

One step synthesis of N-succidimidyl-4-[¹⁸F]-fluorobenzoate ([¹⁸F]SFB)

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SI Table 1: Optimization of time (used conditions for the experiment: K₂CO₃ (1.2 mg) / K₂₂₂ (10 mg)/DMF (0.3 ml), 130°C)

	RCC [%]		
	5 min	10 min	20 min
[¹⁸ F]SFB	9	6	4
[¹⁸ F]3 ([¹⁸ F]FBA)	3	12	11

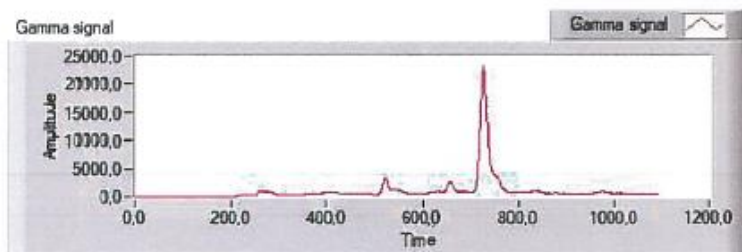
SI Table 2: Experimental conditions used to synthesize [¹⁸F]SFB from 4a (5 mg)

Entry	Base/cryptand	Solvent 0.3 mL	Additives	Temperature	Time [min]	RCC [%]**	Mean ± SD	n
1	1.2 mg K ₂ CO ₃ /10 mg K ₂₂₂	DMF	-	130°C	5	2-33	17.8±17	26
2	1.1 mg KHCO ₃ / 10 mg K ₂₂₂	ACN	t-BuOH (100 µL)	130°C	5	0	-	1
3	1.1 mg KHCO ₃ /10 mg K ₂₂₂	DMF	-	130°C	5	0	-	1
4	1.1 mg KHCO ₃ / 10 mg K ₂₂₂	DMF	Pyridine (100µL)	130°C	5	0	-	1
5	KOTf 10mg /1.2 mg K ₂ CO ₃	DMF	-	130°C	5	0	-	1
6	TEAB 10 mg	DMF	-	130°C	5	0	0	2
7	Cs ₂ CO ₃ 5 mg	DMF	-	130°C	5	0	-	1
8	1.2 mg K ₂ CO ₃ / 10 mg K ₂₂₂	ACN	-	90°C	5	0	-	1
9	1.2 mg K ₂ CO ₃ /10 mg K ₂₂₂	DMSO	-	130°C	5	0	-	1
10	1.1 mg KHCO ₃ / 10 mg K ₂₂₂	DMF	TEMPO 1 mg	130°C	5	0	-	1
11	K ₂ CO ₃ /K ₂₂₂	DMF	TEMPO, 1mg	130°C	5	3-17	12±14	11
12	K ₂ CO ₃ /18-crown-6	DMF		130°C	5	4	-	1
13	1.2 mg K ₂ CO ₃ /10 mg K ₂₂₂	DMF	-	90°C*	10	0	-	1
14	5 mg K ₂ CO ₃ /10 mg K ₂₂₂	DMF		130°C	5	0	-	1
15	1.2 mg K ₂ CO ₃ / 10 mg K ₂₂₂	DMF	-	90°C	5	0	-	1
15	1.2 mg K ₂ CO ₃ / 10 mg K ₂₂₂	DMF	-	140°C	5	0	-	1

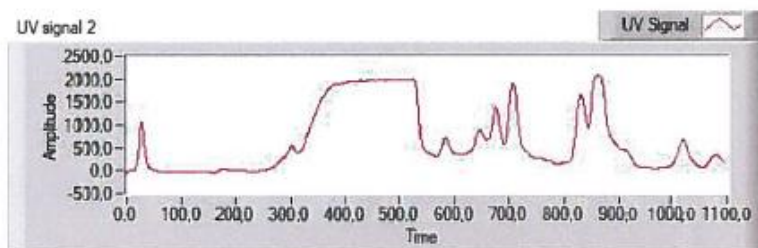
*Microwave heating; ** trace amount of acid sometimes detected if yield was 0.

Example of a semipreparative chromatogram of [^{18}F]SFB

a) Radiochromatogram



b) UV-chromatogram

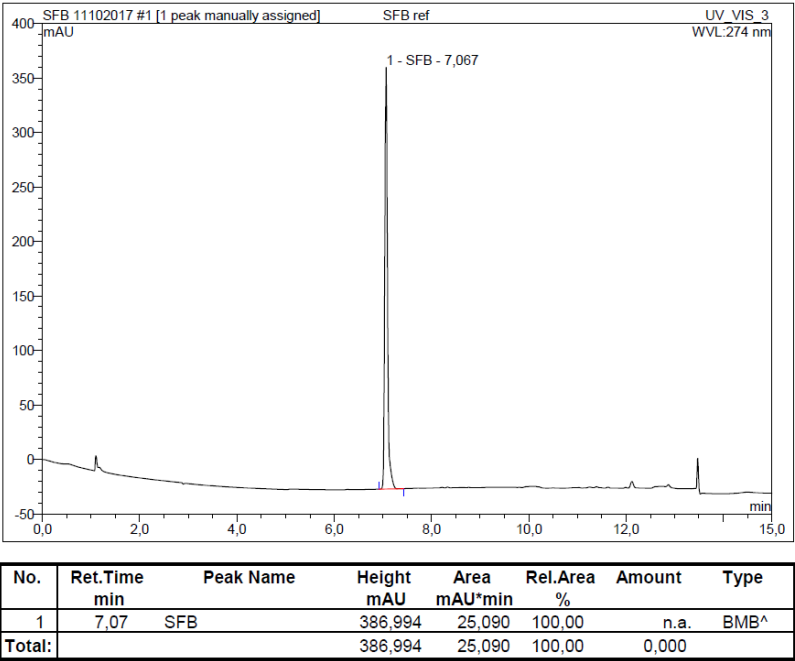


Analytical HPLC chromatograms

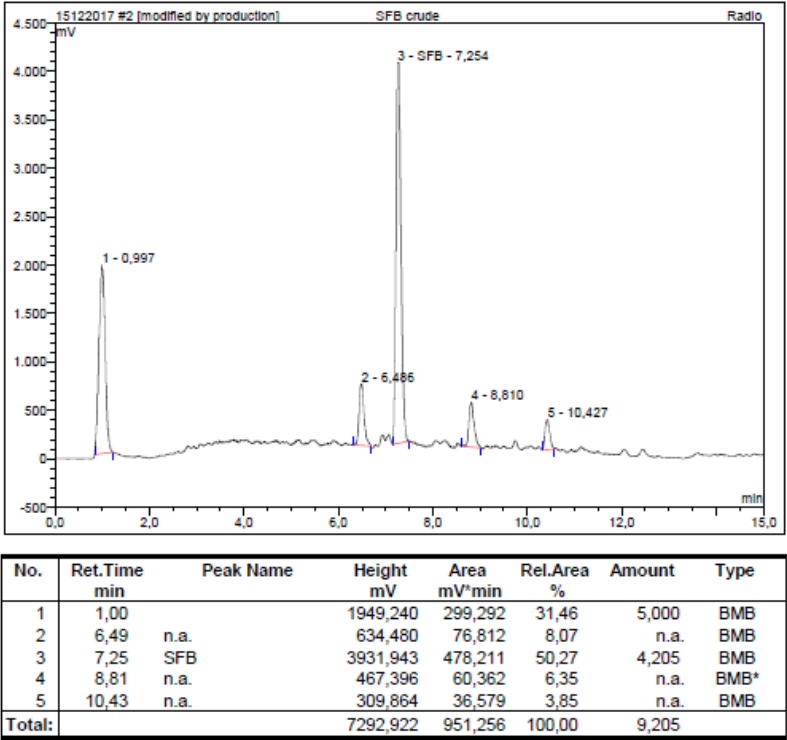
Method 1: *C18 LUNA (phenomenex) column, 5 μm , 250 $\times$ 4.6 mm in 2-mL/min solvent flow. A gradient system with two eluents, A and B, was used, with the fraction of B varying from 0% to 100% over 15 min. A = H_2O , 0.1% TFA; B = MeCN : H_2O , 0.1% TFA*

Method 2: *C4 Grace Vydac column, 5 μm , 250 $\times$ 4.6 mm in 1-mL/min solvent flow. A gradient system with two eluents, A and B, was used, with the fraction of B varying from 0% to 100% over 25 min. A = H_2O , 0.1% TFA; B = MeCN : H_2O , 0.1% TFA*

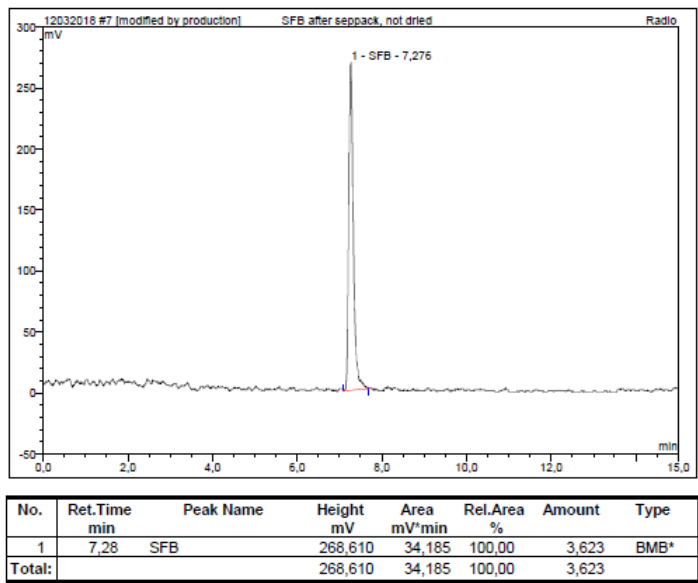
Reference of SFB, Method 1



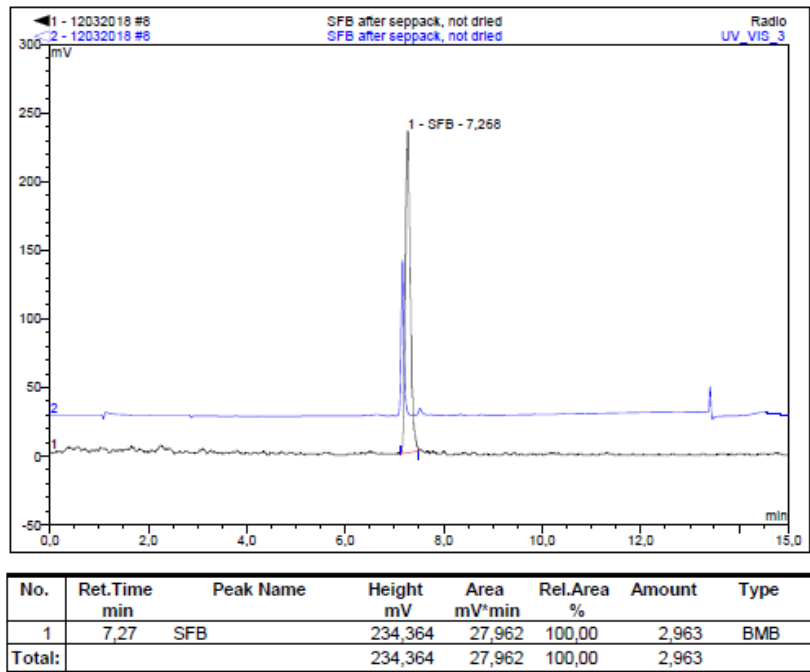
SFB crude reaction mixture, Method 1



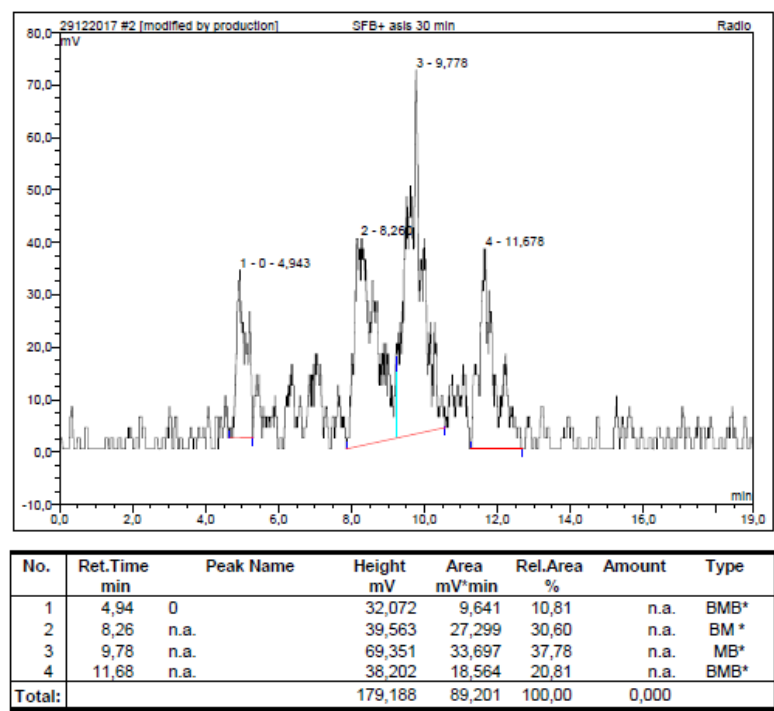
SFB purified, Method 1



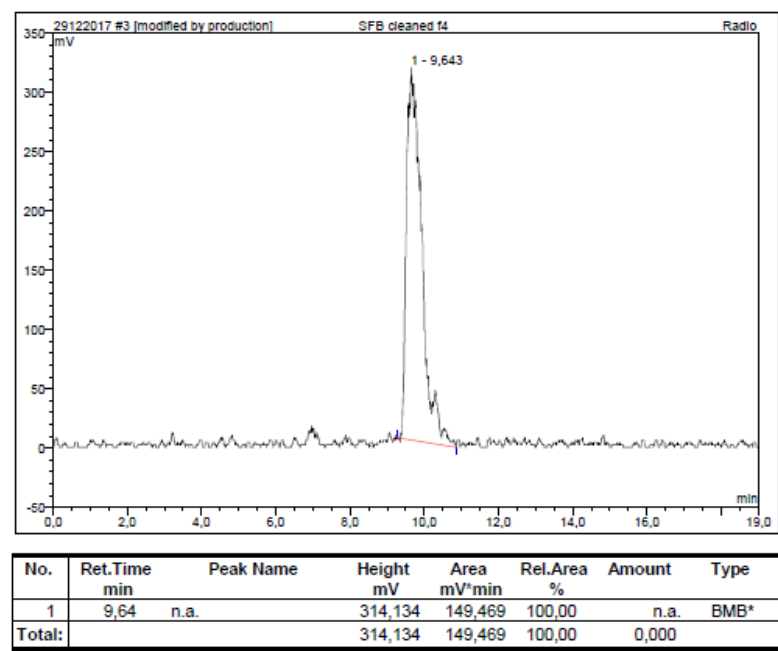
SFB purified and spiked with cold reference, Method 1



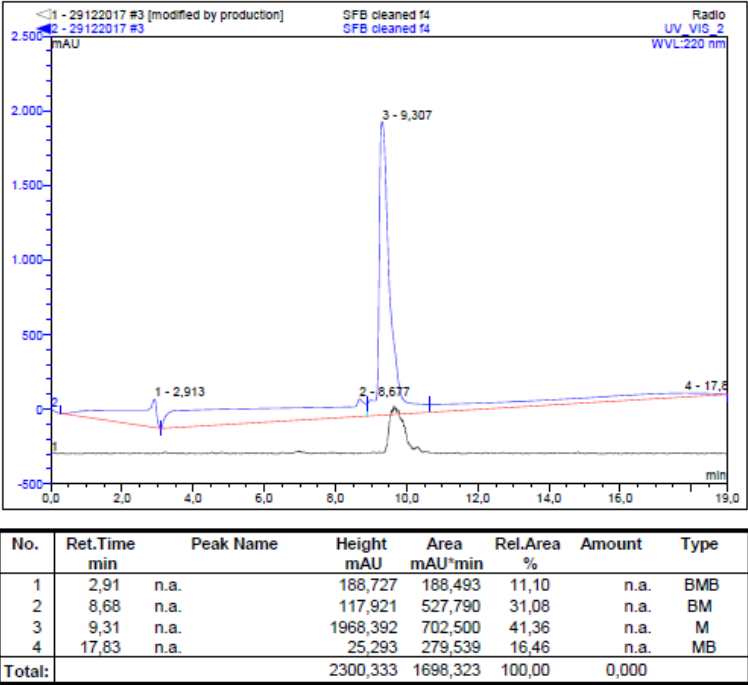
SFB reacted with [¹⁸F]FVIIai, Method 2



SFB reacted with [¹⁸F]FVIIai and purified by PD10, Method 2

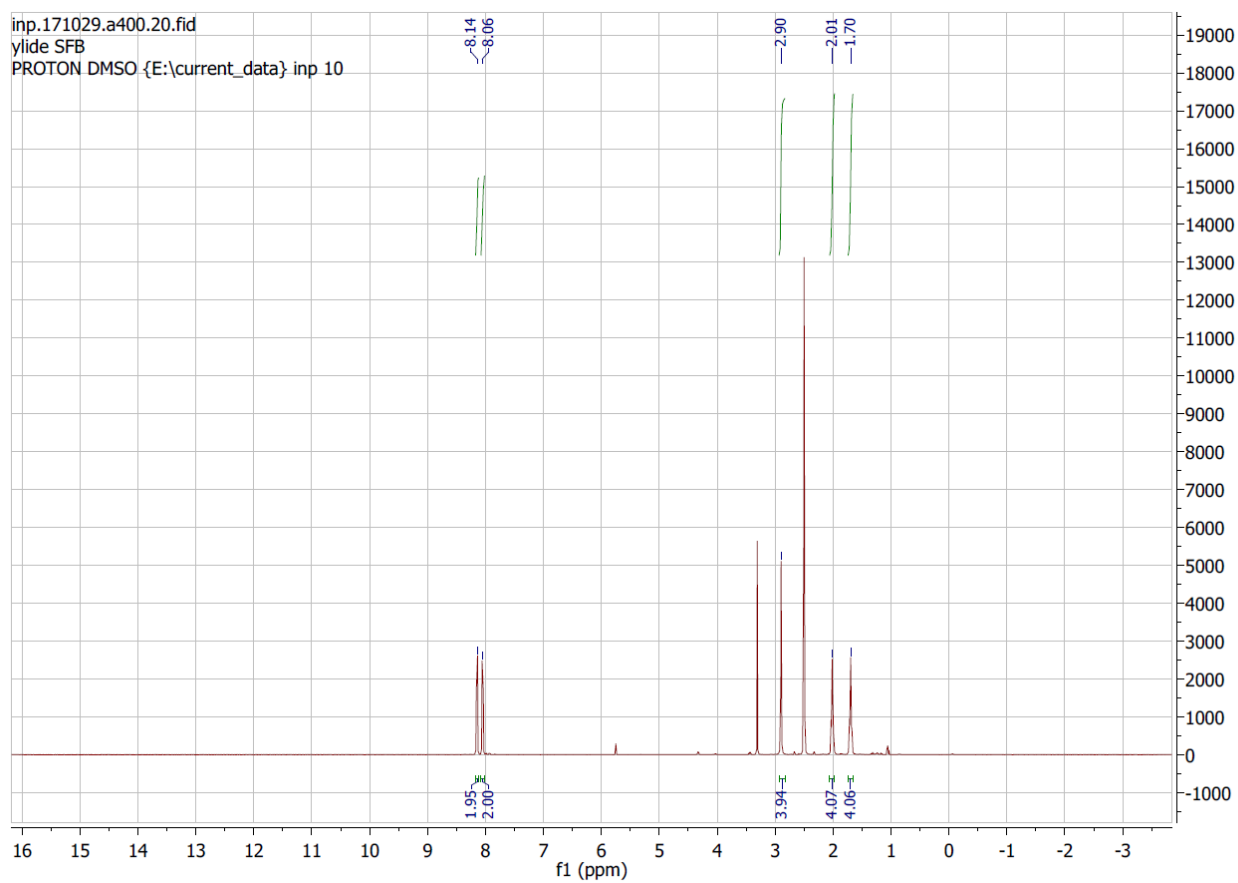


SFB reacted with ASIS and purified by PD10 overlay, Method 2

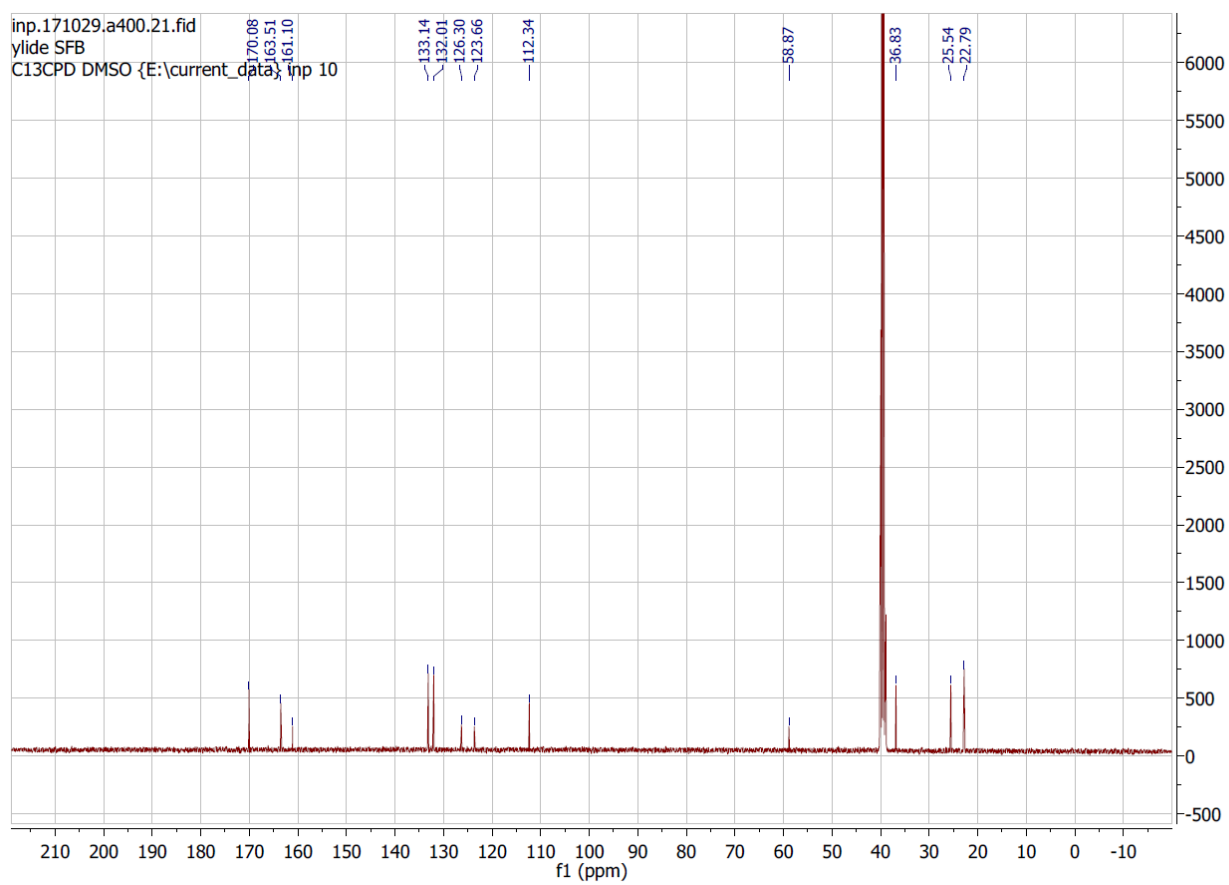


NMR spectra

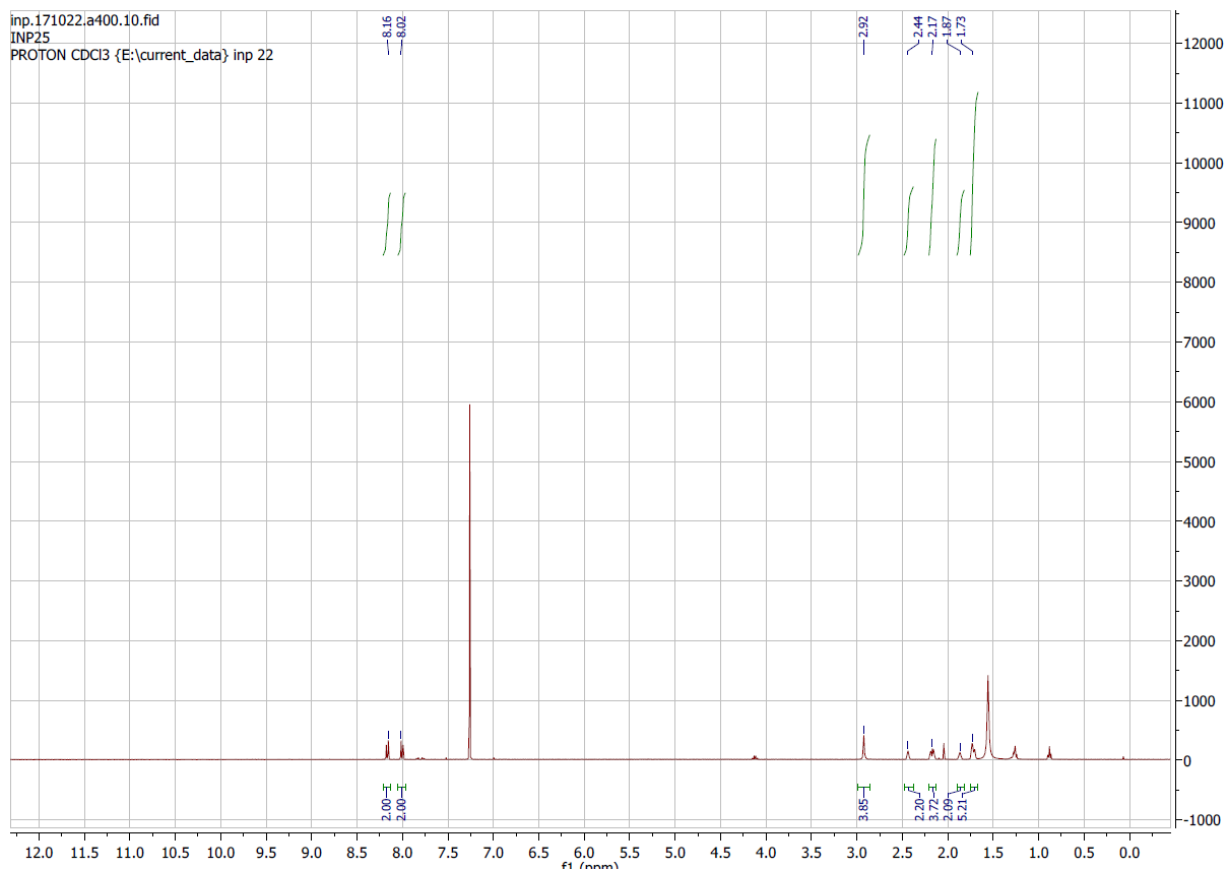
^1H -NMR of SFB-precursor (4a)



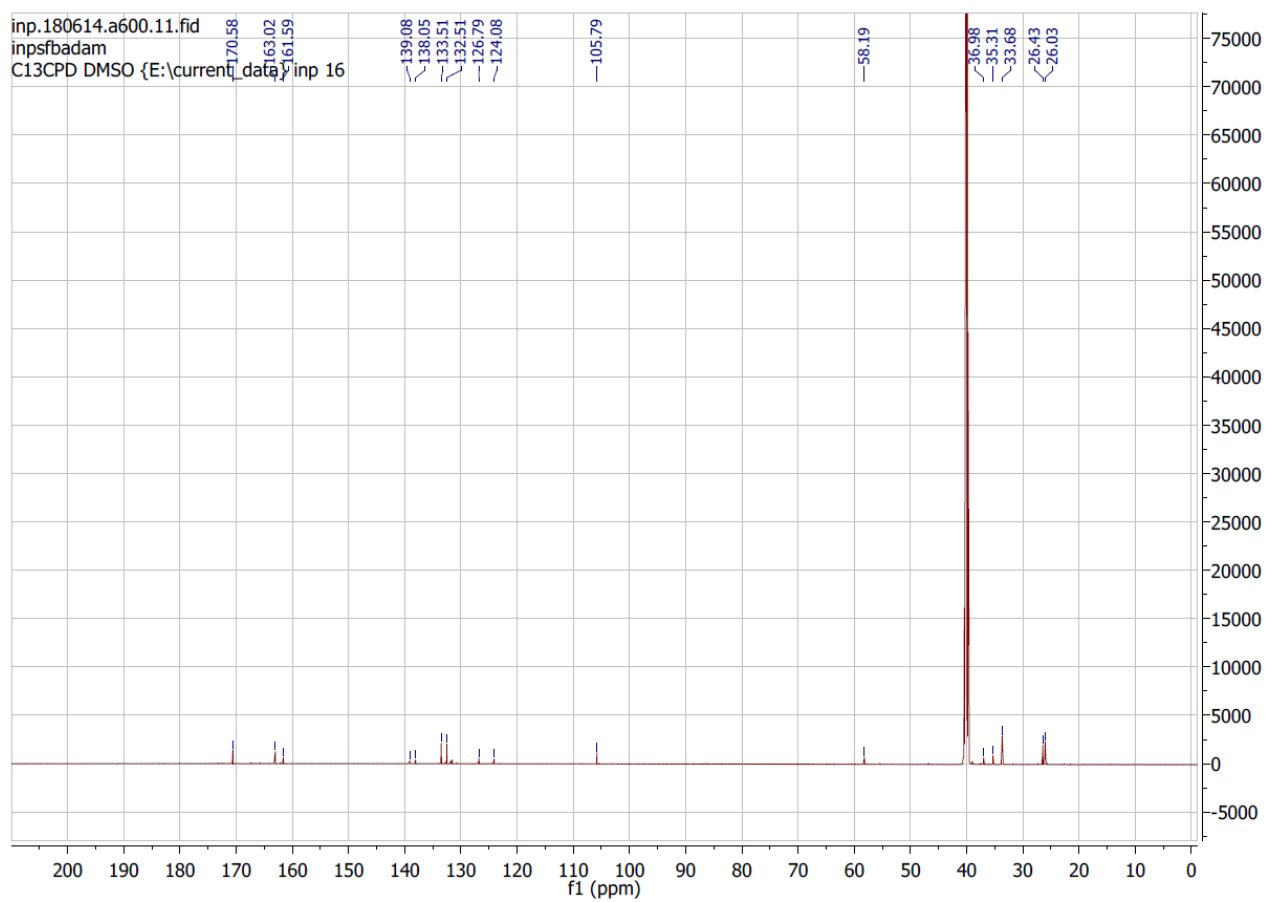
^{13}C -NMR of SFB-precursor (4a)



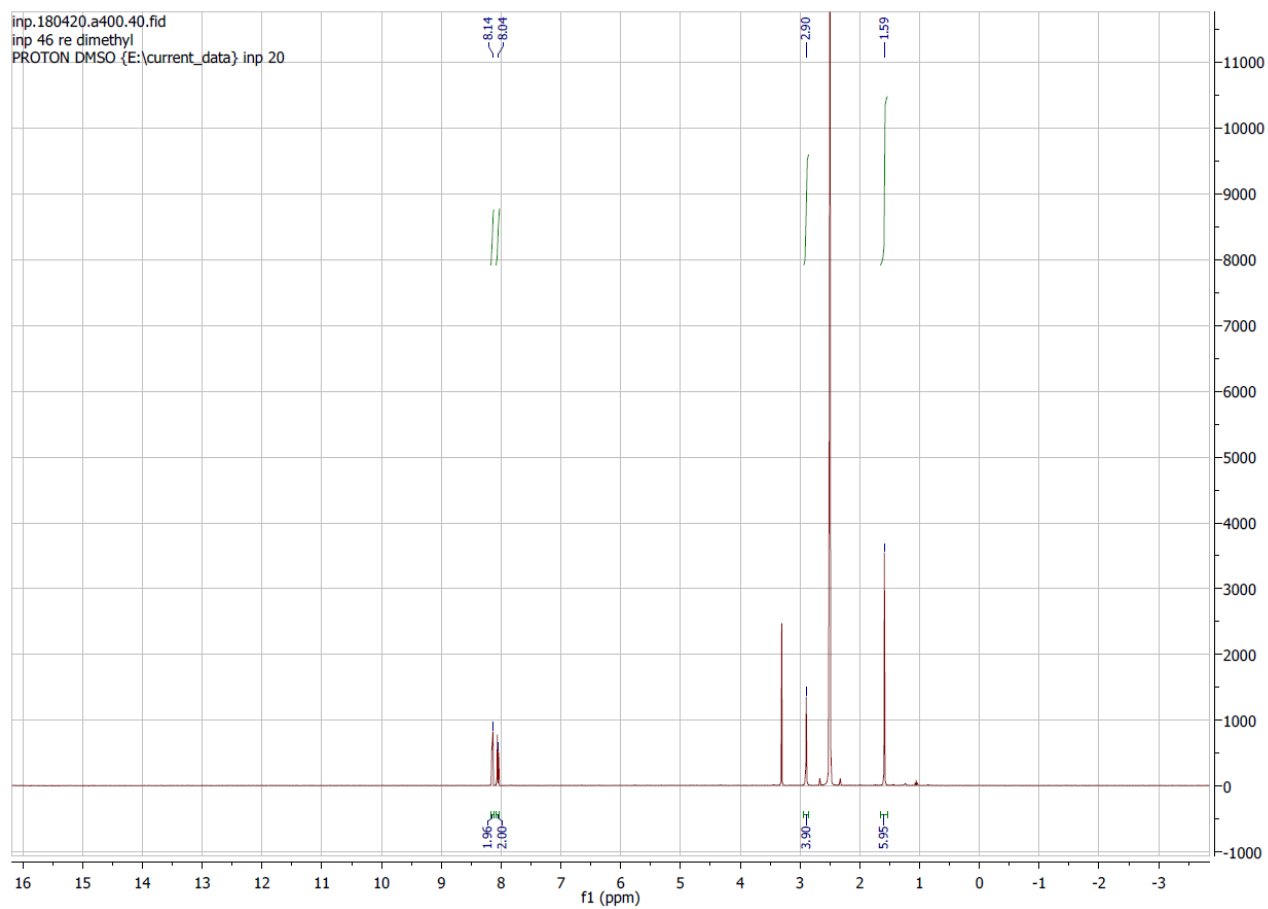
¹H-NMR of SFB-precursor (4b)



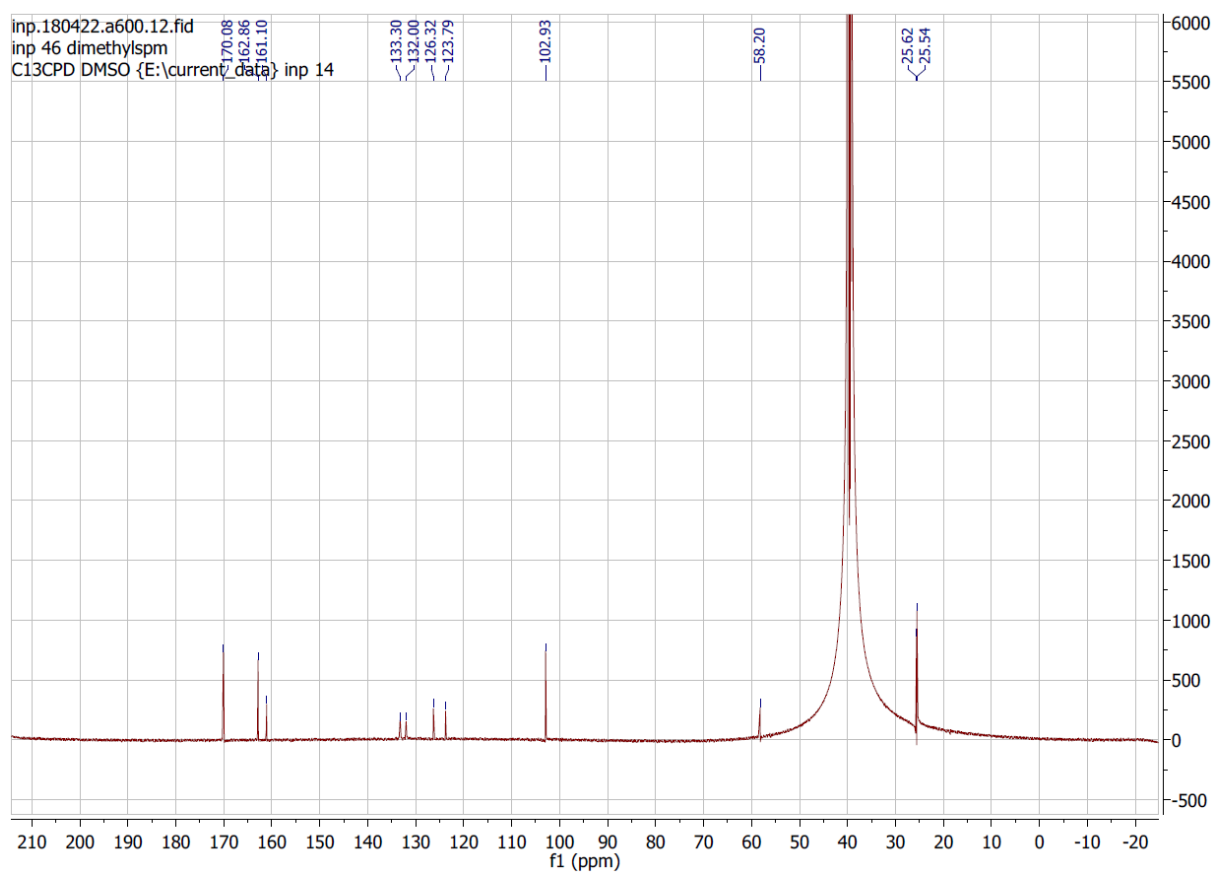
^{13}C -NMR of SFB-precursor (4b)



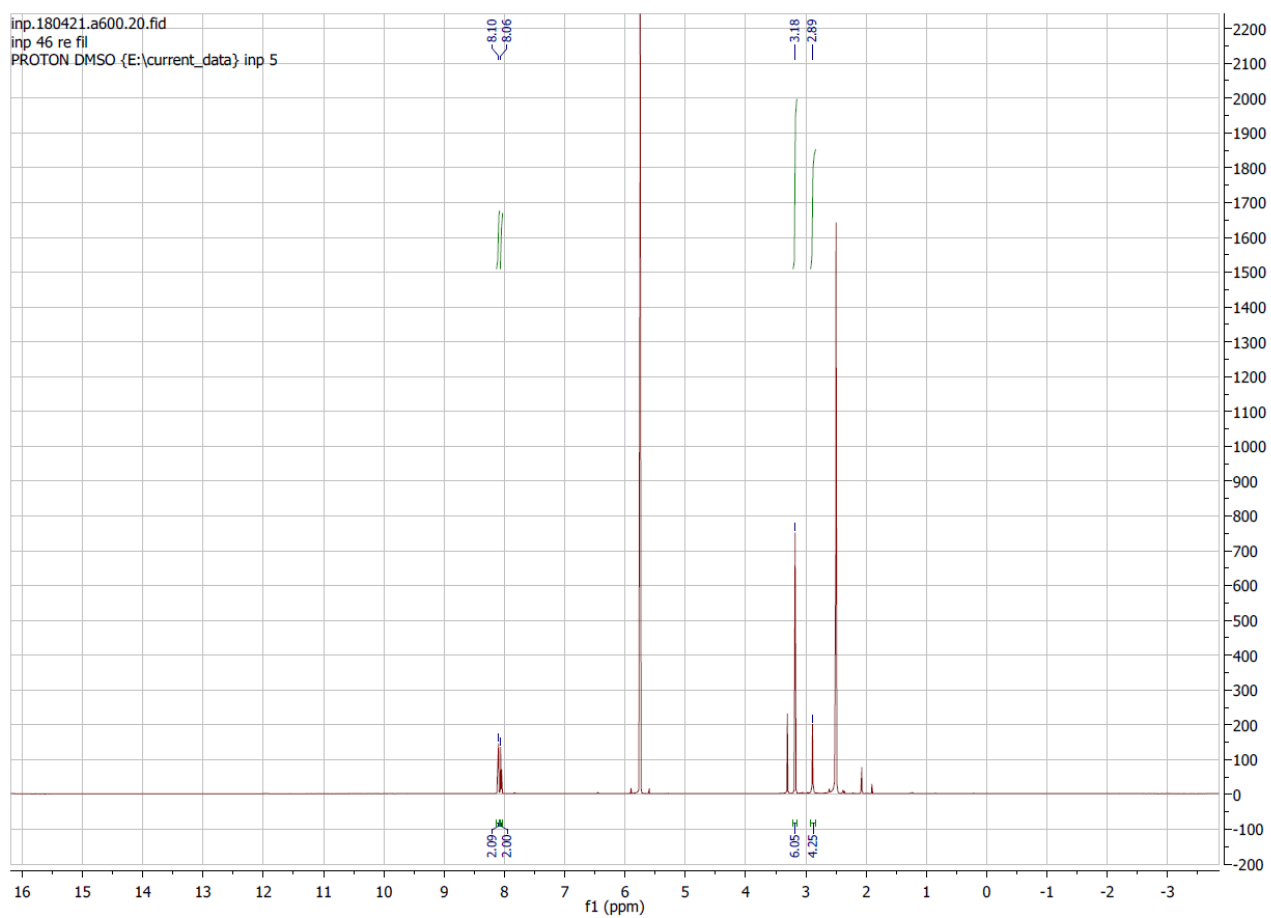
¹H-NMR of SFB-precursor (4c)



^{13}C -NMR of SFB-precursor (4c)



¹H-NMR of SFB-precursor (4d)



^{13}C -NMR of SFB-precursor (4d)

