18(\mathrm{dd}, 2 \mathrm{H}, J=8.2 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, \mathrm{Ph} H), 6.83(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, \mathrm{Ph}$ H), 3.96 (broad s, $1 \mathrm{H}, \mathrm{N} H$ ), $3.25-3.01$ (m, 16H, cationic $\mathrm{N}-\mathrm{CH}_{2}$ ), 1.67-1.50 (m, 16H, cationic $\mathrm{N}-\mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 1.41-1.27 (overlapping m and s, 21 H , cationic $\mathrm{N}-\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ and B-H), $1.04(\mathrm{~s}, 5 \mathrm{H}, \mathrm{B}-\mathrm{H}), 0.95\left(\mathrm{t}, 24 \mathrm{H}, J=7.3 \mathrm{~Hz}\right.$, cationic $\left.\mathrm{CH}_{3}\right), 0.85(\mathrm{~s}, 1 \mathrm{H}$, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $101 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=158.6,142.8,129.5$ (overlapping signals), 121.2, 59.2, 24.3, 20.3, 10.8.
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-5.0(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-15.4(5 \mathrm{~B}, \mathrm{~B}-\mathrm{H}),-16.2(5 \mathrm{~B}$, B-H), -19.3 (1B, B-H).

High-resolution ESI-MS (negative mode, MeOH ): $m / z$ calcd for $\left[\mathrm{C}_{7} \mathrm{H}_{18} \mathrm{~B}_{12} \mathrm{~N}_{2} \mathrm{O}\right]^{2}$ 138.1320. Found: 138.1331.


Synthesis of urea $\left[\mathbf{P P h}_{4}\right]_{2}[\mathbf{7 b}]$ : This product was prepared in a similar manner to $\left[\mathrm{NBu}_{4}\right]_{2}[7 \mathrm{a}]$, using 4-chlorophenyl isocyanate ( $307 \mathrm{mg}, 2.0 \mathrm{mmol}$, 2 equiv) and $\left[\mathrm{PPh}_{4}\right] \mathrm{Br}$ ( $881 \mathrm{mg}, 2.1 \mathrm{mmol}, 2.1$ equiv ). $\left[\mathrm{PPh}_{4}\right]_{2}[7 \mathbf{b}]$ was obtained as a colorless microcrystalline solid ( $869 \mathrm{mg}, 91 \%$ ).
${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=8.59(\mathrm{~s}, 1 \mathrm{H}$, anionic NH$)$, 7.95-7.85 (m, 8 H , cationic H), 7.81-5.58 (overlapping m, 32 H , cationic H), 7.41-7.28 (m, 2H, Ph H ), 7.13-6.96 (m, 2H, Ph H), 4.00 (s, 1H, N-H), 1.33 (broad signal, 5H, B-H), 1.07 (broad signal, $5 \mathrm{H}, \mathrm{B}-\mathrm{H}$ ), 0.88 (broad signal, 1H, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $101 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=158.4,141.7,136.4\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=2.4 \mathrm{~Hz}\right.$, cation $\mathrm{CH}), 135.6\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=10 \mathrm{~Hz}\right.$, cation CH$), 131.3\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=13.0 \mathrm{~Hz}\right.$, cation CH$), 129.2$, 124.9, 119.6, $118.8\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=89 \mathrm{~Hz}\right.$, cation $\left.\mathrm{C}_{\mathrm{q}}\right)$.
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-5.0(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-15.5(5 \mathrm{~B}, \mathrm{~B}-\mathrm{H}),-16.2(5 \mathrm{~B}$, B-H), -19.2 (1B, B-H).
High-resolution ESI-MS (negative mode, MeOH ): $m / z$ calcd for $\left[\mathrm{C}_{7} \mathrm{H}_{17} \mathrm{~B}_{12} \mathrm{~N}_{2} \mathrm{OCl}\right]^{2-}$ 155.1125. Found: 155.1133.


Synthesis of isocyanate $\left[\mathrm{MePPh}_{3}\right]_{2}[8]$ : In a glovebox filled with $\mathrm{N}_{2}$, a 50 mL round-bottom flask was charged with $\mathrm{Cs}\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}_{3}\right](594 \mathrm{mg}, 2.0 \mathrm{mmol}, 1$ equiv), $\mathrm{NaH}(144 \mathrm{mg}, 6.0 \mathrm{mmol}, 3$ equiv) and a stir bar. DMF ( 10 mL ) was added, and the mixture was stirred at $25^{\circ} \mathrm{C}$ for 10 minutes until there was no $\mathrm{H}_{2}$ evolution anymore. Then $\mathrm{ClC}(\mathrm{O}) \mathrm{NMe}_{2}$ ( 6 equiv) diluted in $\mathrm{DMF}(2 \mathrm{~mL}$ ) was slowly added by an Eppendorf pipet. The conversion was complete after stirring for 4 h . The flask was transferred out of the glovebox, and the volatiles were removed under vacuum. The residue was dissolved in $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ at $c a .90^{\circ} \mathrm{C}$, giving a slightly yellow solution. The solution was stirred at $80-100{ }^{\circ} \mathrm{C}$ for 1 h , and $\left[\mathrm{MePPh}_{3}\right] \operatorname{Br}(1.29 \mathrm{~g}, 5 \mathrm{mmol}, 2.5$ equiv) was added. A white precipitate formed, and it was was collected by filtration. Purification by column chromatography (eluent DCM/MeCN 4:3) afforded $\left[\mathrm{MePPh}_{3}\right]_{2}[8]$ as a colorless solid ( $369 \mathrm{mg}, 25 \%$ ). ${ }^{1} \mathrm{H}\left\{{ }^{11} \mathrm{~B}\right\}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=7.90-7.83(\mathrm{~m}, 6 \mathrm{H}$, cationic CH ), 7.76-7.62 (overlapping m, 24 H , cationic CH ), $2.83\left(\mathrm{~d}, J=13.8 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{CH} H_{3}\right.$ ), 1.23 (broad signal, 5H, B-H), 0.97 (broad signal, 5H, B-H), 0.75 (broad signal, 1H, B-H).
${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \operatorname{NMR}\left(101 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}\right): \delta=136.1\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=3.0 \mathrm{~Hz}\right.$, cation CH$)$, $134.2(\mathrm{~d}$, $J_{\mathrm{P}, \mathrm{C}}=11 \mathrm{~Hz}$, cation CH), $131.1\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=13 \mathrm{~Hz}\right.$, cation CH), $120.4\left(\mathrm{~d}, J_{\mathrm{P}, \mathrm{C}}=89 \mathrm{~Hz}\right.$, cation $\left.\mathrm{C}_{\mathrm{q}}\right)$, $9.37\left(\mathrm{~d},{ }^{1} J_{\mathrm{P}, \mathrm{C}}=58 \mathrm{~Hz}\right.$, cation $\left.\mathrm{CH}_{3}\right)$. The $\mathrm{N}=\mathrm{C}=\mathrm{O}$ carbon atom could not be detected unambiguously.
${ }^{11} \mathrm{~B}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $128 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-7.74(1 \mathrm{~B}, \mathrm{~B}-\mathrm{N}),-15.4(5 \mathrm{~B}, \mathrm{~B}-\mathrm{H}),-16.7(5 \mathrm{~B}$, B-H), -19.6 (1B, B-H).

Mass-spectrometric characterization of this product proved difficult; the results that were obtained by negative-mode ESI-MS are shown in Figure S2, along with the IR spectrum in Figure S3.


Figure S2. (-)-ESI Mass spectrum of $\mathbf{8}$ in MeOH .


Figure S3. IR spectrum of $\left[\mathrm{PPh}_{4}\right]_{2}[8]$.

## III X-ray Crystallography

CCDC1861483-1861492 contain the supplementary crystallographic data for this publication. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Crystals of the products $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 b}],\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{3 d} \mathbf{- H}]_{,}\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 e}]$,
$\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{3 e - H}],\left[\mathrm{MePPh}_{3}\right][6 \mathbf{6}],\left[\mathrm{Et}_{3} \mathrm{NH}\right][6 \mathbf{c}]$ and $\left[\mathrm{MePPh}_{3}\right]_{2}[\mathbf{8}]$ were measured at room temperature because the X-ray facility of our department does not routinely offer measurements with nitrogen cooling.

## Crystal structure of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[3 \mathrm{a}](\mathrm{CCDC1861488})$

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 a}](20 \mathrm{mg})$ was dissolved in acetone $/ \mathrm{MeCN}(0.25 \mathrm{~mL} / 0.25 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with hexanes ( 1 mL ). Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{4}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCOPh}\right]_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ suitable for X-ray diffraction grew within 3 d at 25 ${ }^{\circ} \mathrm{C}$.

Bond precision:

Cell: $\quad a=10.3802(8) \quad b=15.9133(12) \quad c=18.0577(15)$ alpha=79.497(7) beta=87.786(7) gamma=87.828(6)
Temperature: 170 K

Calculated Reported
2929.2(4) 2929.2(4)
$\begin{array}{lll}\text { Volume } & 2929.2(4) & 2929 \\ \text { Space group } & \mathrm{P}-1 & \mathrm{P}-1\end{array}$
Hall group $-\mathrm{P} 1 \quad-\mathrm{P} 1$
Moid $2(\mathrm{C} 7 \mathrm{H} 17 \mathrm{~B} 12 \mathrm{~N} \mathrm{O}), 4(\mathrm{C} 6 \quad 2(\mathrm{C} 7 \mathrm{H} 17 \mathrm{~B} 12 \mathrm{~N}$ O), $4(\mathrm{C}$
Sum formula C38 H100 B24 N6 O3
$\mathrm{Mr} \quad 948.68$
Dx,g cm-3 1.076
Z 2
Mu (mm-1) 0.060
F000 1028.0
F000' 1028.24

Tmin' 0.972
h,k,1max $\quad 12,19,21 \quad 19,21$
Nref 1075610623
Tmin,Tmax 0.972,0.977 0.849,1.000
H16 N), H2 O
C38 H100 B24 N6 O3
948.67
1.076

2
0.060
1028.0

Correction method= \# Reported $T$ Limits: Tmin=0.849 Tmax=1.000
AbsCorr $=$ MULTI-SCAN

Data completeness $=0.988 \quad$ Theta $(\max )=25.350$
$R($ reflections $)=0.1239(6692) \quad$ wR2 (reflections $)=0.3535(10623)$
$S=1.034 \quad$ Npar $=655$


Figure S4. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{4}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCOPh}\right]_{2} \cdot \mathrm{H}_{2} \mathrm{O} ; 30 \%$ displacement ellipsoids.

## Crystal structure of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[3 \mathrm{~b}](\mathrm{CCDC1861489})$

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 b}](20 \mathrm{mg})$ was dissolved in acetone $/ \mathrm{MeCN}(0.25 \mathrm{~mL} / 0.25 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with hexanes ( 1 mL ). Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{F}\right] \cdot 0.5 \mathrm{CH}_{3} \mathrm{C}(\mathrm{O}) \mathrm{CH}_{3}$ suitable for X-ray diffraction grew within 1 d at $25^{\circ} \mathrm{C}$.

| Bond precision: | $\mathrm{C}-\mathrm{C}=0.0060 \mathrm{~A}$ | Wavelength $=0.71073$ |
| :---: | :---: | :---: |
| Cell: | $\mathrm{a}=17.517(2) \quad \mathrm{b}=10.7703(8)$ | 8) $\quad \mathrm{C}=35.537(4)$ |
|  | alpha=90 beta=106.010 | 10(11) gamma=90 |
| Temperature: | 293 K |  |
|  | Calculated | Reported |
| Volume | 6444.5(12) | 6444.4(12) |
| Space group | C 2/c | C 1 2/c 1 |
| Hall group | -C 2yc | -C 2yc |
| Moiety formula | $\begin{aligned} & 2(\mathrm{C} 7 \mathrm{H} 16 \mathrm{~B} 12 \mathrm{~F} N \mathrm{~N}), 4(\mathrm{C} 6 \\ & \mathrm{H} 16 \mathrm{~N}), \mathrm{C} 3 \mathrm{H} 6 \mathrm{O} \end{aligned}$ | 2(C7 H16 B12 F N O), C3 H6 0, $4(\mathrm{C} 6 \mathrm{H} 16 \mathrm{~N})$ |
| Sum formula | C41 H102 B24 F2 N6 O3 | C41 H102 B24 F2 N6 O3 |
| Mr | 1024.73 | 1024.72 |
| Dx,g cm-3 | 1.056 | 1.056 |
| Z | 4 | 4 |
| Mu (mm-1) | 0.063 | 0.063 |
| F000 | 2208.0 | 2208.0 |
| F000, | 2208.65 |  |
| h, k, 1 max | 21,12,42 | 21,12,42 |
| Nref | 5906 | 5882 |
| Tmin, Tmax | 0.973,0.992 | 0.780,1.000 |
| Tmin' | 0.970 |  |

Correction method= \# Reported T Limits: Tmin=0.780 Tmax=1.000
AbsCorr $=$ MULTI-SCAN

Data completeness $=0.996$ Theta(max) $=25.350$
$R($ reflections $)=0.0819(3329) \quad$ wR2(reflections) $=0.2399(5882)$
$S=1.015 \quad$ Npar= 351


Figure S5. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{4}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{F}\right]_{2} \cdot \mathrm{CH}_{3} \mathrm{C}(\mathrm{O}) \mathrm{CH}_{3}$; $30 \%$ displacement ellipsoids.

## Crystal structure of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[3 \mathrm{c}](\mathrm{CCDC1861486})$

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 c}](20 \mathrm{mg})$ was dissolved in $\mathrm{MeCN}(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with $\mathrm{Et}_{2} \mathrm{O}(1 \mathrm{~mL})$. Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}$ $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{I}\right]$ suitable for X-ray diffraction grew within 1 d at $25^{\circ} \mathrm{C}$.



Figure S6. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{I}\right] ; 30 \%$ displacement ellipsoids.

## Crystal structure of [ $\left.\mathrm{Et}_{3} \mathrm{NH}\right][3 \mathrm{~d}-\mathrm{H}](\mathrm{CCDC1861491})$

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{3 d}-\mathrm{H}](10 \mathrm{mg})$ was dissolved in acetone $(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into a 18 cm long NMR tube and layered with hexanes ( 1 mL ). Colorless crystals of the composition [ $\mathrm{Et}_{3} \mathrm{NH}$ ] $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHC}(\mathrm{OH})-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{OCH}_{3}\right]$ suitable for X-ray diffraction grew within 5 d at 25 ${ }^{\circ} \mathrm{C}$.

```
Bond precision: B- B = 0.0051 A Wavelength=0.71073
Cell: a=9.0952(6) b=35.086(2) c=14.8327(9)
alpha=90 beta=90 gamma=90
Space group
Hall group
    4733.3(5)
    P C c n
    -P 2ab 2ac -P 2ab 2ac
Moiety formula C8 H20 B12 N 02, C6 H16 N C8 H2O B12 N 02, C6 H16 N
Sum formula C14 H36 B12 N2 O2 C14 H36 B12 N2 O2
Mr
Dx,g cm-3
    394.17
    1.106
394.17
8
Mu (mm-1)
F000
    0.062
    1680.0
    Reported
Volume
gamma=90
Temperature:
293 K
\begin{tabular}{|c|c|c|}
\hline & Calculated & Reported \\
\hline Volume & 4733.3(5) & 4733.4(5) \\
\hline Space group & P c c n & P c c n \\
\hline Hall group & -P 2ab 2ac & -P 2ab 2ac \\
\hline Moiety formula & C8 H20 B12 N O2, C6 H16 & C8 H20 B12 N O2, C6 H16 N \\
\hline Sum formula & C14 H36 B12 N2 O2 & C14 H36 B12 N2 O2 \\
\hline Mr & 394.17 & 394.17 \\
\hline Dx,g cm-3 & 1.106 & 1.106 \\
\hline Z & 8 & 8 \\
\hline Mu (mm-1) & 0.062 & 0.062 \\
\hline F000 & 1680.0 & 1680.0 \\
\hline F000' & 1680.43 & \\
\hline h, k, lmax & 10,42,17 & 10,42,17 \\
\hline Nref & 4329 & 4318 \\
\hline Tmin, Tmax & 0.981,0.989 & 0.935,1.000 \\
\hline Tmin' & 0.971 & \\
\hline
\end{tabular}
Correction method= # Reported T Limits: Tmin=0.935 Tmax=1.000
AbsCorr = MULTI-SCAN
Data completeness= 0.997
Theta(max)= 25.349
R(reflections)= 0.0927( 2910) wR2(reflections)= 0.3020( 4318)
S = 1.042 Npar= 366
```



Figure S7. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHC}(\mathrm{OH})-\mathrm{C}_{6} \mathrm{H}_{4}-\mathrm{OCH}_{3}\right]$; the protonated 4-methoxybenzamide moiety and the triethylammonium cation are both disordered. Only one of the two disordered parts is shown for clarity; $30 \%$ displacement ellipsoids.

## Crystal structure of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[3 \mathrm{e}](\mathrm{CCDC1861492})$

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}[\mathbf{3 e}](20 \mathrm{mg})$ was dissolved in $\mathrm{MeCN}(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with $\mathrm{Et}_{2} \mathrm{O}(1 \mathrm{~mL})$. Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]_{2}$ $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{~N}\right]$ suitable for X-ray diffraction grew within 2 d at $25^{\circ} \mathrm{C}$.

| Bond precision: | $\mathrm{C}-\mathrm{C}=0.0035 \mathrm{~A}$ | A Wavel | $\mathrm{h}=0.71073$ |
| :---: | :---: | :---: | :---: |
| Cell: | $a=31.573$ (2) | $\mathrm{b}=10.9139$ ( 7 ) | $\mathrm{C}=17.2044$ (13) |
|  | alpha=90 | beta=101.056(7) | gamma $=90$ |
| Temperature: | 293 K |  |  |
|  | Calculated | Repo |  |
| Volume | 5818.3(7) | 5818 |  |
| Space group | C 2/c | C 1 | 1 |
| Hall group | -C 2yc | -C |  |
| Moiety formula | C6 H16 B12 N2 O N) | ), 2(C6 H16 C6 N) | $2 \text { N2 O, } 2(\mathrm{C} 6 \mathrm{H} 16$ |
| Sum formula | C18 H48 B12 N4 | $0 \quad \mathrm{C} 18$ | B12 N4 O |
| Mr | 466.32 | 466 |  |
| Dx,g cm-3 | 1.065 | 1.06 |  |
| Z | 8 | 8 |  |
| Mu (mm-1) | 0.059 | 0.05 |  |
| F000 | 2016.0 | 2016 |  |
| F000' | 2016.43 |  |  |
| h, k, 1 max | 38,13,20 | 38,1 |  |
| Nref | 5345 | 5342 |  |
| Tmin, Tmax | 0.972,0.977 | 0.94 | 000 |
| Tmin' | 0.972 |  |  |
| ```Correction method= # Reported T Limits: Tmin=0.949 Tmax=1.000 AbsCorr = MULTI-SCAN``` |  |  |  |
| Data completeness=0.999 |  | Theta $(\max )=25.350$ |  |
| R (reflections) $=0.0605(3486$ ) |  | wR2 (reflections) $=0.1775(5342)$ |  |
| $\mathrm{S}=1.050$ | Npar $=350$ |  |  |



Figure S8. ORTEP representation of $\left[\mathrm{Et} \mathrm{t}_{3} \mathrm{NH}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{~N}\right] ; 30 \%$ displacement ellipsoids.

## Crystal structure of 3e-H (CCDC1861490)

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][3 \mathrm{e}-\mathrm{H}](25 \mathrm{mg})$ was dissolved in $\mathrm{MeOH} / \mathrm{MeCN}(1 \mathrm{~mL} / 1 \mathrm{~mL})$ at ca. $50{ }^{\circ} \mathrm{C}$ in a 4 mL glass vial and allowed to cool to room temperature. Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{~N}-\mathrm{H}\right] \cdot \mathrm{CH}_{3} \mathrm{CN}$ suitable for X-ray diffraction were obtained within 1 d .



Figure S9. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NHCO}-\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{~N}-\mathrm{H}\right] \cdot \mathrm{CH}_{3} \mathrm{CN} ; 30 \%$ displacement ellipsoids.

## Crystal structure of [ $\left.\mathrm{Et}_{3} \mathrm{NH}\right][6 \mathrm{a}]$ (CCDC1861483)

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][6 a](10 \mathrm{mg}, 0.031 \mathrm{mmol})$ was dissolved in acetonitrile $(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into a 18 cm long NMR tube and layered with diethylether ( 1 mL ). Colorless crystals of the composition $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{CH}-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right] \cdot 2 \mathrm{CH}_{3} \mathrm{CN}$ suitable for X-ray diffraction grew within 5 d at $25^{\circ} \mathrm{C}$.



Figure S10. ORTEP representation of $\left[\mathrm{Et}_{3} \mathrm{NH}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{CH}-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right] \cdot 2 \mathrm{CH}_{3} \mathrm{CN}$;
$30 \%$ displacement ellipsoids.

## Crystal structure of [MePPh $\left.{ }_{3}\right][6 \mathrm{a}]$ (CCDC1861484)

Single crystals of $\mathbf{6 a}$ were also obtained with the $\left[\mathrm{MePPh}_{3}\right]^{+}$cation, and the structure is similar to that of $\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{6 a}] .\left[\mathrm{Et}_{3} \mathrm{NH}\right][\mathbf{6 a}](30 \mathrm{mg})$ was suspended in water $(1 \mathrm{~mL})$, and NaOH (2 equiv) was added to form the $\mathrm{Na}^{+}$salt. To this solution $\left[\mathrm{MePPh}_{3}\right] \mathrm{Br}(2$ equiv) was added to give $\left[\mathrm{MePPh}_{3}\right][\mathbf{6 a}]$ as a colorless precipitate. $\left[\mathrm{MePPh}_{3}\right][\mathbf{6 a}](20$ $\mathrm{mg})$ was dissolved in acetone $(0.5 \mathrm{~mL})$. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with $\mathrm{Et}_{2} \mathrm{O}(1 \mathrm{~mL})$. Colorless crystals of the composition $\left[\mathrm{MePPh}_{3}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{CH}-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right]$ suitable for X-ray diffraction grew within 2 d at $25^{\circ} \mathrm{C}$.



Figure S11. ORTEP representation of $\left[\mathrm{MePPh}_{3}\right]\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{CH}-\mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right] ; 30 \%$ displacement ellipsoids.

## Crystal structure of [ $\left.\mathrm{Et}_{3} \mathrm{NH}\right][6 \mathrm{c}]$ (CCDC1861485)

Compound $\left[\mathrm{Et}_{3} \mathrm{NH}\right][6 \mathrm{c}](10 \mathrm{mg})$ was dissolved in acetonitrile $(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into a 18 cm long NMR tube and layered with diethylether $(1 \mathrm{~mL})$. Colorless crystals of the composition [ $\mathrm{Et}_{3} \mathrm{NH}$ ] $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{C}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)\left(\mathrm{NH}-\mathrm{C}_{6} \mathrm{H}_{5}\right)\right] \cdot \mathrm{H}_{2} \mathrm{O}$ suitable for X-ray diffraction grew within 5 d at $25^{\circ} \mathrm{C}$.

```
Bond precision: C-C = 0.0059 A Wavelength=0.71073
Cell: a=10.8688(7) b=12.0426(6) c=13.1037(9)
    alpha=66.402(5) beta=67.187(6) gamma=85.080(5)
Temperature: 293 K
```



```
Correction method= # Reported T Limits: Tmin=0.985 Tmax=1.000
AbsCorr = MULTI-SCAN
Data completeness=0.986 Theta(max)=25.350
R(reflections)=0.0962( 3419) wR2(reflections)= 0.3064( 5202)
S = 1.050 Npar= 329
```



Figure S12. ORTEP representation of $\left[\mathrm{B}_{12} \mathrm{H}_{11} \mathrm{NH}=\mathrm{C}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)\left(\mathrm{NH}-\mathrm{C}_{6} \mathrm{H}_{5}\right)\right] \cdot \mathrm{H}_{2} \mathrm{O} ; 30 \%$ displacement ellipsoids.

## Crystal structure of $\left[\mathrm{MePPh}_{3}\right]_{2}[8](\mathbf{C C D C 1 8 6 1 4 8 7})$

$\left[\mathrm{MePPh}_{3}\right]_{2}[8](10 \mathrm{mg})$ was dissolved in acetone $(0.5 \mathrm{~mL})$ in a 1 mL glass vial. The resulting colorless solution was filtered into an 18 cm long NMR tube and layered with $\mathrm{Et}_{2} \mathrm{O}(1 \mathrm{~mL})$. Colorless crystals of the composition $\left[\mathrm{MePPh}_{3}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{~N}=\mathrm{C}=\mathrm{O}\right]$ suitable for X-ray diffraction grew within 2 d at $25^{\circ} \mathrm{C}$. Single crystals could also be obtained by recrystallization from aceotone.

```
Bond precision: C-C = 0.0041 A Wavelength=0.71073
Cell: a=11.3939(14) b=13.1505(15) c=14.8700(15)
    alpha=89.844(9) beta=81.969(9) gamma=71.540(11)
Temperature: 293 K
\begin{tabular}{|c|c|c|}
\hline & Calculated & Reported \\
\hline Volume & 2090.6(4) & 2090.6(4) \\
\hline Space group & P -1 & P -1 \\
\hline Hall group & -P 1 & -P 1 \\
\hline Moiety formula & \[
2(\mathrm{C} 19 \mathrm{H} 18 \mathrm{P}), \mathrm{C} \text { H11 B12 N }
\] & \[
\begin{aligned}
& 2(\text { C19 H18 P), C H11 B12 N } \\
& \text { O }
\end{aligned}
\] \\
\hline Sum formula & C39 H47 B12 N O P2 & C39 H47 B12 N O P2 \\
\hline Mr & 737.44 & 737.44 \\
\hline Dx,g cm-3 & 1.171 & 1.171 \\
\hline Z & 2 & 2 \\
\hline Mu (mm-1) & 0.137 & 0.137 \\
\hline F000 & 772.0 & 772.0 \\
\hline F000' & 772.62 & \\
\hline h, k, 1max & 13,15,17 & 13,15,17 \\
\hline Nref & 7655 & 7633 \\
\hline Tmin, Tmax & 0.952,0.973 & 0.575,1.000 \\
\hline Tmin' & 0.952 & \\
\hline
\end{tabular}
Correction method= # Reported T Limits: Tmin=0.575 Tmax=1.000
AbsCorr = MULTI-SCAN
Data completeness= 0.997 Theta(max)= 25.350
R(reflections)= 0.0507( 4888) wR2(reflections)= 0.1366( 7633)
S = 0.961 Npar= 498
```



Figure S13. ORTEP representation of $\left[\mathrm{MePPh}_{3}\right]_{2}\left[\mathrm{~B}_{12} \mathrm{H}_{11} \mathrm{~N}=\mathrm{C}=\mathrm{O}\right] ; 30 \%$ displacement ellipsoids.

## IV References

[1] V. Geis, K. Guttsche, C. Knapp, H. Scherer, R. Uzun, Dalton Trans. 2009, 2687-2694.
[2] O. Bondarev, A. A. Khan, X. Tu, Y. V. Sevrugina, S. S. Jalisatgi, M. F. Hawthorne, J. Am. Chem. Soc. 2013, 135, 13204-13211.
[3] Y. Sun, J. Zhang, Y. Zhang, J. Liu, S. van der Veen, S. Duttwyler, Chem. Eur. J. 2018, 24, 10364-10371.


11B NMR 126 MHz CD3CN







20180602 ［NBu4］2［B12H11NHCONHPh］40mg dissolved in CD3CN 13C\｛1H\} NMR 101MHz

（17

$\left[\mathrm{NB}_{4}^{\oplus} \mathrm{u}_{4}\right]$
$\left[\mathrm{NB}_{4}^{\oplus}\right]$



 13C\｛1H\} NMR 101MHz
$\stackrel{\text { n }}{\sim}$




140





2016062920 mg [MePPh3]2[B12H11NCO] dissolved in 0.6 mL CD3CN, 1H NMR, 400MHz signal of NCO not detected

## $6 Z^{\circ} 811$ 06.611 $6 L^{\circ} 021$ $01^{\circ} 1 \varepsilon 1$ $\varepsilon \iota^{\circ} 1 \varepsilon 1$ $\angle l^{\circ} \hbar \varepsilon$ $8 Z^{\circ} \downarrow \varepsilon$ $90^{\circ} 9 \varepsilon$ $60^{\circ} 9 \varepsilon$ 







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${ }^{\oplus}$
$\mathrm{Et}_{3} \mathrm{NH}^{\oplus}$
20150313－syj－0077－2，Et3NHB12H11NHCHNMe2
20150316， $160 \mathrm{MHz}, 11 \mathrm{~B}, 12.4 \mathrm{mg}$ in 0.6 ml CD3CN

| Current <br> NAME <br> EXPNO <br> PROCNO | $\begin{gathered} \text { Data Parameters } \\ 20150313-\mathrm{syj}-0078-1 \\ 3 \\ 1 \end{gathered}$ |
| :---: | :---: |
| F2－Acquisition |  |
| Date | 20150316 |
| Time | 18.38 |
| INSTRUM | spect |
| PROBHD | 5 mm PABBO ${ }^{\text {BB－}}$ |
| PULPROG | zg30 |
| TD | 65536 |
| SOLVENT | CD3CN |
| NS | 16 |
| DS |  |
| SWH | 32051.281 Hz |
| FIDRES | 0.489064 Hz |
| AQ | 1.0223616 sec |
| RG | 203 |
| DW | 15.600 usec |
| DE | 6.50 usec |
| TE | 293.6 |
| D1 | 2.00000000 sec |
|  | CHANNEL f1＝＝ |
| ${ }_{\text {P1 }}{ }^{\text {N }}$ | 11B |
| ${ }_{\text {P1 }}^{\text {PLW1 }}$ | 75.000000000 W |
| SFO1 | 160.4615792 MHz |
| F2－Processing parame |  |
| SI | 32768 |
| SF $\quad 160.4615790 \mathrm{MH}$ |  |
| WDW EM |  |
| SSB | 0 － |
| LB $\quad 10.00 \mathrm{~Hz}$ |  |
| $\begin{aligned} & \text { GB } \\ & \text { PC } \end{aligned}$ | 0 1．40 |
| PC | 1.40 |


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20160228-syj-0196-1, Et3NHB12NHC(C6H5)NHC6H5
20160228, $128 \mathrm{MHz}, 11 \mathrm{~B}$ NMR, 6.2 mg in 0.6 ml CD2CI2

$\angle 8^{\circ} \mathrm{G}$





$$
\begin{array}{llll}
-10 & -15 & -20
\end{array}
$$

20160228－syj－0196－1，Et3NHB12NHC（C6H5）NHC6H5
20160228， $128 \mathrm{MHz}, 11 \mathrm{~B}\{1 \mathrm{H}\}$ NMR， 6.2 mg in 0.6 ml CD2

\&がLL-
\&がLL-
ع9Gl-
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S8＇s－

Current
Nata Parameters
NAME
EXPNO
PROCNO

$*$
20160223－syj－0193－1，B12NHC（C6H3Cl2）NH（CH2）2NH（CH3）2

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26.6 \varepsilon
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