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# A Self-assembled Zn<sup>II</sup>-Nd<sup>III</sup> Heterohexanuclear Dimer Based on a Hexadentate N<sub>2</sub>O<sub>4</sub>-type Ligand and Terephthalic Acid: Synthesis, Structure, and Fluorescence Properties

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## Supporting Information

**Table S1** Selected bond lengths (Å) and angles (°) for the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

**Table S2** Hydrogen bonding interactions [Å, deg] for the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

**Fig. S1.** View of the dihedral angles between the benzene rings of terephthalic acid and the basal planes (N<sub>2</sub>O<sub>2</sub> planes) of the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

**Fig. S2.** Intermolecular hydrogen bonding interactions of the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound (hydrogen atoms, except those forming hydrogen bonds, are omitted for clarity).

**Fig. S3.** View of the 1D supramolecular structure of the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound showing the C-H ··· O hydrogen bondings.

**Fig. S4.** IR spectra of H<sub>2</sub>L and its corresponding Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

**Fig. S5.** UV/Vis absorption spectra of H<sub>2</sub>L and its Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

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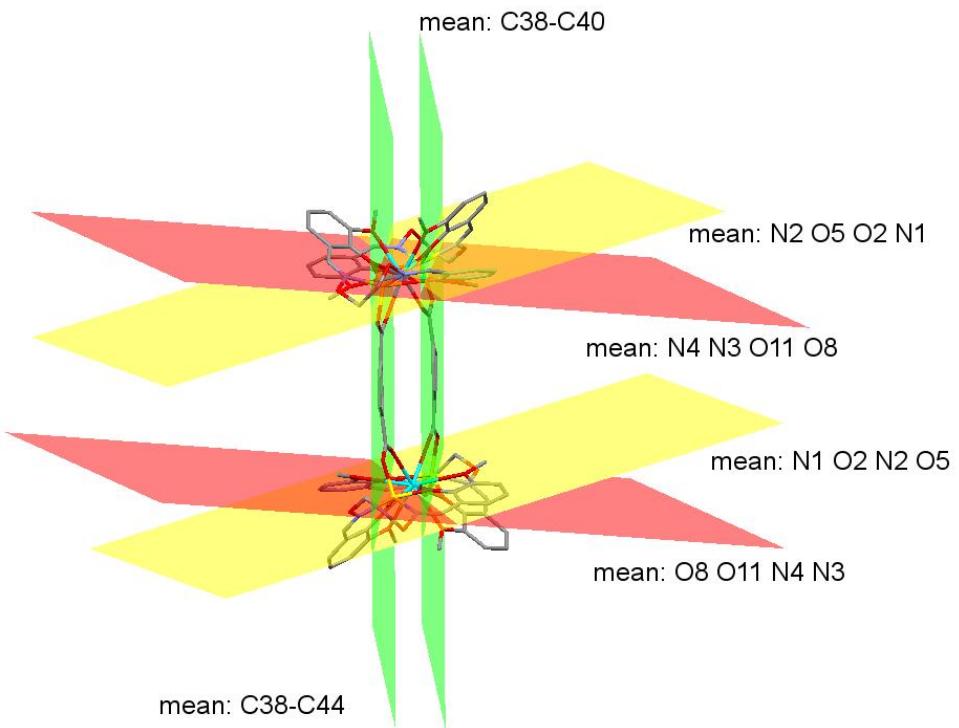
**Table S1** Selected bond lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for the  $\text{Zn}^{\text{II}}\text{-Nd}^{\text{III}}$  coordination compound.

Bonds Lengths ( $\text{\AA}$ )	Bonds Lengths ( $\text{\AA}$ )	Bonds Lengths ( $\text{\AA}$ )
Zn(1)-O(8) <b>2.014(3)</b>	Zn(1)-O(11) <b>2.062(3)</b>	Zn(1)-O(16) <b>1.988(3)</b>
Zn(1)-N(3) <b>2.043(4)</b>	Zn(1)-N(4) <b>2.139(4)</b>	Zn(2)-N(1) <b>2.119(4)</b>
Zn(2)-N(2) <b>2.043(4)</b>	Zn(2)-O(2) <b>2.004(3)</b>	Zn(2)-O(5) <b>2.067(3)</b>
Zn(2)-O(13) <b>1.979(3)</b>	Nd(1)-O(15) <b>2.422(3)</b>	Nd(1)-O(14) <b>2.443(3)</b>
Nd(1)-O(2) <b>2.454(3)</b>	Nd(1)-O(11) <b>2.466(3)</b>	Nd(1)-O(8) <b>2.454(3)</b>
Nd(1)-O(5) <b>2.479(3)</b>	Nd(1)-O(1) <b>2.647(3)</b>	Nd(1)-O(7) <b>2.683(3)</b>
Nd(1)-O(6) <b>2.781(3)</b>	Nd(1)-O(12) <b>2.802(3)</b>	
Angles ( $^\circ$ )	Angles ( $^\circ$ )	Angles ( $^\circ$ )
O(16)-Zn(1)-O(8) <b>112.59(12)</b>	O(16)-Zn(1)-N(3) <b>120.06(14)</b>	O(8)-Zn(1)-N(3) <b>126.84(13)</b>
O(16)-Zn(1)-O(11) <b>97.99(11)</b>	O(8)-Zn(1)-O(11) <b>126.84(13)</b>	N(3)-Zn(1)-O(11) <b>86.75(13)</b>
O(16)-Zn(1)-N(4) <b>97.95(17)</b>	O(8)-Zn(1)-N(4) <b>86.75(13)</b>	N(3)-Zn(1)-N(4) <b>92.70(15)</b>
O(11)-Zn(1)-N(4) <b>161.97(14)</b>	O(13)-Zn(2)-O(2) <b>111.87(12)</b>	O(13)-Zn(2)-N(2) <b>117.47(15)</b>
O(2)-Zn(2)-N(2) <b>130.04(14)</b>	O(13)-Zn(2)-O(5) <b>97.38(12)</b>	O(2)-Zn(2)-O(5) <b>79.50(11)</b>
N(2)-Zn(2)-O(5) <b>86.84(15)</b>	O(13)-Zn(2)-N(1) <b>98.33(14)</b>	O(2)-Zn(2)-N(1) <b>87.43(13)</b>
N(2)-Zn(2)-N(1) <b>92.79(16)</b>	O(5)-Zn(2)-N(1) <b>162.46(13)</b>	O(15)-Nd(1)-O(14) <b>73.41(9)</b>
O(15)-Nd(1)-O(2) <b>150.62(9)</b>	O(14)-Nd(1)-O(2) <b>77.31(9)</b>	O(15)-Nd(1)-O(11) <b>70.90(9)</b>
O(14)-Nd(1)-O(11) <b>107.98(9)</b>	O(2)-Nd(1)-O(11) <b>117.05(9)</b>	O(15)-Nd(1)-O(8) <b>77.39(9)</b>
O(14)-Nd(1)-O(8) <b>150.70(10)</b>	O(2)-Nd(1)-O(8) <b>131.95(9)</b>	O(11)-Nd(1)-O(8) <b>63.62(10)</b>
O(15)-Nd(1)-O(5) <b>107.68(10)</b>	O(14)-Nd(1)-O(5) <b>70.88(10)</b>	O(2)-Nd(1)-O(5) <b>63.87(10)</b>
O(11)-Nd(1)-O(5) <b>178.47(9)</b>	O(8)-Nd(1)-O(5) <b>116.84(10)</b>	O(15)-Nd(1)-O(1) <b>139.04(9)</b>
O(14)-Nd(1)-O(1) <b>125.60(9)</b>	O(2)-Nd(1)-O(1) <b>60.44(8)</b>	O(11)-Nd(1)-O(1) <b>68.64(9)</b>
O(8)-Nd(1)-O(1) <b>79.32(9)</b>	O(5)-Nd(1)-O(1) <b>112.83(9)</b>	O(15)-Nd(1)-O(7) <b>125.36(10)</b>
O(14)-Nd(1)-O(7) <b>140.48(9)</b>	O(2)-Nd(1)-O(7) <b>80.01(9)</b>	O(11)-Nd(1)-O(7) <b>111.09(9)</b>
O(8)-Nd(1)-O(7) <b>59.18(9)</b>	O(5)-Nd(1)-O(7) <b>70.15(9)</b>	O(1)-Nd(1)-O(7) <b>66.12(9)</b>
O(15)-Nd(1)-O(6) <b>66.54(11)</b>	O(14)-Nd(1)-O(6) <b>96.94(10)</b>	O(2)-Nd(1)-O(6) <b>120.23(10)</b>
O(11)-Nd(1)-O(6) <b>121.04(9)</b>	O(8)-Nd(1)-O(6) <b>68.70(9)</b>	O(5)-Nd(1)-O(6) <b>58.40(10)</b>
O(1)-Nd(1)-O(6) <b>132.67(9)</b>	O(7)-Nd(1)-O(6) <b>67.69(10)</b>	O(15)-Nd(1)-O(12) <b>97.33(10)</b>
O(14)-Nd(1)-O(12) <b>67.57(9)</b>	O(2)-Nd(1)-O(12) <b>68.80(9)</b>	O(11)-Nd(1)-O(12) <b>58.03(8)</b>
O(8)-Nd(1)-O(12) <b>119.41(9)</b>	O(5)-Nd(1)-O(12) <b>121.95(9)</b>	O(1)-Nd(1)-O(12) <b>66.18(9)</b>
O(7)-Nd(1)-O(12) <b>131.35(9)</b>	O(6)-Nd(1)-O(12) <b>160.92(9)</b>	

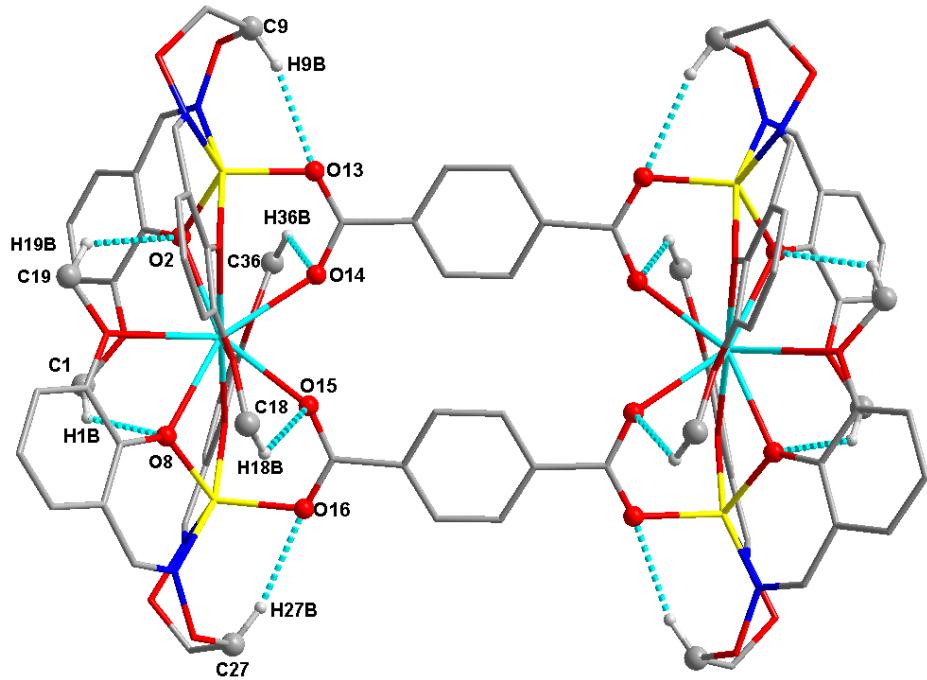
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**Table S2** Hydrogen bonding interactions [Å, deg] for the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound.

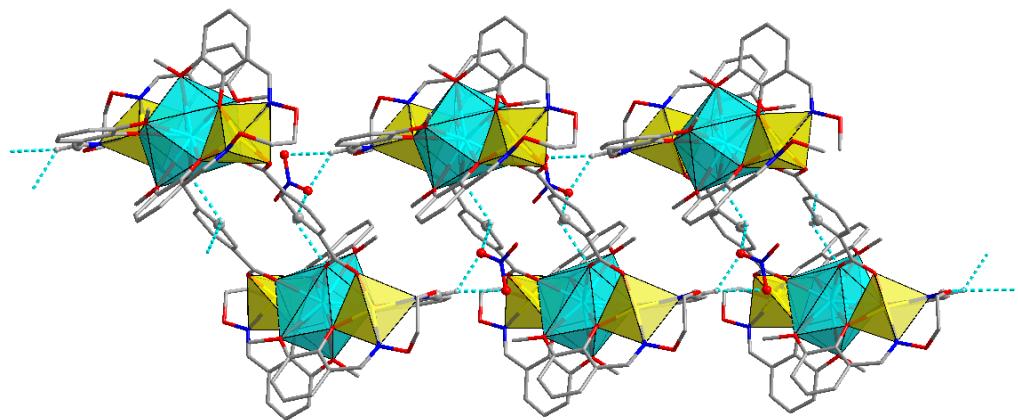
D-H ·· A	d(D-H)	d(H ·· A)	d(D ·· A)	∠DHA
C1-H1B ·· O8	0.98	2.58	3.328(5)	133
C9-H9B ·· O13	0.99	2.39	3.315(6)	155
C18-H18B ·· O15	0.98	2.51	3.095(7)	118
C19-H19B ·· O2	0.98	2.55	3.319(6)	135
C27-H27B ·· O16	0.99	2.41	3.313(5)	152
C36-H36B ·· O14	0.98	2.50	3.127(6)	122
C8-H8 ·· O17	0.95	2.07	2.822(10)	135
C8-H8 ·· O19	0.95	2.50	3.402(9)	159
C39-H39 ·· O17	0.95	2.58	3.317(10)	135



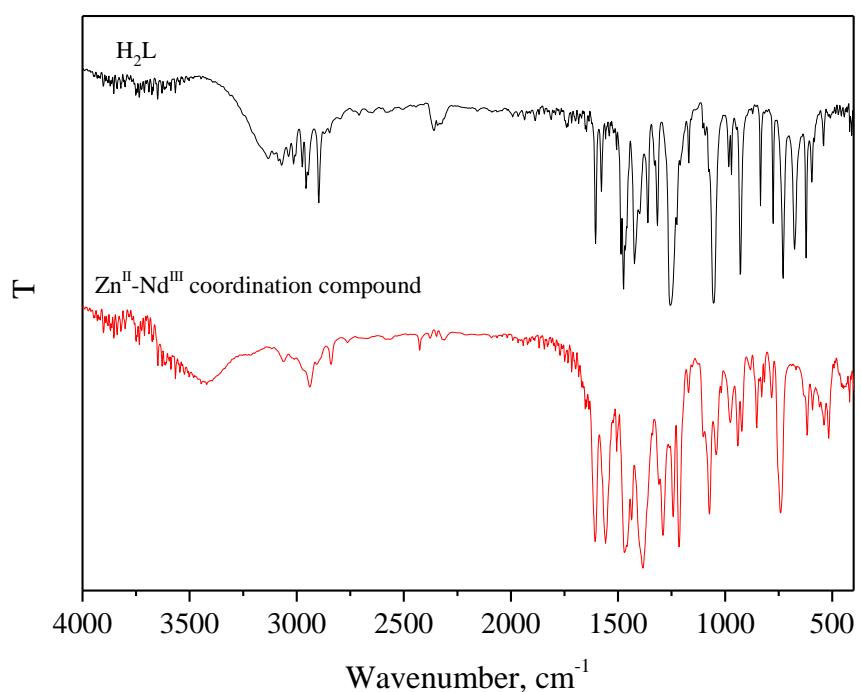
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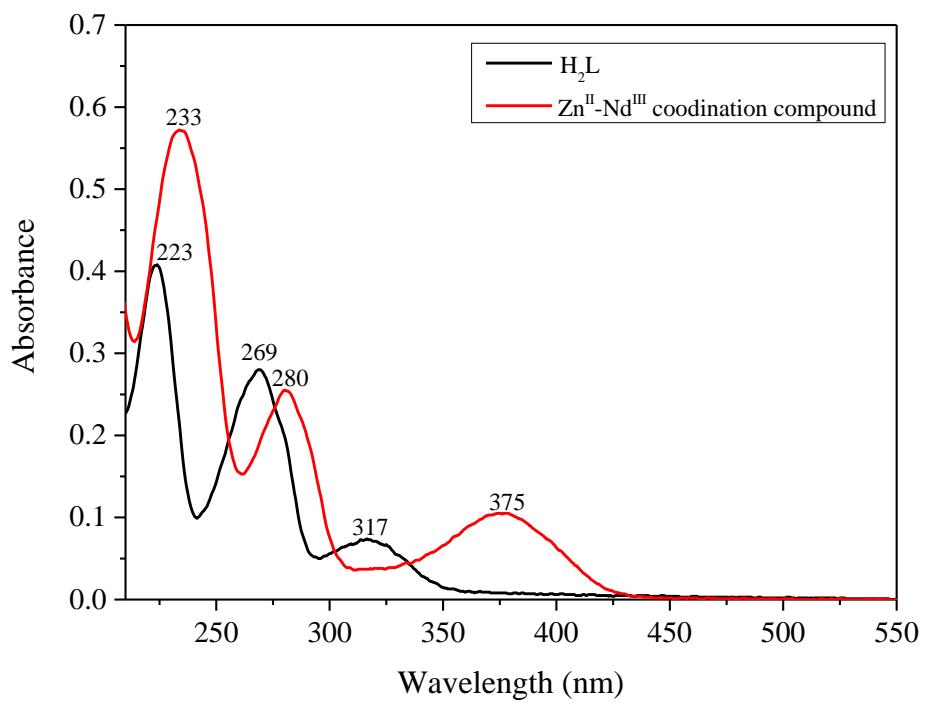
**Fig. S2.** Intermolecular hydrogen bonding interactions of the Zn<sup>II</sup>-Nd<sup>III</sup> coordination compound (hydrogen atoms, except those forming hydrogen bonds, are omitted for clarity).



**Fig. S3.** View of the 1D supramolecular structure of the  $\text{Zn}^{\text{II}}\text{-Nd}^{\text{III}}$  coordination compound showing the  $\text{C-H} \cdots \text{O}$  hydrogen bondings.



**Fig. S4.** IR spectra of  $\text{H}_2\text{L}$  and its corresponding  $\text{Zn}^{\text{II}}\text{-Nd}^{\text{III}}$  coordination compound.



**Fig. S5.** UV/Vis absorption spectra of  $\text{H}_2\text{L}$  and its  $\text{Zn}^{\text{II}}\text{-Nd}^{\text{III}}$  coordination compound.