

# **Silyl Ketene Acetals/B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>Lewis Pair-Catalyzed Living Group Transfer Polymerization of Renewable Cyclic Acrylic Monomers**

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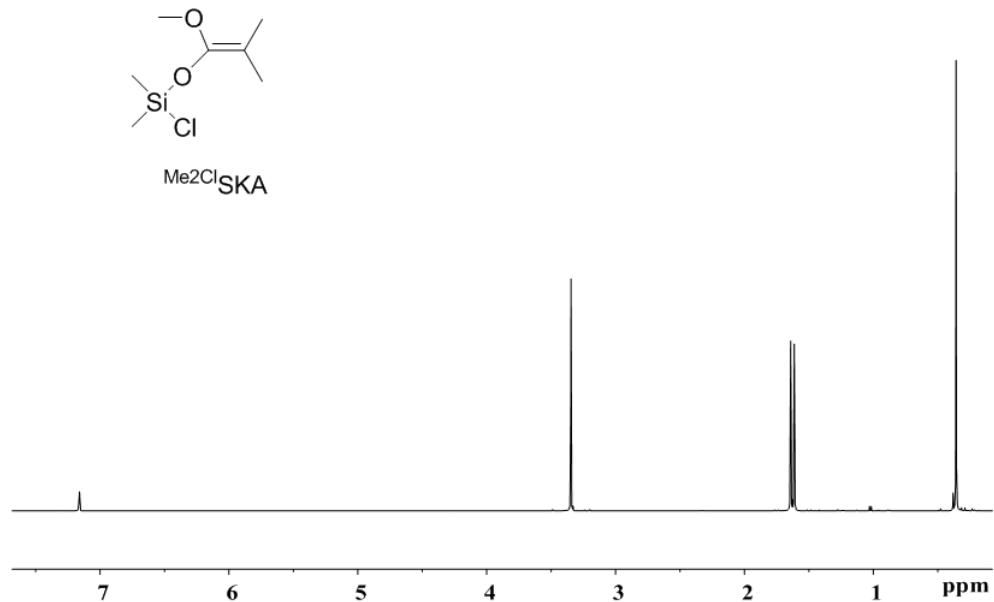
## 1. Selected polymerization data

**Table S1.**  $B(C_6F_5)_3$ -catalyzed MMA polymerization <sup>a</sup>

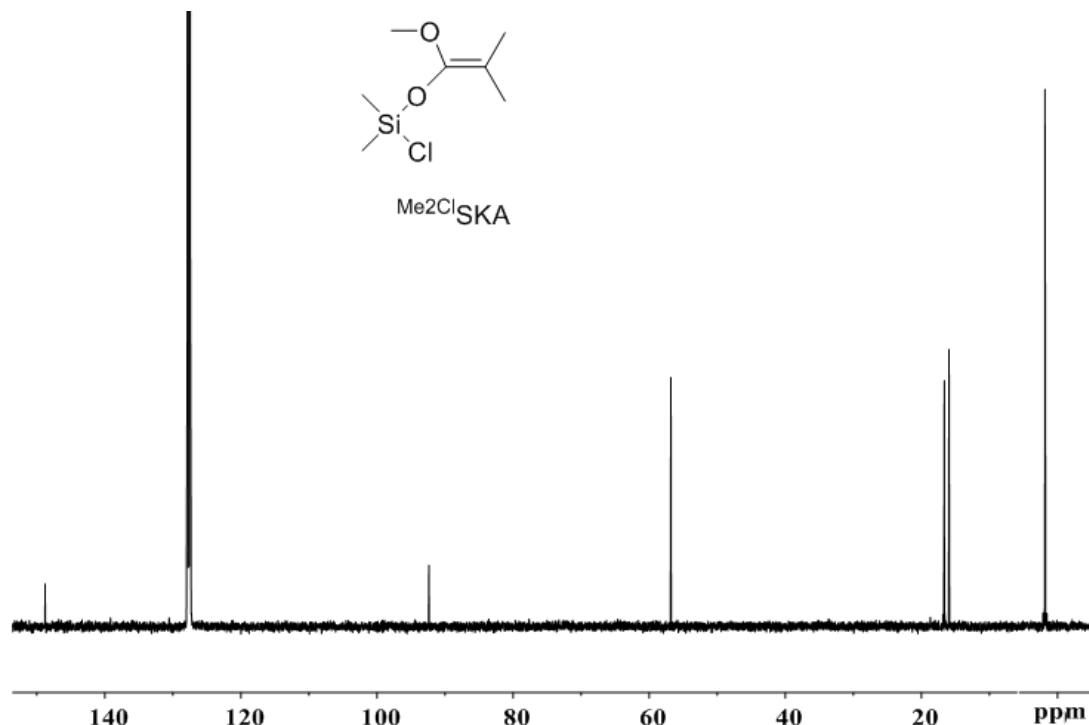
Run No.	Initiator (I)	[M]:[I]:[B] <sup>b</sup>	Time (h)	Conv. <sup>c</sup> (%)	$M_n^d$ (Kg·mol <sup>-1</sup> )	D	$I^{*e}$ (%)
1	Me <sub>2</sub> PhSiH	200:1:1	24	4.46	n.d.	n.d.	n.d.
2	Me <sub>2</sub> PhSiH	50:1:1	24	18.1	104	1.24	1
3 <sup>e</sup>	Me <sub>2</sub> PhSiH	200:1:1	24	12.9	296	1.55	1
4	Me <sub>2</sub> EtSiH	200:1:1	24	5.67	n.d.	n.d.	n.d.
5	Me <sub>2</sub> EtSiH	50:1:1	24	51.9	3	1.06	88
6 <sup>e</sup>	Me <sub>2</sub> EtSiH	200:1:1	24	12.1	428	1.33	0.6
7	Et <sub>3</sub> SiH	200:1:1	24	6.24	n.d.	n.d.	n.d.
8	Et <sub>3</sub> SiH	50:1:1	24	18.37	94	1.31	1
9 <sup>e</sup>	Et <sub>3</sub> SiH	200:1:1	24	9.24	305	1.55	0.6
10	Ph <sub>3</sub> SiH	200:1:1	24	4.34	n.d.	n.d.	n.d.
11	Ph <sub>3</sub> SiH	50:1:1	24	10.7	192	1.55	0.3
12 <sup>e</sup>	Ph <sub>3</sub> SiH	200:1:1	24	6.05	691	1.14	0.2
13	iBu <sub>3</sub> SiH	200:1:1	24	7.13	n.d.	n.d.	n.d.
14	iBu <sub>3</sub> SiH	50:1:1	24	13.9	234	1.53	0.3
15 <sup>e</sup>	iBu <sub>3</sub> SiH	200:1:1	24	8.88	668	1.09	0.3
16	Me <sub>2</sub> ClSiH	200:1:1	24	4.24	n.d.	n.d.	n.d.
17	Me <sub>2</sub> ClSiH	50:1:1	24	11.3	101	1.26	0.6
18 <sup>e</sup>	Me <sub>2</sub> ClSiH	200:1:1	24	9.7	367	1.43	0.5
19	MeSKA	200:1:1	24	5	n.d.	n.d.	n.d.
20	MeSKA	50:1:1	24	47.2	3.1	1.05	77
21 <sup>e</sup>	MeSKA	200:1:1	24	13.2	312	1.49	1
22	EtSKA	200:1:1	24	5.59	n.d.	n.d.	n.d.
23	EtSKA	50:1:1	24	18.2	106	1.25	1
24 <sup>e</sup>	EtSKA	200:1:1	24	11.2	358	1.4	0.6
25	iBuSKA	200:1:1	24	6.36	n.d.	n.d.	n.d.
26	iBuSKA	50:1:1	24	29.2	2.8	1.01	53
27 <sup>e</sup>	iBuSKA	200:1:1	24	69.4	705	1.08	0.2
28	PhSKA	200:1:1	24	4.59	n.d.	n.d.	n.d.
29	PhSKA	50:1:1	24	11.9	74.3	1.53	0.8
30 <sup>e</sup>	PhSKA	200:1:1	24	12.2	377	1.35	0.7
31	Me <sub>2</sub> ClSKA	200:1:1	24	6.10	n.d.	n.d.	n.d.
32	Me <sub>2</sub> ClSKA	50:1:1	24	13.7	88.5	1.33	0.8
33 <sup>e</sup>	Me <sub>2</sub> ClSKA	200:1:1	24	8.77	427	1.35	0.4
34	Me <sub>2</sub> (EtO)SKA	200:1:1	24	23.5	145	1.25	3
35	Me <sub>2</sub> (EtO)SKA	50:1:1	24	54.6	3.1	1.08	89
36 <sup>e</sup>	Me <sub>2</sub> (EtO)SKA	200:1:1	24	30.8	246	1.51	3
37	Me <sub>2</sub> PhSKA	200:1:1	24	4.26	n.d.	n.d.	n.d.
38	Me <sub>2</sub> PhSKA	50:1:1	24	14.8	78.8	1.48	1
39 <sup>e</sup>	Me <sub>2</sub> PhSKA	200:1:1	24	9.79	315	1.48	0.6

<sup>a</sup> Carried out in 9 mL CH<sub>2</sub>Cl<sub>2</sub> at room temperature, where [MMA]<sub>0</sub> = 0.943 M, n.d. = not determined. <sup>b</sup> [M] = [Monomer], [I] = [Initiator], and [B] = [B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>]. <sup>c</sup> Monomer conversions measured by <sup>1</sup>H NMR. <sup>d</sup> M<sub>n</sub> and D determined by GPC relative to PMMA standards in DMF. <sup>e</sup> Initiator efficiency (I\*)% = M<sub>n</sub>(calcd)/M<sub>n</sub>(exptl) × 100, where M<sub>n</sub>(calcd) = [MW(MMA)] × ([MMA]<sub>0</sub>/[I]<sub>0</sub>) (conversion) + MW of chain-end groups. <sup>f</sup> [MMA]<sub>0</sub> = 3.77 M.

## 2. NMR spectrum of <sup>Me<sub>2</sub>Cl</sup>SKA

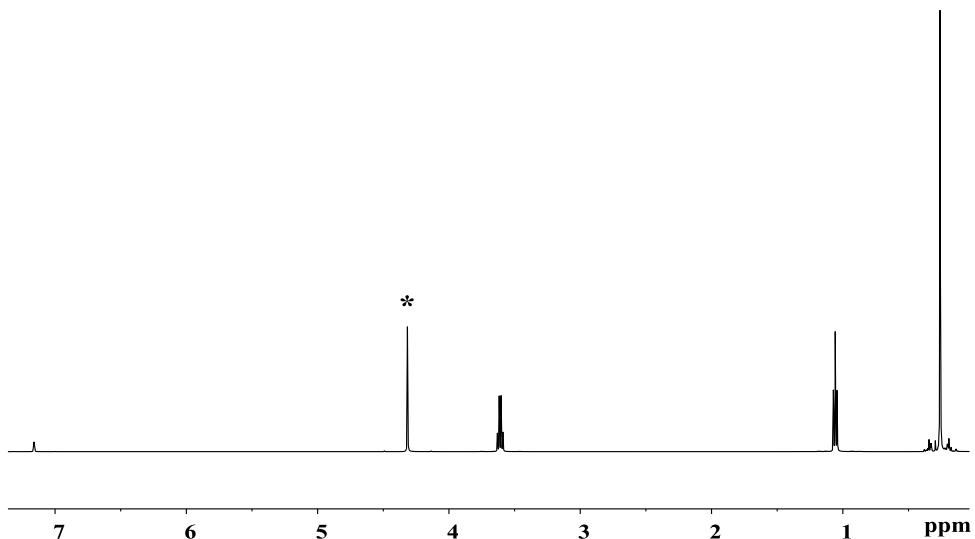


**Figure S1.** <sup>1</sup>H NMR spectrum (benzene-*d*<sub>6</sub>, 500 MHz) of <sup>Me<sub>2</sub>Cl</sup>SKA.

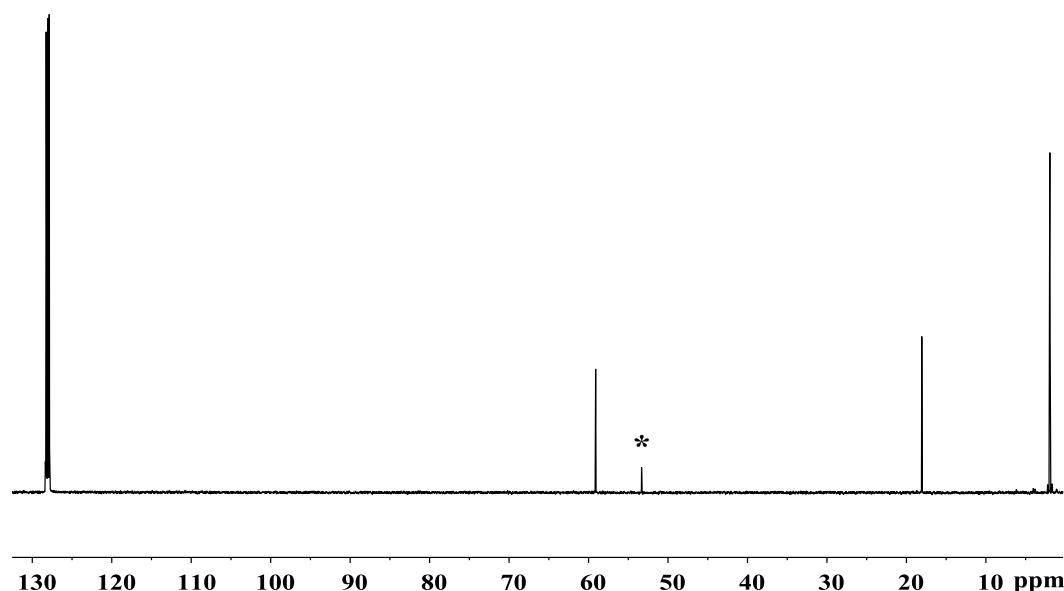


**Figure S2.** <sup>13</sup>C NMR spectrum (benzene-*d*<sub>6</sub>, 126 MHz) of <sup>Me<sub>2</sub>Cl</sup>SKA.

**3. NMR spectrum of Me<sub>2</sub>EtOSiCl**

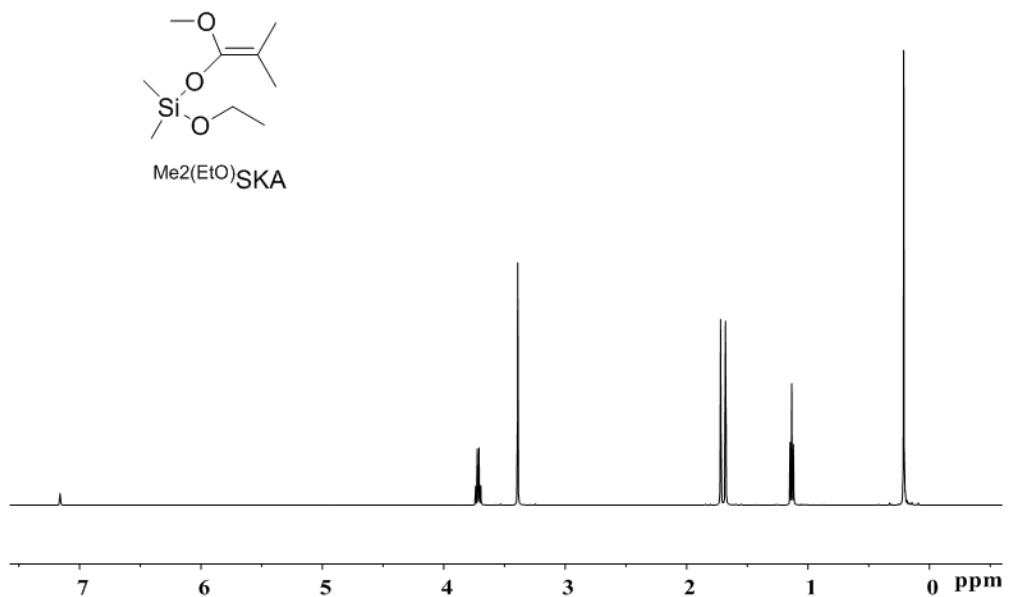


**Figure S3.** <sup>1</sup>H NMR spectrum of (benzene-*d*<sub>6</sub>, 500 MHz) Me<sub>2</sub>EtOSiCl. This spectrum also contain CH<sub>2</sub>Cl<sub>2</sub> (peak marked with an \*)

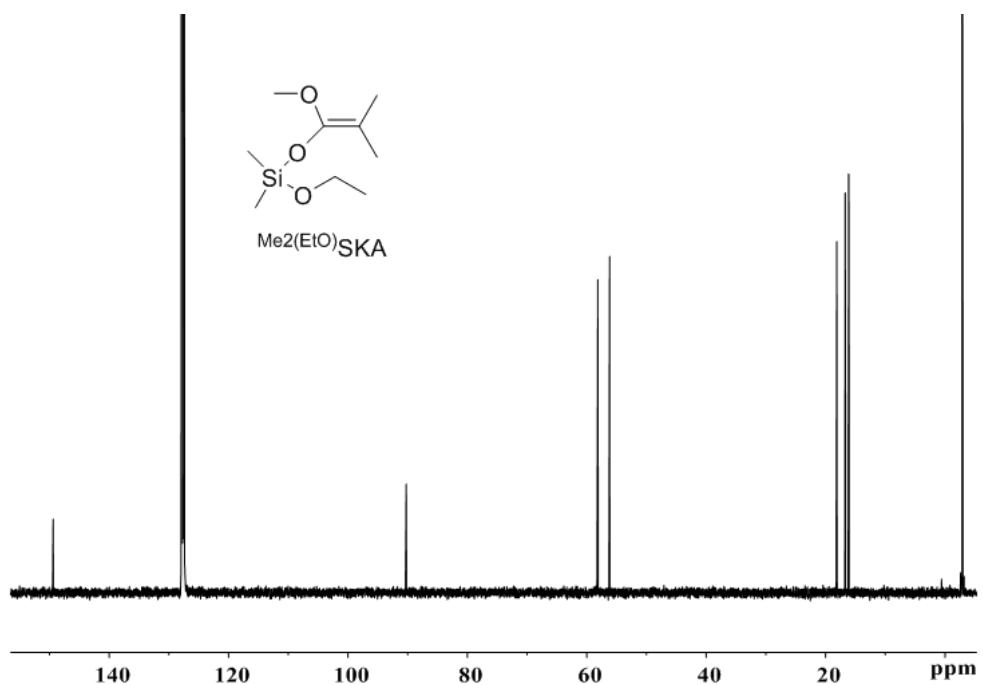


**Figure S4.** <sup>13</sup>C NMR spectrum (benzene-*d*<sub>6</sub>, 126 MHz) of Me<sub>2</sub>EtOSiCl. This spectrum also contain CH<sub>2</sub>Cl<sub>2</sub> (peak marked with an \*)

**4. NMR spectrum of  $^{Me_2(EtO)}SKA$**

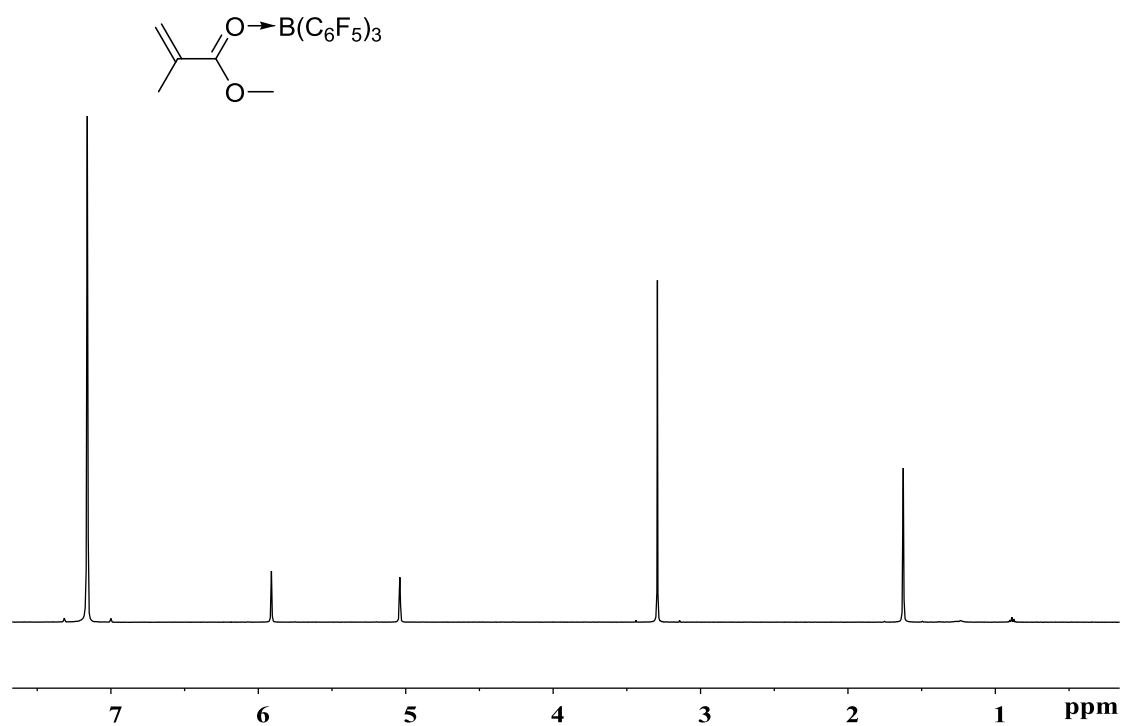


**Figure S5.**  $^1H$  NMR spectrum (benzene- $d_6$ , 500 MHz) of  $^{Me_2(EtO)}SKA$ .

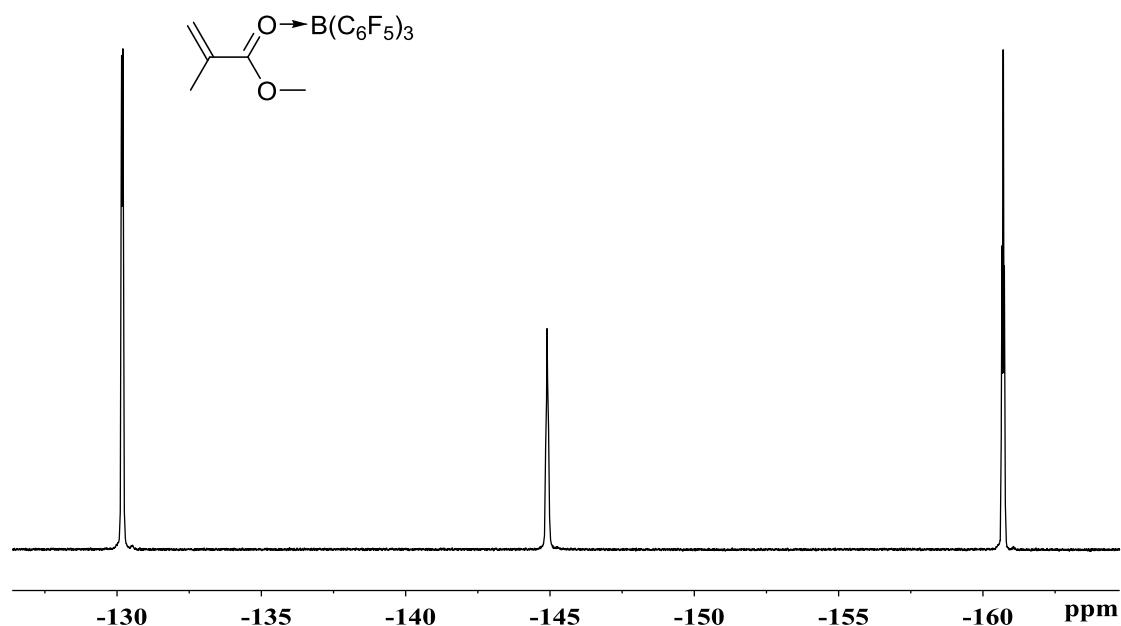


**Figure S6.**  $^{13}C$  NMR spectrum (benzene- $d_6$ , 126 MHz) of  $^{Me_2EtOSKA}$

**5. NMR spectrum of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMA}$**

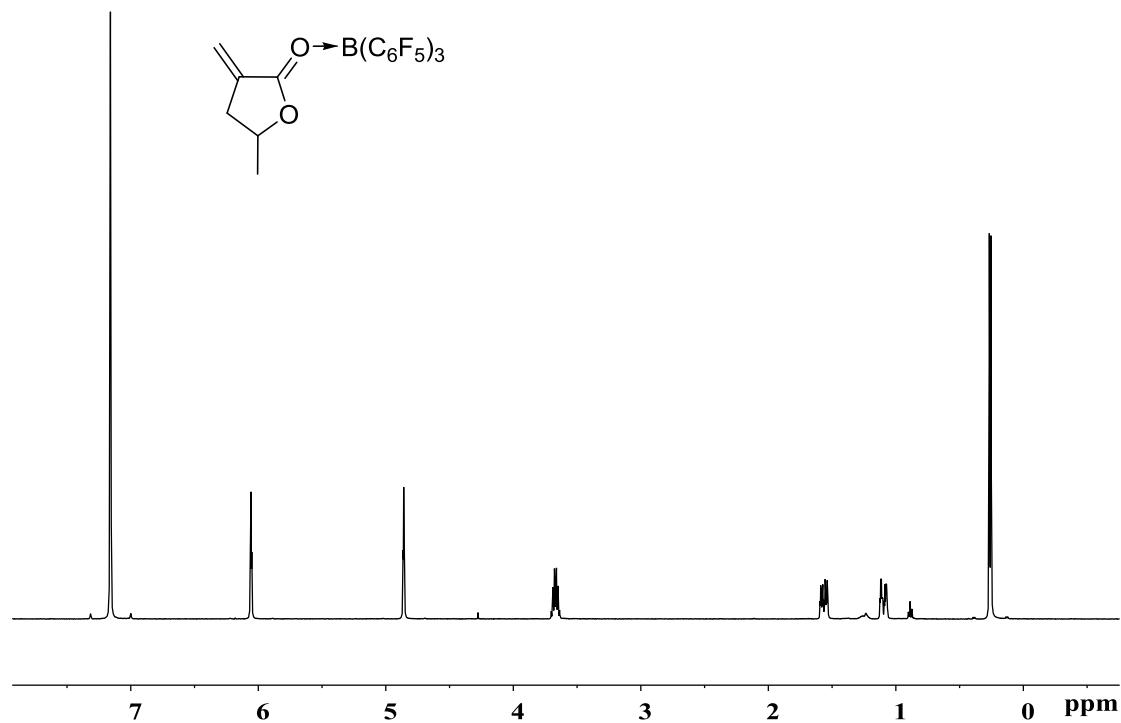


**Figure S7.**  $^1\text{H}$  NMR spectrum (benzene- $d_6$ , 500 MHz) of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMA}$ .

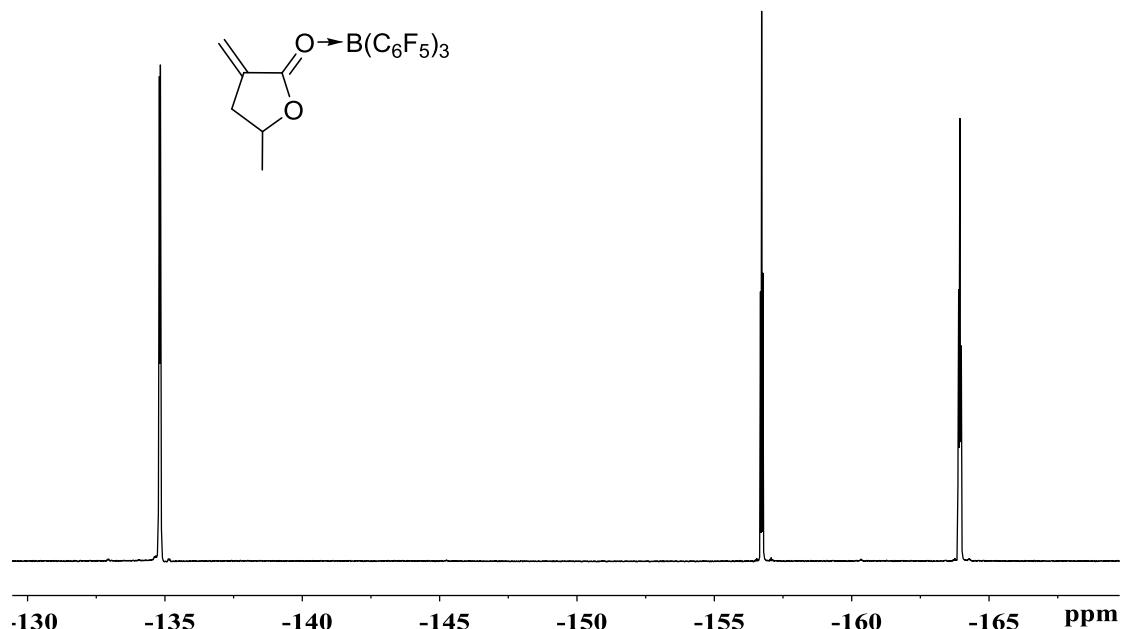


**Figure S8.**  $^{19}\text{F}$  NMR spectrum (benzene- $d_6$ , 471 MHz) of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMA}$ .

**6. NMR spectrum of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMBL}$**

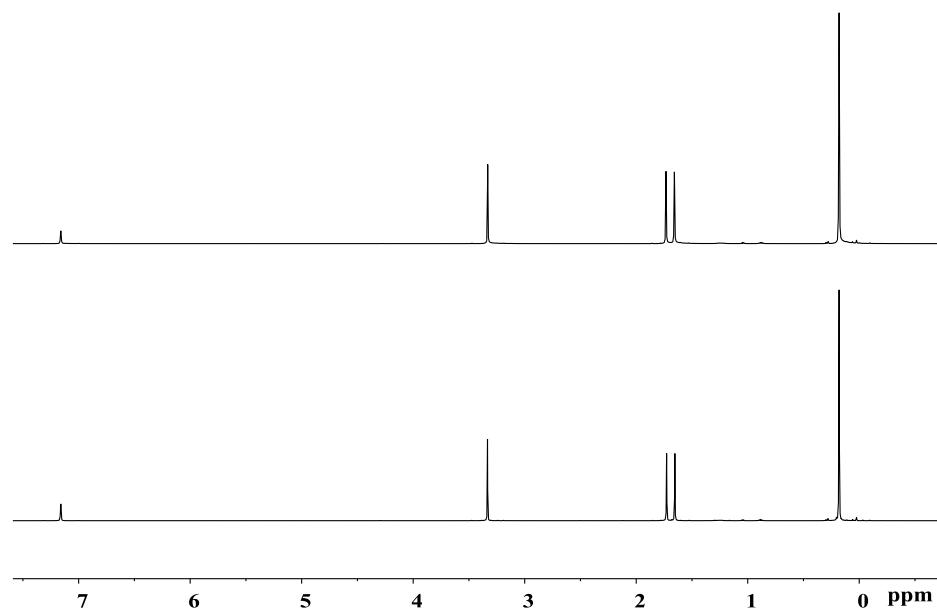


**Figure S9.** <sup>1</sup>H NMR spectrum (benzene-*d*<sub>6</sub>, 500 MHz) of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMBL}$ .

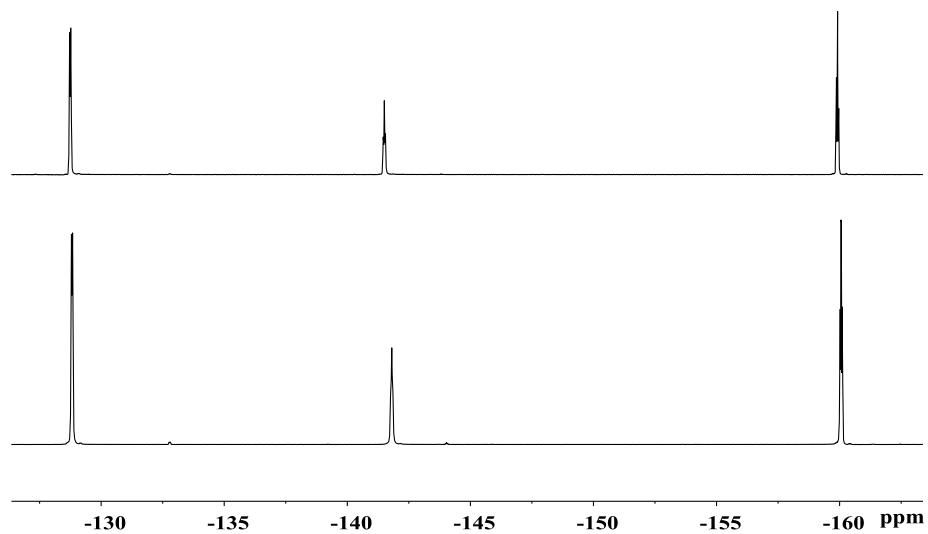


**Figure S10.** <sup>19</sup>F NMR spectrum (benzene-*d*<sub>6</sub>, 471 MHz) of  $\text{B}(\text{C}_6\text{F}_5)_3\cdot\text{MMBL}$

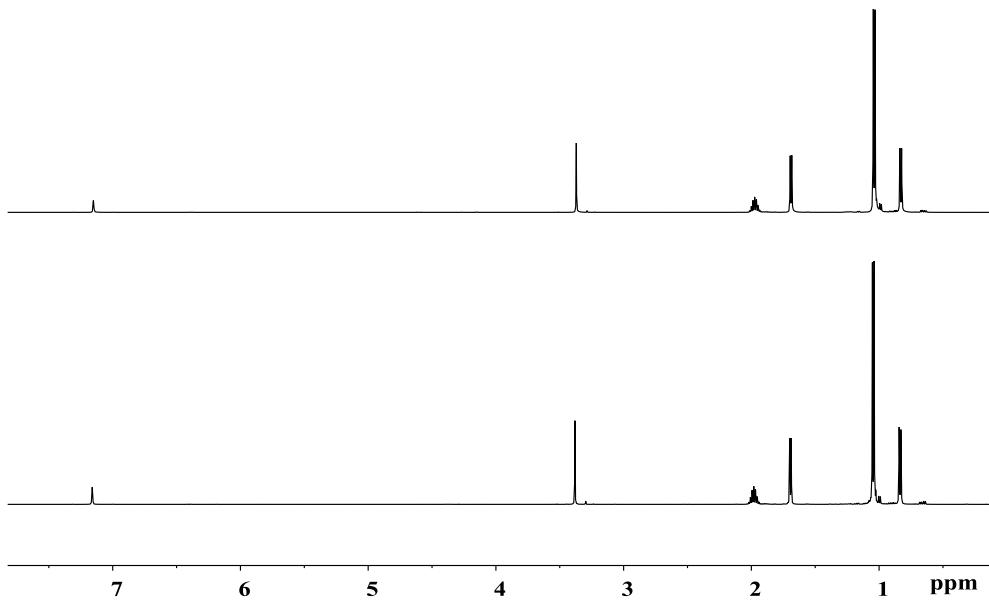
**7. NMR spectrum of the reaction of SKA with  $B(C_6F_5)_3$  in 1:1 ratio**



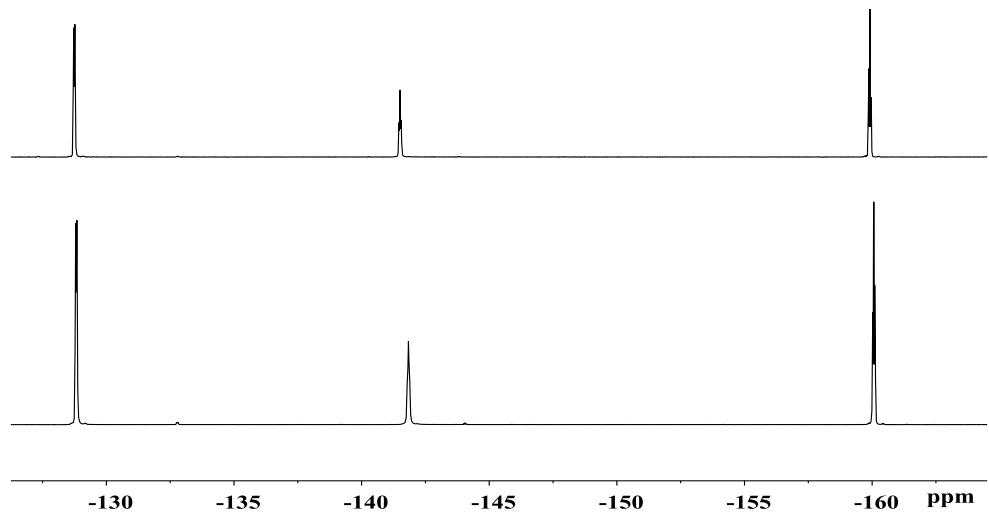
**Figure S11.**  $^1H$  NMR spectrum (benzene-*d*<sub>6</sub>, 500 MHz) of <sup>Me</sup>SKA (Top) and the reaction with <sup>Me</sup>SKA/ $B(C_6F_5)_3$ = 1:1ratio at RT (Bottom).



**Figure S12.**  $^{19}F$  NMR spectrum (benzene-*d*<sub>6</sub>, 471 MHz) of  $B(C_6F_5)_3$  (Top) and the reaction with <sup>Me</sup>SKA/ $B(C_6F_5)_3$ = 1:1ratio at RT (Bottom).

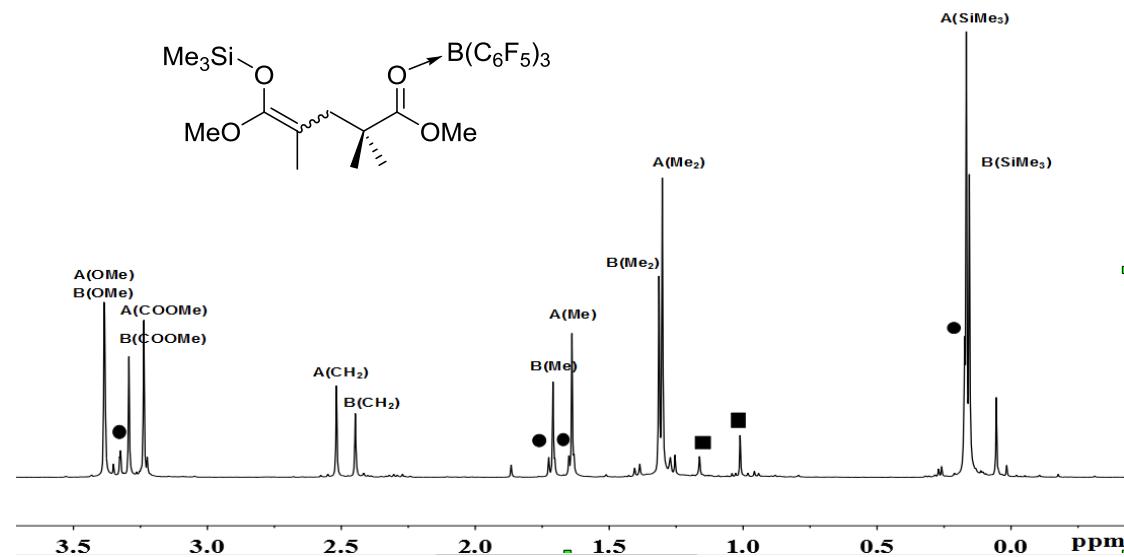


**Figure S13.** <sup>1</sup>H NMR spectrum (benzene-*d*<sub>6</sub>, 500 MHz) of <sup>i</sup>BuSKA (Top) and the reaction with <sup>i</sup>BuSKA/B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> = 1:1ratio at RT (Bottom).

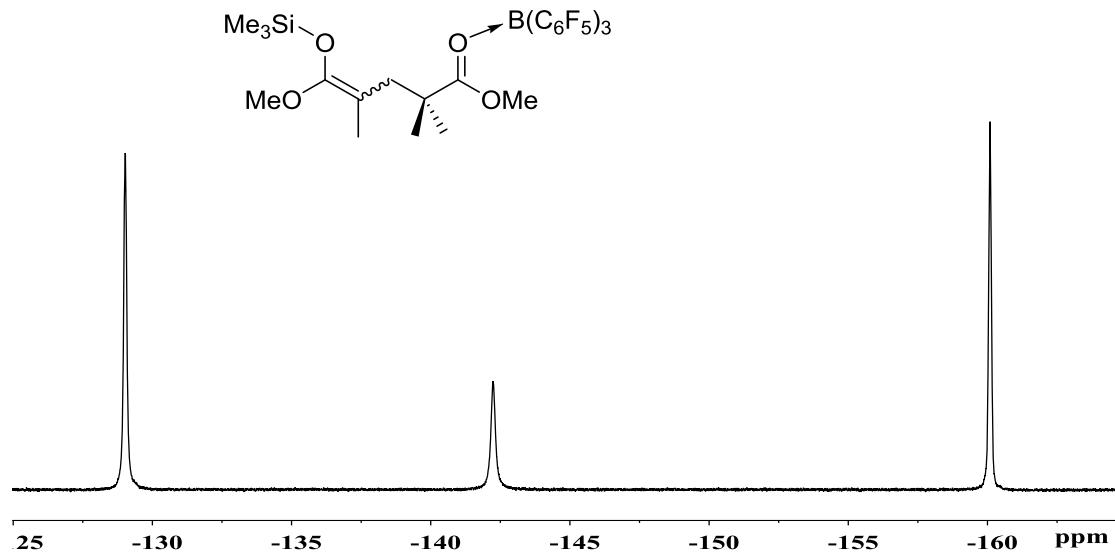


**Figure S14.** <sup>19</sup>F NMR spectrum (benzene-*d*<sub>6</sub>, 471 MHz) of B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> (Top) and the reaction with <sup>i</sup>BuSKA/B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> = 1:1ratio at RT (Bottom).

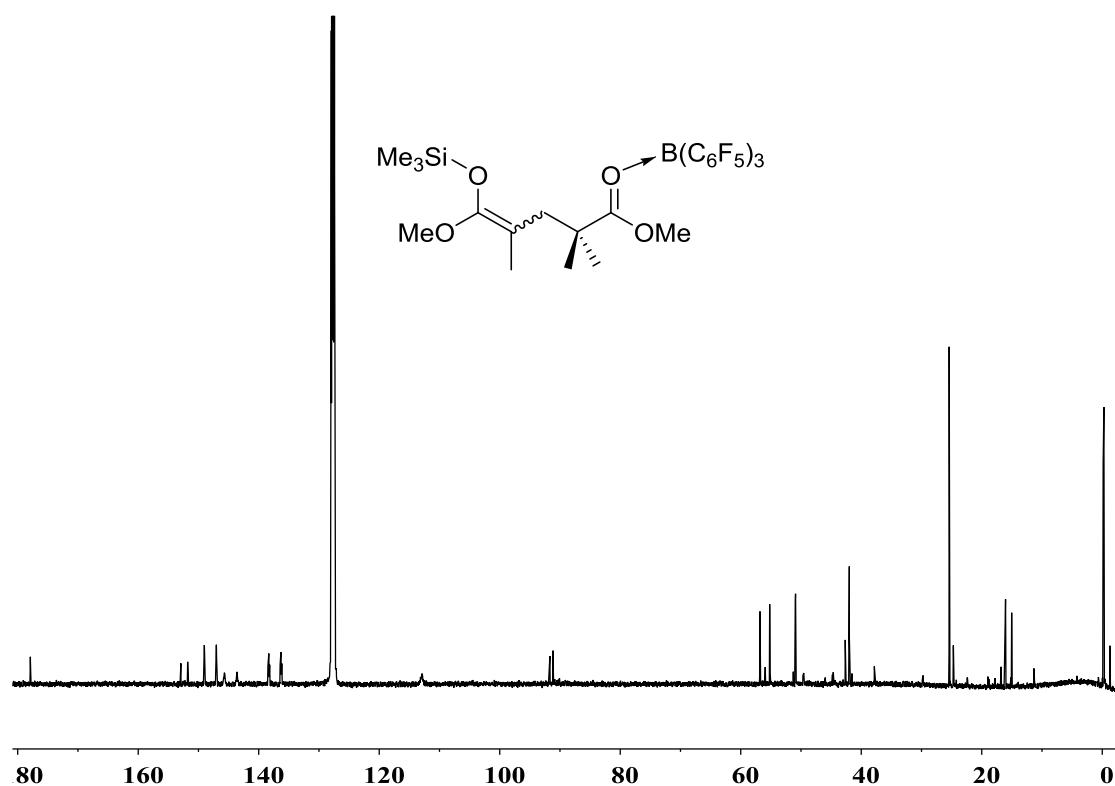
**8. NMR spectrum of the reaction of SKA with  $B(C_6F_5)_3\cdot MMA$  in 1:1 ratio**



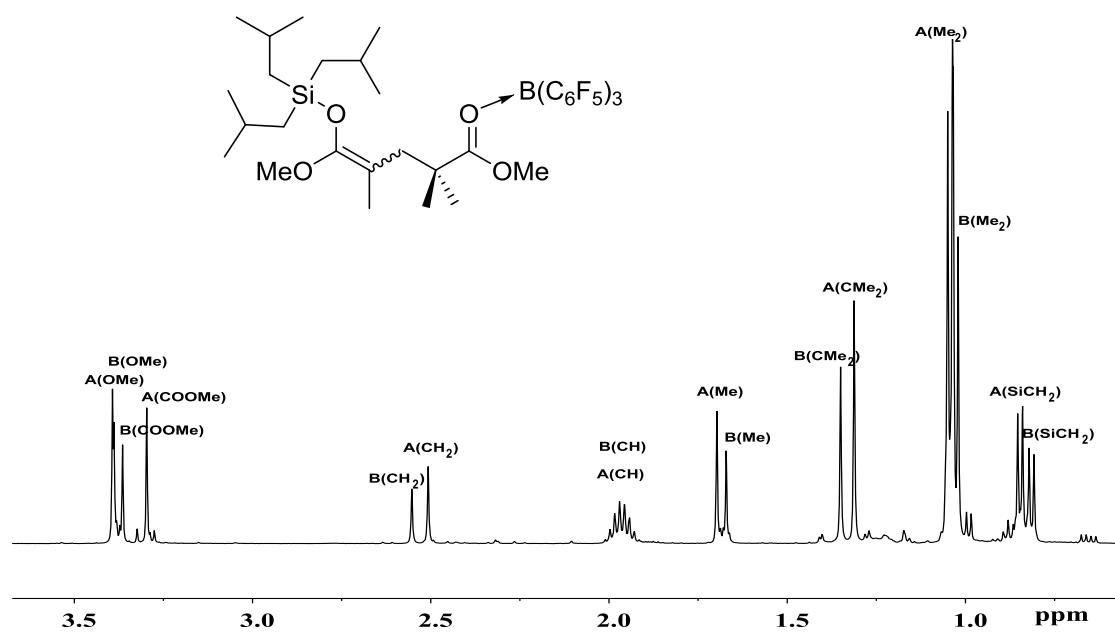
**Figure S15.**  $^1H$  NMR spectrum (benzene- $d_6$ , 500 MHz) of the reaction with  $^{Me}SKA/B(C_6F_5)_3\cdot MMA = 1:1$  ratio at RT. (major isomer **A** and minor isomer **B** in 3:2 ratio, the spectrum also contains a small amount of  $^{Me}SKA$  (peaks marked with circle)



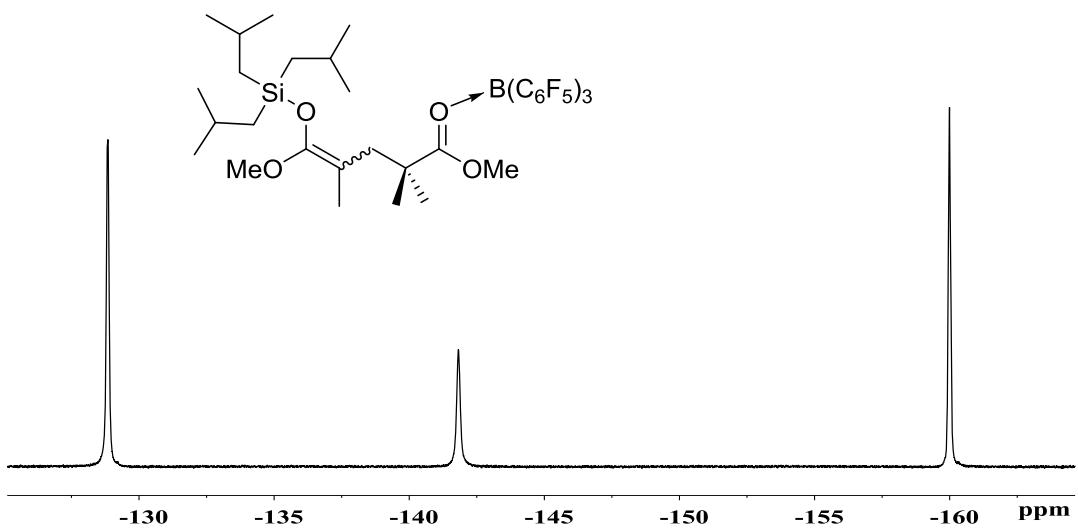
**Figure S16.**  $^{19}F$  NMR spectrum (benzene- $d_6$ , 471 MHz) of the reaction with  $^{Me}SKA/B(C_6F_5)_3\cdot MMA = 1:1$  ratio at RT.



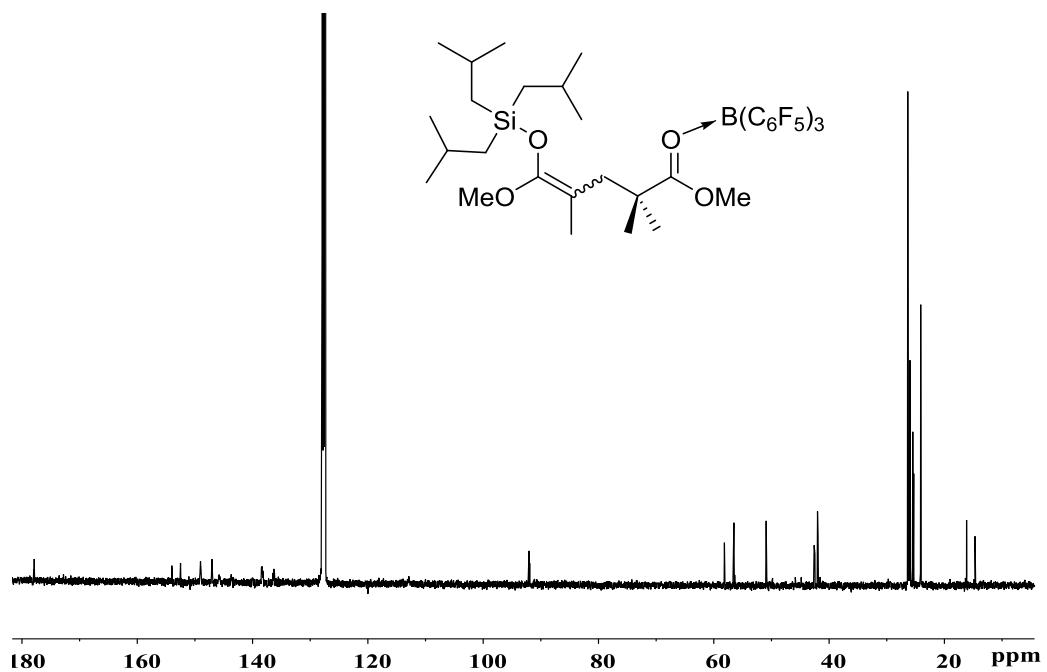
**Figure S17.**  $^{13}\text{C}$  NMR spectrum (benzene- $d_6$ , 126 MHz) of the reaction with  $^{\text{Me}}\text{SKA}/\text{B}(\text{C}_6\text{F}_5)_3\text{-MMA}=1:1$  ratio at RT.



**Figure S18.**  $^1\text{H}$  NMR spectrum (benzene- $d_6$ , 500 MHz) of the reaction with  $^{\text{iBu}}\text{SKA}/\text{B}(\text{C}_6\text{F}_5)_3\text{-MMA}=1:1$  ratio at RT.

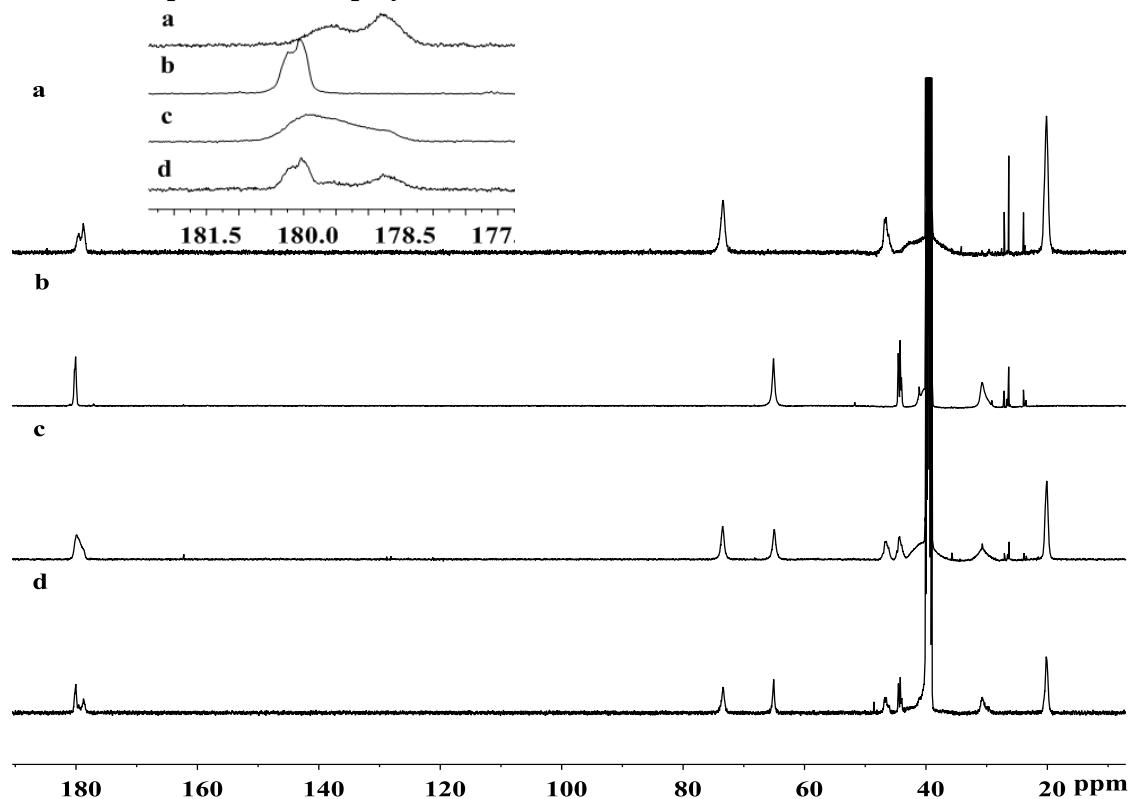


**Figure S19.**  $^{19}F$  NMR spectrum (benzene- $d_6$ , 471 MHz) of the reaction with  $i^3\text{BuSKA}/B(C_6F_5)_3\text{-MMA}$ =1:1 ratio at RT.



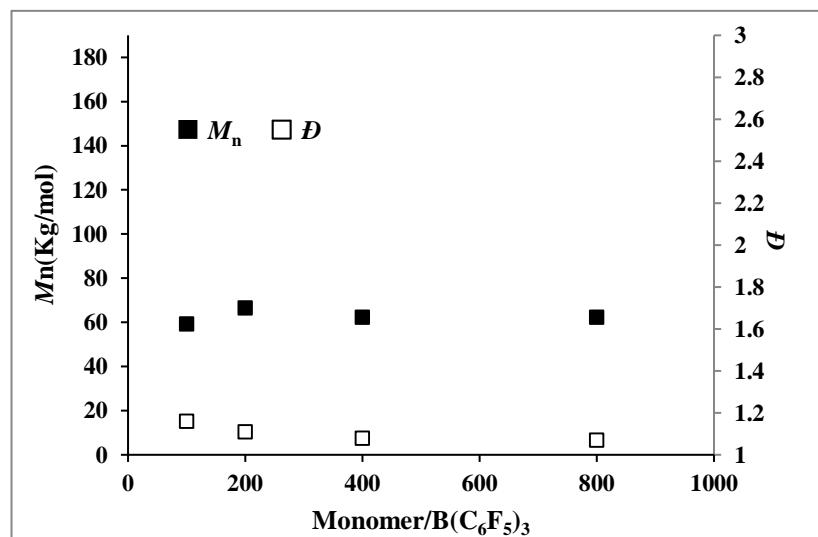
**Figure S20.**  $^{13}C$  NMR spectrum (benzene- $d_6$ , 126 MHz) of the reaction with  $i^3\text{BuSKA}/B(C_6F_5)_3\text{-MMA}$ =1:1ratio at RT.

**9.  $^{13}\text{C}$  NMR spectrum of (co)polymers.**



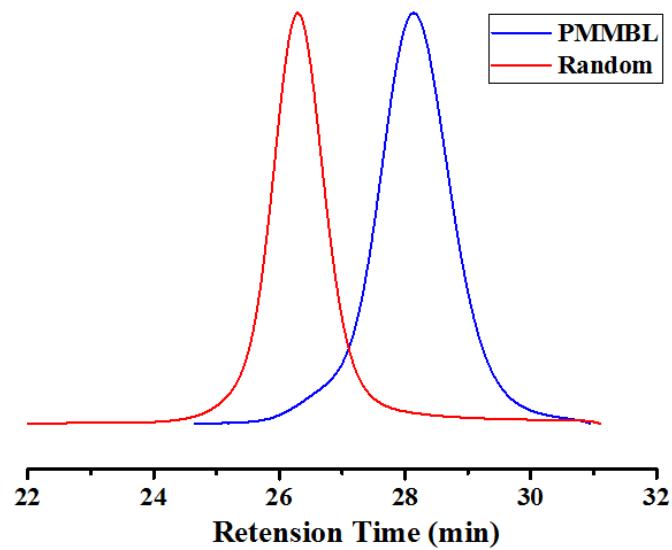
**Figure S21.**  $^{13}\text{C}$  NMR spectrum ( $\text{DMSO}-d_6$ , 126 MHz) of (a) PMMBL, (b) PMLB, (c) random PMMBL-*r*-PMLB, (d) diblock PMMBL-*b*-PMLB and enlarged carbonyl signals (inset).

**10. Plots of  $M_n$  and  $D$  values of PMMBL samples vs  $[\text{MMBL}]_0/[\text{B}(\text{C}_6\text{F}_5)_3]_0$  ratio**



**Figure S22.** Plots of  $M_n$  and  $D$  values of PMMBL samples vs  $[\text{MMBL}]_0/[\text{B}(\text{C}_6\text{F}_5)_3]_0$  ratio at RT. Condition:  $[\text{MMBL}]_0/[{}^{\text{i}}\text{BuSKA}]_0/[\text{B}(\text{C}_6\text{F}_5)_3]_0 = 400:1:0.5, 400:1:1, 400:1:2, 400:1:4$ , R.T.  $[\text{MMBL}]_0 = 0.936\text{M}$ .

### 11. The GPC traces of PMMBL-*r*-PMBL



**Figure S23.** The GPC traces of homopolymer PMMBL (blue), and PMMBL-*r*-PMBL (red) produced by *i*BuSKA/B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> at RT.