
Supporting information

**Zn(OAc)₂-Catalyzing Ring-Opening Polymerization of *N*-
Carboxy-Anhydrides for Synthesis of Well-Defined Polypeptides**

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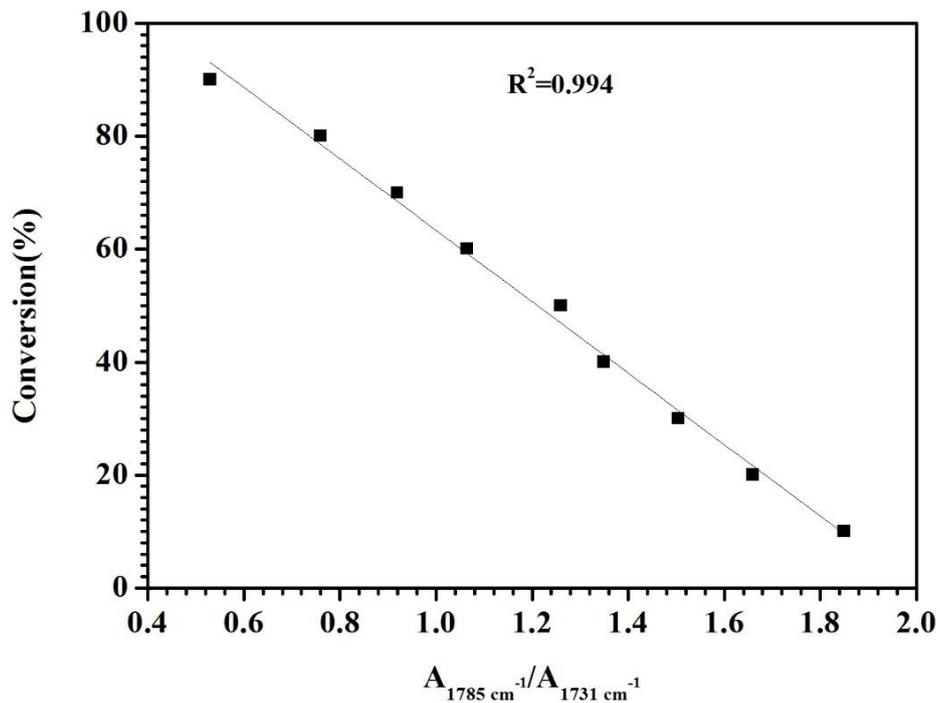
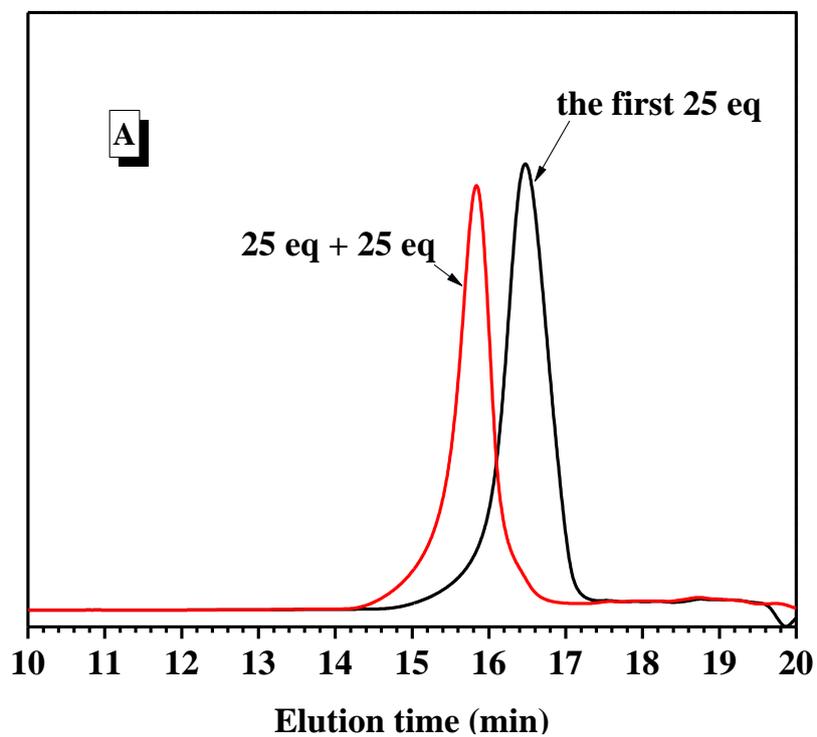


Figure S1. Calibrated curve of BLG-NCA conversion vs the peak intensity ratio at 1785 cm^{-1} and 1731 cm^{-1} .



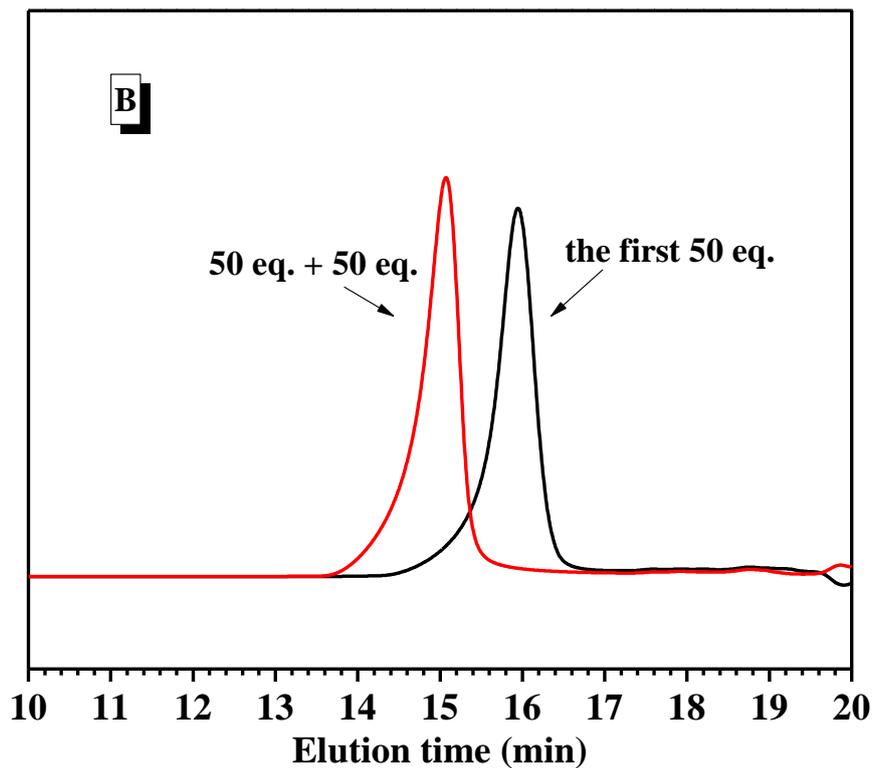


Figure S2. GPC curves of PBLG prepared in the sequential addition. (A) 25/25; (B) 50/50. $[\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}]/[\text{aniline}] = 1/1$, $[\text{BLG-NCA}] = 0.75 \text{ M}$, at $25 \text{ }^\circ\text{C}$ in CH_2Cl_2 .

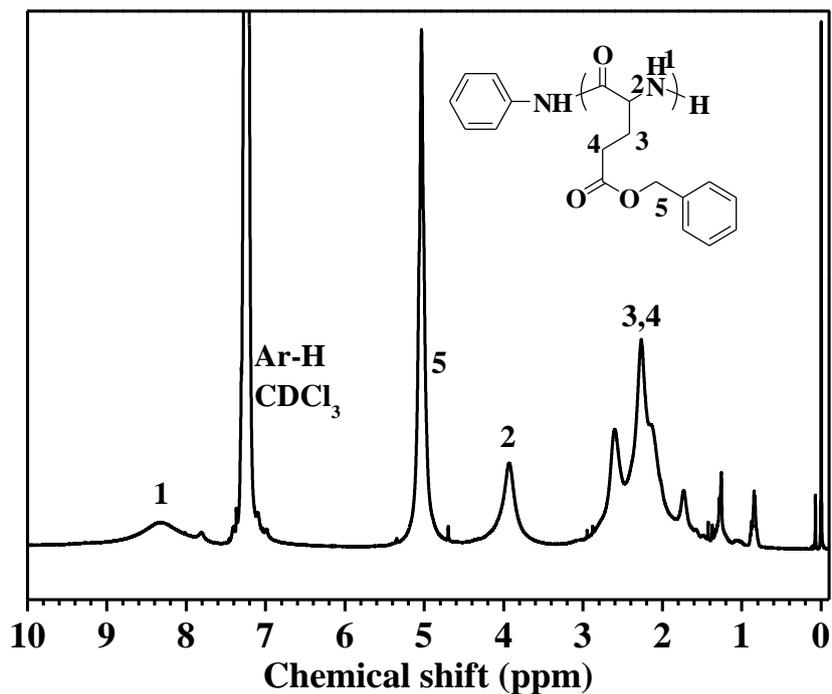


Figure S3. ^1H NMR spectrum of PBLG catalyzed by Lewis pair of $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$ and aniline

Table S1. Polymerization results of BLG-NCA catalyzed by various aniline analogues without $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$ in CH_2Cl_2 .^a

Run	Ana	Ana:M	Time (h) ^b	$M_{n,\text{cal}} \times 10^{-4}$ ^c	$M_{n,\text{mea}} \times 10^{-4}$ ^d	\mathcal{D} ^d
1	Ana-1	1:50	1.5	1.10	2.72	1.28
2	Ana -2	1:50	4.0	1.10	6.00	1.42
3	Ana -3	1:50	1.0	1.10	1.32	1.32
4	Ana -4	1:50	2.5	1.10	3.31	1.54
5	Ana -5	1:50	2.0	1.10	1.50	1.38
6	Ana -6	1:50	7.0	1.10	1.93	1.63
7	Ana -7	1:25	3.5	0.57	2.97	1.38
8	Ana -8	1:50	2.5	1.10	1.43	1.42
9	Ana -9	1:25	1.0	0.57	2.39	1.33
10	Ana -10	1:50	1.0	1.10	4.29	1.48

^a Performed by at 25 °C. ^b The polymerization time for 99% monomer conversion. ^c

Calculated by $([\text{Ana}]-1)+[\text{BLG-NCA}]/[\text{Ana}] \times (M_{\text{NCA}}-44) \times \text{monomer conversion}$. ^d Determined

by GPC, \mathcal{D} represents molecular weight distribution.

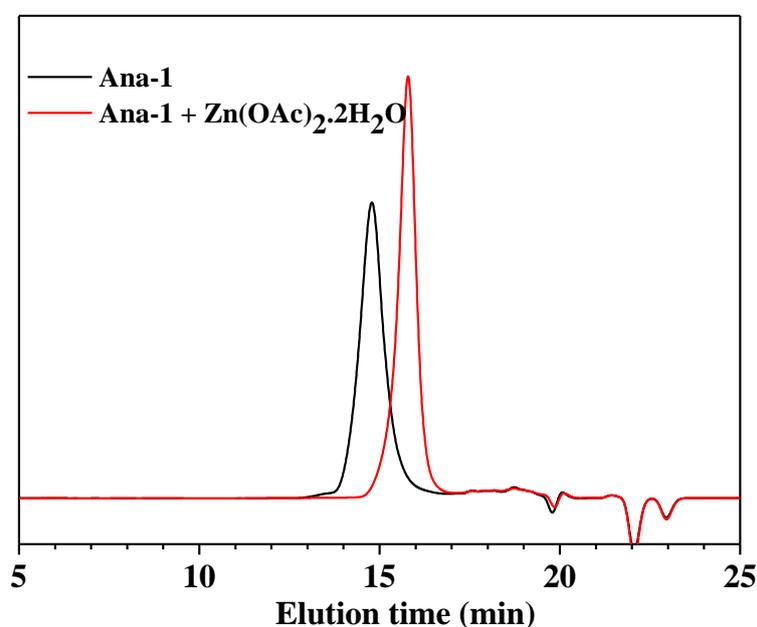


Figure S4. GPC profiles of PBLG initiated by Ana-1 with or without $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$. $[\text{BLG-NCA}]/[\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}]/[\text{Ana -1}] = 50/1/1$, $[\text{BLG-NCA}] = 0.75 \text{ M}$, at 25 °C in CH_2Cl_2 .

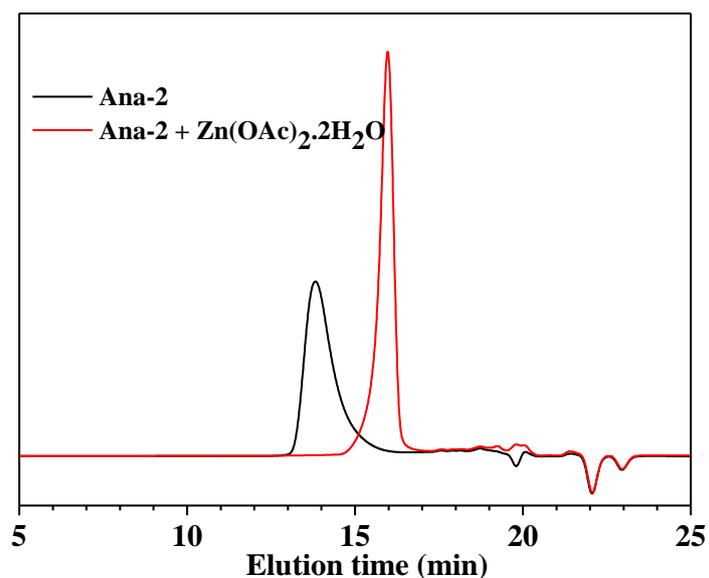


Figure S5. GPC profiles of PBLG initiate by Ana-2 with or without Zn(OAc)₂·2H₂O. [BLG-NCA]/[Zn(OAc)₂·2H₂O]/[Ana-2] = 50/1/1, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

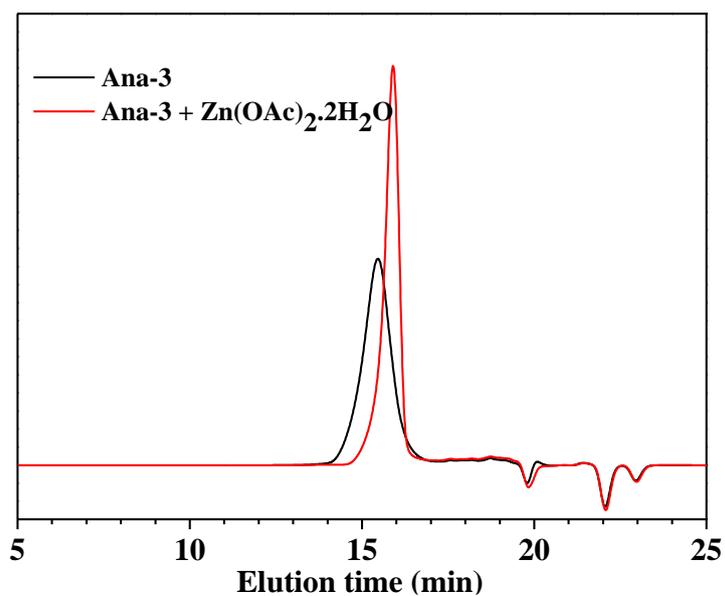


Figure S6. GPC profiles of PBLG initiate by Ana-3 with or without Zn(OAc)₂·2H₂O. [BLG-NCA]/[Zn(OAc)₂·2H₂O]/[Ana-2] = 50/1/1, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

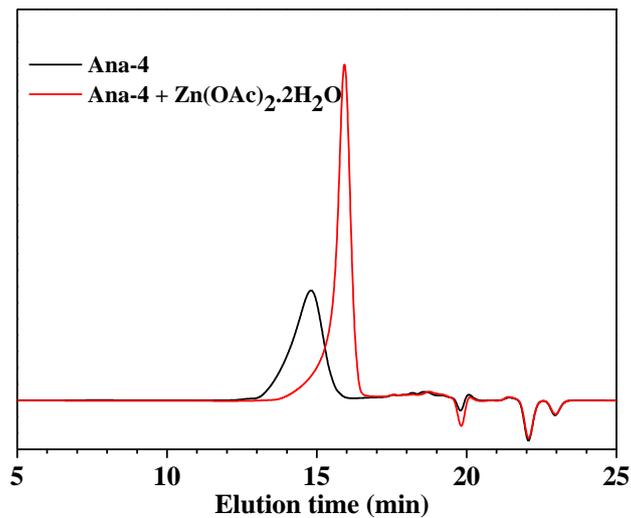


Figure S7. GPC profiles of PBLG initiate by Ana-4 with or without Zn(OAc)₂·2H₂O. [BLG-NCA]/[Zn(OAc)₂·2H₂O]/[Ana-4] = 50/1/1, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

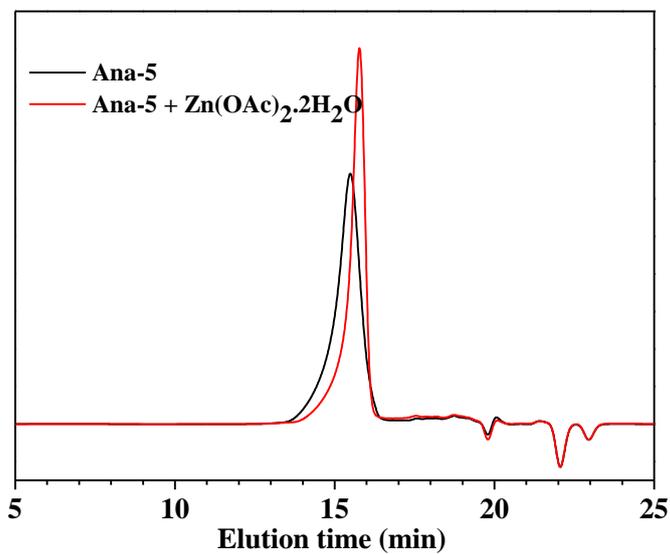


Figure S8. GPC profiles of PBLG initiate by Ana-5 with or without Zn(OAc)₂·2H₂O. [BLG-NCA]/[Zn(OAc)₂·2H₂O]/[Ana-5] = 50/1/1, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

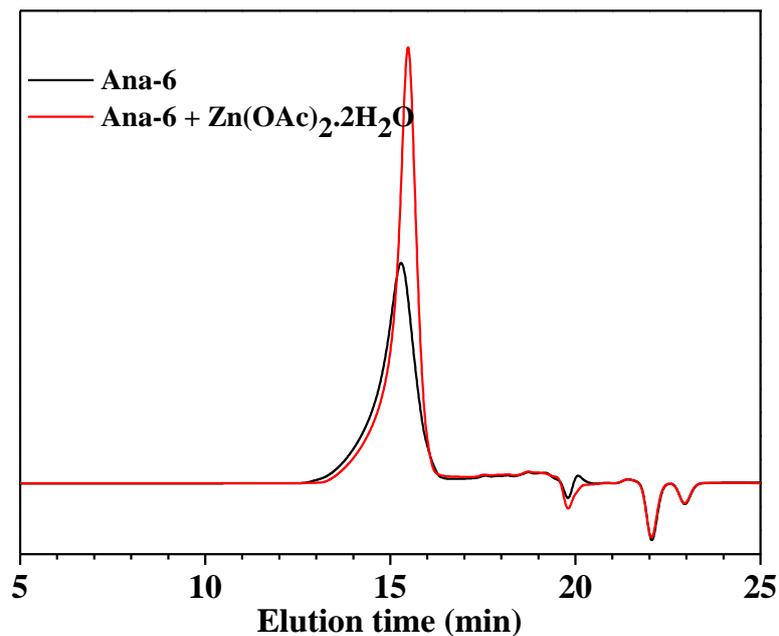


Figure S9. GPC profiles of PBLG initiate by Ana-6 with or without $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$. $[\text{BLG-NCA}]/[\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}]/[\text{Ana-6}] = 50/1/1$, $[\text{BLG-NCA}] = 0.75 \text{ M}$, at $25 \text{ }^\circ\text{C}$ in CH_2Cl_2 .

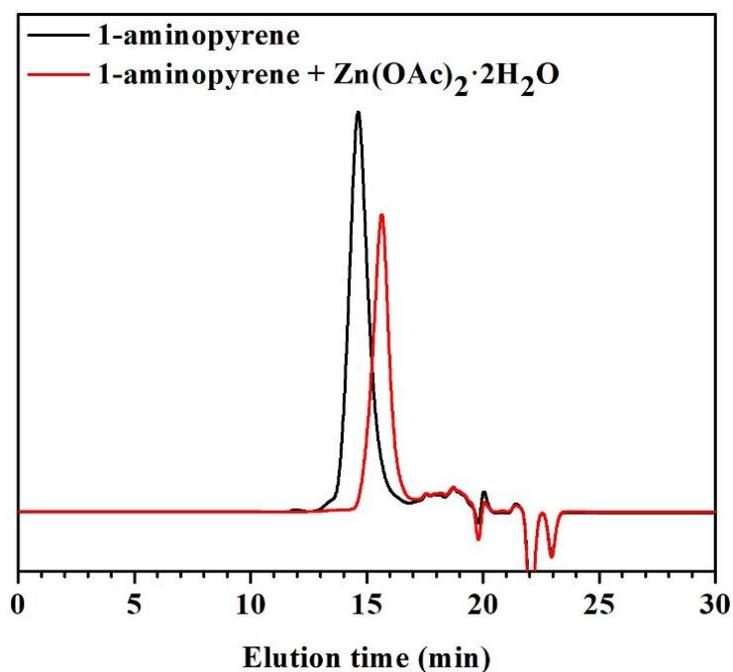


Figure S10. GPC curves of PBLG with Ana-7 with or without $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$. $[\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}]/[\text{Ana-7}]/[\text{BLG-NCA}] = 1/1/25$, $[\text{BLG-NCA}] = 0.75 \text{ M}$, at $25 \text{ }^\circ\text{C}$ in CH_2Cl_2 .

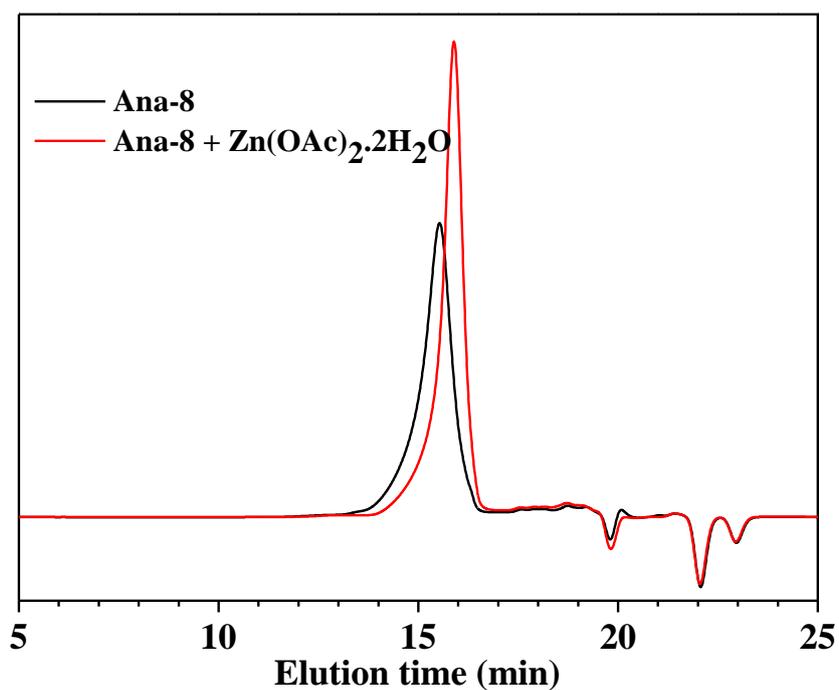


Figure S11. GPC curves of PBLG with Ana-8 with or without Zn(OAc)₂·2H₂O. [Zn(OAc)₂·2H₂O]/[Ana-8]/[BLG-NCA] = 1/1/25, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

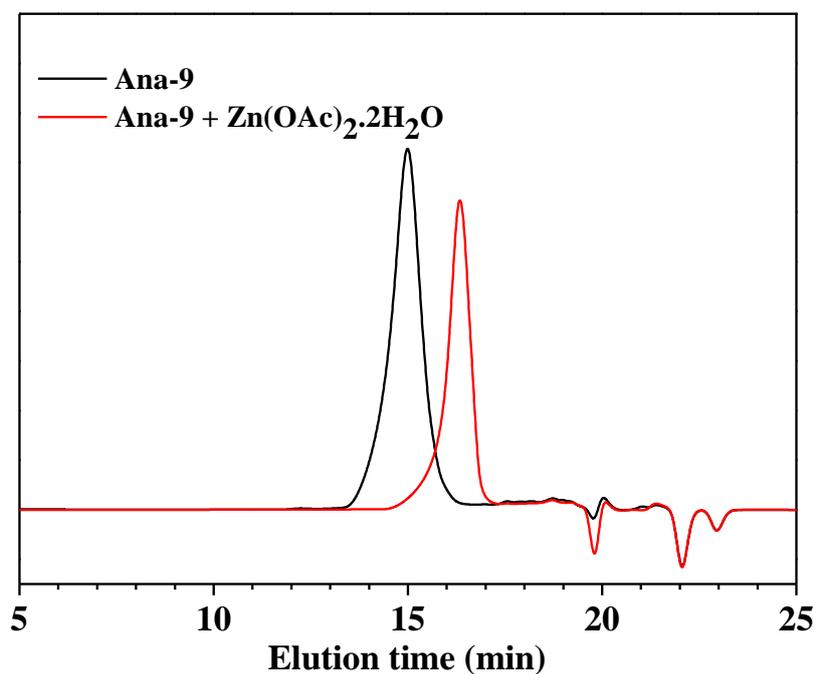


Figure S12. GPC curves of PBLG with Ana-9 with or without Zn(OAc)₂·2H₂O. [Zn(OAc)₂·2H₂O]/[Ana-9]/[BLG-NCA] = 1/1/25, [BLG-NCA] = 0.75 M, at 25 °C in CH₂Cl₂.

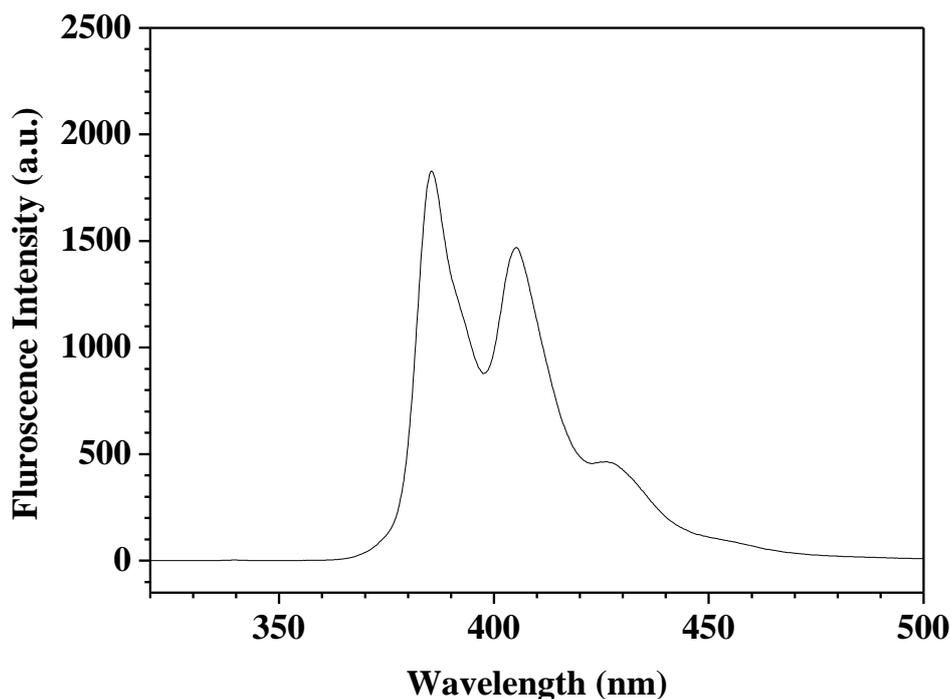
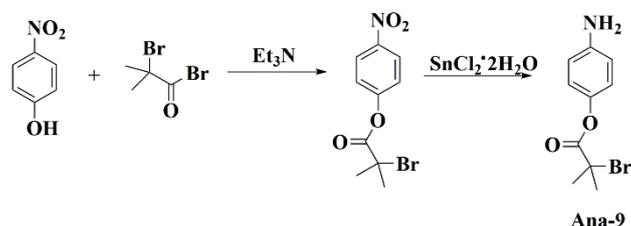


Figure S13. Fluorescent spectrum of PBLG initiated by a combination of $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$ with 1-aminopyrene.



Scheme S1. Synthesis route of Ana-9.

The synthesis of Ana-9.

4-Nitrophenol (2.78 g, 0.02 mol), triethylamine (4.04 g, 0.04 mol) and THF 200 mL were placed in one three-neck round bottomed flask. Bromoisobutyryl bromide (4.28 g, 0.02 mol) was added slowly with stirring. After 6 hours, the reaction was filtered and THF was removed in vacuum to obtain 2-bromo-2-methylpropionic acid 4-nitrophenyl ester. The 2-bromo-2-methylpropionic acid 4-nitrophenyl ester (1.44 g, 0.005 mol) and $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$ (0.025 mol) were dissolved in ethyl acetate (100 mL). The mixture was heated under reflux for 1 h at 80 °C, cooled, and made basic (pH 8-9) using 5% sodium bicarbonate aqueous solution. Distilled water (200 mL) was added and the ethyl acetate layer separated. The organic layer was washed with saturated brine solution (3 × 100 mL) followed by distilled water (2 × 100 mL). The organic layer was dried with magnesium sulfate, and the solvent was removed in vacuo. This gave a slightly brown crystalline product **9**.

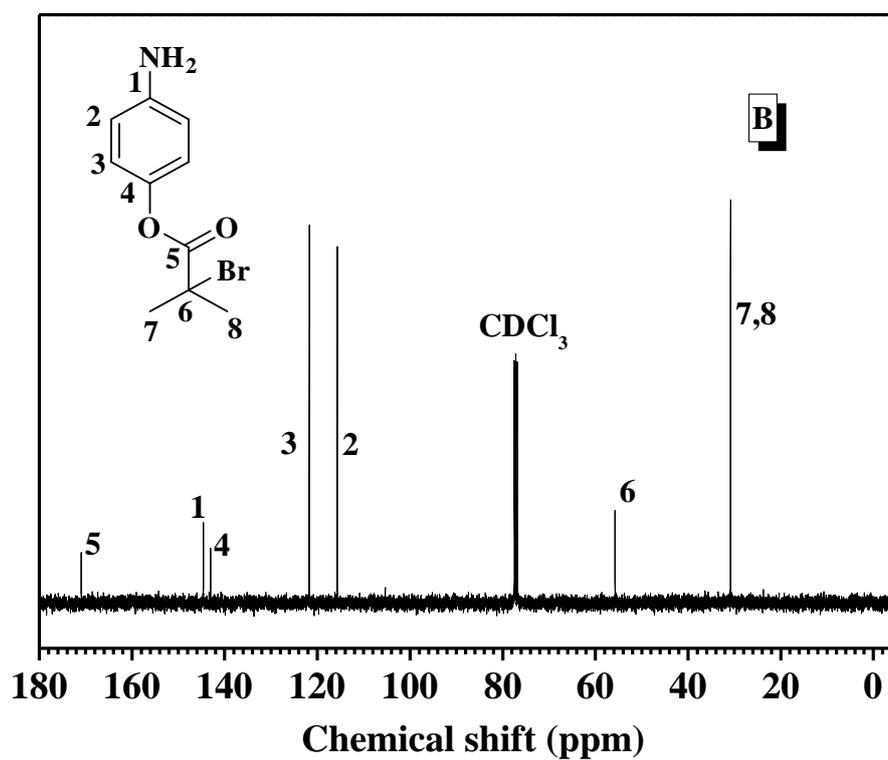
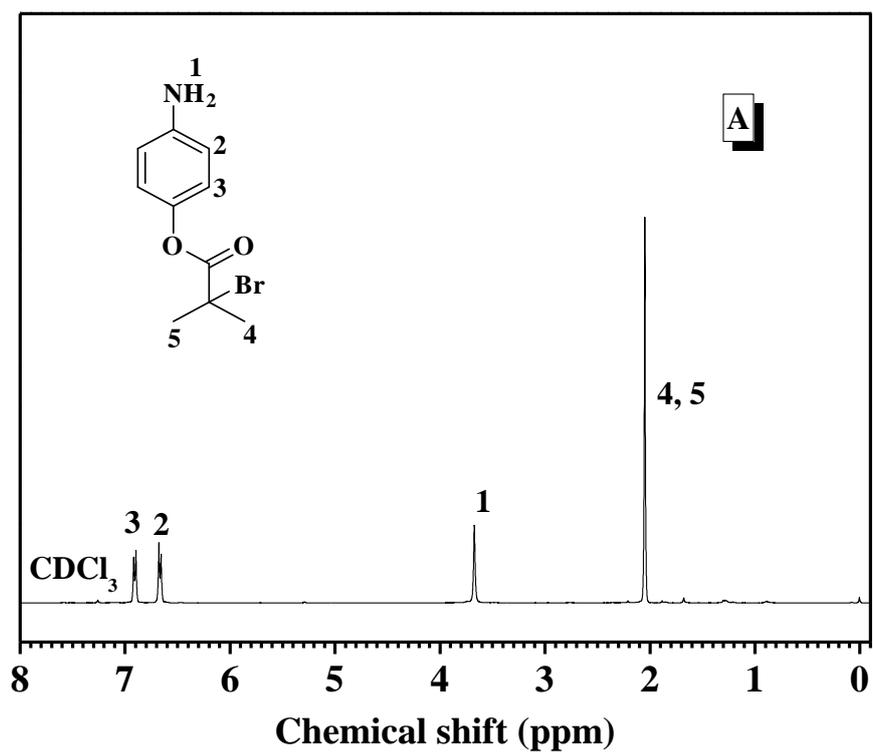


Figure S14. ^1H NMR (A) and ^{13}C NMR (B) spectra of 2-bromo-2-methylpropionic acid 4-nitrophenyl ester.

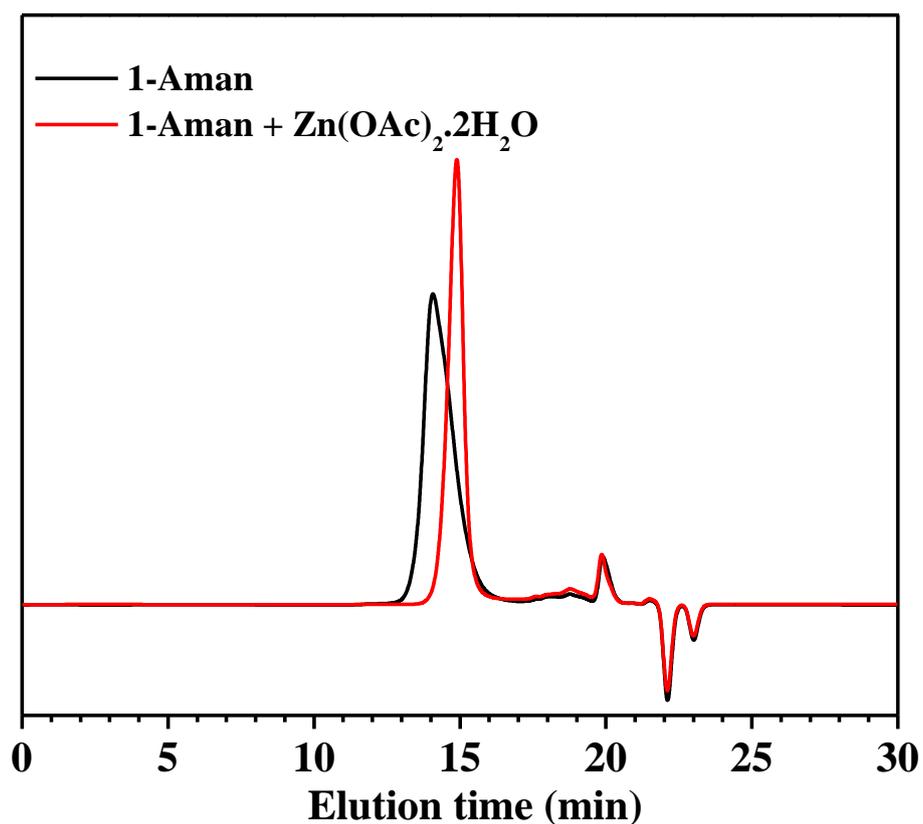


Figure S15. GPC profiles of PBLG initiate by Ana-10 with or without $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$. $[\text{BLG-NCA}]/[\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}]/[\text{Ana-10}] = 50/1/1$, $[\text{BLG-NCA}] = 0.75 \text{ M}$, at 25°C in CH_2Cl_2 .

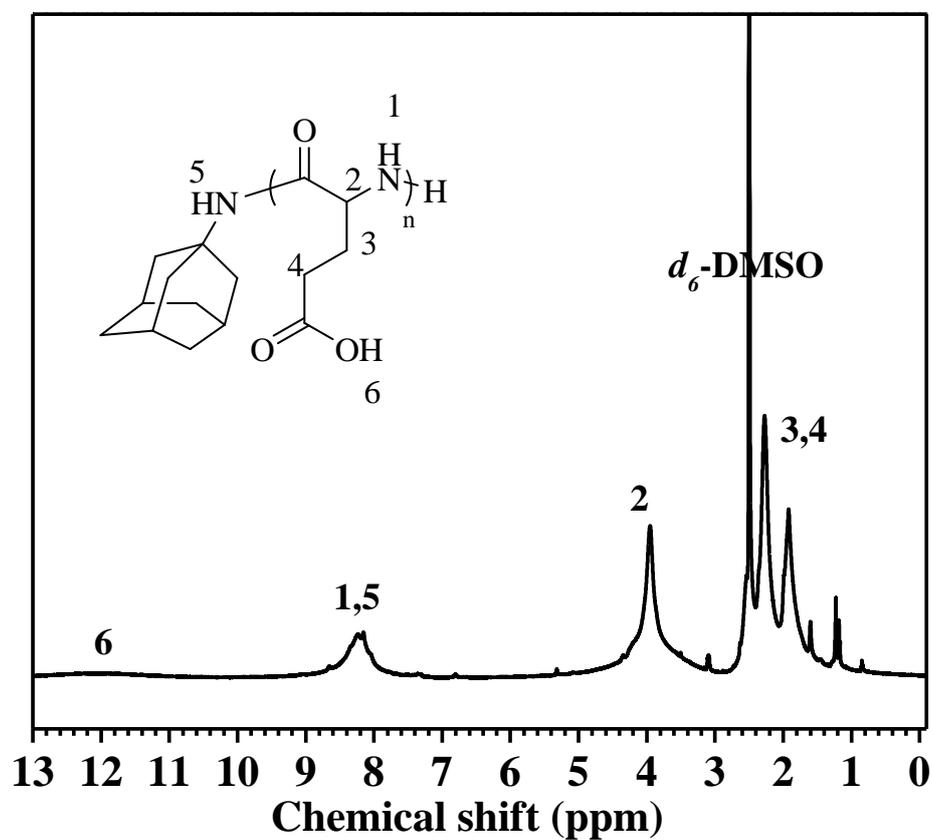


Figure S16. ^1H NMR spectrum of Aman-capped PLG.