

Supporting Information

Hydro/Deutero Deamination of Arylazo Sulfones under Metal- and (Photo)Catalyst-Free Conditions

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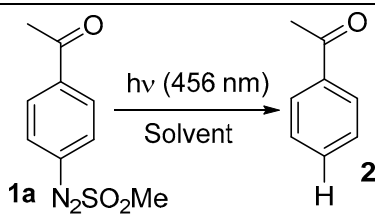
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1 Optimization of the procedure for the reductive dediazonation of arylazo sulfones.

Table S1. Irradiation of arylazosulfone **1a** in different solvents.^a

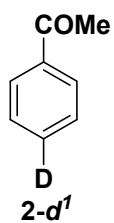
		
Entry	Conditions	2 (% yield)
1	1a (0.025 M), MeOH	48
2	1a (0.025 M), MeOH-H ₂ O 9:1	50
3	1a (0.025 M), MeOH-H ₂ O 4:1	51
4	1a (0.025 M), MeCN	30
5	1a (0.025 M), MeCN-H ₂ O 9:1	38
6	1a (0.025 M), MeCN-H ₂ O 4:1	43
7	1a (0.025 M), Acetone	54 ^b
8	1a (0.05 M), Acetone	49 ^b
9	1a (0.1 M), Acetone	34 ^b
10	1a (0.025 M), THF	59
11	1a (0.025 M), THF-H ₂ O 9:1	56
12	1a (0.025 M), THF-H ₂ O 4:1	61
13	1a (0.025 M), <i>i</i> PrOH-H ₂ O 9:1	76
14	1a (0.025 M), <i>i</i> PrOH-H ₂ O 4:1	71
15 ^c	1a (0.025 M), <i>i</i> PrOH-H ₂ O 9:1	^c

^a A solution of **1a** in the chosen medium (1 mL) was nitrogen purged for 5 min, then irradiated by means of a 34 W Kessil Lamp ($\lambda_{em} = 456$ nm). The reaction course was monitored by GC analyses and the amount of **2** quantified by a calibration curve. ^b 4-Acetylphenyl methylsulfone (<10%) detected by GC-MS ^c Blank experiment, no consumption of **1a** observed.

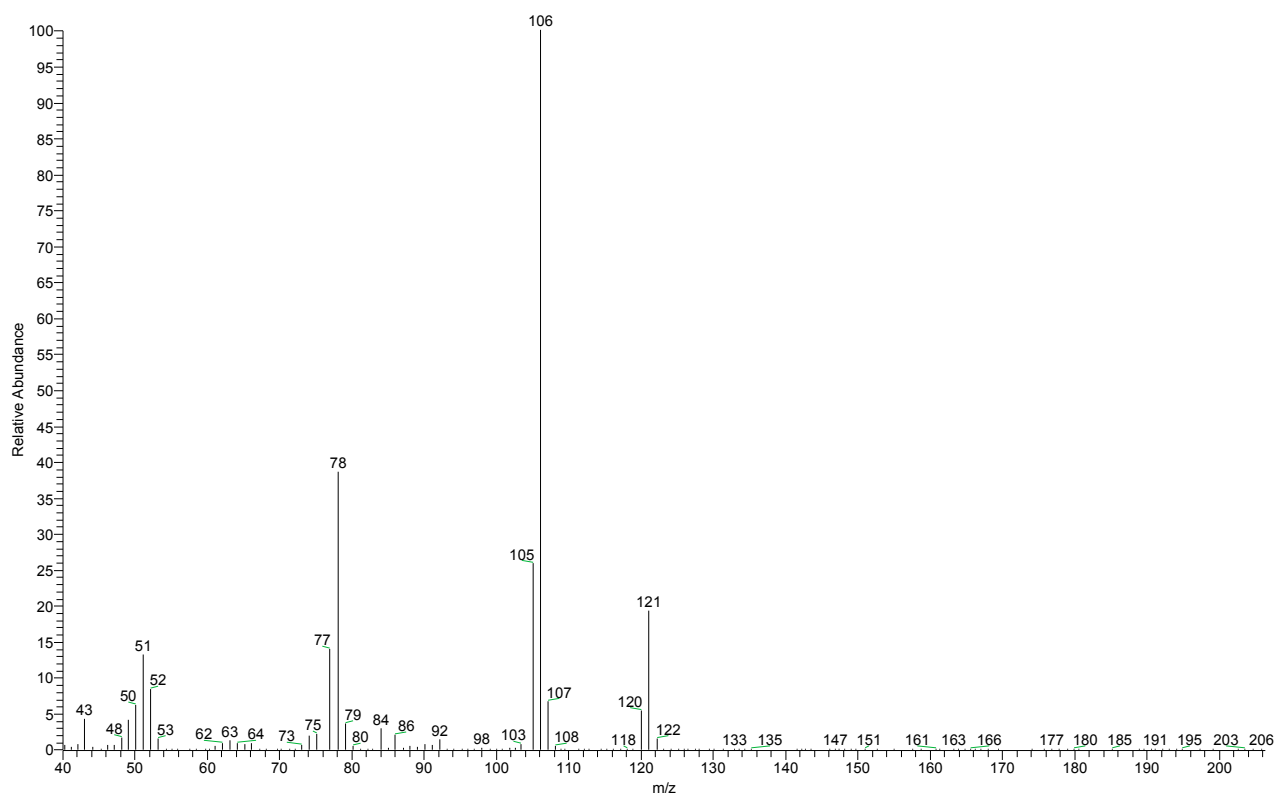
As depicted in Table S1, irradiation of a 0.025 M solution of **1a** in methanol as well as in different methanol-water mixtures (entries 1-3) afforded a moderate yield of acetophenone **2**. A similar behavior was observed in acetonitrile containing mixtures (entries 4-6), while when moving to acetone, a noxious effect on the yield of **2** was observed when increasing the concentration of **1a** (entries 7-9). Furthermore, small amounts (<10% yield) of the by-product 4-acetylphenyl methyl sulfone (GC-MS: 198 (6), 183 (100), 121 (54))^{S1} were detected. Better yields were obtained by choosing THF as the (co)solvent (up to 61% yield, entries 10-12). However, when shifting to aqueous isopropanol (entries 13,14), conversion of **1a** to **2** was found to occur in a satisfactory yield (76% in

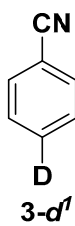
*i*PrOH-H₂O 9:1 mixture). Finally, no consumption of **1a** took place in the absence of irradiation (entry 15). The protocol described in entry 13 was thus adopted to investigate the versatility of the reaction.

2. MS spectra of compounds 2-*d*^I-14-*d*^I and ¹H and ¹³C NMR of 13-*d*^I

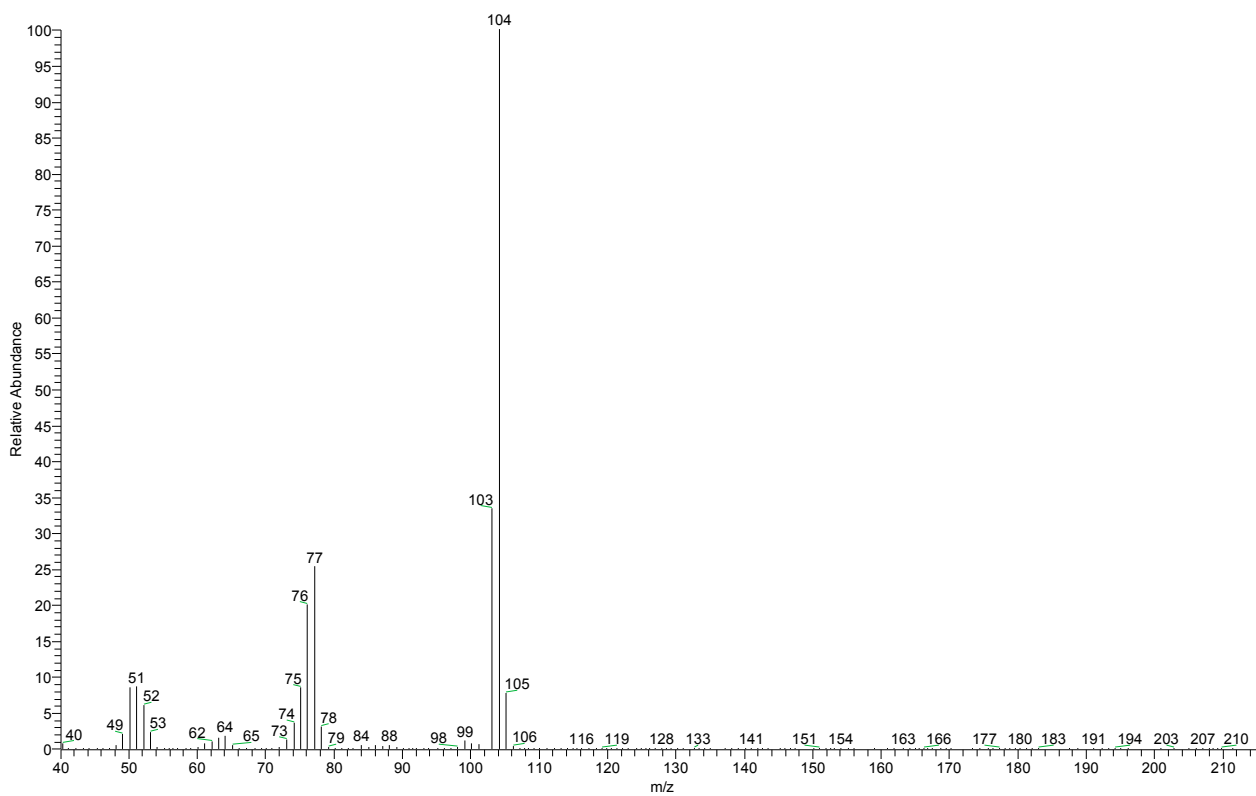


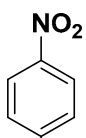
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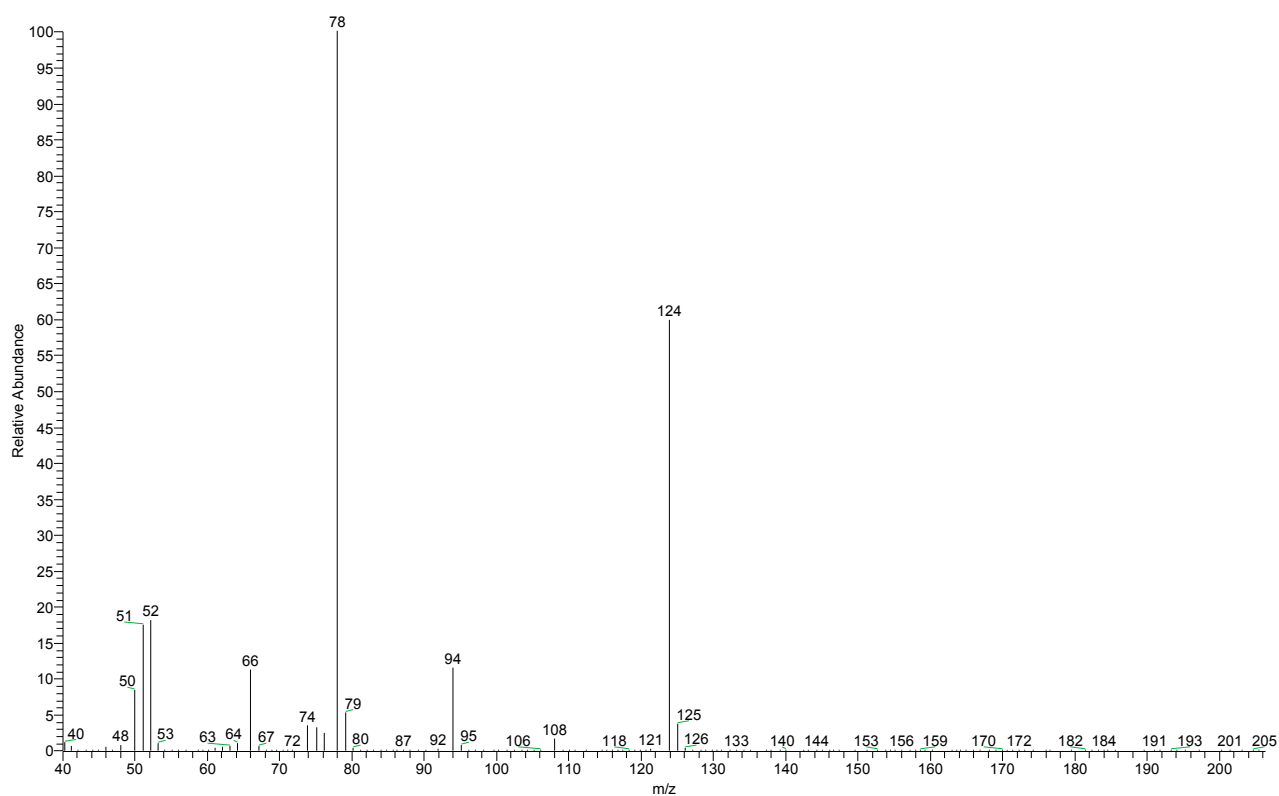
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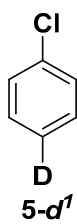




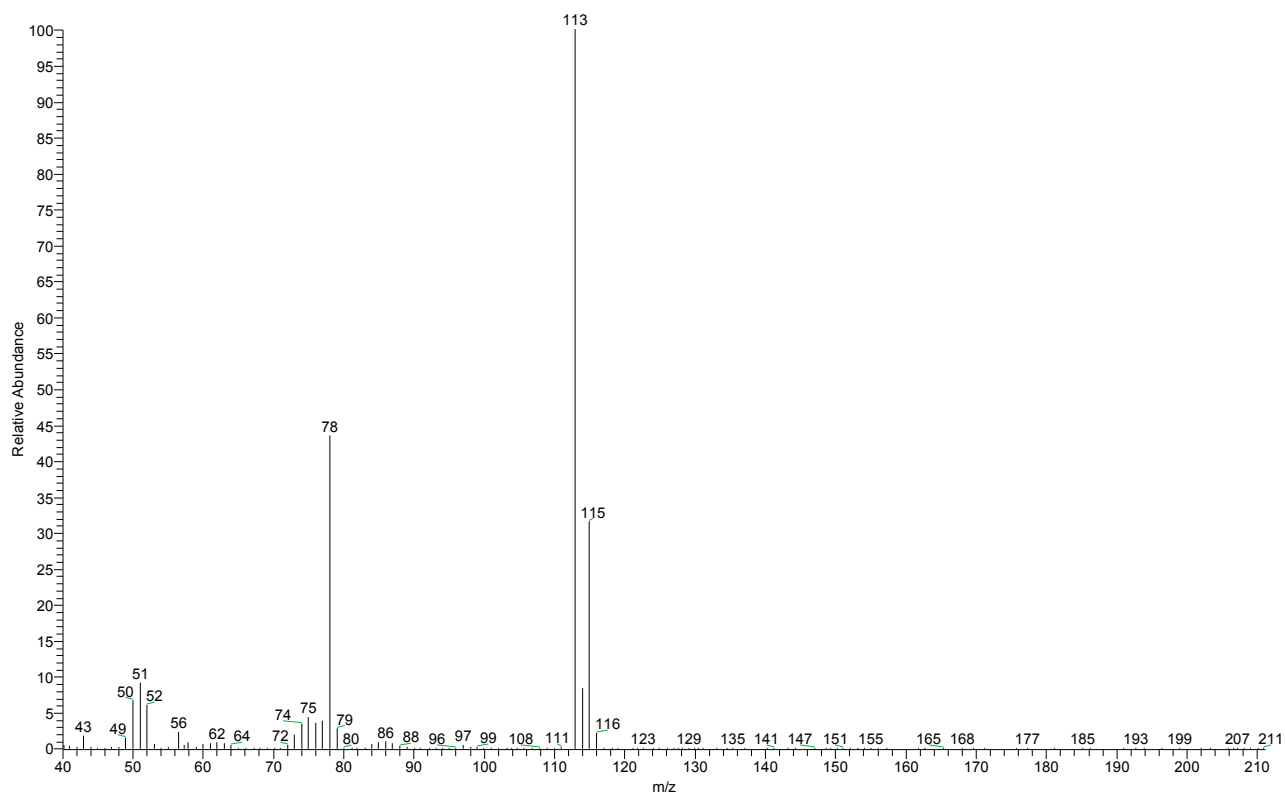
D
4-d¹

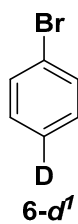
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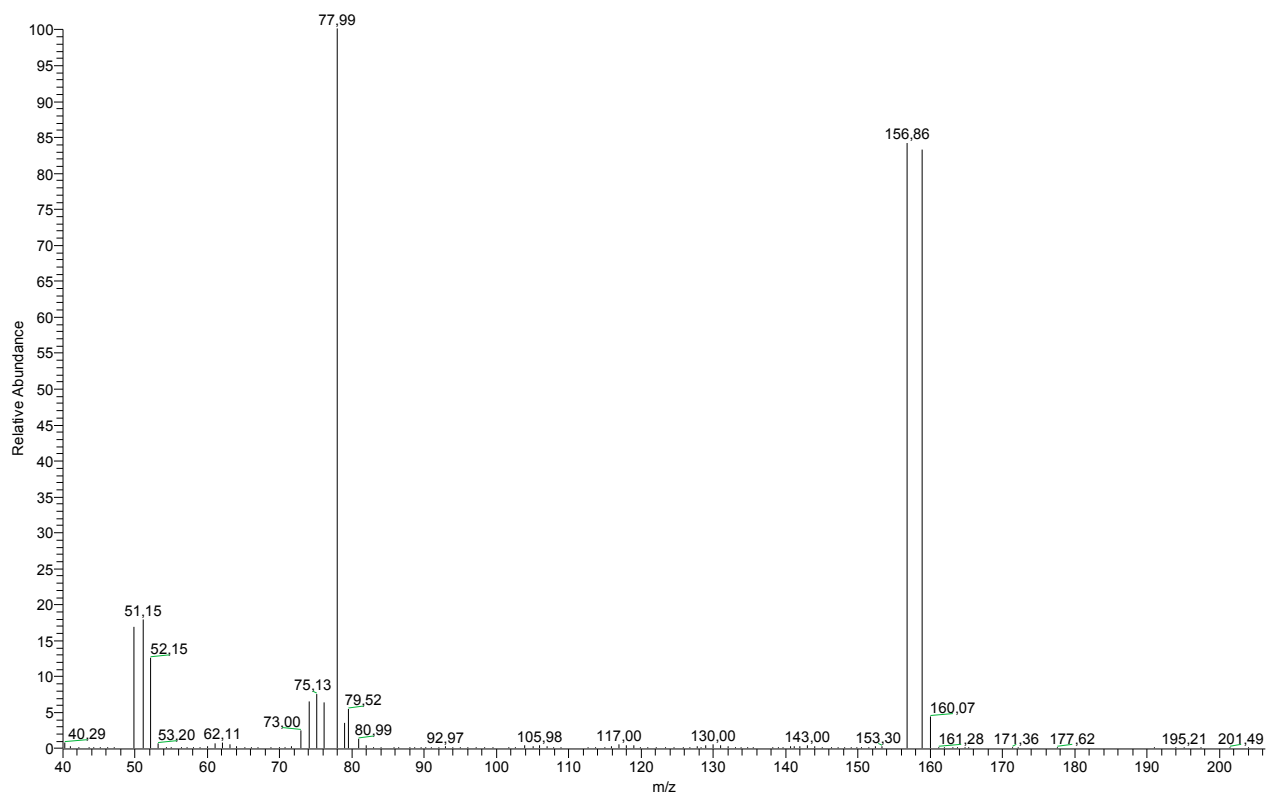


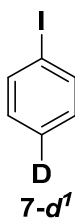
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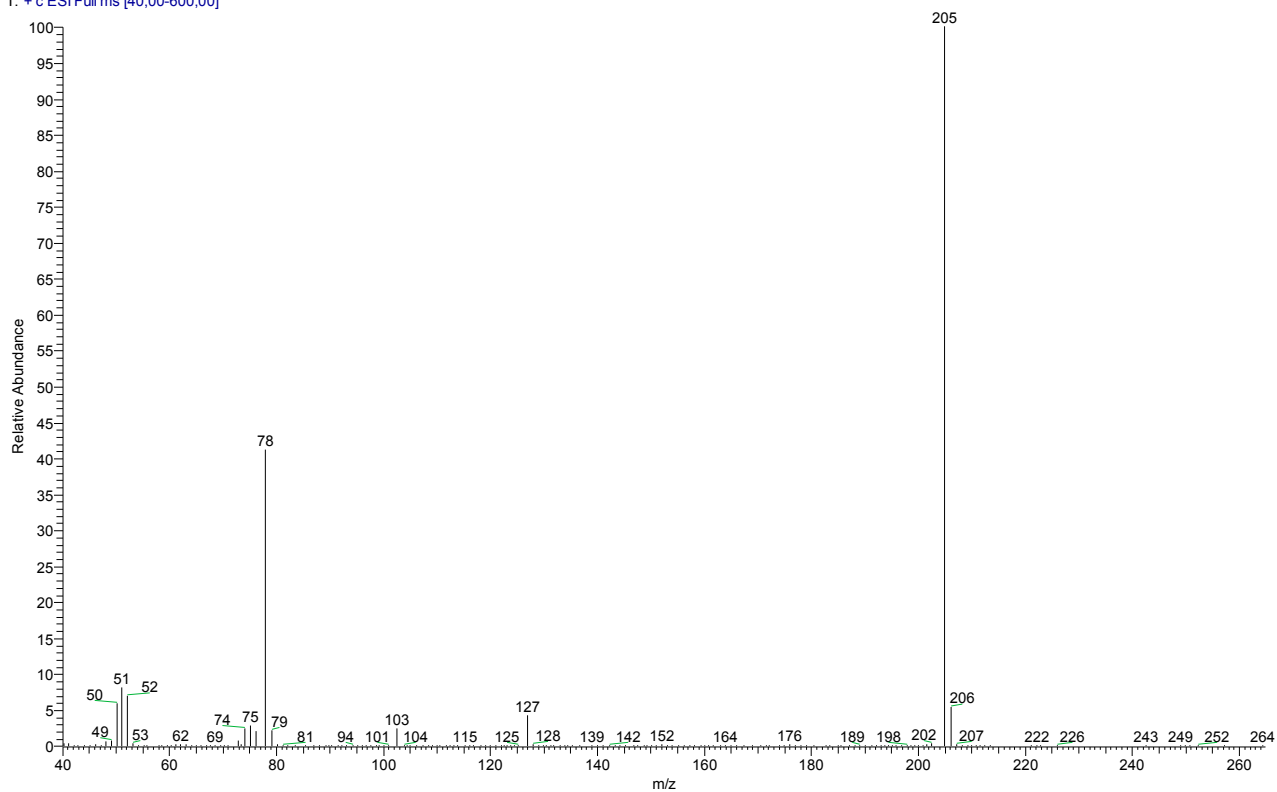
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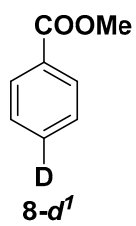




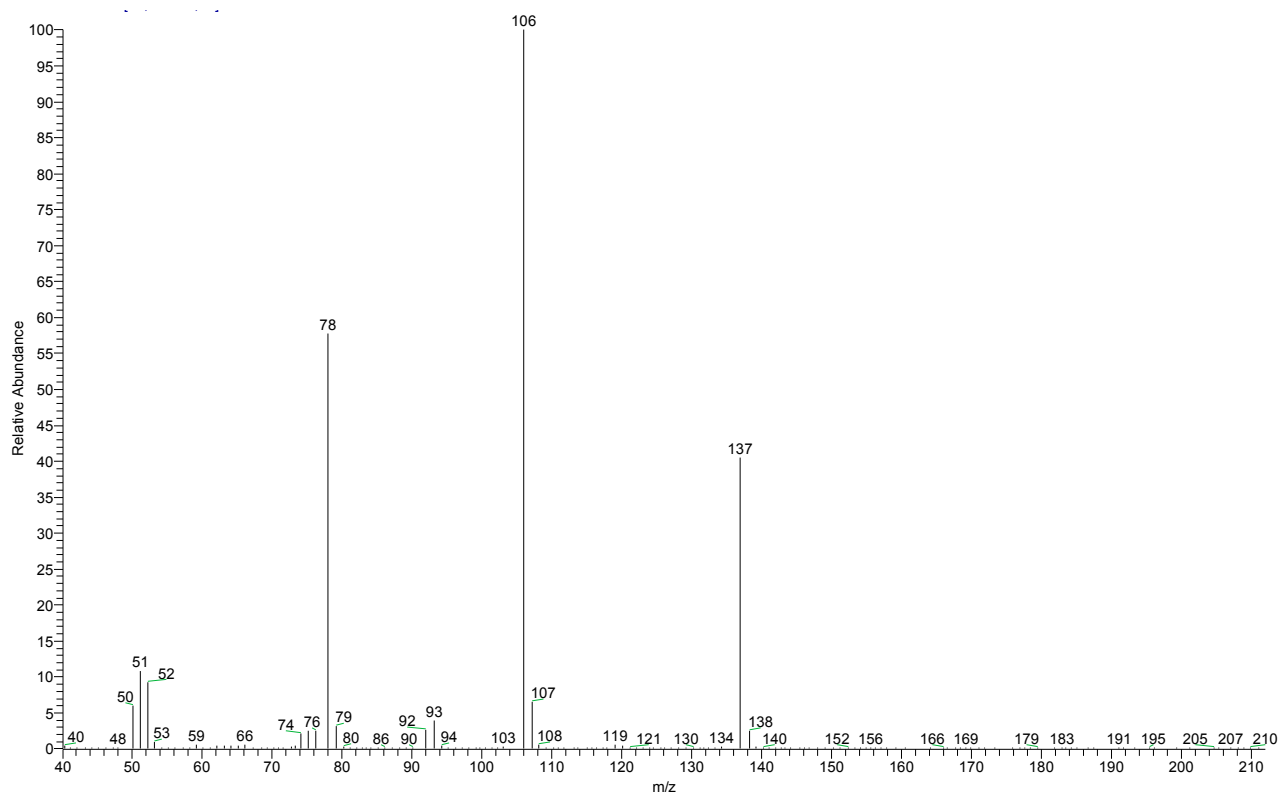
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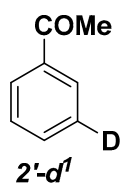
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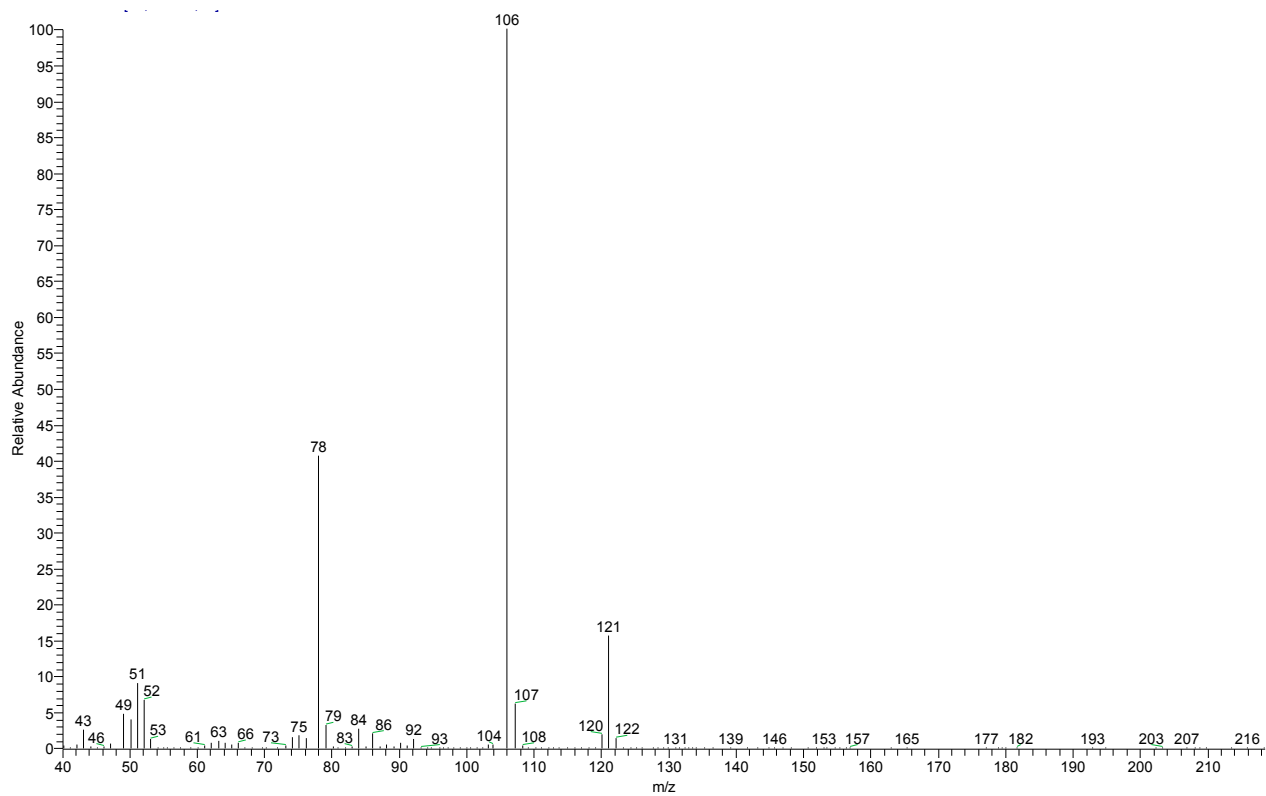


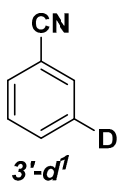
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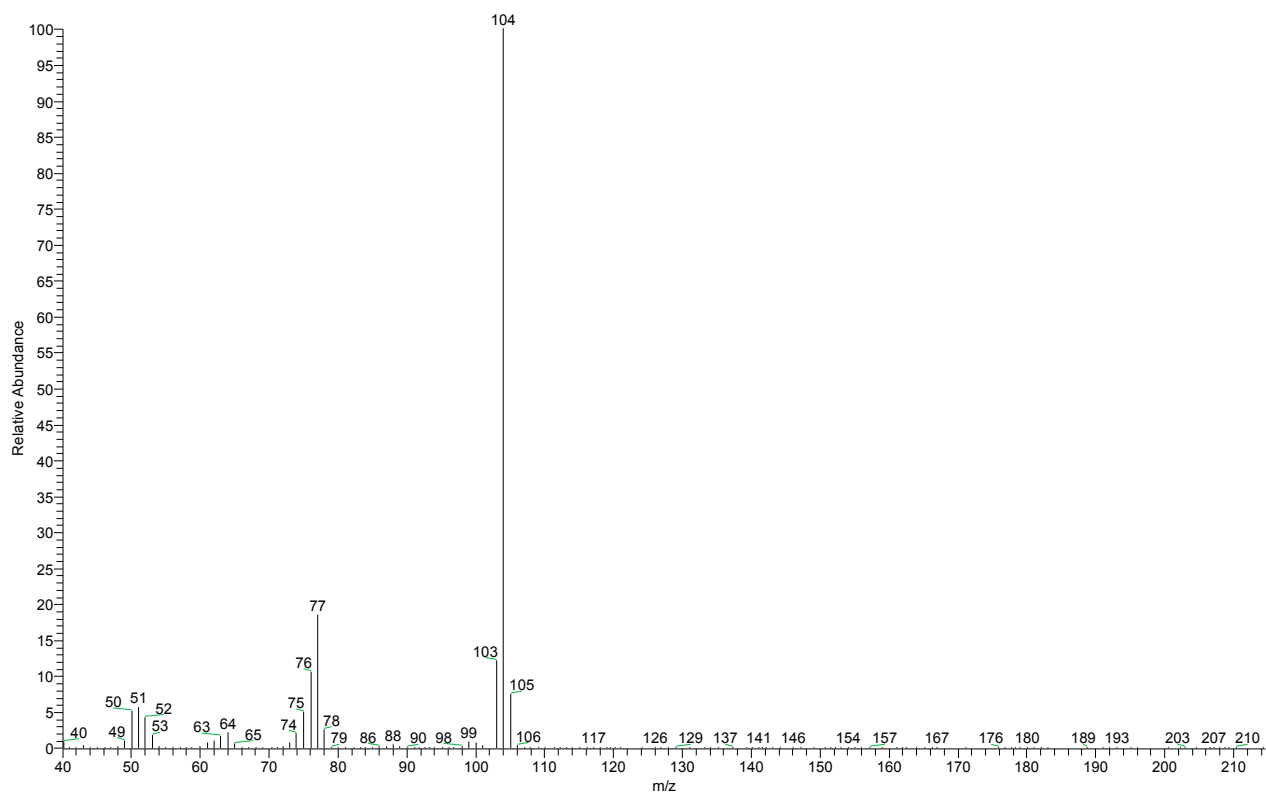


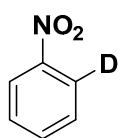
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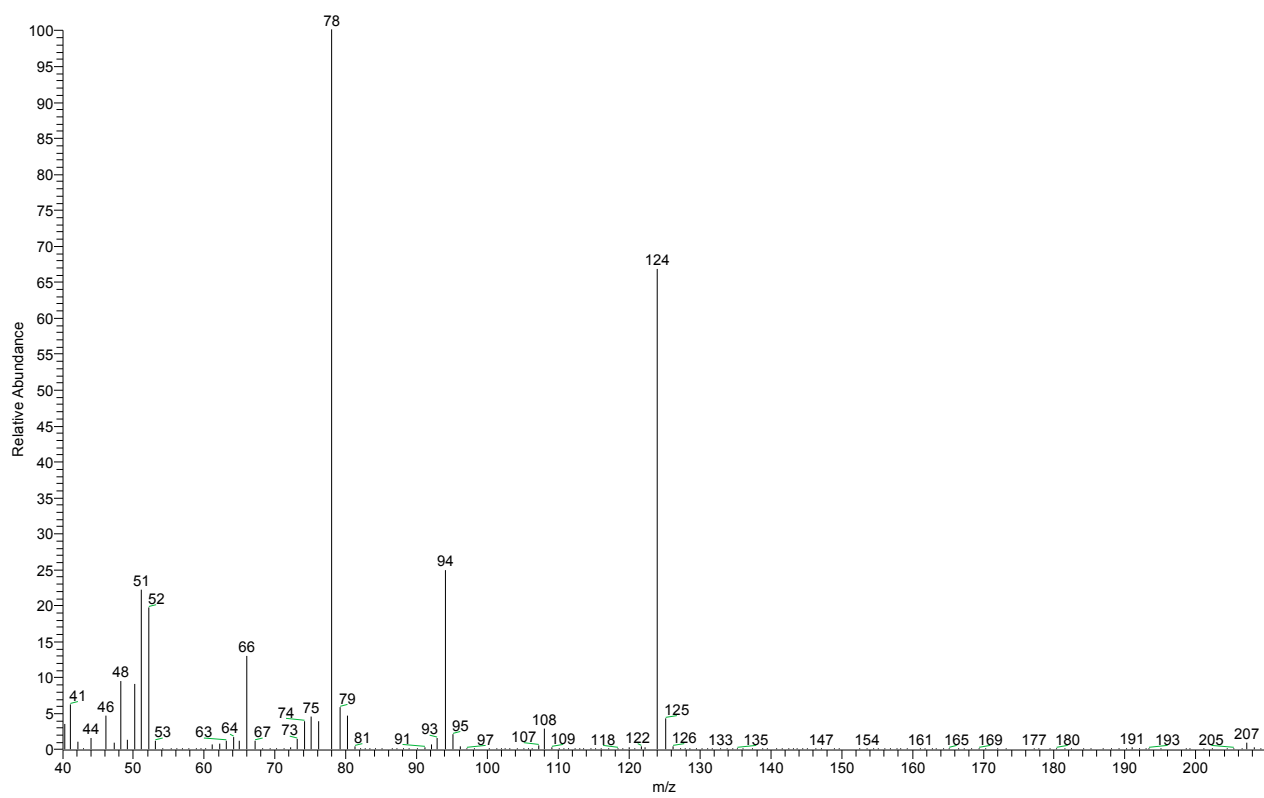
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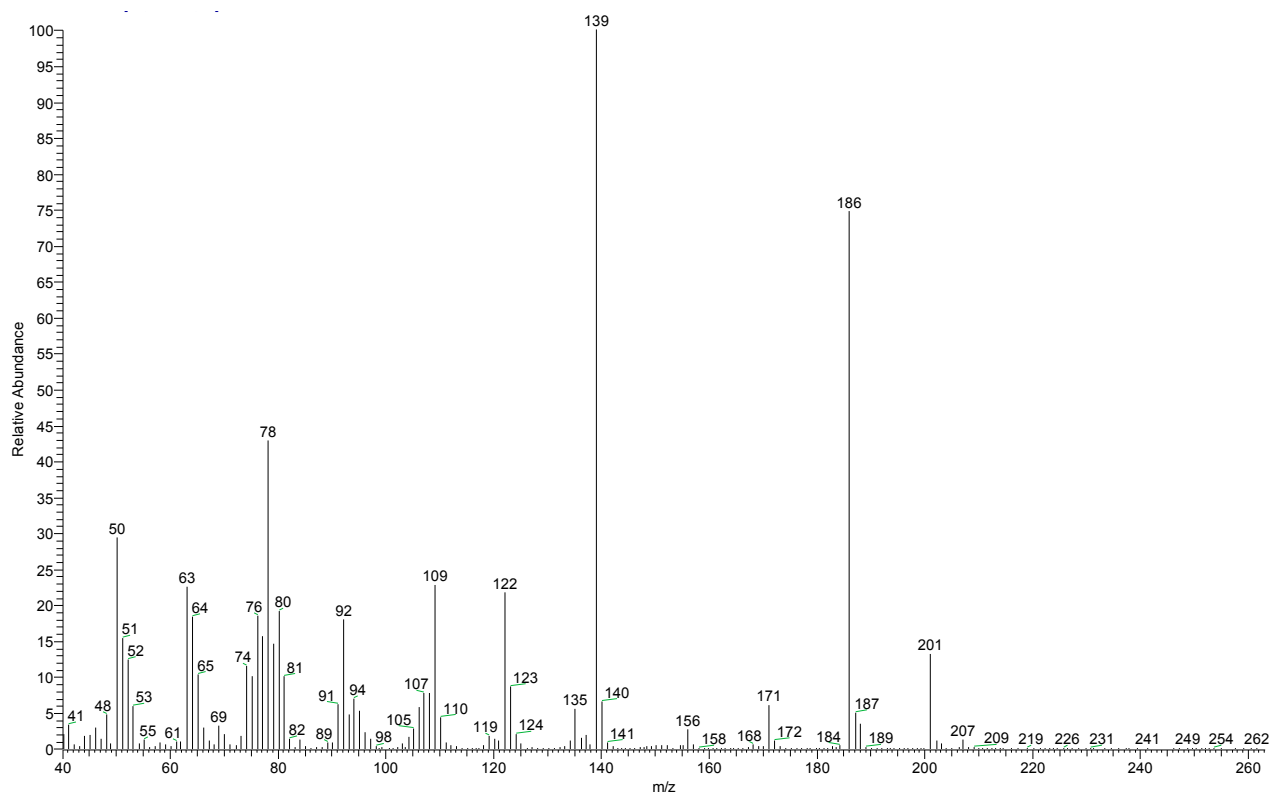
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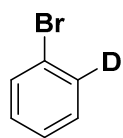
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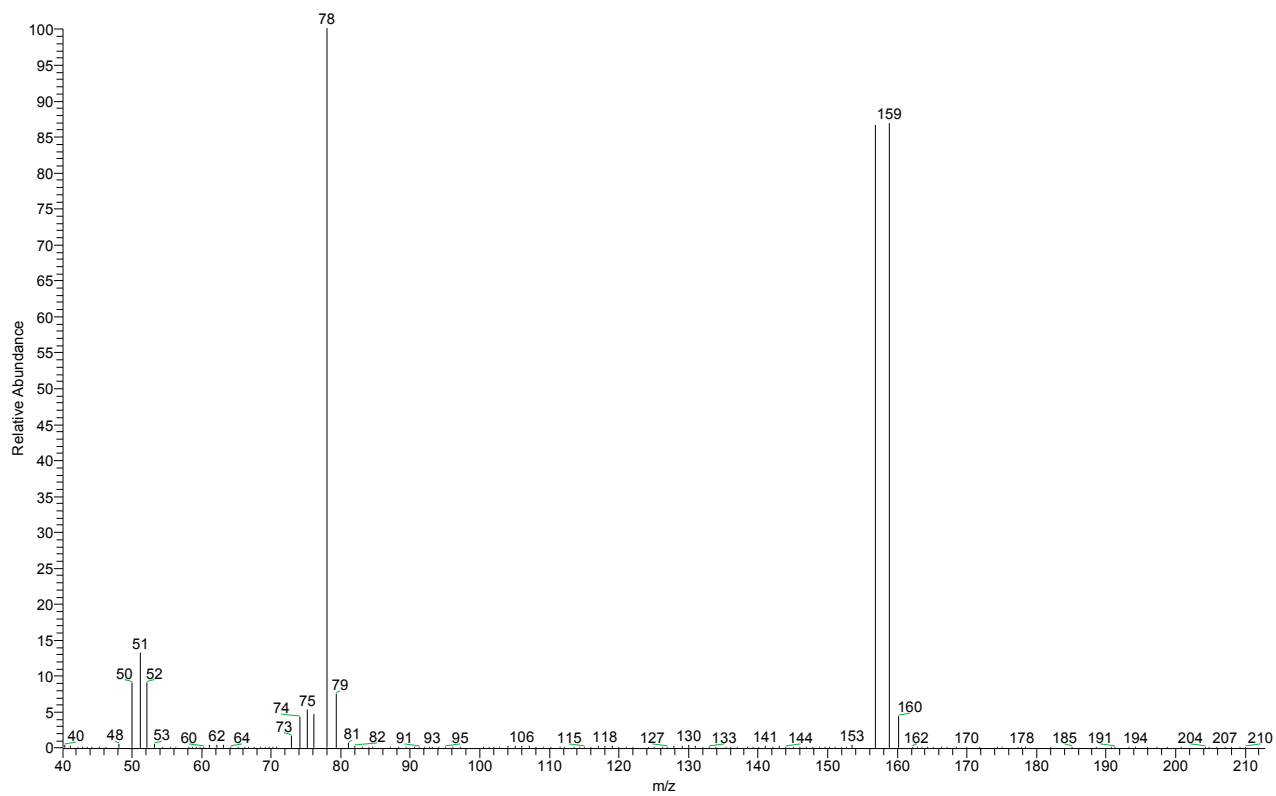
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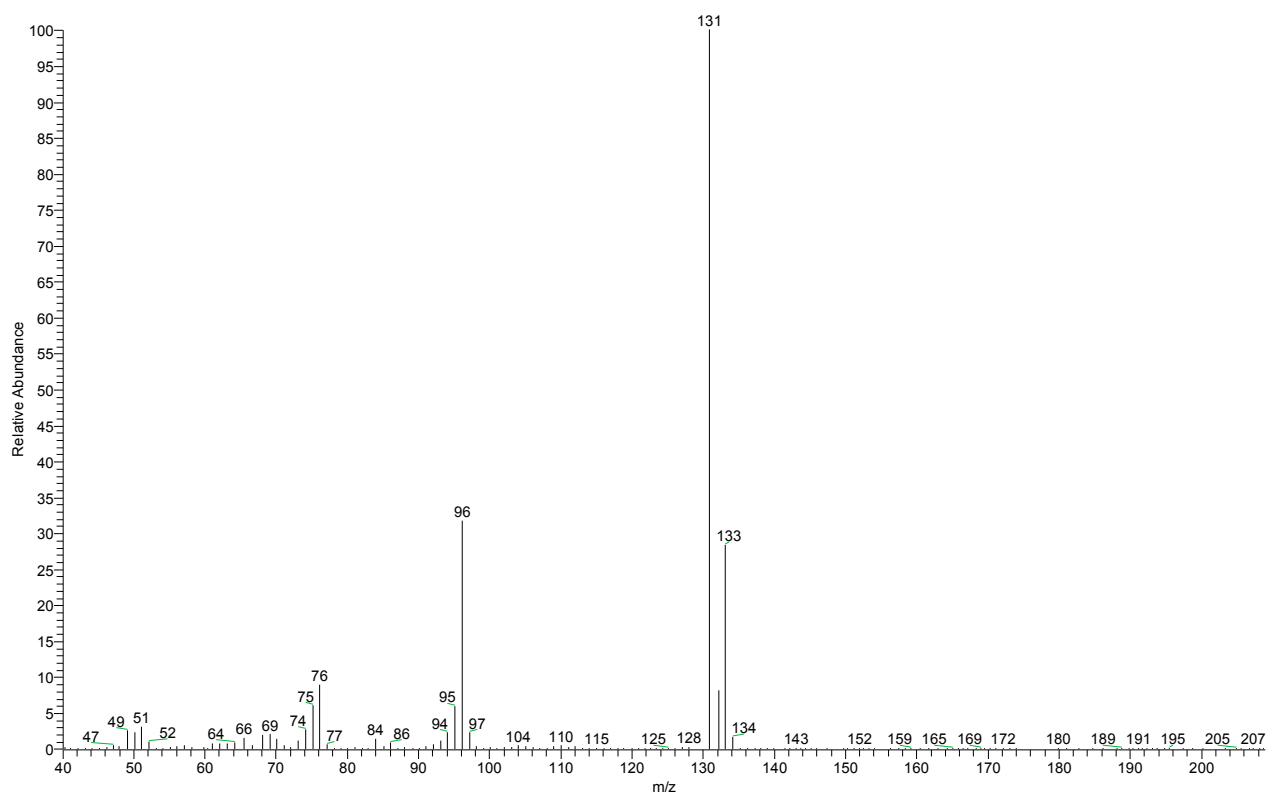
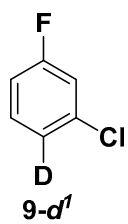


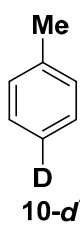


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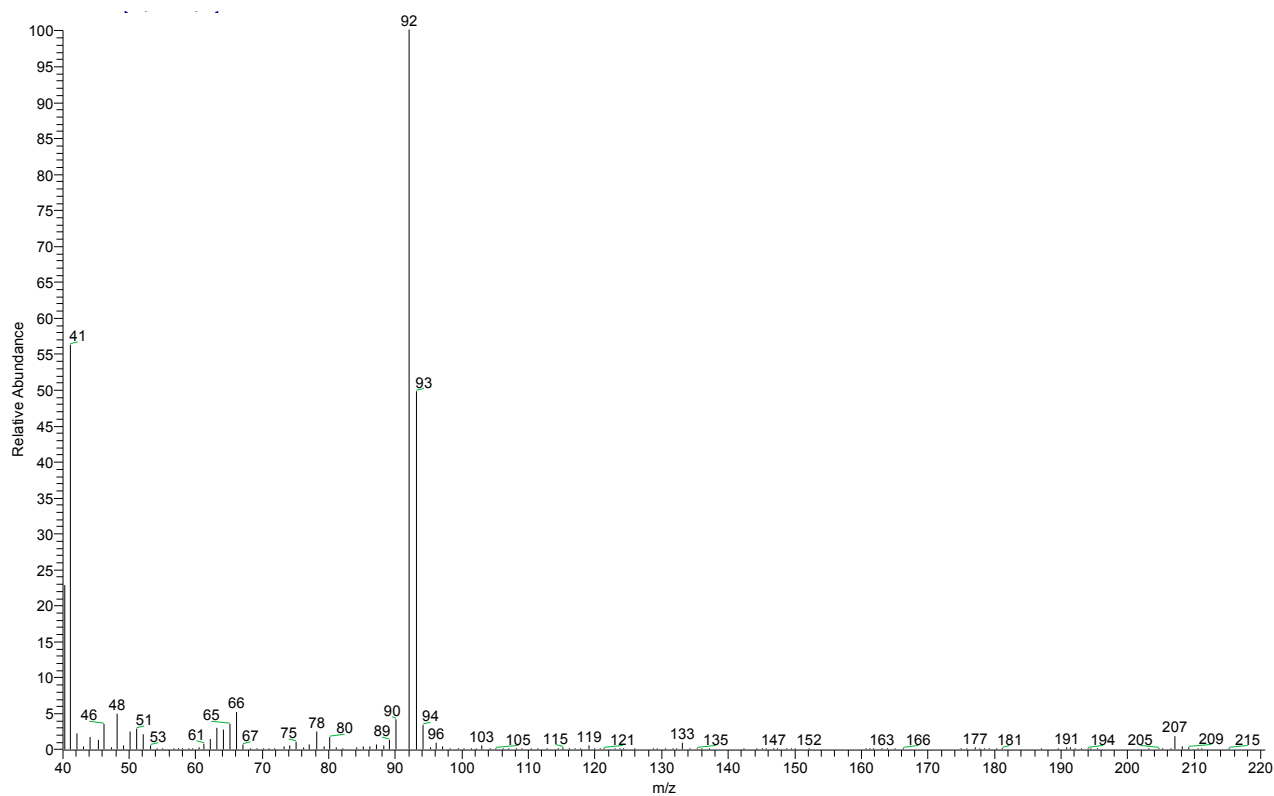
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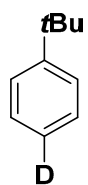






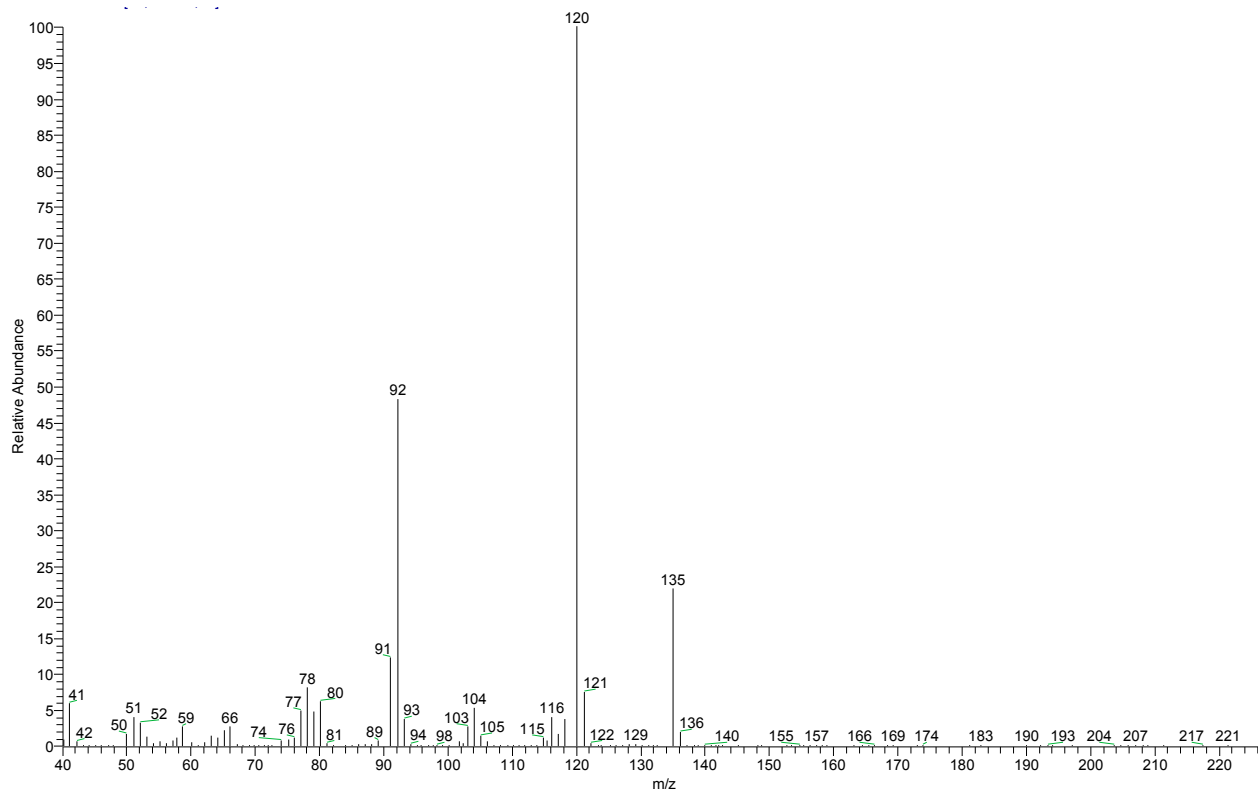
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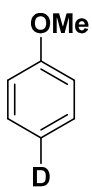




11-d'

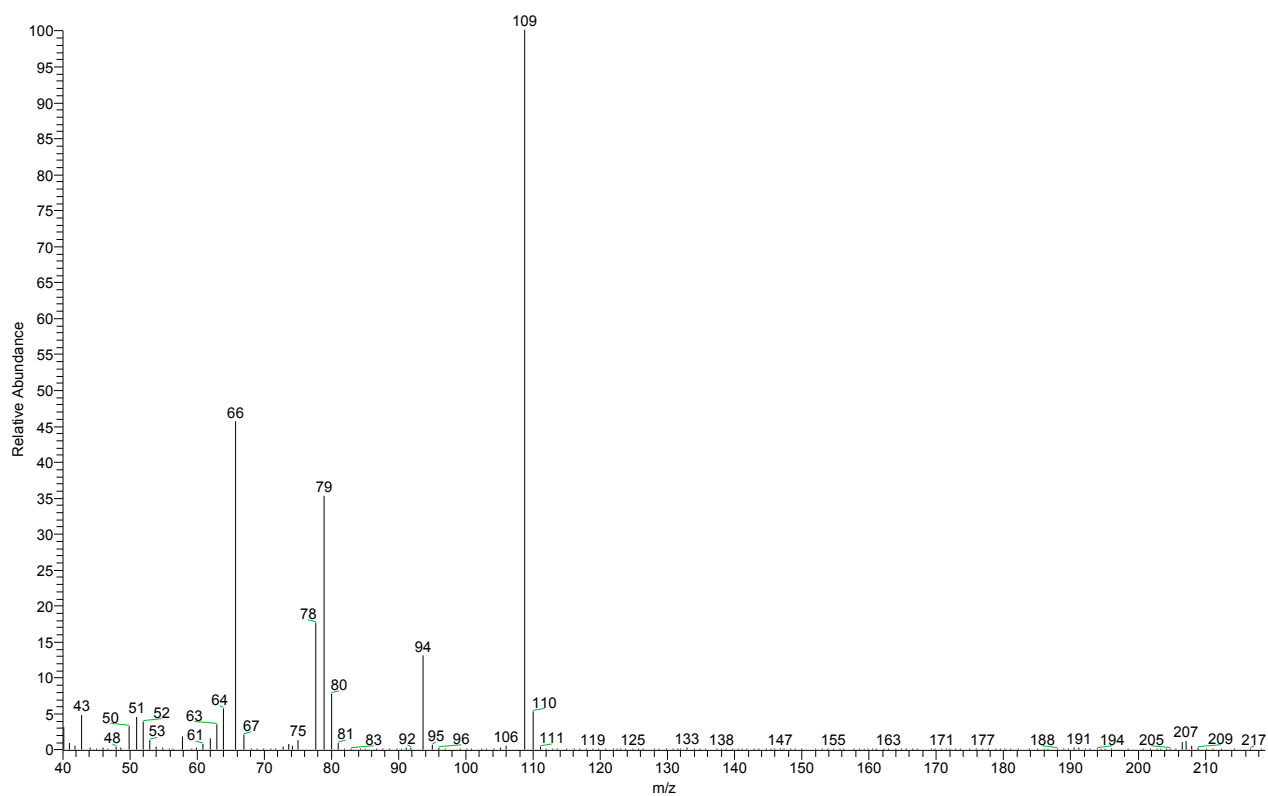
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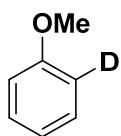




12-d'

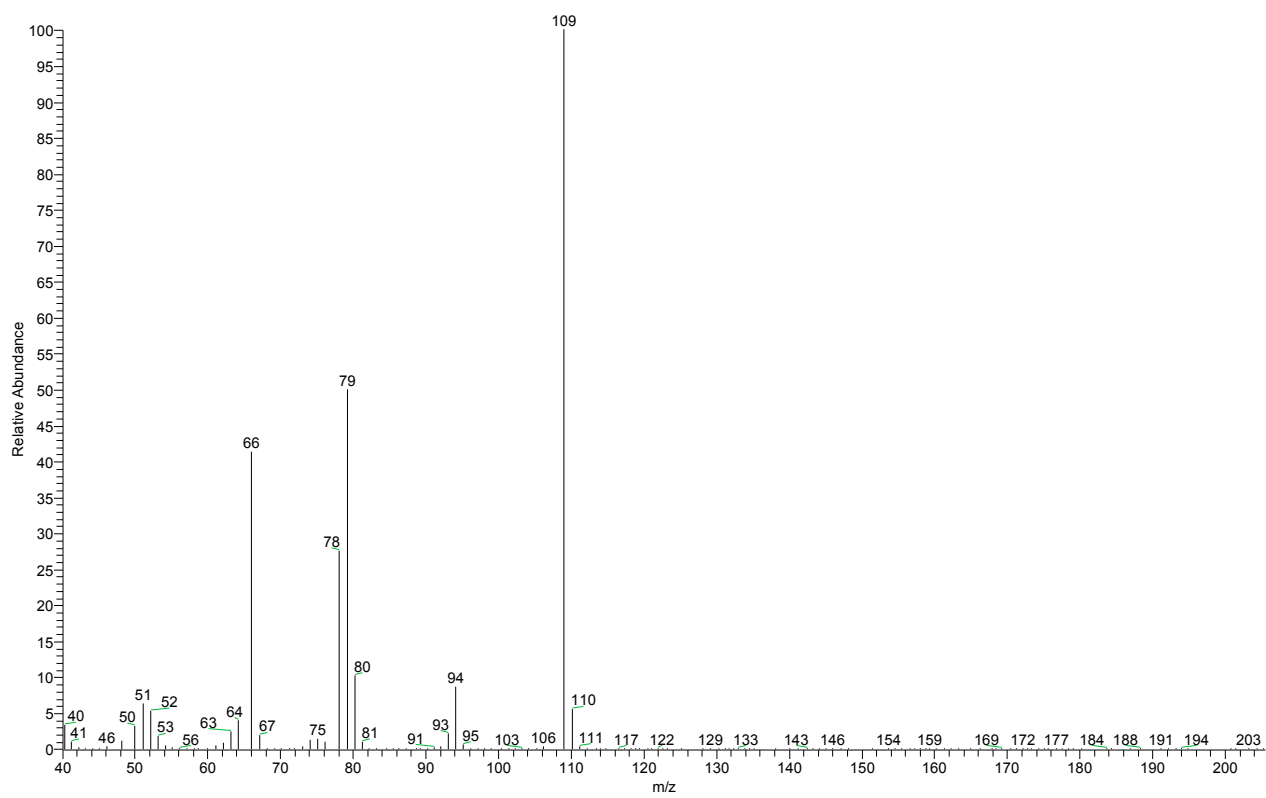
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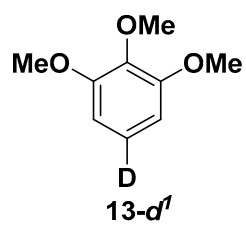




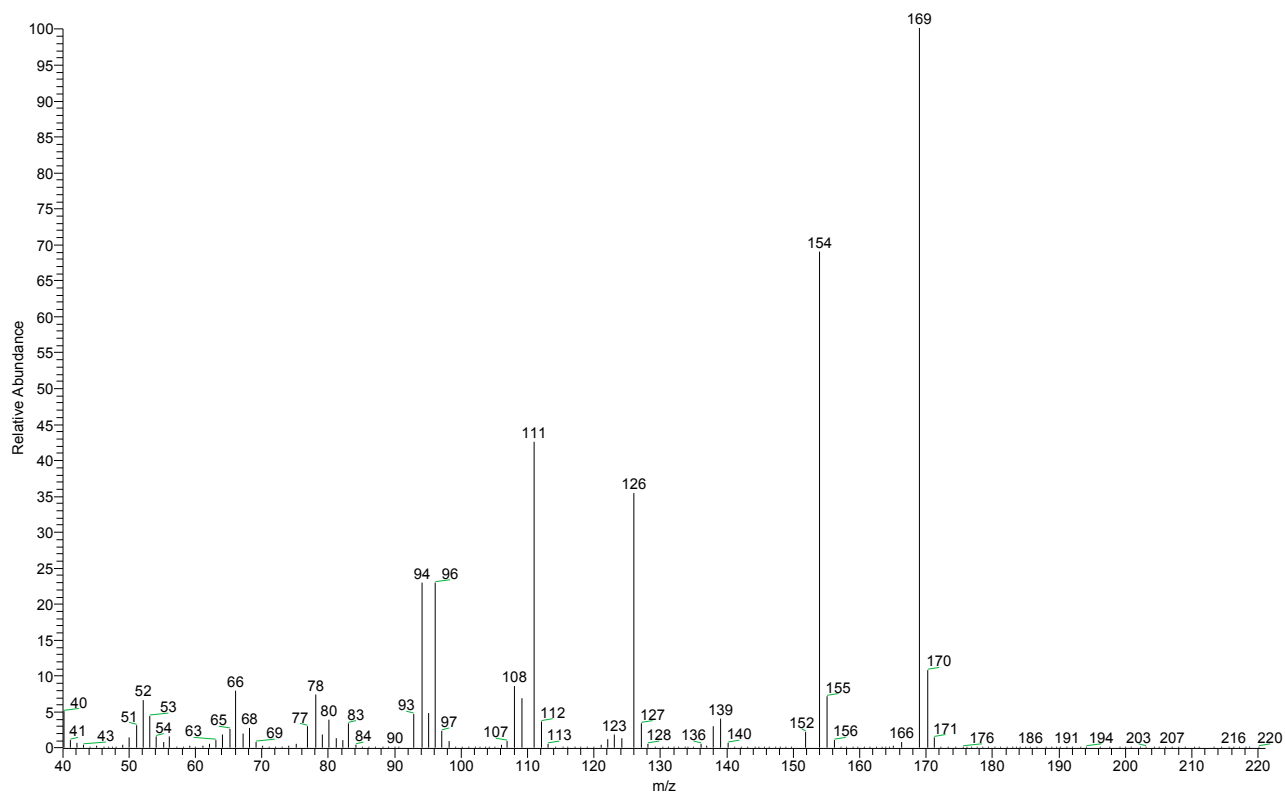
12'-d¹

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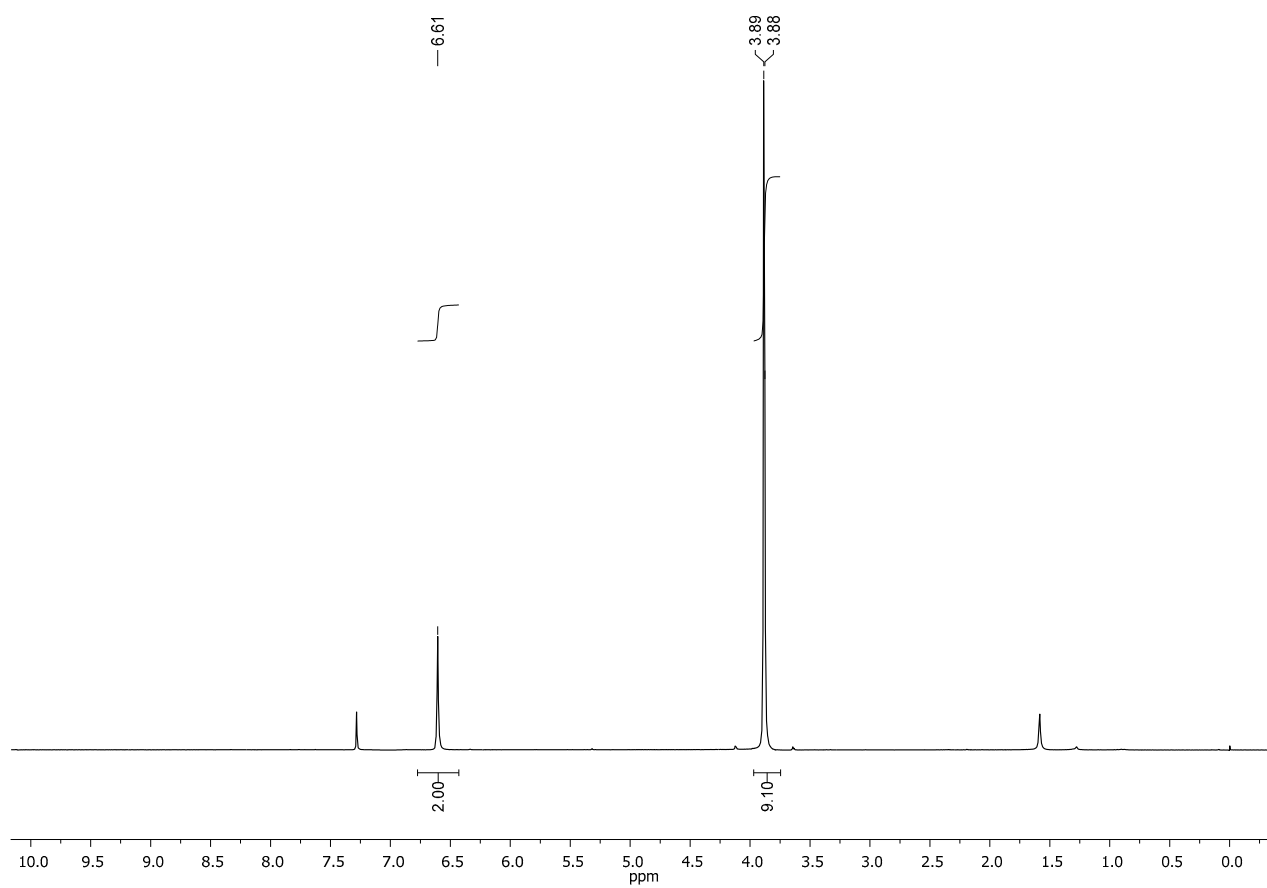




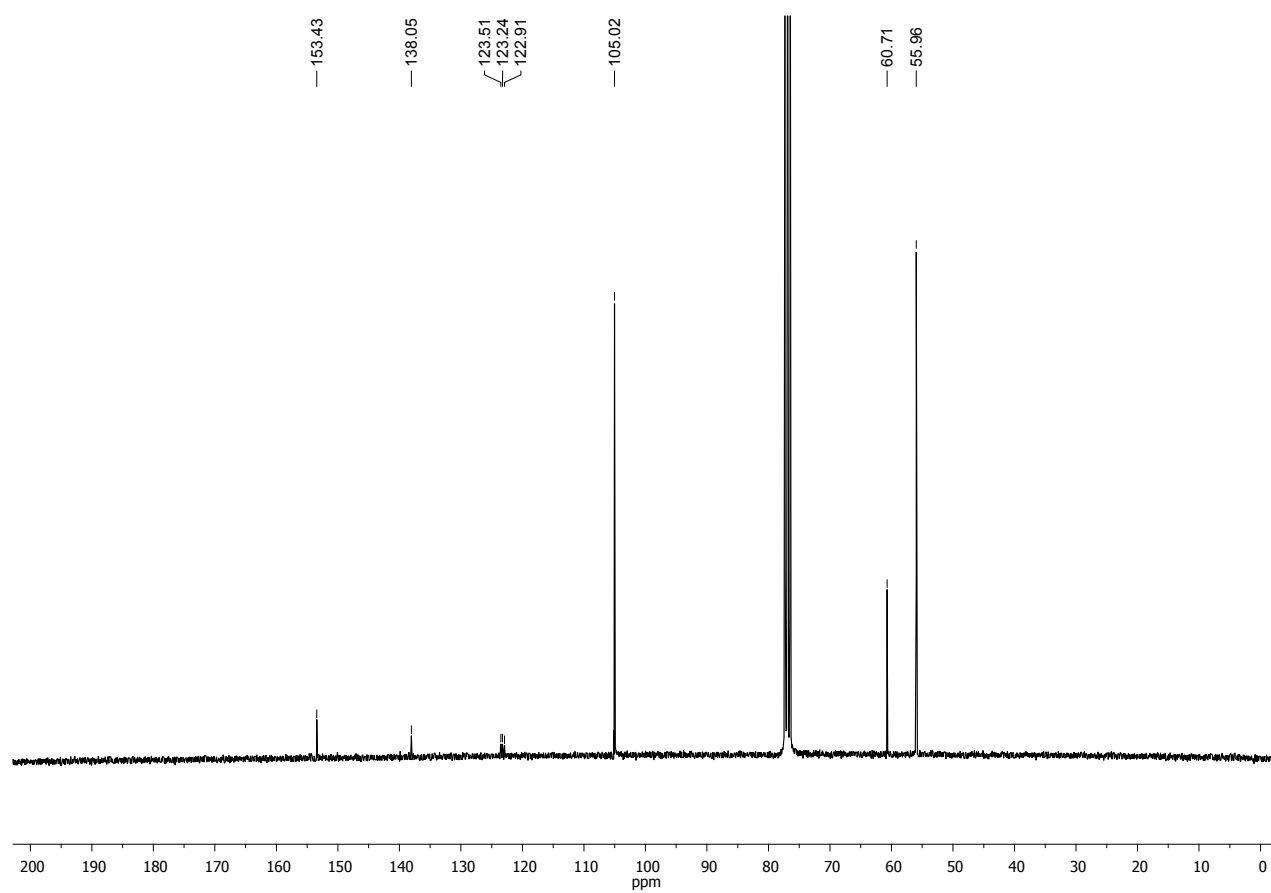
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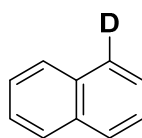


13-*d*^I (¹H NMR, 300 MHz)



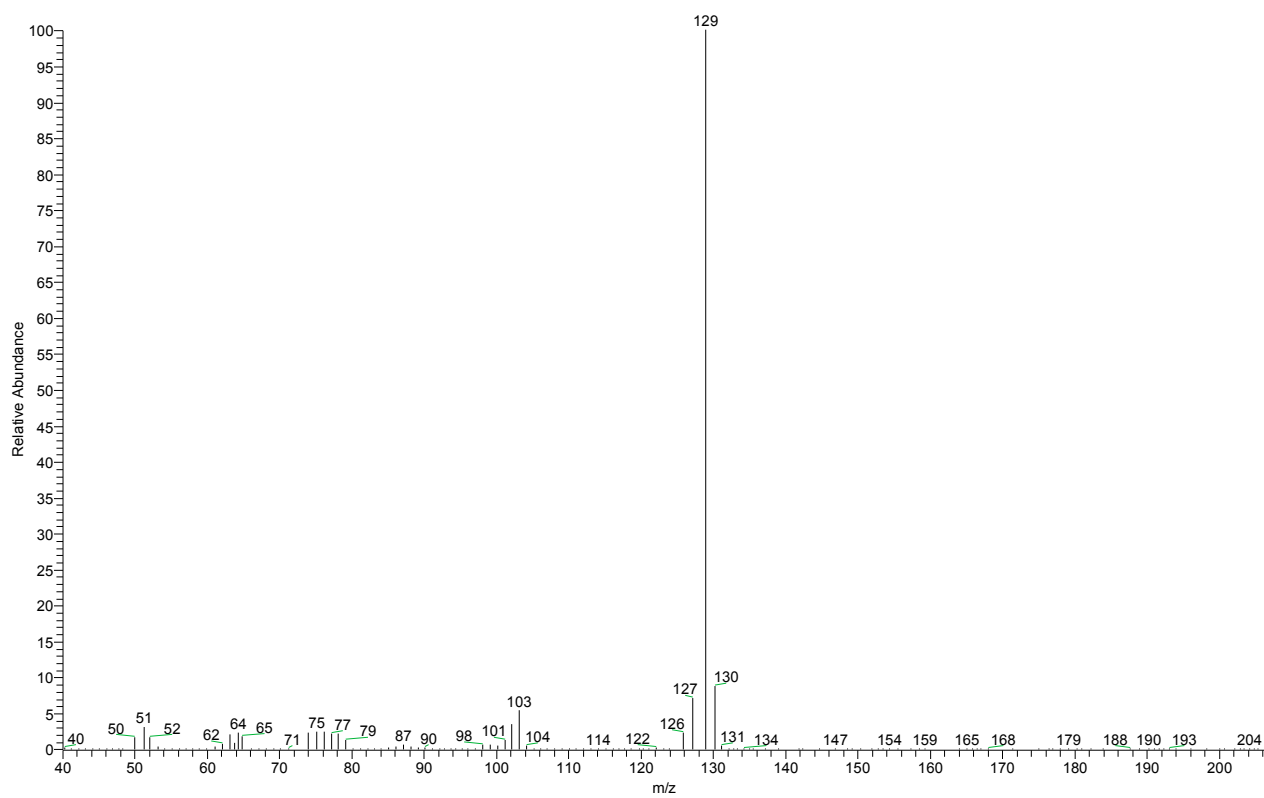
13-*d*^I (¹³C NMR, 75 MHz)



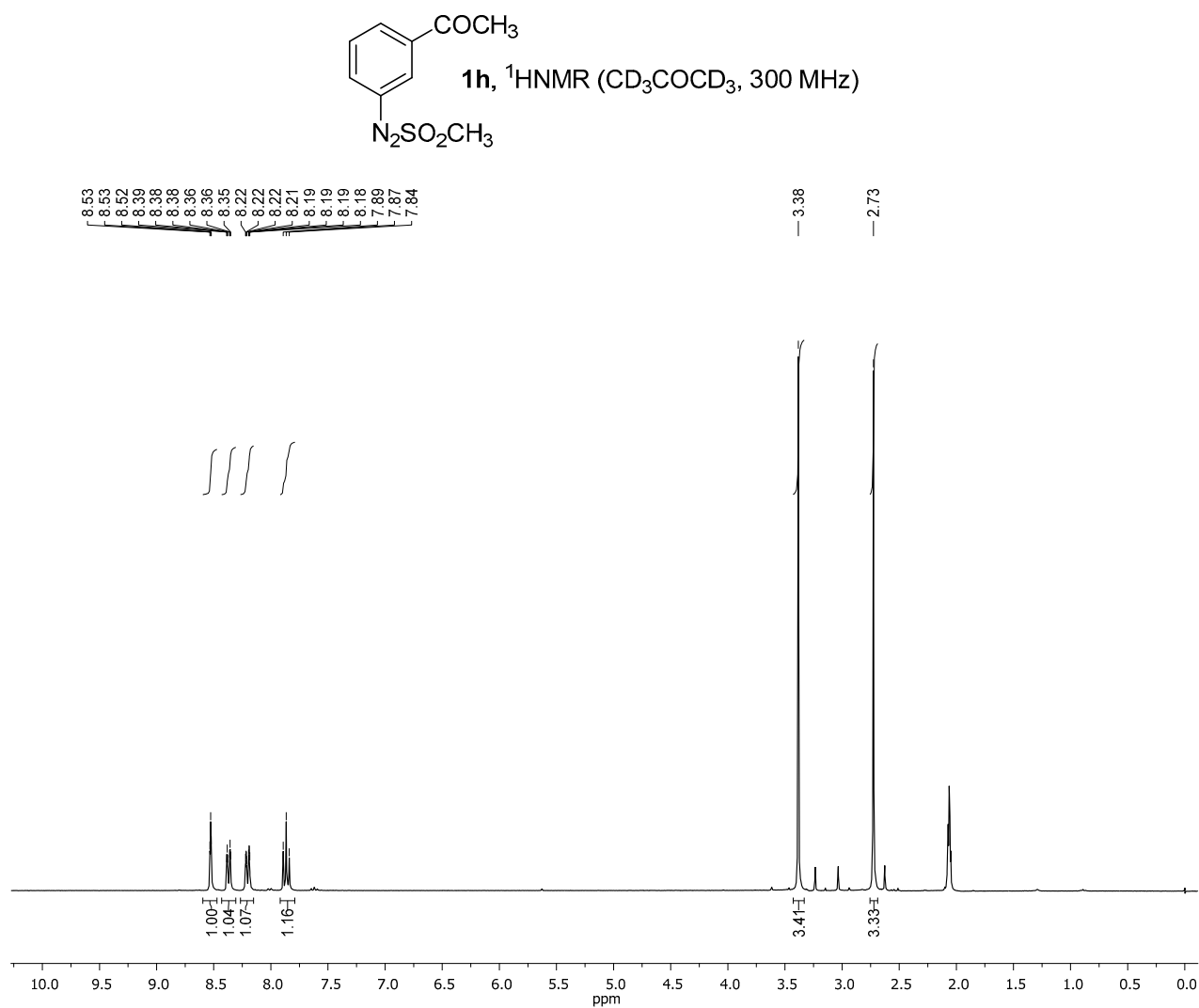


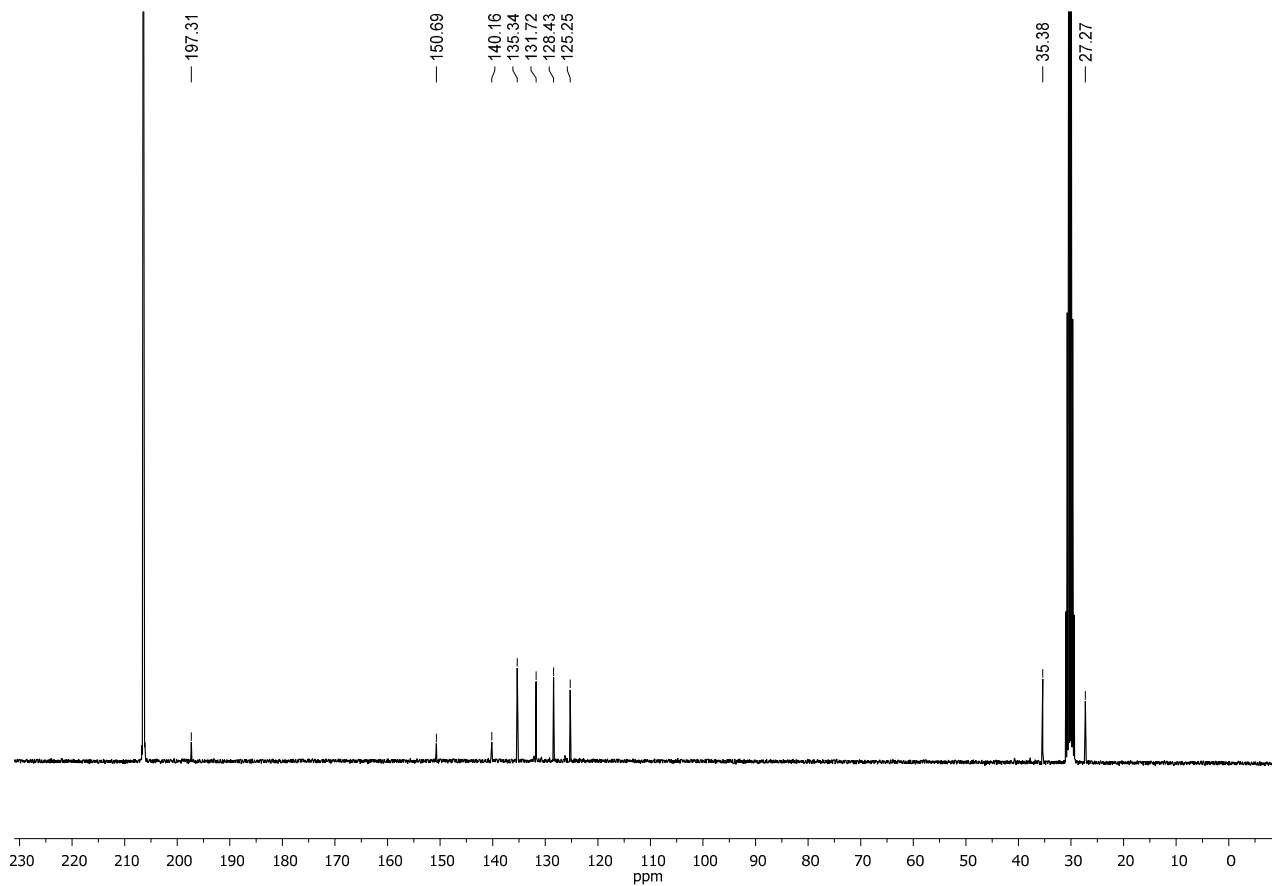
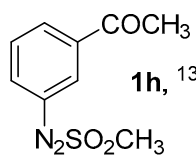
14-*d*¹

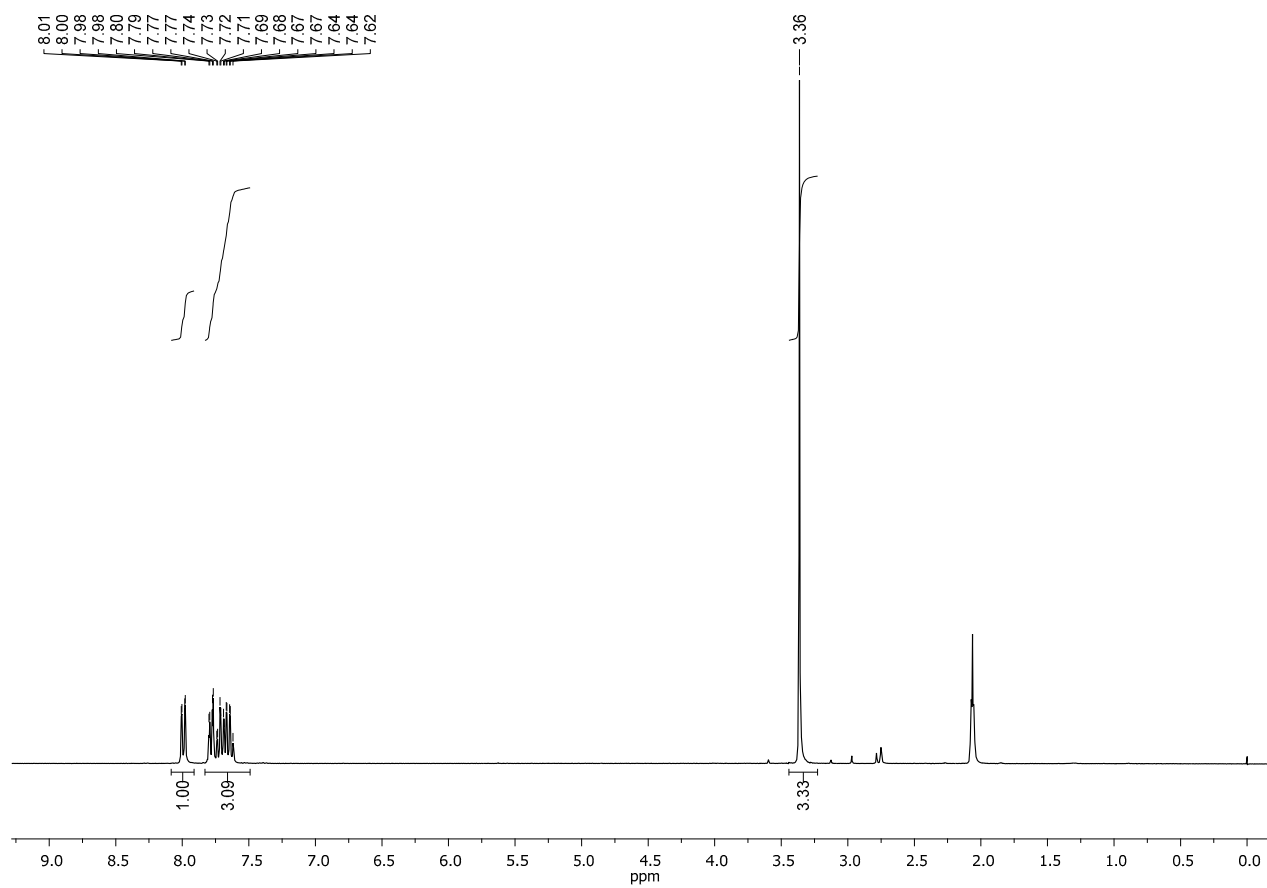
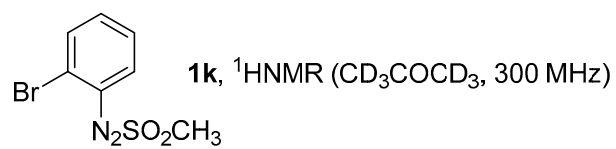
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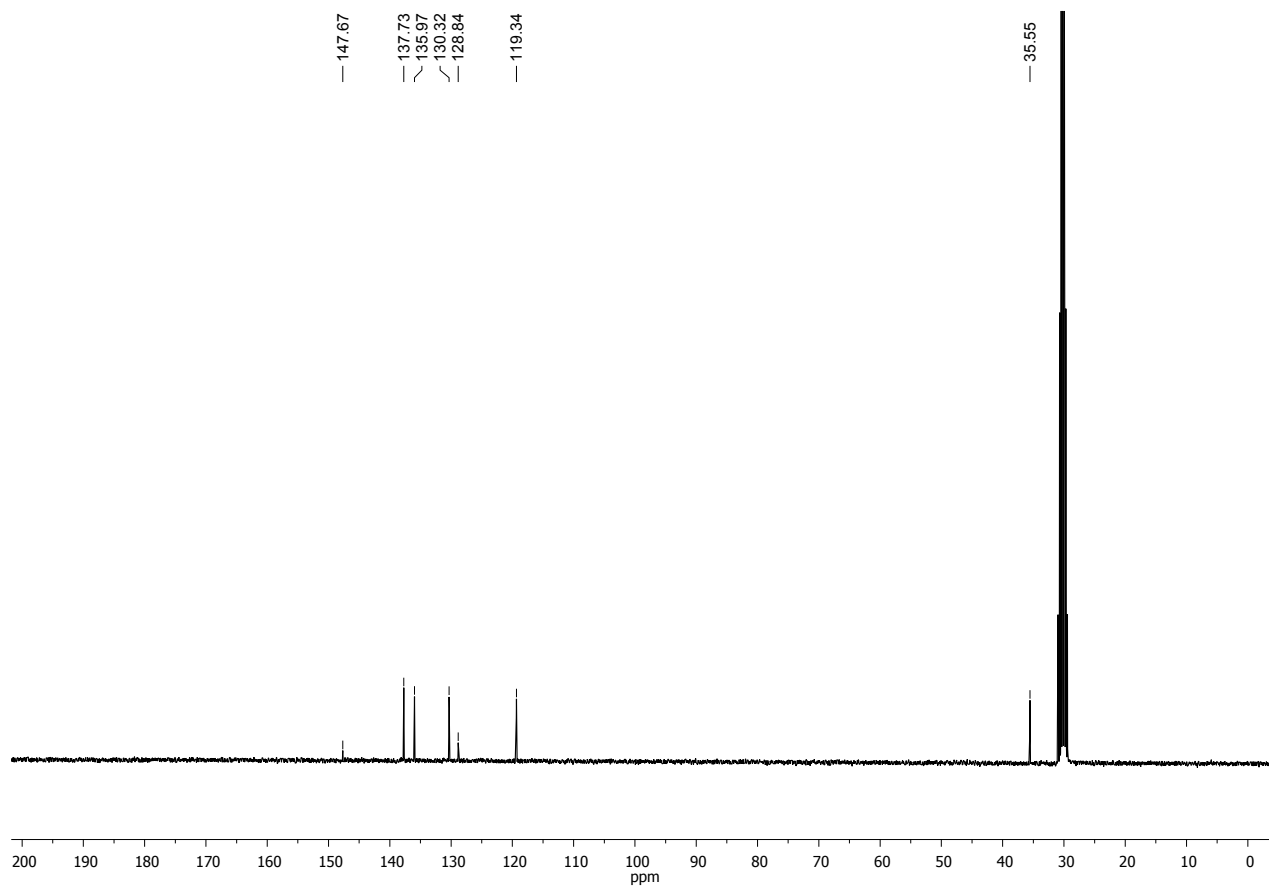
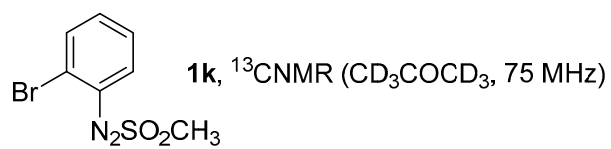


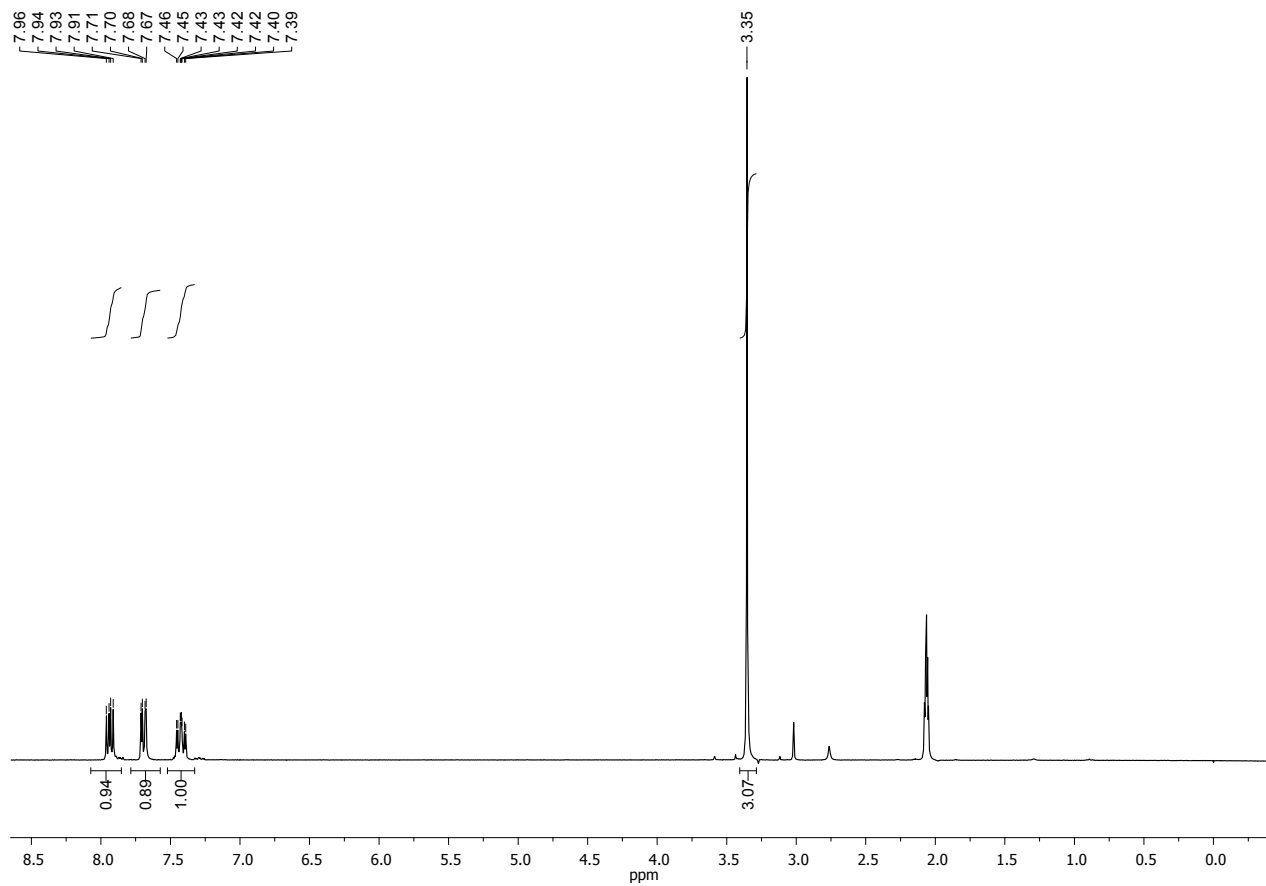
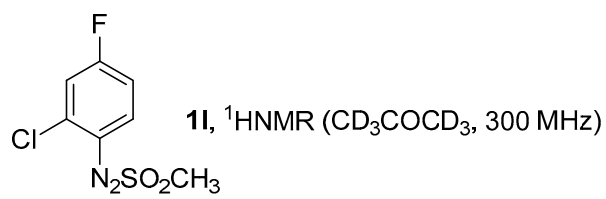
3 ^1H and ^{13}C NMR spectra of compounds 1h, 1i, 1k, 1l, 1q

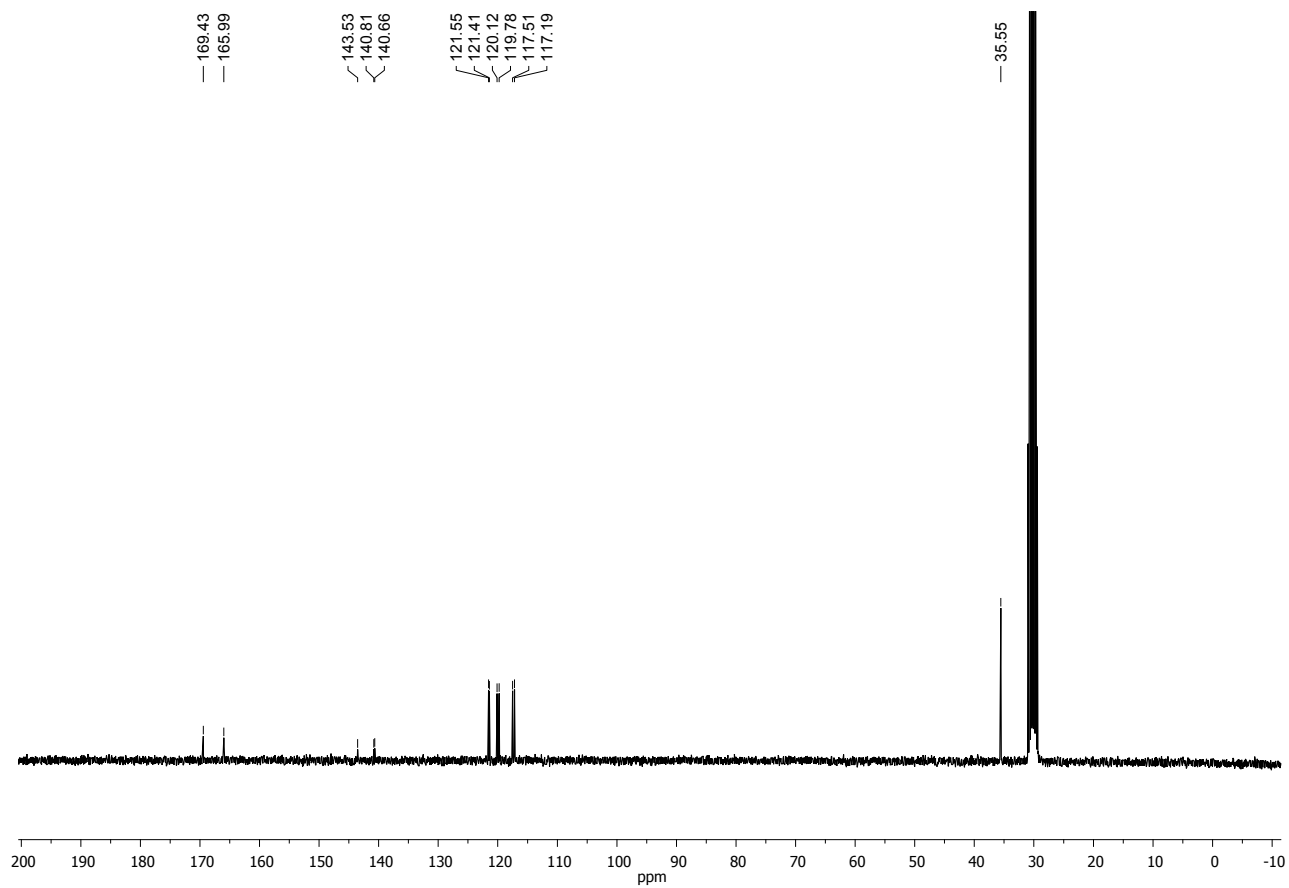
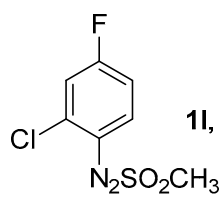


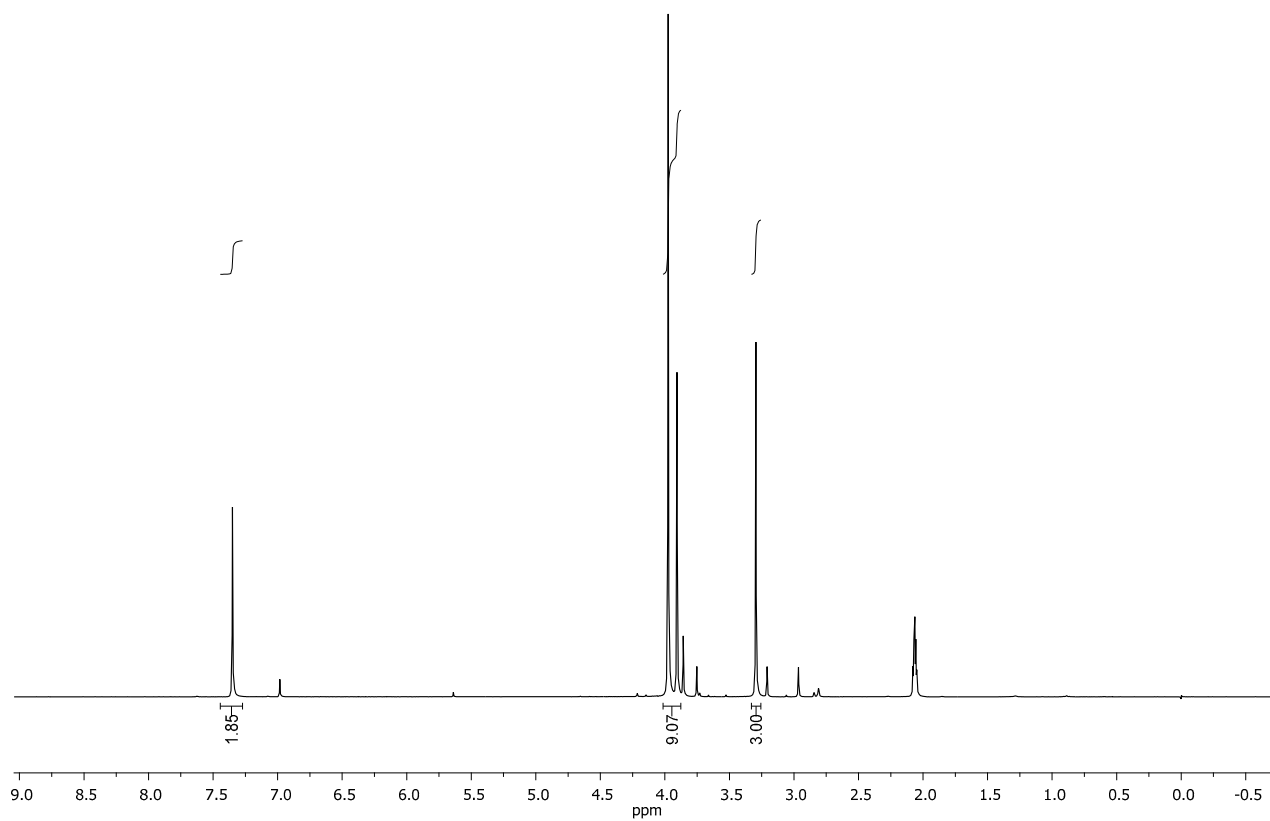
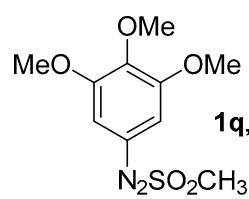


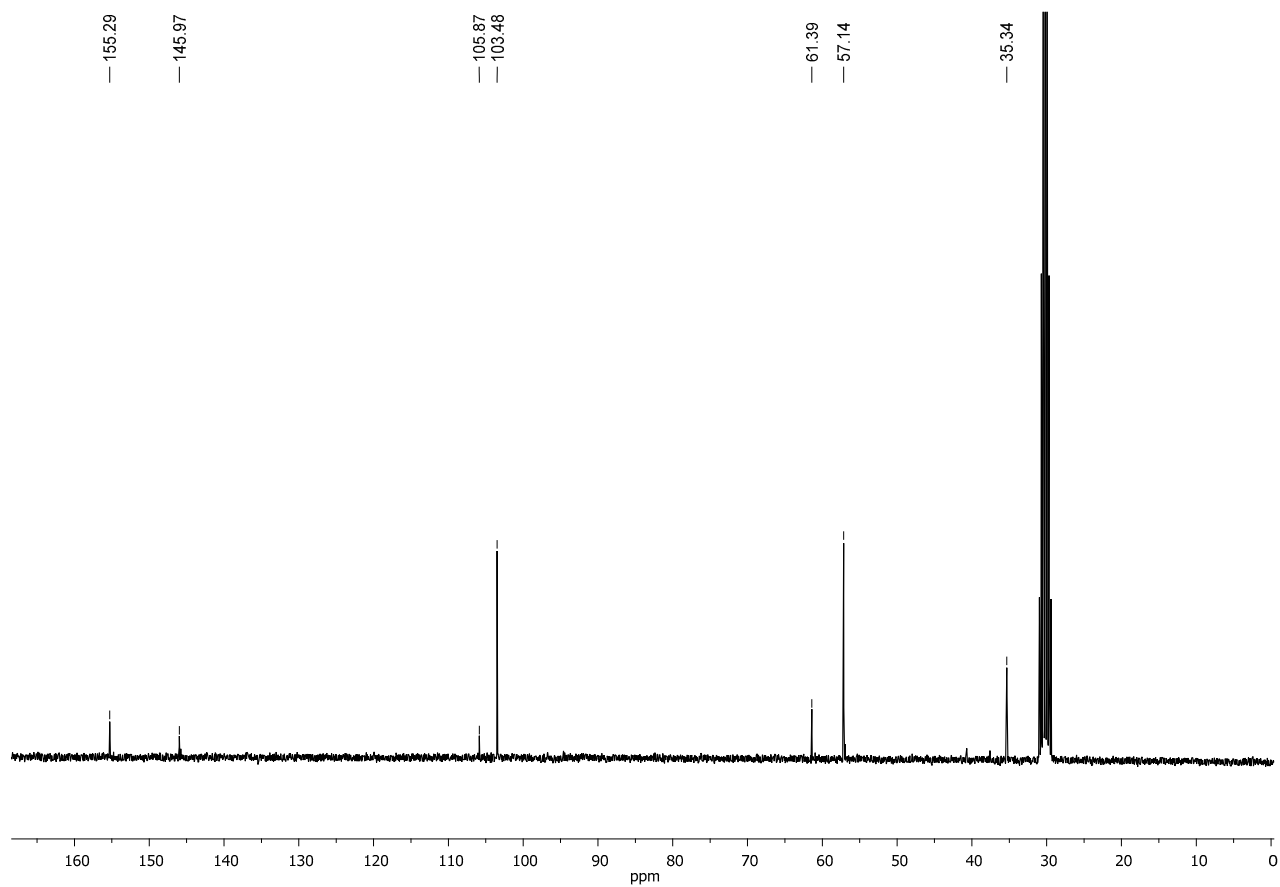
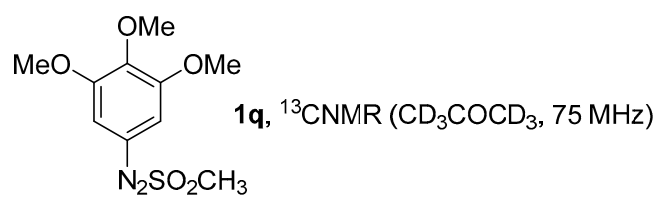












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