

Formation, Characterization, and Bonding of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂], *cis-trans*-[Pt₃Cl₆{Te(CH₂)₆}₄], and *cis-trans*-[Pt₄Cl₈{Te(CH₂)₆}₄]: Experimental and DFT Study [†]

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[†] Dedicated to Professor J. Derek Woollins on the occasion of his retirement.

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Supporting Information

1. NMR spectroscopy

1.1 ¹H NMR Spectra of the mixture of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂]

Figure 1S. ¹H NMR spectra of the mixture of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂] (a) 30 s and (b) 1.2 h after the dissolution of *cis*-[PtCl₂{Te(CH₂)₆}₂].

Figure 2S. Fluxionality of the Te(CH₂)₆ ligand in *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂].

1.2 ¹²⁵Te{¹H} and ¹⁹⁵Pt{¹H} NMR Spectra of the reaction mixture of *cis*-[PtCl₂(NCPh)₂] and Te(CH₂)₆

Figure 3S. (a) The ¹²⁵Te{¹H} NMR and (b) the ¹⁹⁵Pt{¹H} NMR spectra from the reaction solution of *cis*-[PtCl₂(NCPh)₂] and Te(CH₂)₆.

2. X-ray crystallography

Table 1S. Crystal data and refinement details for the X-ray structure determinations of *cis*-[PtCl₂{Te(CH₂)₆}₂] (**1_{cis}**), *cis-trans*-[Pt₃Cl₆{Te(CH₂)₆}₄] · 1¼CH₂Cl₂ (**2** · 1¼CH₂Cl₂), *cis-trans*-[Pt₄Cl₈{Te(CH₂)₆}₄] · 4CDCl₃ (**3** · 4CDCl₃), and [PtCl₂{S(O)(CD₃)₂}{Te(CH₂)₆}] (**4**).

Table 2S. Selected bond lengths (Å) and angles (°) in *cis-trans*-[Pt₃Cl₆{Te(CH₂)₆}₄] · 1¼CH₂Cl₂ (**2** · 1¼CH₂Cl₂) and *cis-trans*-[Pt₄Cl₈{Te(CH₂)₆}₄] · 4CDCl₃ (**3** · 4CDCl₃).

3. DFT Computations

3.1 Optimum geometries

Table 3S. Atomic coordinates (Å) of the PBE0-D3/def2-TZVP optimized species discussed in this contribution.

Table 4S. PBE0-D3/def2-TZVP optimized geometries of the [Pt_nCl_{2n}{Te(CH₂)₆}_m] (*n* = 1-4; *m* = 2-4).

Table 5S. Total energies of optimized species at PBE0-D3/def2-TZVP level of theory in vacuum (Hartree).

Table 6S. Total energies of optimized species at PBE0-D3/def2-TZVP level of theory in dichloromethane (Hartree).

3.2 Secondary bonding interactions

Figure 4S. The Pt...Pt interactions result in the square-planar coordination plane to become slightly concave in (a) **1_{cis}**, (b) **2**, and (c) **3**.

3.3 Formation energetics

Table 7S. Gibbs formation energies calculated at PBE0-D3/def2-TZVP level of theory (kJ mol⁻¹) in dichloromethane.

1. NMR spectroscopy

1.1 ^1H NMR spectra of the mixture of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂]

The ^1H spectra of the mixture of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂] has been shown in Figure 1S both after 30 s and 1.2 h after the dissolution of *cis*-[PtCl₂{Te(CH₂)₆}₂]. The multiplets were assigned to the *cis*- and *trans*-form based on relative intensities.

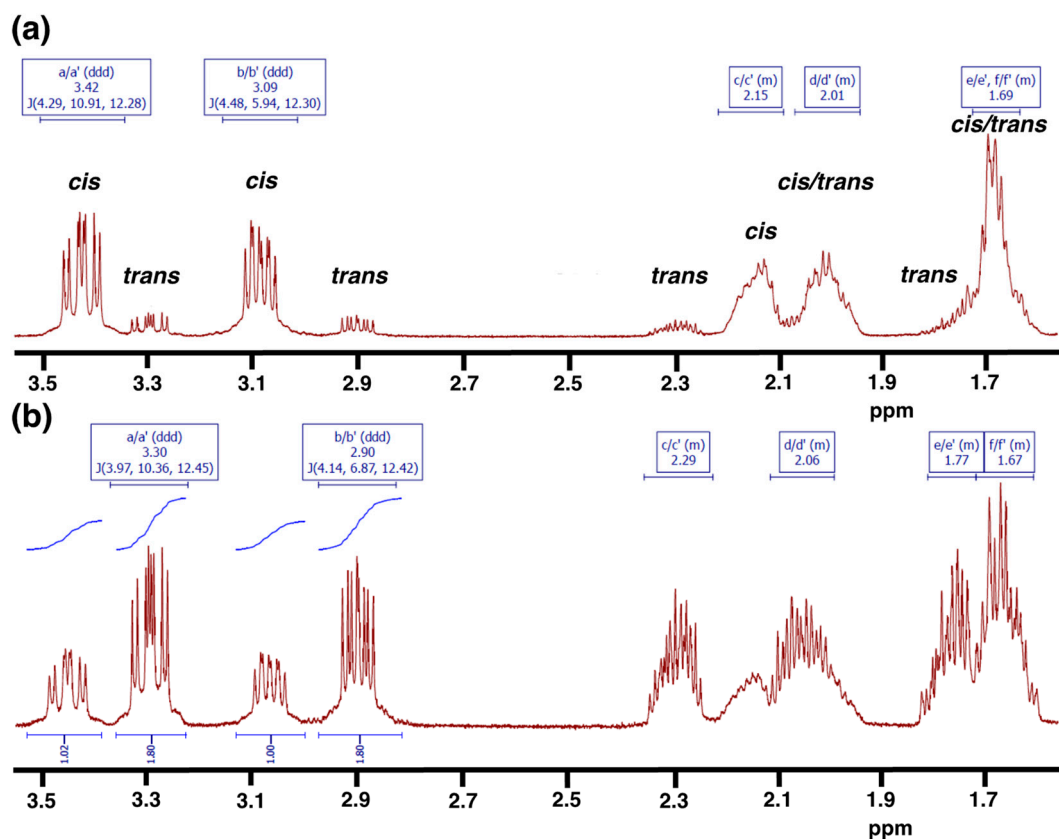


Figure S1. ^1H NMR spectra of the mixture of *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂] (a) 30 s and (b) 1.2 h after the dissolution of *cis*-[PtCl₂{Te(CH₂)₆}₂].

The ^1H NMR spectra of both species exhibit very complex multiplets. Since tellurium donates its *np* lone pair to the Pt-Te bond, the C-Te-C plane will be approximately perpendicular to this bond. The Te(CH₂)₆ ring is fluxional interconverting between two conformations, appearing planar in the NMR time scale (see Figure 2S). Therefore, the protons on the side of the ring pointing towards the platinum center in Te(CH₂)₆ aren't chemically equivalent to the ones pointing away from the metal center giving rise to a complicated [AA'BB'CC'DD'EE'FF'] spin system. In case of the free ligand, significantly simpler [ABC]₄ spin system is observed [1].

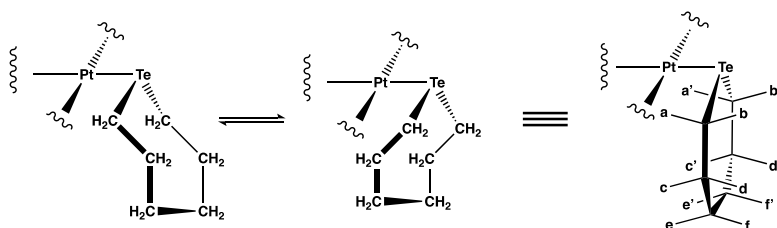


Figure 2S. Fluxionality of the Te(CH₂)₆ ligand in *cis*- and *trans*-[PtCl₂{Te(CH₂)₆}₂].

The chemical shifts of the geminal methylene protons closest to tellurium lie at lowest field and are sufficiently different for the second order effects to play only a minor role, which can therefore be ignored in the estimation of coupling constants between the protons *a* and *b* (closest to tellurium), and between the protons of position *a/b* and *c/d* (see Figure 1S). The geminal coupling constants of $|^2J_{aa'}|=12.3$ Hz (*cis* isomer) and $|^2J_{aa'}|=12.4$ Hz (*trans* isomer) are rather typical for cycloalkane derivatives. The signal assignment to hydrogens *a* and *a'* is not unambiguous but that does not preclude the assignment of these two multiplets to the pair *a/a'*. This model is consistent with the slight lowfield shift all α -methylene signals experience compared to the free ligand [1].

It can clearly be seen in Figure 1S(a) that immediately after the dissolution, *cis*-[PtCl₂{Te(CH₂)₆}₂] is the major species, but the relative concentration of *trans*-[PtCl₂{Te(CH₂)₆}₂] rapidly increases as a function of time [Figure 1S(b)].

1.2 $^{125}\text{Te}\{^1\text{H}\}$ and $^{195}\text{Pt}\{^1\text{H}\}$ NMR spectra of the reaction mixture of *cis*-[PtCl₂(NPh)₂] and Te(CH₂)₆

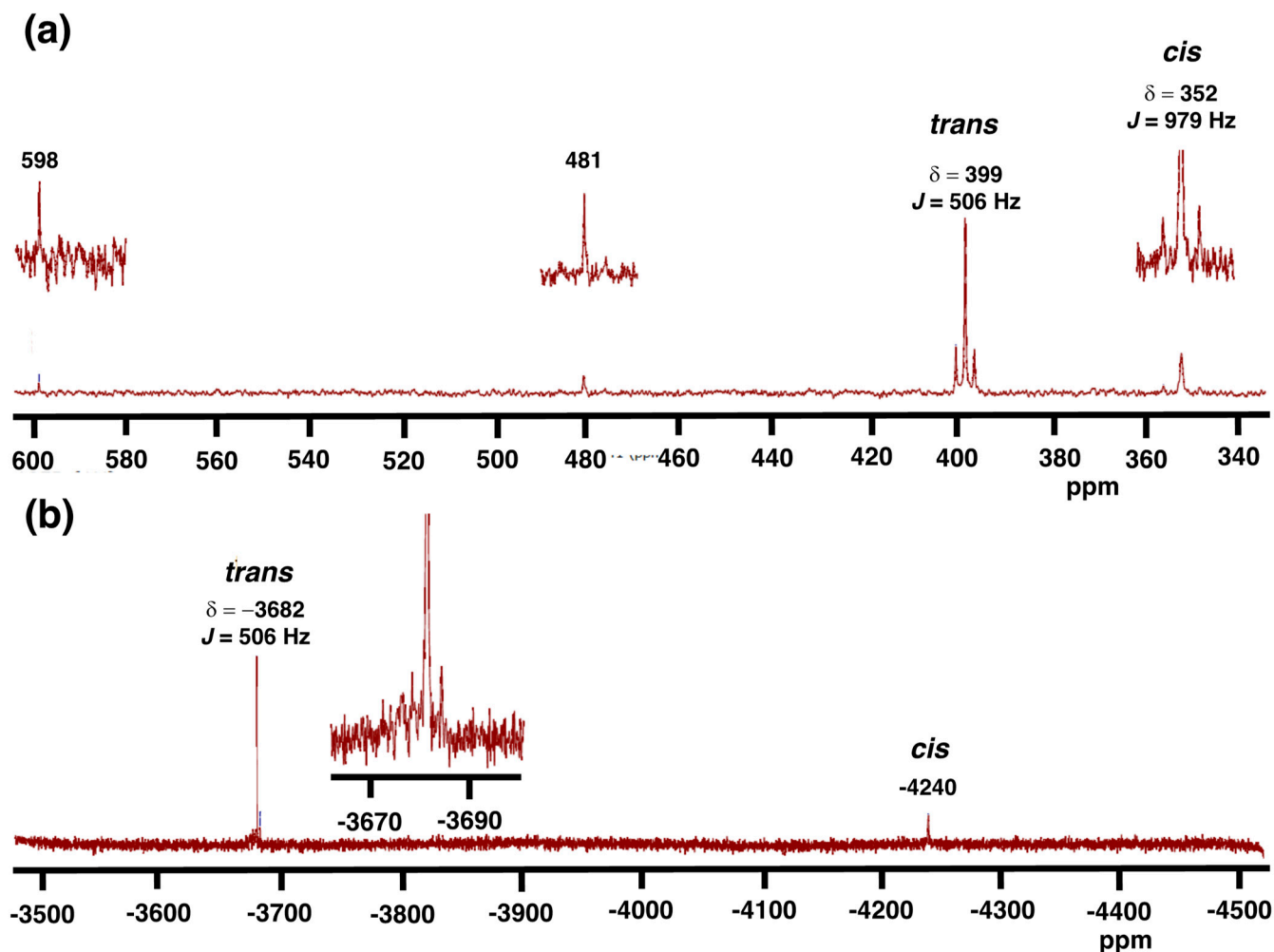


Figure S3. (a) The $^{125}\text{Te}\{^1\text{H}\}$ NMR and (b) the $^{195}\text{Pt}\{^1\text{H}\}$ NMR spectra from the reaction solution of *cis*-[PtCl₂(NPh)₂] and Te(CH₂)₆.

2. X-ray crystallography

Table S1. Crystal data and refinement details for the X-ray structure determinations of *cis*-[PtCl₂{Te(CH₂)₆}₂] (**1_{cis}**), *cis-trans*-[Pt₃Cl₆{Te(CH₂)₆}₄] · 1¼CH₂Cl₂ (**2** · 1¼CH₂Cl₂), *cis-trans*-[Pt₄Cl₈{Te(CH₂)₆}₄] · 4CDCl₃ (**3** · 4CDCl₃), and [PtCl₂{S(O)(CD₃)₂}{Te(CH₂)₆}] (**4**).

| Compound | 1_{cis} | 2 · 1¼CH ₂ Cl ₂ | 3 · 4CDCl ₃ ^a | 4 ^a |
|---|---|---|--|--|
| formula | C ₁₂ H ₂₄ Cl ₂ PtTe ₂ | C _{25.5} H ₅₁ Cl ₉ Pt ₃ Te ₄ | C ₂₈ H ₅₂ Cl ₂₀ Pt ₄ Te ₄ | C ₈ H ₁₈ Cl ₂ OPtS Te |
| fw (g·mol ⁻¹) | 689.50 | 1772.38 | 2388.45 | 555.87 |
| T/°C | -140(2) | -140(2) | -140(2) | -140(2) |
| crystal system | orthorhombic | triclinic | monoclinic | monoclinic |
| space group | <i>Pbca</i> | <i>Pī</i> | <i>C2/c</i> | <i>P2₁/c</i> |
| <i>a</i> /Å | 12.4443(3) | 12.5826(4) | 12.7358(3) | 9.9274(2) |
| <i>b</i> /Å | 12.8290(3) | 12.6959(4) | 18.6821(5) | 16.3202(4) |
| <i>c</i> /Å | 21.2611(4) | 26.7884(8) | 24.5098(6) | 9.0309(2) |
| <i>α</i> /° | 90 | 93.341(1) | 90 | 90 |
| <i>β</i> /° | 90 | 98.757(1) | 97.857(1) | 108.8660(10) |
| <i>γ</i> /° | 90 | 90.571(1) | 90 | 90 |
| <i>V</i> /Å ³ | 3394.29(13) | 4221.5(2) | 5776.9(3) | 1384.56(5) |
| <i>Z</i> | 8 | 4 | 4 | 4 |
| <i>ρ</i> (g·cm ⁻³) | 2.699 | 2.789 | 2.746 | 2.667 |
| <i>μ</i> (cm ⁻¹) | 119.33 | 132.15 | 125.85 | 127.07 |
| measured data | 19066 | 35285 | 32246 | 15959 |
| data with <i>I</i> > 2σ(<i>I</i>) | 3337 | 14448 | 6312 | 3092 |
| unique data (<i>R</i> _{int}) | 3887/0.0784 | 18271/0.0417 | 6614/0.0381 | 3167/0.0263 |
| <i>wR</i> ₂ (all data, on <i>F</i> ²) ^b | 0.0825 | 0.2461 | 0.0939 | 0.0339 |
| <i>R</i> ₁ (<i>I</i> > 2σ(<i>I</i>)) ^b | 0.0433 | 0.0787 | 0.0411 | 0.0150 |
| <i>S</i> ^c | 1.093 | 1.067 | 1.114 | 1.119 |
| Res. Dens./e·Å ⁻³ | 1.993/-1.591 | 6.215/-1.735 | 5.661/-2.911 | 0.612/-0.903 |
| absorpt method | multi-scan | multi-scan | multi-scan | multi-scan |
| absorpt corr <i>T</i> _{min} / _{max} | 0.3791/0.7456 | 0.2703/0.7456 | 0.4244/0.7456 | 0.5337/0.7456 |
| CCDC No. | 2298160 | 2298161 | 2298162 | 2301073 |

^a D atoms have been designated as H. ^b Definition of the *R* indices: $R_1 = (\sum ||F_o| - |F_c||) / \sum |F_o|$; $wR_2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$ with $w^{-1} = 2(F_o^2) + (aP)^2 + bP$; $P = [2F_c^2 + \text{Max}(F_o^2)]/3$; ^c $S = \{\sum [w(F_o^2 - F_c^2)^2] / (N_o - p..)\}^{1/2}$.

Table S2. Selected bond lengths (Å) and angles (°) in *cis-trans*-[Pt₃Cl₆{Te(CH₂)₆}₄] · 1¼CH₂Cl₂ (**2** · 1¼CH₂Cl₂), and *cis-trans*-[Pt₄Cl₈{Te(CH₂)₆}₄] · 4CDCl₃ (**3** · 4CDCl₃).

| 2 · 1¼CH ₂ Cl ₂ | | | | | | 3 · 4CDCl ₃ | | | |
|---------------------------------------|----------------|----------------|----------------|----------------|----------------|------------------------|-----------|-------------|-----------|
| A ^a | | | B ^a | | | | | | |
| | A ^a | B ^a | | A ^a | B ^a | | | | |
| Pt1-Te1 | 2.5178(15) | 2.5140(15) | Te1-Pt1-Te4 | 92.26(5) | 92.13(5) | Pt1-Te1 | 2.5045(7) | Te1-Pt1-Te2 | 94.86(2) |
| Pt1-Te4 | 2.5170(16) | 2.5219(16) | Te1-Pt1-Cl1 | 178.57(13) | 178.57(14) | Pt1-Te2 | 2.5226(6) | Te1-Pt1-Cl1 | 88.16(6) |
| Pt2-Te1 | 2.5774(15) | 2.5768(17) | Te1-Pt1-Cl2 | 88.88(13) | 88.249(2) | Pt2-Te1 | 2.5577(6) | Te1-Pt1-Cl2 | 175.20(7) |
| Pt2-Te2 | 2.5560(15) | 2.568(2) | Te4-Pt1-Cl1 | 89.16(14) | 89.35(14) | Pt2-Te2 | 2.5546(6) | Te2-Pt1-Cl1 | 173.72(7) |
| Pt3-Te3 | 2.5608(17) | 2.5626(15) | Te4-Pt1-Cl2 | 177.75(14) | 178.45(15) | Pt1-Cl1 | 2.321(2) | Te2-Pt1-Cl2 | 87.06(7) |
| Pt3-Te4 | 2.5640(15) | 2.5635(14) | Cl1-Pt1-Cl2 | 88.88(14) | 88.28(15) | Pt1-Cl2 | 2.309(3) | Cl1-Pt1-Cl2 | 90.35(9) |
| Pt1-Cl1 | 2.331(5) | 2.330(5) | Te1-Pt2-Te2 | 176.56(5) | 173.87(6) | Pt2-Cl3 | 2.305(2) | Te1-Pt2-Te2 | 178.92(2) |
| Pt1-Cl2 | 2.311(6) | 2.328(6) | Te1-Pt2-Cl3 | 84.33(13) | 83.57(13) | Pt2-Cl4 | 2.302(2) | Te1-Pt2-Cl3 | 93.43(6) |
| Pt2-Cl3 | 2.307(5) | 2.313(5) | Te1-Pt2-Cl4 | 95.79(14) | 94.97(13) | Pt2···Pt2 | 3.0765(6) | Te1-Pt2-Cl4 | 86.29(6) |
| Pt2-Cl4 | 2.305(6) | 2.316(5) | Te2-Pt2-Cl3 | 94.74(13) | 94.32(13) | Te1···Cl3 | 3.527(2) | Te2-Pt2-Cl3 | 85.52(6) |
| Pt3-Cl5 | 2.309(5) | 2.325(6) | Te2-Pt2-Cl4 | 85.16(13) | 87.29(13) | Te2···Cl4 | 3.556(2) | Te2-Pt2-Cl4 | 94.77(6) |
| Pt3-Cl6 | 2.313(6) | 2.320(6) | Cl3-Pt2-Cl4 | 179.8(2) | 178.3(2) | | | Cl3-Pt2-Cl4 | 175.77(7) |
| Pt2···Pt3 | 3.1499(13) | 3.1170(11) | Te3-Pt3-Te4 | 178.14(6) | 179.28(5) | | | | |
| Te1···Cl6 | 3.481(7) | 3.512(5) | Te3-Pt3-Cl5 | 85.47(14) | 86.53(15) | | | | |
| Te2···Cl5 | 3.922(6) | 3.964(6) | Te3-Pt3-Cl6 | 95.62(16) | 95.26(13) | | | | |
| Te3···Cl4 | 3.594(6) | 3.435(6) | Te4-Pt3-Cl5 | 92.72(14) | 92.78(15) | | | | |
| Te4···Cl3 | 3.590(5) | 3.590(5) | Te4-Pt3-Cl6 | 86.16(16) | 85.44(13) | | | | |
| | | | Cl5-Pt3-Cl6 | 174.70(14) | 175.3(2) | | | | |

^a The asymmetric unit contains two independent molecules that have been denoted by A and B.

3. DFT Computations¹

3.1 Optimum geometries

Table S3. Atomic coordinates (Å) of the PBE0-D3/def2-TZVP optimized species discussed in this contribution.

PhCN C_{2v} E=-324.228470092 a.u.

| | | | | | | | |
|---|-------------|--------------|--------------|---|-------------|--------------|--------------|
| C | 0.000000000 | 0.000000000 | 0.604835000 | H | 0.000000000 | 2.141688000 | -2.014339000 |
| C | 0.000000000 | 1.209525000 | -0.089890000 | H | 0.000000000 | 0.000000000 | -3.248749000 |
| C | 0.000000000 | 1.203159000 | -1.473355000 | H | 0.000000000 | -2.141688000 | -2.014339000 |
| C | 0.000000000 | 0.000000000 | -2.165113000 | H | 0.000000000 | -2.141617000 | 0.461061000 |
| C | 0.000000000 | -1.203159000 | -1.473355000 | C | 0.000000000 | 0.000000000 | 2.031910000 |
| C | 0.000000000 | -1.209525000 | -0.089890000 | N | 0.000000000 | 0.000000000 | 3.183492000 |
| H | 0.000000000 | 2.141617000 | 0.461061000 | | | | |

Te(CH₂)₆ C₁ E=-503.704254999 a.u.

| | | | | | | | |
|----|--------------|--------------|--------------|---|--------------|--------------|--------------|
| Te | -1.210147000 | -0.092563000 | -0.101255000 | C | 2.574441000 | -0.570581000 | -0.086381000 |
| C | -0.142466000 | 1.767622000 | 0.146456000 | H | 3.411818000 | -0.895108000 | -0.712723000 |
| H | -0.249894000 | 2.288014000 | -0.807522000 | H | 2.966064000 | -0.528994000 | 0.937879000 |
| H | -0.672345000 | 2.335706000 | 0.910062000 | C | 1.501412000 | -1.646643000 | -0.166658000 |
| C | 1.320721000 | 1.610532000 | 0.506850000 | H | 1.189195000 | -1.776675000 | -1.207944000 |
| H | 1.411655000 | 1.145454000 | 1.493685000 | H | 1.962004000 | -2.595596000 | 0.137101000 |
| H | 1.726403000 | 2.622252000 | 0.625499000 | C | 0.273519000 | -1.425347000 | 0.698719000 |
| C | 2.172907000 | 0.840225000 | -0.509266000 | H | 0.529626000 | -1.090026000 | 1.704770000 |
| H | 1.652078000 | 0.801754000 | -1.472985000 | H | -0.295924000 | -2.351006000 | 0.806406000 |
| H | 3.093740000 | 1.402635000 | -0.687287000 | | | | |

cis-[PtCl₂(NCPH)₂] C_s E=-1688.06794514 a.u.

| | | | | | | | |
|----|--------------|--------------|-------------|----|--------------|-------------|-------------|
| C | -1.448801000 | -3.655410000 | 0.000000000 | Cl | 2.748639000 | 1.625916000 | 0.000000000 |
| C | -2.843133000 | -3.708134000 | 0.000000000 | N | -0.148166000 | 1.446649000 | 0.000000000 |
| C | -3.472666000 | -4.938992000 | 0.000000000 | C | -0.765694000 | 2.413804000 | 0.000000000 |
| C | -2.721177000 | -6.106566000 | 0.000000000 | C | -1.448598000 | 3.655540000 | 0.000000000 |
| C | -1.333906000 | -6.049577000 | 0.000000000 | C | -2.842946000 | 3.708161000 | 0.000000000 |
| C | -0.687992000 | -4.827309000 | 0.000000000 | C | -0.687892000 | 4.827497000 | 0.000000000 |
| H | -3.417083000 | -2.790020000 | 0.000000000 | C | -3.472576000 | 4.938963000 | 0.000000000 |
| H | -4.554686000 | -4.988661000 | 0.000000000 | H | -3.416816000 | 2.789996000 | 0.000000000 |
| H | -3.220818000 | -7.068098000 | 0.000000000 | C | -1.333906000 | 6.049717000 | 0.000000000 |
| H | -0.751248000 | -6.962619000 | 0.000000000 | H | 0.393210000 | 4.759767000 | 0.000000000 |
| H | 0.393104000 | -4.759480000 | 0.000000000 | C | -2.721176000 | 6.106599000 | 0.000000000 |
| C | -0.765938000 | -2.413644000 | 0.000000000 | H | -4.554600000 | 4.988549000 | 0.000000000 |
| N | -0.148356000 | -1.446524000 | 0.000000000 | H | -0.751316000 | 6.962802000 | 0.000000000 |
| Pt | 1.166735000 | -0.000007000 | 0.000000000 | H | -3.220893000 | 7.068092000 | 0.000000000 |
| Cl | 2.748412000 | -1.626185000 | 0.000000000 | | | | |

trans-[PtCl₂(NCPH)₂] D_{2h} E=-1688.07709399 a.u.

| | | | | | | | |
|---|-------------|--------------|-------------|----|-------------|--------------|--------------|
| C | 0.000000000 | 0.000000000 | 4.484879000 | C | 0.000000000 | 0.000000000 | 3.068206000 |
| C | 0.000000000 | 1.214614000 | 5.173137000 | N | 0.000000000 | 0.000000000 | 1.922951000 |
| C | 0.000000000 | 1.205326000 | 6.555534000 | Pt | 0.000000000 | 0.000000000 | 0.000000000 |
| C | 0.000000000 | 0.000000000 | 7.244835000 | Cl | 0.000000000 | -2.307595000 | 0.000000000 |
| C | 0.000000000 | -1.205326000 | 6.555534000 | Cl | 0.000000000 | 2.307595000 | 0.000000000 |
| C | 0.000000000 | -1.214614000 | 5.173137000 | N | 0.000000000 | 0.000000000 | -1.922951000 |
| H | 0.000000000 | 2.144470000 | 4.618226000 | C | 0.000000000 | 0.000000000 | -3.068206000 |
| H | 0.000000000 | 2.142823000 | 7.097943000 | C | 0.000000000 | 0.000000000 | -4.484879000 |
| H | 0.000000000 | 0.000000000 | 8.328483000 | C | 0.000000000 | -1.214614000 | -5.173137000 |
| H | 0.000000000 | -2.142823000 | 7.097943000 | C | 0.000000000 | 1.214614000 | -5.173137000 |
| H | 0.000000000 | -2.144470000 | 4.618226000 | C | 0.000000000 | -1.205326000 | -6.555534000 |

| | | | |
|---|-------------|--------------|--------------|
| H | 0.000000000 | -2.144470000 | -4.618226000 |
| C | 0.000000000 | 1.205326000 | -6.55534000 |
| H | 0.000000000 | 2.144470000 | -4.618226000 |
| C | 0.000000000 | 0.000000000 | -7.244835000 |

| | | | |
|---|-------------|--------------|--------------|
| H | 0.000000000 | -2.142823000 | -7.097943000 |
| H | 0.000000000 | 2.142823000 | -7.097943000 |
| H | 0.000000000 | 0.000000000 | -8.328483000 |

cis-[PtCl₂{Te(CH₂)₆}₂] C₁ E=-2047.06703240 a.u.

| | | | |
|----|--------------|--------------|--------------|
| Pt | -0.146560000 | -1.095428000 | 0.099196000 |
| Te | 1.532599000 | 0.680053000 | -0.550223000 |
| Te | -2.219050000 | -0.074417000 | -0.933730000 |
| Cl | -1.724300000 | -2.744233000 | 0.613361000 |
| Cl | 1.557618000 | -2.281023000 | 1.147292000 |
| C | 2.318033000 | 1.174401000 | 1.378314000 |
| H | 1.462031000 | 1.550590000 | 1.939703000 |
| H | 2.576882000 | 0.203201000 | 1.797820000 |
| C | 3.464858000 | 2.172227000 | 1.332542000 |
| H | 4.025965000 | 2.079756000 | 2.270148000 |
| H | 3.054136000 | 3.186251000 | 1.322355000 |
| C | 4.421598000 | 2.047322000 | 0.153133000 |
| H | 3.863665000 | 2.241474000 | -0.772732000 |
| H | 5.148153000 | 2.862974000 | 0.219700000 |
| C | 5.192048000 | 0.740076000 | 0.012122000 |
| H | 5.845451000 | 0.609768000 | 0.881429000 |
| H | 5.859166000 | 0.849909000 | -0.850067000 |
| C | 4.372891000 | -0.536219000 | -0.164548000 |
| H | 5.052469000 | -1.331090000 | -0.493132000 |
| H | 3.967926000 | -0.891236000 | 0.783303000 |
| C | 3.252532000 | -0.469616000 | -1.182676000 |

| | | | |
|---|--------------|--------------|--------------|
| H | 2.839265000 | -1.458772000 | -1.370607000 |
| H | 3.569919000 | -0.015892000 | -2.124201000 |
| C | -1.805695000 | 2.019445000 | -1.171011000 |
| H | -1.004865000 | 2.017458000 | -1.914054000 |
| H | -2.684170000 | 2.458469000 | -1.648274000 |
| C | -1.401339000 | 2.790625000 | 0.074782000 |
| H | -0.815591000 | 2.151988000 | 0.746843000 |
| H | -0.729820000 | 3.597640000 | -0.239314000 |
| C | -2.557514000 | 3.414072000 | 0.845472000 |
| H | -3.207767000 | 3.942179000 | 0.136931000 |
| H | -2.144831000 | 4.186098000 | 1.502725000 |
| C | -3.402168000 | 2.486623000 | 1.707071000 |
| H | -2.764176000 | 2.008478000 | 2.459752000 |
| H | -4.099736000 | 3.117589000 | 2.266281000 |
| C | -4.212003000 | 1.404125000 | 0.995034000 |
| H | -4.620035000 | 1.788384000 | 0.052997000 |
| H | -5.086396000 | 1.177388000 | 1.615270000 |
| C | -3.492531000 | 0.084668000 | 0.792957000 |
| H | -4.188977000 | -0.738043000 | 0.635485000 |
| H | -2.862215000 | -0.181446000 | 1.639656000 |

trans-[PtCl₂{Te(CH₂)₆}₂] C₁ E=-2047.08131913 a.u.

| | | | |
|----|--------------|--------------|--------------|
| Pt | -0.000041000 | -0.741429000 | 0.000040000 |
| Te | 2.532226000 | -0.744071000 | 0.507497000 |
| Te | -2.532314000 | -0.744031000 | -0.507416000 |
| Cl | -0.231568000 | -0.765587000 | 2.303753000 |
| Cl | 0.231469000 | -0.765956000 | -2.303673000 |
| C | 2.657679000 | 1.086279000 | 1.621284000 |
| H | 2.349058000 | 0.748860000 | 2.611282000 |
| H | 3.716294000 | 1.352034000 | 1.662675000 |
| C | 1.774308000 | 2.223153000 | 1.144918000 |
| H | 1.651074000 | 2.909937000 | 1.991249000 |
| H | 0.778943000 | 1.822027000 | 0.933380000 |
| C | 2.275387000 | 3.029405000 | -0.042640000 |
| H | 1.553719000 | 3.835630000 | -0.211601000 |
| H | 3.217208000 | 3.524702000 | 0.226473000 |
| C | 2.474635000 | 2.281258000 | -1.356944000 |
| H | 2.543436000 | 3.029510000 | -2.151192000 |
| H | 1.590285000 | 1.679013000 | -1.590091000 |
| C | 3.725693000 | 1.397534000 | -1.417757000 |
| H | 4.228643000 | 1.550030000 | -2.379606000 |
| H | 4.447951000 | 1.719635000 | -0.660058000 |
| C | 3.476059000 | -0.092378000 | -1.307263000 |

| | | | |
|---|--------------|--------------|--------------|
| H | 4.405765000 | -0.660004000 | -1.364909000 |
| H | 2.796224000 | -0.441359000 | -2.086707000 |
| C | -3.476057000 | -0.092042000 | 1.307286000 |
| H | -4.405786000 | -0.659616000 | 1.365060000 |
| H | -2.796200000 | -0.440928000 | 2.086753000 |
| C | -3.725578000 | 1.397903000 | 1.417533000 |
| H | -4.447712000 | 1.719968000 | 0.659700000 |
| H | -4.228623000 | 1.550603000 | 2.379299000 |
| C | -2.474393000 | 2.281446000 | 1.356699000 |
| H | -1.590143000 | 1.679076000 | 1.589905000 |
| H | -2.543101000 | 3.029761000 | 2.150895000 |
| C | -2.274995000 | 3.029457000 | 0.042342000 |
| H | -3.216730000 | 3.524885000 | -0.226831000 |
| H | -1.553196000 | 3.835576000 | 0.211247000 |
| C | -1.774030000 | 2.223015000 | -1.145136000 |
| H | -0.778723000 | 1.821770000 | -0.933544000 |
| H | -1.650693000 | 2.909690000 | -1.991539000 |
| C | -2.657556000 | 1.086211000 | -1.621384000 |
| H | -2.348972000 | 0.748639000 | -2.611343000 |
| H | -3.716136000 | 1.352095000 | -1.662807000 |

[Pt₂Cl₄{Te(CH₂)₆}₃] C₁ E=-3590.43986476 a.u.

| | | | |
|----|--------------|--------------|--------------|
| Pt | 2.623285000 | 1.047304000 | 0.075026000 |
| Pt | -1.599956000 | -0.241632000 | -0.400069000 |
| Te | 2.249045000 | -1.444715000 | -0.326524000 |
| Te | -3.473540000 | -2.000594000 | -0.639439000 |
| Te | 0.188951000 | 1.573352000 | -0.140589000 |

| | | | |
|----|--------------|--------------|--------------|
| Cl | 2.845130000 | 3.353633000 | 0.420918000 |
| Cl | 4.871370000 | 0.717947000 | 0.462798000 |
| Cl | -2.410751000 | 0.616068000 | -2.383382000 |
| Cl | -0.727353000 | -1.107002000 | 1.567083000 |
| C | 4.055866000 | -2.014874000 | -1.322886000 |

| | | | | | | | |
|---|--------------|--------------|--------------|---|--------------|--------------|--------------|
| H | 4.024304000 | -1.483250000 | -2.275060000 | H | -4.682453000 | -0.380681000 | 3.493586000 |
| H | 4.853595000 | -1.567493000 | -0.734271000 | H | -3.470772000 | -0.356918000 | 2.238167000 |
| C | 4.182958000 | -3.518191000 | -1.506898000 | C | -4.779341000 | -2.078819000 | 2.187258000 |
| H | 5.231884000 | -3.736226000 | -1.741291000 | H | -4.795021000 | -2.606133000 | 3.147811000 |
| H | 3.604955000 | -3.826567000 | -2.383324000 | H | -5.778977000 | -2.221256000 | 1.763660000 |
| C | 3.739180000 | -4.376090000 | -0.328619000 | C | -3.745224000 | -2.788661000 | 1.338471000 |
| H | 2.654025000 | -4.266941000 | -0.200376000 | H | -3.996630000 | -3.840211000 | 1.193942000 |
| H | 3.885269000 | -5.425825000 | -0.601331000 | H | -2.749400000 | -2.718446000 | 1.779210000 |
| C | 4.428225000 | -4.135458000 | 1.009867000 | C | -0.078362000 | 3.171994000 | -1.523538000 |
| H | 5.503976000 | -4.311034000 | 0.903126000 | H | 0.939572000 | 3.422008000 | -1.820214000 |
| H | 4.065478000 | -4.905553000 | 1.699403000 | H | -0.623595000 | 2.712868000 | -2.349636000 |
| C | 4.220288000 | -2.768649000 | 1.665093000 | C | -0.813571000 | 4.354949000 | -0.932956000 |
| H | 4.490819000 | -2.856728000 | 2.723526000 | H | -0.193537000 | 4.820693000 | -0.161824000 |
| H | 4.902419000 | -2.017500000 | 1.263098000 | H | -0.883907000 | 5.093571000 | -1.738910000 |
| C | 2.804186000 | -2.236366000 | 1.606210000 | C | -2.219027000 | 4.060494000 | -0.390459000 |
| H | 2.663173000 | -1.387634000 | 2.272366000 | H | -2.594927000 | 3.123719000 | -0.818727000 |
| H | 2.049551000 | -2.991355000 | 1.833644000 | H | -2.895947000 | 4.838519000 | -0.752326000 |
| C | -5.171637000 | -0.689656000 | -0.711354000 | C | -2.337022000 | 4.030385000 | 1.131886000 |
| H | -5.177585000 | -0.411215000 | -1.765567000 | H | -1.832752000 | 4.913643000 | 1.541770000 |
| H | -6.045506000 | -1.315146000 | -0.515905000 | H | -3.393863000 | 4.138803000 | 1.396197000 |
| C | -5.110505000 | 0.544654000 | 0.167806000 | C | -1.822320000 | 2.787232000 | 1.842064000 |
| H | -5.831592000 | 1.261782000 | -0.242771000 | H | -2.393521000 | 1.908877000 | 1.520923000 |
| H | -4.126552000 | 1.007074000 | 0.049021000 | H | -2.024014000 | 2.898232000 | 2.913681000 |
| C | -5.426328000 | 0.356271000 | 1.642999000 | C | -0.342535000 | 2.495255000 | 1.702261000 |
| H | -5.383122000 | 1.346004000 | 2.109161000 | H | -0.010816000 | 1.739332000 | 2.415100000 |
| H | -6.465566000 | 0.021355000 | 1.752467000 | H | 0.311396000 | 3.364206000 | 1.784592000 |
| C | -4.523906000 | -0.587177000 | 2.431969000 | | | | |

[Pt₃Cl₆{Te(CH₂)₆}₄] C₁ E=-5133.82851344 a.u.

| | | | | | | | |
|----|--------------|--------------|--------------|---|--------------|--------------|--------------|
| Pt | 3.320924000 | -0.226714000 | 0.008568000 | C | 1.569889000 | -1.709282000 | 3.111812000 |
| Pt | -0.869579000 | -1.519234000 | -0.060186000 | H | 2.158306000 | -0.826126000 | 3.355465000 |
| Pt | -0.643947000 | 1.638979000 | 0.021760000 | H | 0.538816000 | -1.558335000 | 3.436483000 |
| Te | 1.482135000 | -1.648664000 | 0.975738000 | C | -2.907273000 | -1.879476000 | -3.225953000 |
| Te | -3.203688000 | -1.426418000 | -1.152774000 | H | -3.594027000 | -1.211088000 | -3.746995000 |
| Te | -2.993066000 | 1.833247000 | 1.072322000 | H | -1.885507000 | -1.550825000 | -3.418564000 |
| Te | 1.715626000 | 1.436328000 | -0.981840000 | C | -3.119186000 | -3.325585000 | -3.623769000 |
| Cl | 5.001649000 | 0.966308000 | -1.047450000 | H | -4.159014000 | -3.621609000 | -3.447720000 |
| Cl | 4.838720000 | -1.632045000 | 1.049603000 | H | -2.990674000 | -3.358186000 | -4.711829000 |
| Cl | 0.309214000 | -1.633752000 | -2.049264000 | C | -2.168350000 | -4.338940000 | -2.976814000 |
| Cl | -2.025225000 | -1.450154000 | 1.953295000 | H | -1.835774000 | -5.035661000 | -3.750997000 |
| Cl | -1.753484000 | 1.786782000 | -2.009437000 | H | -1.266148000 | -3.825366000 | -2.629410000 |
| Cl | 0.498442000 | 1.547345000 | 2.035481000 | C | -2.765533000 | -5.170959000 | -1.846638000 |
| C | 2.099841000 | -3.643601000 | 0.579349000 | H | -2.112554000 | -6.033062000 | -1.676234000 |
| H | 2.242952000 | -3.648475000 | -0.501642000 | H | -3.725549000 | -5.585908000 | -2.179973000 |
| H | 3.081257000 | -3.710661000 | 1.045638000 | C | -2.958611000 | -4.481135000 | -0.505741000 |
| C | 1.101583000 | -4.687963000 | 1.044216000 | H | -3.320189000 | -5.229008000 | 0.210314000 |
| H | 1.628435000 | -5.648845000 | 1.083546000 | H | -1.994931000 | -4.130833000 | -0.120534000 |
| H | 0.324763000 | -4.792258000 | 0.281232000 | C | -3.940258000 | -3.325603000 | -0.478946000 |
| C | 0.413158000 | -4.421381000 | 2.376660000 | H | -4.222405000 | -3.081463000 | 0.546679000 |
| H | -0.257105000 | -3.560233000 | 2.265762000 | H | -4.852913000 | -3.527993000 | -1.044275000 |
| H | -0.255310000 | -5.263797000 | 2.579692000 | C | -3.415723000 | 3.868644000 | 0.545255000 |
| C | 1.295604000 | -4.220560000 | 3.603814000 | H | -3.658094000 | 3.755369000 | -0.513219000 |
| H | 1.915973000 | -5.108829000 | 3.764168000 | H | -4.329632000 | 4.148737000 | 1.074270000 |
| H | 0.624960000 | -4.161741000 | 4.467395000 | C | -2.293717000 | 4.870448000 | 0.737301000 |
| C | 2.202929000 | -2.988022000 | 3.624024000 | H | -2.497830000 | 5.716487000 | 0.070000000 |
| H | 2.501171000 | -2.803085000 | 4.662019000 | H | -1.357583000 | 4.423314000 | 0.386879000 |
| H | 3.139199000 | -3.166671000 | 3.090944000 | C | -2.111908000 | 5.416797000 | 2.144188000 |

| | | | |
|---|--------------|-------------|--------------|
| H | -1.356214000 | 6.206940000 | 2.089461000 |
| H | -3.039000000 | 5.911959000 | 2.461006000 |
| C | -1.684689000 | 4.427745000 | 3.223297000 |
| H | -1.332447000 | 5.013100000 | 4.077039000 |
| H | -0.821039000 | 3.849050000 | 2.879604000 |
| C | -2.776717000 | 3.473386000 | 3.718326000 |
| H | -2.704172000 | 3.383582000 | 4.808392000 |
| H | -3.766255000 | 3.904236000 | 3.530261000 |
| C | -2.702045000 | 2.058894000 | 3.182715000 |
| H | -3.467553000 | 1.417685000 | 3.621662000 |
| H | -1.725921000 | 1.607031000 | 3.360685000 |
| C | 1.806736000 | 1.393621000 | -3.109744000 |
| H | 2.504913000 | 0.582044000 | -3.312535000 |
| H | 0.799368000 | 1.093232000 | -3.400628000 |
| C | 2.232162000 | 2.704603000 | -3.731434000 |

| | | | |
|---|-------------|-------------|--------------|
| H | 3.256471000 | 2.943354000 | -3.430613000 |
| H | 2.279638000 | 2.518178000 | -4.810177000 |
| C | 1.298821000 | 3.892879000 | -3.465923000 |
| H | 0.305622000 | 3.527798000 | -3.178051000 |
| H | 1.154273000 | 4.430257000 | -4.406771000 |
| C | 1.797781000 | 4.906316000 | -2.439177000 |
| H | 2.838205000 | 5.164686000 | -2.670785000 |
| H | 1.223109000 | 5.829230000 | -2.565508000 |
| C | 1.696419000 | 4.509632000 | -0.974147000 |
| H | 0.650005000 | 4.311575000 | -0.715254000 |
| H | 1.998502000 | 5.369894000 | -0.365951000 |
| C | 2.555384000 | 3.340041000 | -0.539043000 |
| H | 2.635036000 | 3.274727000 | 0.547588000 |
| H | 3.561347000 | 3.328251000 | -0.958496000 |

[Pt₄Cl₈{Te(CH₂)₆]₄ D₂ E=-6173.45185839 a.u.

| | | | |
|----|--------------|--------------|--------------|
| Pt | -4.082078000 | 0.000000000 | 0.000000000 |
| Pt | 0.000000000 | 1.537514000 | 0.000000000 |
| Te | -2.376037000 | -1.574905000 | 0.953694000 |
| Te | -2.376037000 | 1.574905000 | -0.953694000 |
| Cl | -5.673083000 | -1.342847000 | 1.007258000 |
| Cl | -5.673083000 | 1.342847000 | -1.007258000 |
| Cl | -1.044504000 | 1.595352000 | 2.063720000 |
| Cl | 1.044504000 | 1.595352000 | -2.063720000 |
| C | -2.470431000 | -1.573703000 | 3.081067000 |
| H | -1.490885000 | -1.197948000 | 3.378847000 |
| H | -3.233489000 | -0.824932000 | 3.291917000 |
| C | -2.792164000 | -2.925414000 | 3.677689000 |
| H | -3.793588000 | -3.239632000 | 3.369283000 |
| H | -2.857911000 | -2.758402000 | 4.758545000 |
| C | -1.766944000 | -4.032153000 | 3.400610000 |
| H | -0.800364000 | -3.585871000 | 3.138732000 |
| H | -1.596200000 | -4.577805000 | 4.331988000 |
| C | -2.170020000 | -5.059142000 | 2.345457000 |
| H | -1.525597000 | -5.935697000 | 2.463068000 |
| H | -3.189339000 | -5.404725000 | 2.555607000 |
| C | -2.080245000 | -4.628476000 | 0.889135000 |
| H | -2.296454000 | -5.500281000 | 0.261410000 |
| H | -1.051492000 | -4.334381000 | 0.649695000 |
| C | -3.034708000 | -3.534092000 | 0.456990000 |
| H | -4.043896000 | -3.618958000 | 0.860323000 |
| H | -3.101099000 | -3.454079000 | -0.629550000 |
| C | -2.470431000 | 1.573703000 | -3.081067000 |
| H | -1.490885000 | 1.197948000 | -3.378847000 |
| H | -3.233489000 | 0.824932000 | -3.291917000 |
| C | -2.792164000 | 2.925414000 | -3.677689000 |
| H | -3.793588000 | 3.239632000 | -3.369283000 |
| H | -2.857911000 | 2.758402000 | -4.758545000 |
| C | -1.766944000 | 4.032153000 | -3.400610000 |
| H | -1.596200000 | 4.577805000 | -4.331988000 |
| H | -0.800364000 | 3.585871000 | -3.138732000 |
| C | -2.170020000 | 5.059142000 | -2.345457000 |
| H | -1.525597000 | 5.935697000 | -2.463068000 |
| H | -3.189339000 | 5.404725000 | -2.555607000 |
| C | -2.080245000 | 4.628476000 | -0.889135000 |
| H | -2.296454000 | 5.500281000 | -0.261410000 |

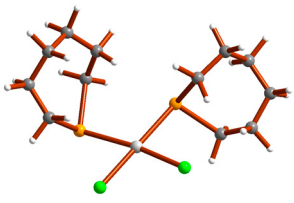
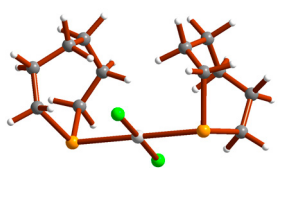
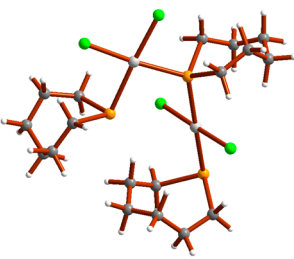
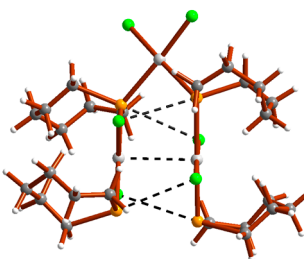
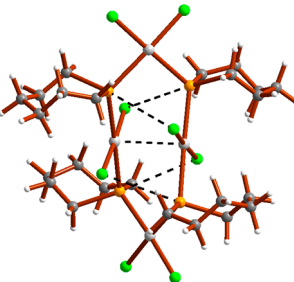
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|----|--------------|--------------|--------------|
| H | -1.051492000 | 4.334381000 | -0.649695000 |
| C | -3.034708000 | 3.534092000 | -0.456990000 |
| H | -4.043896000 | 3.618958000 | -0.860323000 |
| H | -3.101099000 | 3.454079000 | 0.629550000 |
| Pt | 4.082078000 | 0.000000000 | 0.000000000 |
| Pt | 0.000000000 | -1.537514000 | 0.000000000 |
| Te | 2.376037000 | 1.574905000 | 0.953694000 |
| Te | 2.376037000 | -1.574905000 | -0.953694000 |
| Cl | 5.673083000 | 1.342847000 | 1.007258000 |
| Cl | 5.673083000 | -1.342847000 | -1.007258000 |
| Cl | 1.044504000 | -1.595352000 | 2.063720000 |
| Cl | -1.044504000 | -1.595352000 | -2.063720000 |
| C | 2.470431000 | 1.573703000 | 3.081067000 |
| H | 1.490885000 | 1.197948000 | 3.378847000 |
| H | 3.233489000 | 0.824932000 | 3.291917000 |
| C | 2.792164000 | 2.925414000 | 3.677689000 |
| H | 3.793588000 | 3.239632000 | 3.369283000 |
| H | 2.857911000 | 2.758402000 | 4.758545000 |
| C | 1.766944000 | 4.032153000 | 3.400610000 |
| H | 0.800364000 | 3.585871000 | 3.138732000 |
| H | 1.596200000 | 4.577805000 | 4.331988000 |
| C | 2.170020000 | 5.059142000 | 2.345457000 |
| H | 1.525597000 | 5.935697000 | 2.463068000 |
| H | 3.189339000 | 5.404725000 | 2.555607000 |
| C | 2.080245000 | 4.628476000 | 0.889135000 |
| H | 2.296454000 | 5.500281000 | 0.261410000 |
| H | 1.051492000 | 4.334381000 | 0.649695000 |
| C | 3.034708000 | 3.534092000 | 0.456990000 |
| H | 4.043896000 | 3.618958000 | 0.860323000 |
| H | 3.101099000 | 3.454079000 | -0.629550000 |
| C | 2.470431000 | -1.573703000 | -3.081067000 |
| H | 1.490885000 | -1.197948000 | -3.378847000 |
| H | 3.233489000 | -0.824932000 | -3.291917000 |
| C | 2.792164000 | -2.925414000 | -3.677689000 |
| H | 3.793588000 | -3.239632000 | -3.369283000 |
| H | 2.857911000 | -2.758402000 | -4.758545000 |
| C | 1.766944000 | -4.032153000 | -3.400610000 |
| H | 1.596200000 | -4.577805000 | -4.331988000 |
| H | 0.800364000 | -3.585871000 | -3.138732000 |
| C | 2.170020000 | -5.059142000 | -2.345457000 |

| | | | | | | | |
|---|-------------|--------------|--------------|---|-------------|--------------|--------------|
| H | 1.525597000 | -5.935697000 | -2.463068000 | H | 1.051492000 | -4.334381000 | -0.649695000 |
| H | 3.189339000 | -5.404725000 | -2.555607000 | C | 3.034708000 | -3.534092000 | -0.456990000 |
| C | 2.080245000 | -4.628476000 | -0.889135000 | H | 4.043896000 | -3.618958000 | -0.860323000 |
| H | 2.296454000 | -5.500281000 | -0.261410000 | H | 3.101099000 | -3.454079000 | 0.629550000 |

[PtCl₂{Te(CH₂)₆}₂]₂ dimer, cis conformation C; E=-4094.19210670 a.u.

| | | | | | | | |
|----|--------------|--------------|--------------|---|--------------|--------------|--------------|
| Pt | 1.137936000 | -0.240996000 | -1.241450000 | H | 3.583609000 | 4.392421000 | 0.574166000 |
| Te | 2.689410000 | -1.742826000 | 0.089688000 | H | 5.130206000 | 1.843897000 | 0.237009000 |
| Te | 1.758327000 | 1.875665000 | 0.027578000 | H | 5.522670000 | 3.080346000 | -0.925286000 |
| Cl | -0.173537000 | 1.250464000 | -2.517272000 | H | 3.737048000 | 1.536795000 | -1.754688000 |
| Cl | 0.562846000 | -2.015503000 | -2.664614000 | H | 3.220190000 | 3.222257000 | -1.719337000 |
| C | 1.667062000 | -3.620496000 | 0.195201000 | H | 0.849006000 | -3.420738000 | 0.886199000 |
| C | 2.510801000 | -4.788858000 | 0.660670000 | H | 1.262070000 | -3.753923000 | -0.808790000 |
| C | 4.920755000 | -4.205315000 | -0.038368000 | H | 2.837993000 | -4.626104000 | 1.693577000 |
| C | 3.724127000 | -5.153198000 | -0.191472000 | H | 1.837038000 | -5.653700000 | 0.695245000 |
| C | 5.124989000 | -3.192227000 | -1.159949000 | H | 4.035901000 | -6.159356000 | 0.100641000 |
| C | 3.897977000 | -2.395242000 | -1.561332000 | H | 3.428281000 | -5.227446000 | -1.243998000 |
| C | 3.537367000 | 2.382366000 | -1.098090000 | H | 5.840772000 | -4.789314000 | 0.051657000 |
| C | 4.733927000 | 2.743975000 | -0.242773000 | H | 4.826187000 | -3.669066000 | 0.913988000 |
| C | 4.487979000 | 3.822690000 | 0.822062000 | H | 5.925767000 | -2.508096000 | -0.862557000 |
| C | 4.399006000 | 3.306390000 | 2.257091000 | H | 5.488567000 | -3.712829000 | -2.054258000 |
| C | 3.116913000 | 2.590545000 | 2.658237000 | H | 3.193490000 | -2.955781000 | -2.176380000 |
| C | 2.762108000 | 1.358552000 | 1.844332000 | H | 4.154761000 | -1.492450000 | -2.115707000 |
| Pt | -1.137936000 | 0.240996000 | 1.241450000 | H | -3.630342000 | -0.740735000 | -1.613709000 |
| Te | -1.758327000 | -1.875665000 | -0.027578000 | H | -2.021037000 | -0.740792000 | -2.353670000 |
| Cl | 0.173537000 | -1.250464000 | 2.517272000 | H | -3.215965000 | -2.275244000 | -3.702787000 |
| Cl | -0.562846000 | 2.015503000 | 2.664614000 | H | -2.272859000 | -3.287989000 | -2.645941000 |
| C | -2.762108000 | -1.358552000 | -1.844332000 | H | -5.253088000 | -2.641098000 | -2.435035000 |
| C | -3.116913000 | -2.590545000 | -2.658237000 | H | -4.532043000 | -4.152256000 | -2.938918000 |
| C | -4.399006000 | -3.306390000 | -2.257091000 | H | -5.307630000 | -4.544953000 | -0.782170000 |
| C | -4.487979000 | -3.822690000 | -0.822062000 | H | -3.583609000 | -4.392421000 | -0.574166000 |
| C | -4.733927000 | -2.743975000 | 0.242773000 | H | -5.130206000 | -1.843897000 | -0.237009000 |
| C | -3.537367000 | -2.382366000 | 1.098090000 | H | -5.522670000 | -3.080346000 | 0.925286000 |
| Te | -2.689410000 | 1.742826000 | -0.089688000 | H | -3.737048000 | -1.536795000 | 1.754688000 |
| C | -3.897977000 | 2.395242000 | 1.561332000 | H | -3.220190000 | -3.222257000 | 1.719337000 |
| C | -5.124989000 | 3.192227000 | 1.159949000 | H | -4.154761000 | 1.492450000 | 2.115707000 |
| C | -4.920755000 | 4.205315000 | 0.038368000 | H | -3.193490000 | 2.955781000 | 2.176380000 |
| C | -3.724127000 | 5.153198000 | 0.191472000 | H | -5.925767000 | 2.508096000 | 0.862557000 |
| C | -2.510801000 | 4.788858000 | -0.660670000 | H | -5.488567000 | 3.712829000 | 2.054258000 |
| C | -1.667062000 | 3.620496000 | -0.195201000 | H | -2.837993000 | 4.626104000 | -1.693577000 |
| H | 2.021037000 | 0.740792000 | 2.353670000 | H | -1.837038000 | 5.653700000 | -0.695245000 |
| H | 3.630342000 | 0.740735000 | 1.613709000 | H | -1.262070000 | 3.753923000 | 0.808790000 |
| H | 3.215965000 | 2.275244000 | 3.702787000 | H | -0.849006000 | 3.420738000 | -0.886199000 |
| H | 2.272859000 | 3.287989000 | 2.645941000 | H | -3.428281000 | 5.227446000 | 1.243998000 |
| H | 5.253088000 | 2.641098000 | 2.435035000 | H | -4.035901000 | 6.159356000 | -0.100641000 |
| H | 4.532043000 | 4.152256000 | 2.938918000 | H | -4.826187000 | 3.669066000 | -0.913988000 |
| H | 5.307630000 | 4.544953000 | 0.782170000 | H | -5.840772000 | 4.789314000 | -0.051657000 |

Table S4. PBE0-D3/def2-TZVP optimized geometries of the $[\text{Pt}_n\text{Cl}_{2n}\{\text{Te}(\text{CH}_2)_6\}_m]$ ($n = 1-4$; $m = 2-4$).

| | |
|--|---|
| <p><i>cis</i>-$[\text{PtCl}_2\{\text{Te}(\text{CH}_2)_6\}_2]$</p>  <p>Pt-Te 2.528-2.531 Pt-Cl 2.326-2.339</p> | <p><i>trans</i>-$[\text{PtCl}_2\{\text{Te}(\text{CH}_2)_6\}_2]$</p>  <p>Pt-Te 2.583 Pt-Cl 2.315</p> |
| <p>$[\text{Pt}_2\text{Cl}_4\{\text{Te}(\text{CH}_2)_6\}_3]$</p>  <p>Pt-Te(<i>cis</i>) 2.501-2.543 Pt-Te(<i>trans</i>) 2.565-2.580 Pt-Cl(<i>cis</i>) 2.307-2.339 Pt-Cl(<i>trans</i>) 2.301-2.325 Te...Cl 3.636</p> | <p>$[\text{Pt}_3\text{Cl}_6\{\text{Te}(\text{CH}_2)_6\}_4]$</p>  <p>Pt-Te(<i>cis</i>) 2.515-2.518 Pt-Te(<i>trans</i>) 2.572-2.581 Pt-Cl(<i>cis</i>) 2.316 Pt-Cl(<i>trans</i>) 2.315-2.323 Pt...Pt 3.167 Te...Cl 3.508-3.628</p> |
| <p>$[\text{Pt}_4\text{Cl}_8\{\text{Te}(\text{CH}_2)_6\}_4]$</p>  <p>Pt-Te(<i>cis</i>) 2.510 Pt-Te(<i>trans</i>) 2.561 Pt-Cl(<i>cis</i>) 2.313 Pt-Cl(<i>trans</i>) 2.314 Pt...Pt 3.075 Te...Cl 3.613</p> | |

3.2 Secondary bonding interactions

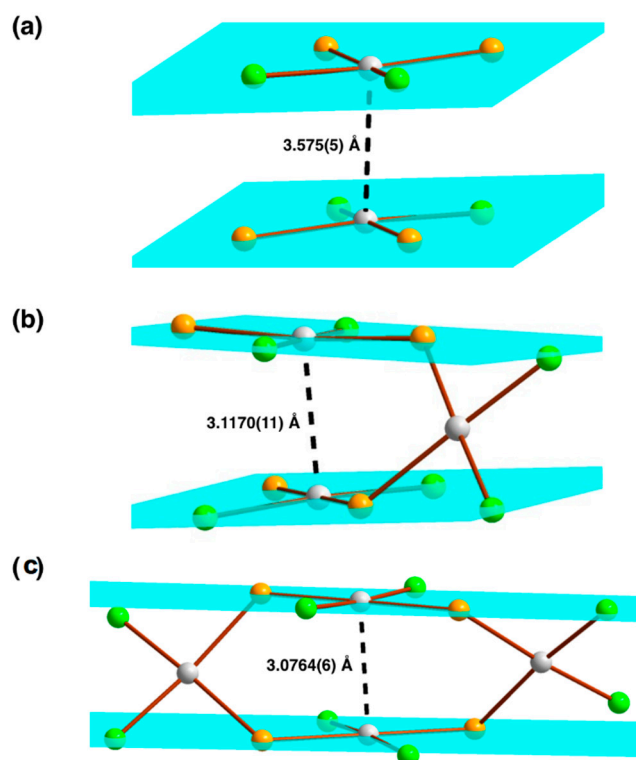


Figure S4. The Pt...Pt interactions lead to the square-planar coordination plane to become slightly concave in (a) **1_{cis}**, (b) **2**, and (c) **3**.

3.3 Formation energetics

Table S5. Total energies of optimized species at PBE0-D3/def2-TZVP level of theory in vacuum (Hartree).

| Species | E | H | G_{298} |
|---|-------------|-------------|-------------|
| <i>cis</i> -[PtCl ₂ (NCPH) ₂] | -1688.06795 | -1687.84233 | -1687.91708 |
| <i>trans</i> -[PtCl ₂ (NCPH) ₂] | -1688.07709 | -1687.85113 | -1687.92435 |
| PhCN | -324.22847 | -324.12182 | -324.15846 |
| Te(CH ₂) ₆ | -503.70425 | -503.52586 | -503.56845 |
| <i>cis</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ | -2047.06703 | -2046.69856 | -2046.77953 |
| <i>trans</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ | -2047.08132 | -2046.71280 | -2046.79224 |
| [Pt ₂ Cl ₄ {Te(CH ₂) ₆ }] ₃ | -3590.44077 | -3589.88192 | -3589.99880 |
| [Pt ₃ Cl ₆ {Te(CH ₂) ₆ }] ₄ | -5133.82851 | -5133.07979 | -5133.23107 |
| [Pt ₄ Cl ₈ {Te(CH ₂) ₆ }] ₄ | -6173.45186 | -6172.69360 | -6172.85587 |
| [PtCl ₂ (NCPH){Te(CH ₂) ₆ }] | -1867.57576 | -1867.27869 | -1867.35713 |

Table S6. Total energies of optimized species at PBE0-D3/def2-TZVP level of theory in dichloromethane (Hartree).

| | E | H | G_{298} |
|---|-------------|-------------|-------------|
| <i>cis</i> -[PtCl ₂ (NCPH) ₂] | -1688.09774 | -1687.87197 | -1687.94624 |
| <i>trans</i> -[PtCl ₂ (NCPH) ₂] | -1688.10200 | -1687.87599 | -1687.94883 |
| PhCN | -324.23565 | -324.12893 | -324.16554 |
| Te(CH ₂) ₆ | -503.70723 | -503.52904 | -503.57170 |
| <i>cis</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ | -2047.09060 | -2046.72240 | -2046.80539 |
| <i>trans</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ | -2047.09323 | -2046.72499 | -2046.80445 |
| [Pt ₂ Cl ₄ {Te(CH ₂) ₆ }] ₃ | -3590.46723 | -3589.90901 | -3590.02745 |
| [Pt ₃ Cl ₆ {Te(CH ₂) ₆ }] ₄ | -5133.85842 | -5133.11156 | -5133.26138 |
| [Pt ₄ Cl ₈ {Te(CH ₂) ₆ }] ₄ | -6173.49395 | -6172.73679 | -6172.90005 |
| [PtCl ₂ (NCPH){Te(CH ₂) ₆ }] | -1867.59427 | -1867.29725 | -1867.37482 |

Table S7. Gibbs PBE0-D3/def2-TZVP formation energies of **1_{cis}**, **1_{trans}**, **2**, and **3** from *cis*-[PtCl₂(NCPH)₂] and Te(CH₂)₆. In dichloromethane (kJ mol⁻¹)-

| Reaction | ΔE | ΔH | ΔG_{298} | $\Delta G/\text{Pt}$ | $\Delta G/\text{Te}$ |
|---|------------|------------|------------------|----------------------|----------------------|
| <i>cis</i> -[PtCl ₂ (NCPH) ₂] + 2 Te(CH ₂) ₆ \rightleftharpoons <i>cis</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ (1_{cis}) + 2 PhCN | -130 | -132 | -123 | -123 | -61 |
| <i>cis</i> -[PtCl ₂ (NCPH) ₂] + 2 Te(CH ₂) ₆ \rightleftharpoons <i>trans</i> -[PtCl ₂ {Te(CH ₂) ₆ }] ₂ (1_{trans}) + 2 PhCN | -137 | -139 | -121 | -121 | -60 |
| 2 <i>cis</i> -[PtCl ₂ (NCPH) ₂] + 3 Te(CH ₂) ₆ \rightleftharpoons [Pt ₂ Cl ₄ {Te(CH ₂) ₆ }] ₃ + 4 PhCN | -243 | -246 | -215 | -107 | -71 |
| 3 <i>cis</i> -[PtCl ₂ (NCPH) ₂] + 4 Te(CH ₂) ₆ \rightleftharpoons [Pt ₃ Cl ₆ {Te(CH ₂) ₆ }] ₄ (2) + 6 PhCN | -394 | -402 | -339 | -113 | -85 |
| 4 <i>cis</i> -[PtCl ₂ (NCPH) ₂] + 4 Te(CH ₂) ₆ \rightleftharpoons [Pt ₄ Cl ₈ {Te(CH ₂) ₆ }] ₄ (3) + 8 PhCN | -418 | -431 | -401 | -100 | -100 |

4. References

- [1] Rodewald, M.; Rautiainen, J. M.; Niksch, T.; Görls, H.; Oilunkaniemi, R.; Weigand, W.; Laitinen, R. S. Chalcogen-bonding interactions in telluroether heterocycles $[\text{Te}(\text{CH}_2)_m]_n$ ($n = 1-4$; $m = 3-7$). *Chem. Eur. J.* **2020**, *26*, 13806- 13818. doi.org/10.1002/chem.202002510