

SUPPLEMENTARY INFORMATION

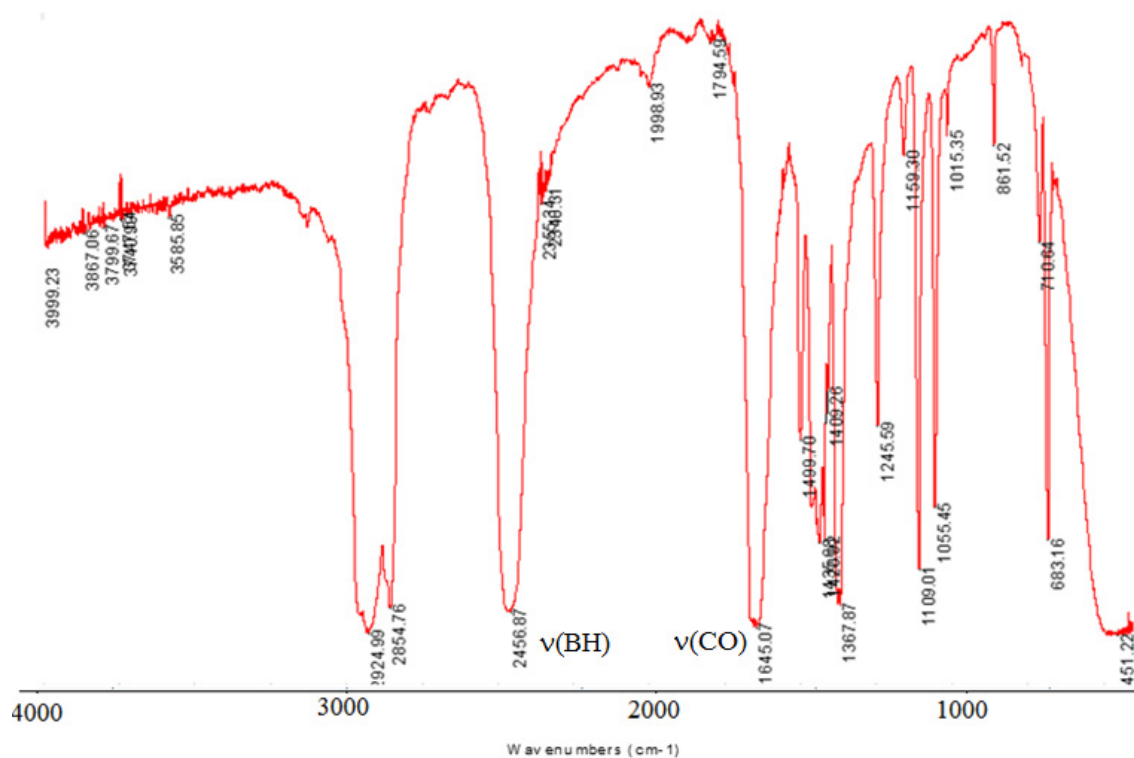
for

A new approach to the synthesis of nanocrystalline cobalt boride in the course of the thermal reduction of cobalt complexes with boron cluster anions

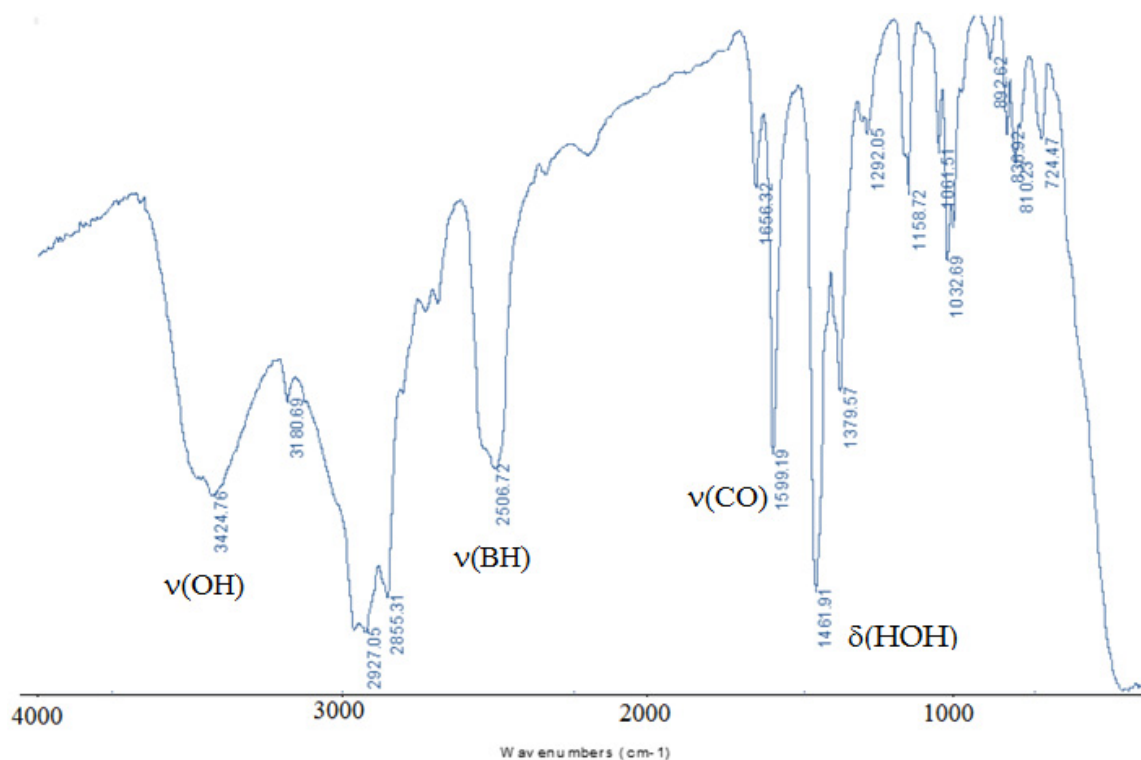
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Table S1. Crystal data and structure refinement for complex **1**.

Compound	1
Empirical formula	C ₁₈ H ₅₄ B ₁₂ CoN ₆ O ₆
Formula weight	639.32
Temperature/K	150.15
Crystal system	triclinic
Space group	P-1
a/Å	9.5857(5)
b/Å	18.9576(11)
c/Å	19.1748(11)
α/°	89.9010(10)
β/°	85.0580(10)
γ/°	89.9280(10)
Volume/Å ³	3471.5(3)
Z	4
ρ _{calc} /cm ³	1.223
μ/mm ⁻¹	0.535
F(000)	1356.0
Radiation	MoKα (λ = 0.71073)
Index ranges	-13 ≤ h ≤ 13 -25 ≤ k ≤ 25 -25 ≤ l ≤ 26
Reflections collected	38452
Independent reflections	18096 [R _{int} = 0.0531, R _{sigma} = 0.0906]
Data/restraints/parameters	18096/0/770
Goodness-of-fit on F ²	0.984
Final R indexes [I ≥ 2σ (I)]	R ₁ = 0.0510, wR ₂ = 0.1001
Final R indexes [all data]	R ₁ = 0.1092, wR ₂ = 0.1201



(a)



(b)

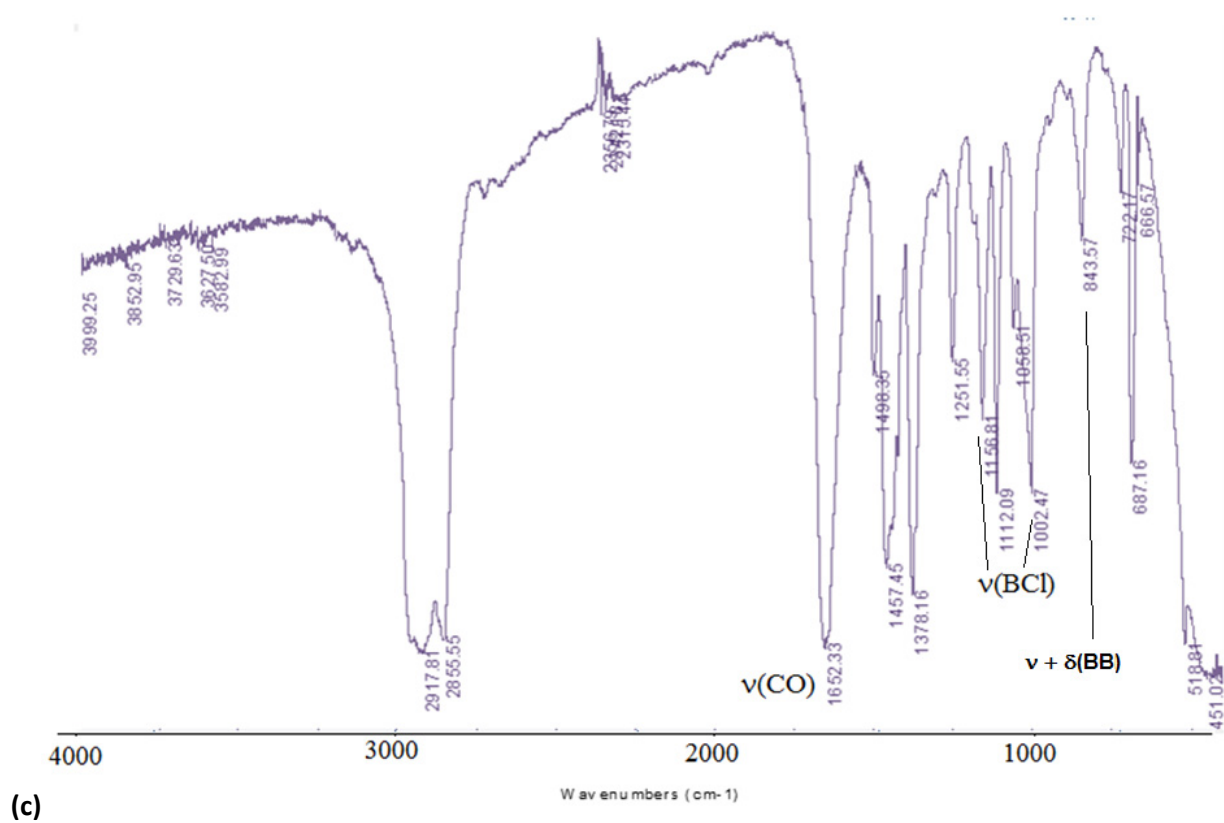


Figure S1. IR spectra of compounds (a) **1**, (b) **2**·3H₂O, and (c) **3**.

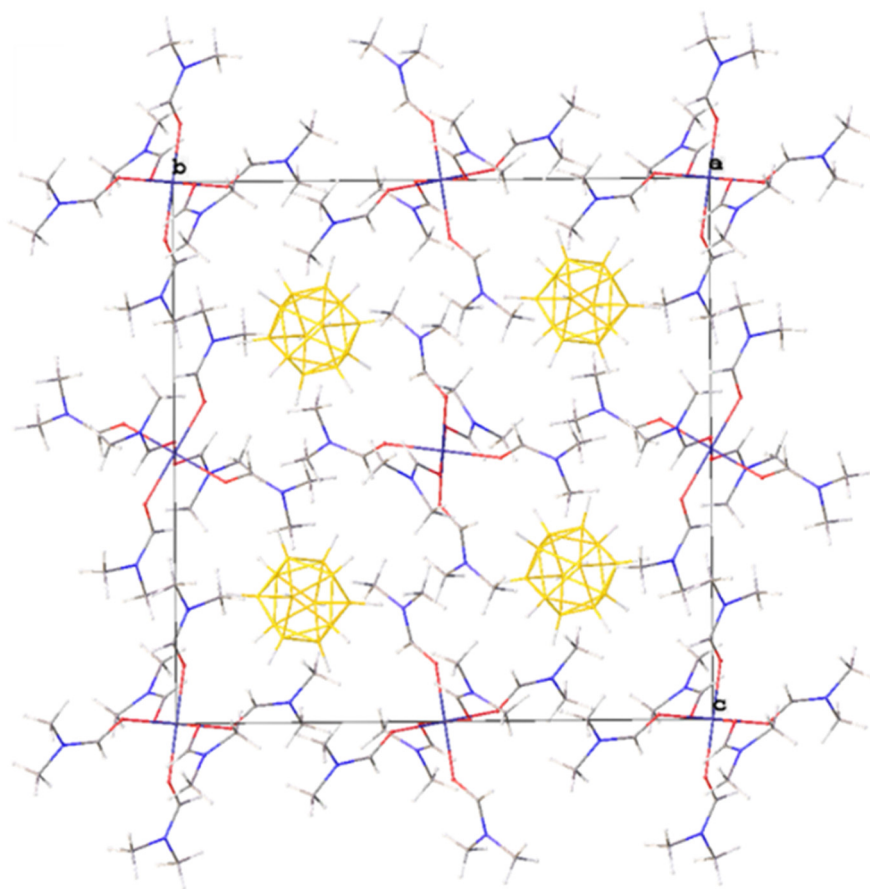
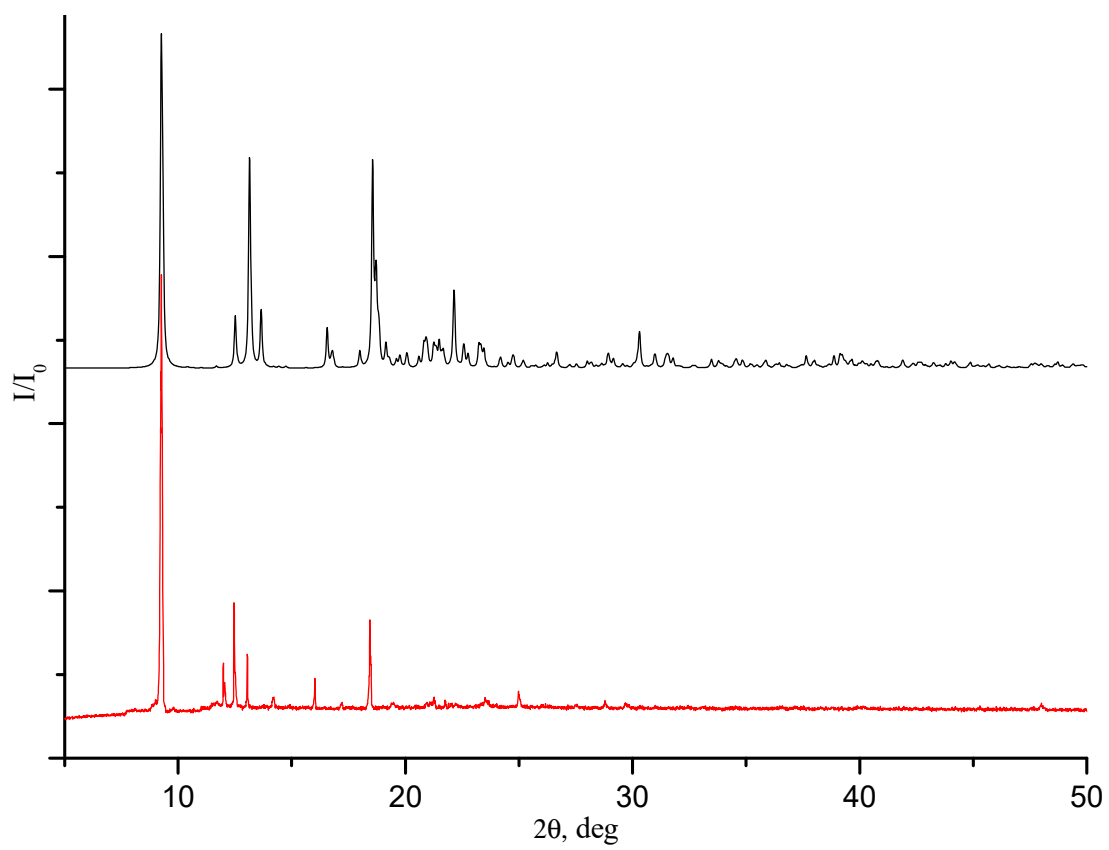
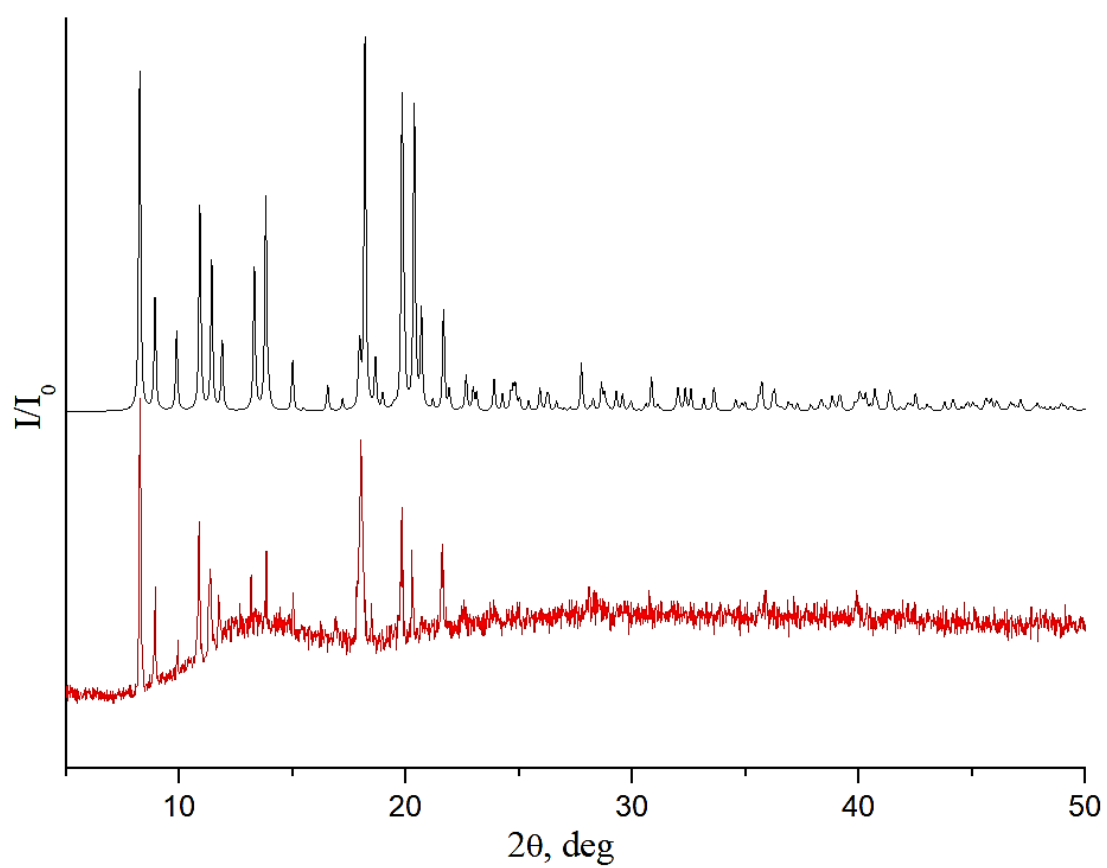


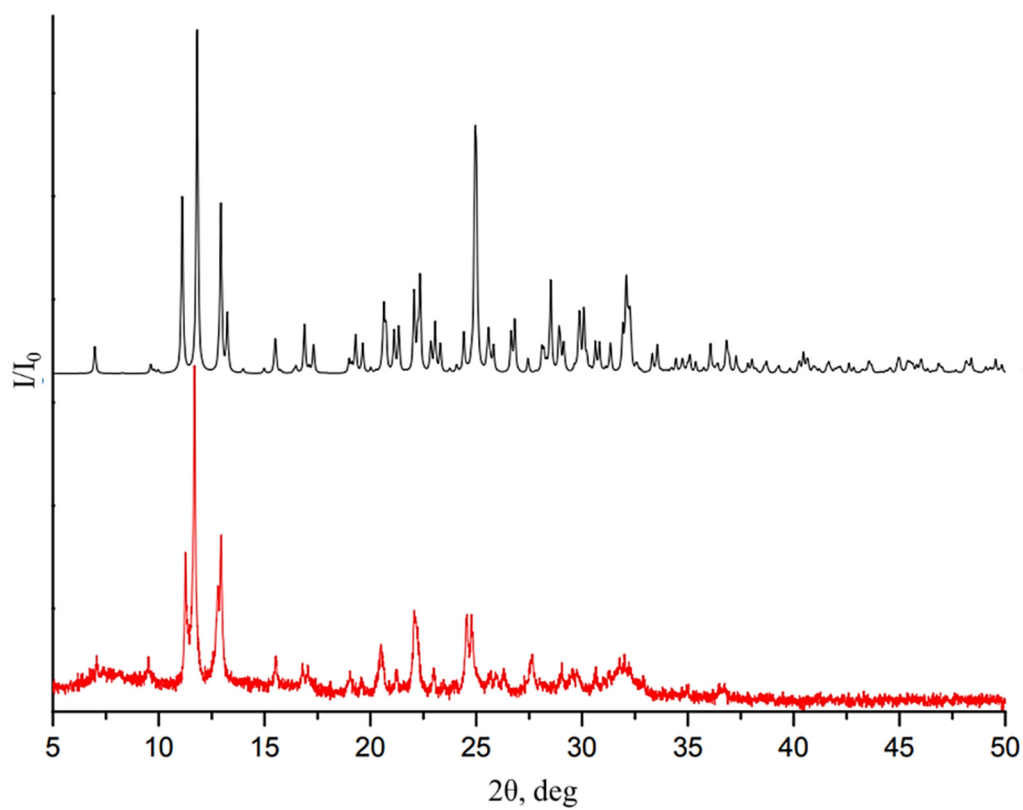
Figure S2. Projection of structure of complex **1** along the *a* axis.



(a)



(b)



(c)

Figure S3. Calculated (black) and experimental (red) X-ray powder diffraction patterns for (a) compound 1, (b) compound 2, and (c) compound 3.

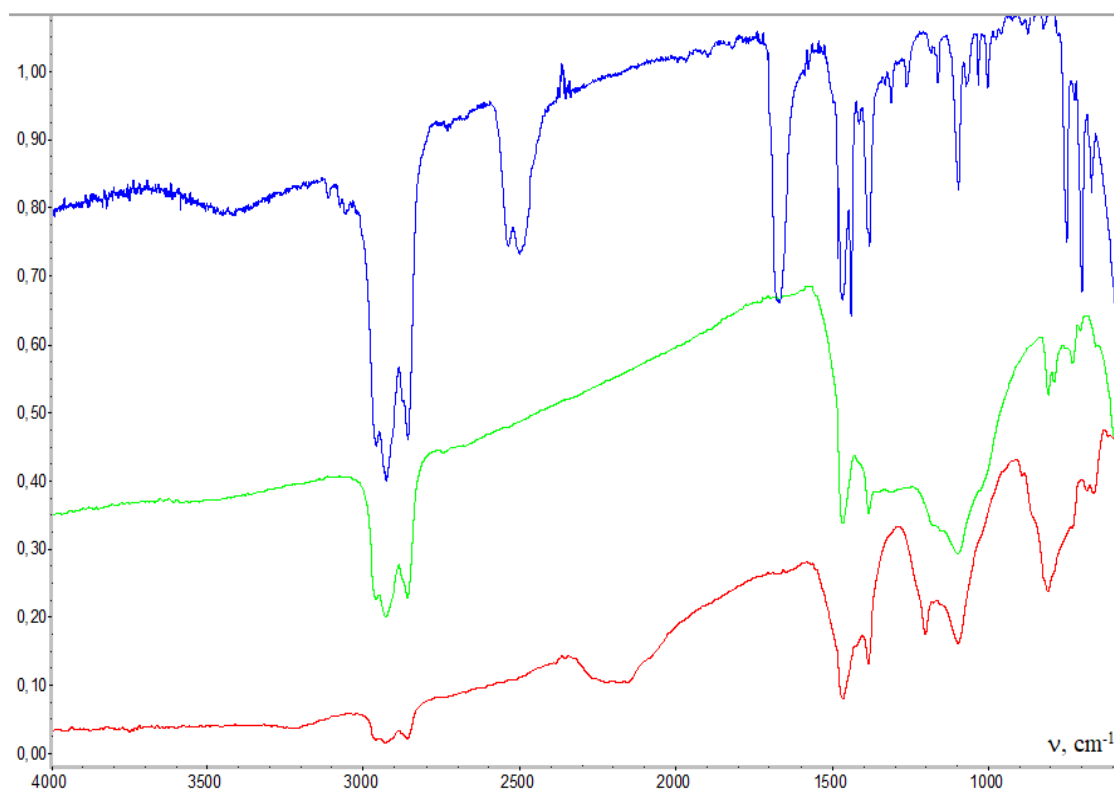


Figure S4. IR spectra of complex $2 \cdot n\text{H}_2\text{O}$ (blue), sample 2a (green) and amorphous boron (red).

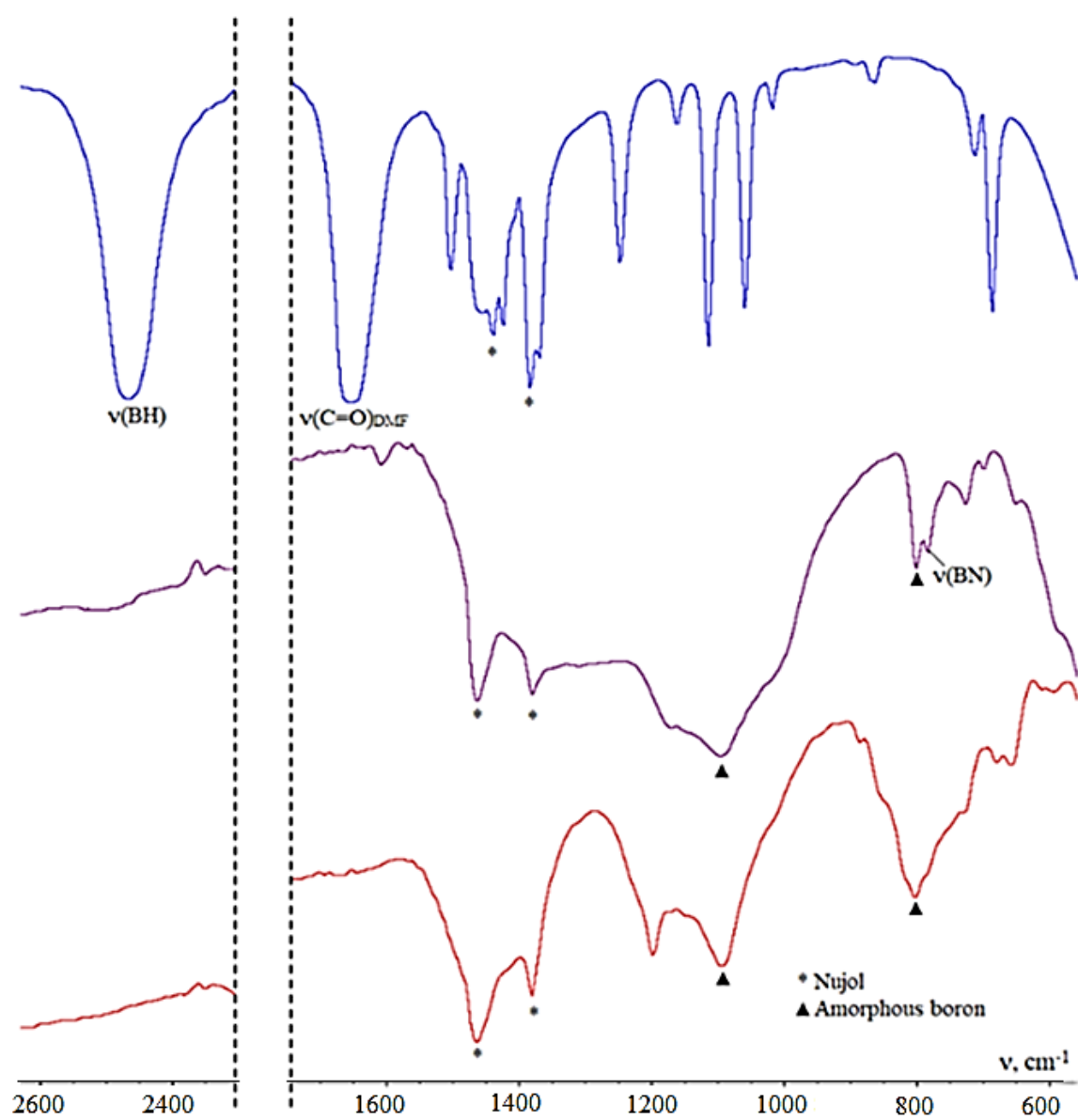


Figure S5. Fragments of IR spectra of complex **1** (blue), sample **1a** (purple) and amorphous boron (red).

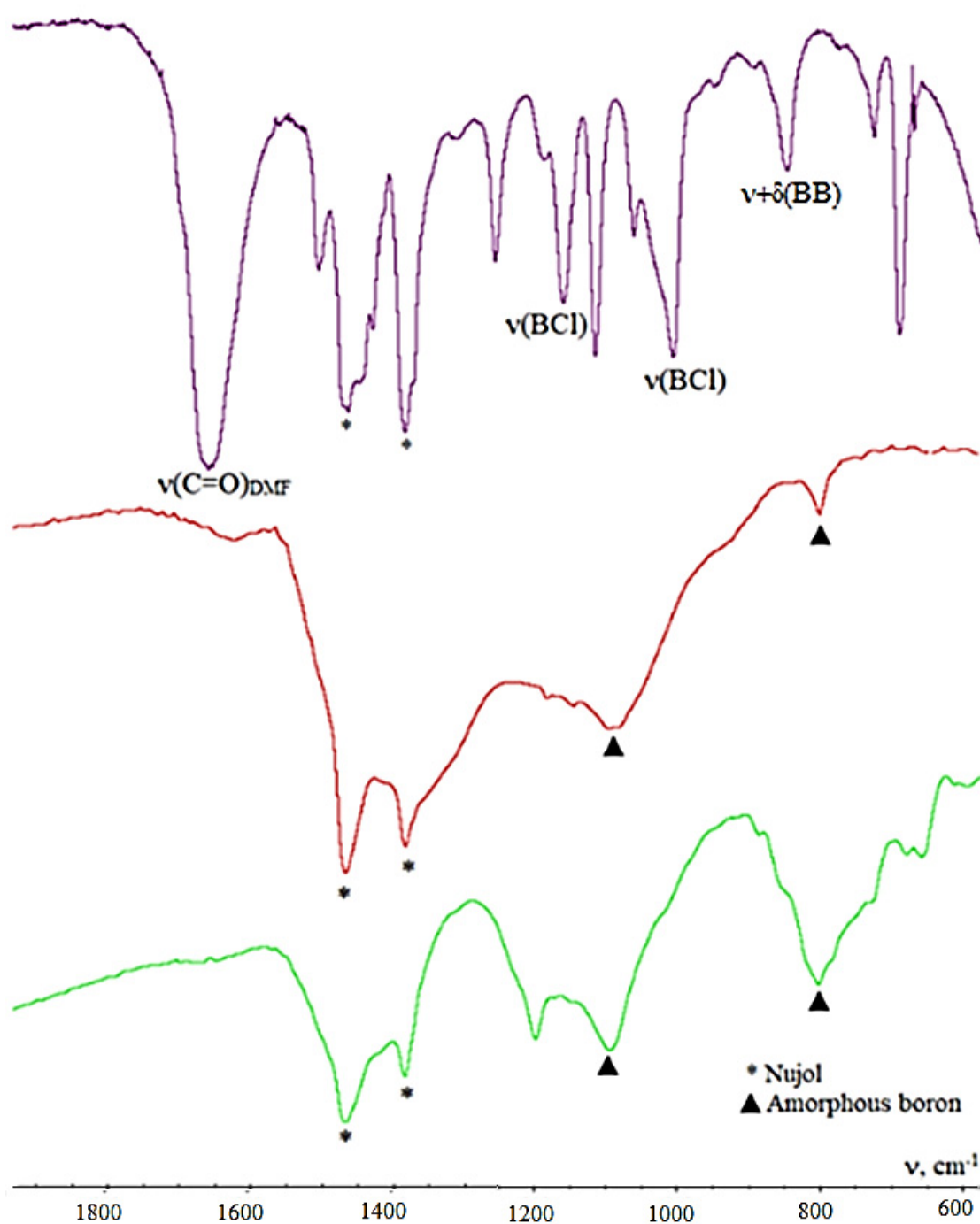


Figure S6. Fragments of IR spectra of complex **3** (purple), sample **3a** (red) and amorphous boron (green).

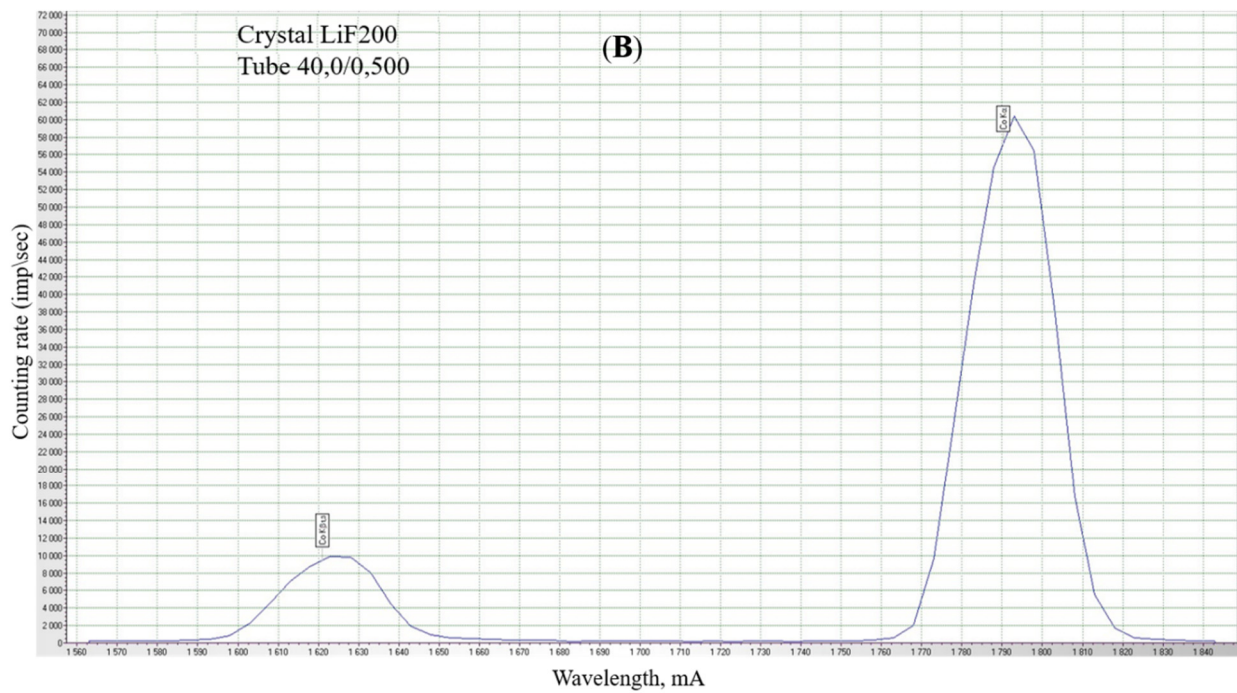
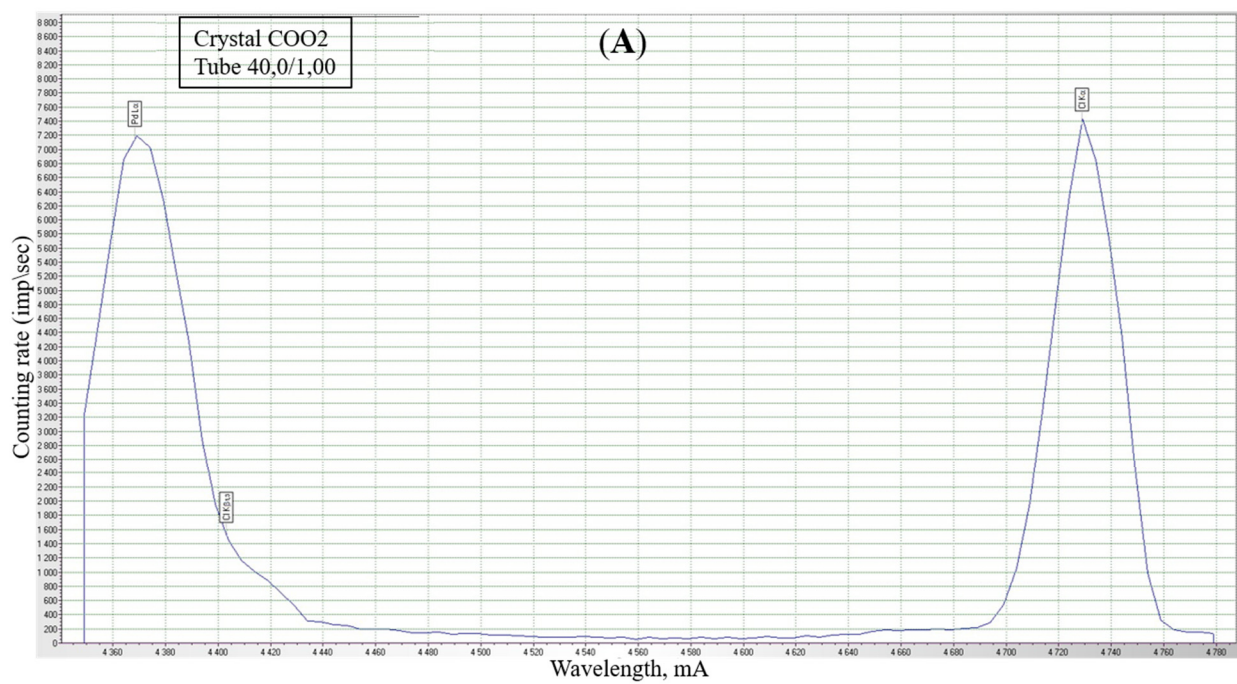


Figure S7. X-ray spectral fluorescence spectrum of sample **3a**: (A) determination of Cl, (B) determination of Co.