

Functionalization of Polylactide with Multiple Tetraphenylethane Inifer Groups to Form PLA Block Copolymers with Vinyl Monomers

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1. Determination of molecular weights of synthesized PLA diols

Molecular weights of PLA polymers were calculated by the comparison of the intensities of signals (in DMSO-d₆) corresponding to $\text{-CH(CH}_3\text{)-OH}$ end groups at 4.20 ppm and corresponding to $\text{-CH(CH}_3\text{)-}$ groups from PLA backbone at 5.12 and 5.20 ppm (Figure 1 in the main text).

2. Analysis of PLA-based polyurethane (PLA-PU)

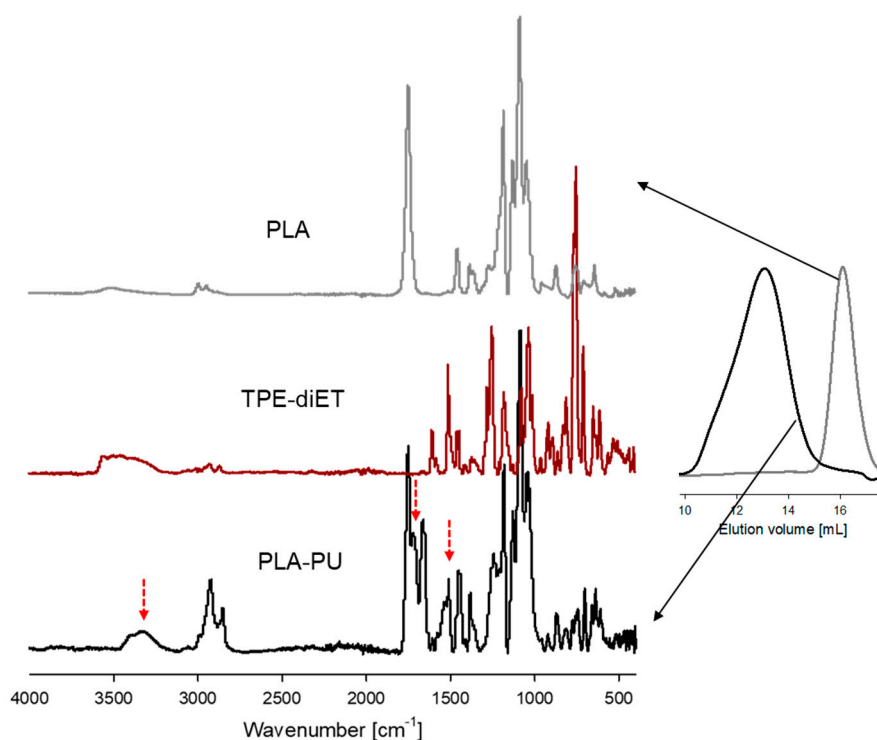


Figure S1. FTIR spectra of PLA-PU and the initial components (red arrows indicate newly formed urethane linkages) and SEC curves for starting PLA and PLA-PU (DMF as an eluent).

3. Calculation of the relative decomposition rate of TPE groups present in PLA-PU polymer on the basis of ESR spectra

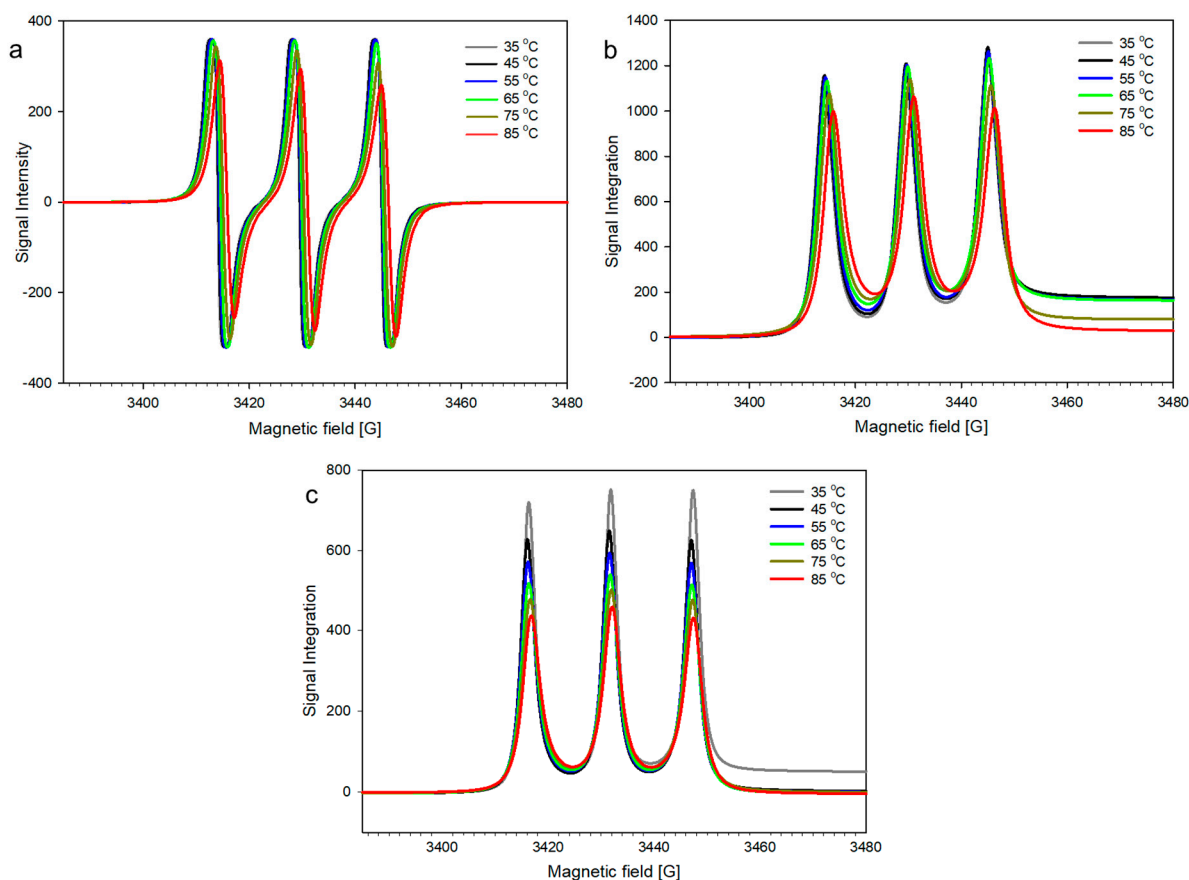


Figure S2. a) ESR spectra of TEMPO, b) normalized spectra for plots in Fig. S2a, c) normalized spectra for PLA-PU (+TEMPO) shown in Fig. 2a in the main text.

The concentration of the formed radicals was measured indirectly, by the measurement the concertation of TEMPO radicals which remained after catching ...-TPE/2•. To calculate TPE decomposition ratio, the formula was applied:

$$Y = x_1(I_1 - I_2)/2x_2I_1 \quad (S1a)$$

x_1 – quantity of TEMPO (mols)

x_2 - quantity of TPE groups in PLA-PU (mols)

I_1 – integral area from the normalized absorption curve for TEMPO

I_2 – integral area from the normalized absorption curve for PLA-PU with TEMPO

Two solutions were prepared and ESR spectra for them were recorded at 35, 45, 55, 65, 75 °C:

1. 0.8 mg of TEMPO in 1 mL of DMF, i.e. 5.12×10^{-3} mmol

2. 80 mg of PLA-PU + 0.8 mg of TEMPO in 1 mL of DMF, i.e. 46×10^{-3} mmol + 5.12×10^{-3} mmol

After inserting the values into the equation (S1a):

$$Y = 0.55(I_1 - I_2)/I_1 \quad (\text{S1b})$$

ESR spectra presented in Fig. 2a for PU-PLA (with TEMPO) and Fig. S2a for TEMPO were transformed into normalized spectra presented in Fig. S2b and S2c using Origin Software (Integration of signals from the original ESR).

The I_1 and I_2 values were found from normalized spectra and were put into the equation 1b to find Y values at different temperatures. The dependence $Y = f(T)$ is shown in Fig. 2b in the main text.

4. FTIR analyses of PLA/PST and PLA/PAN copolymers

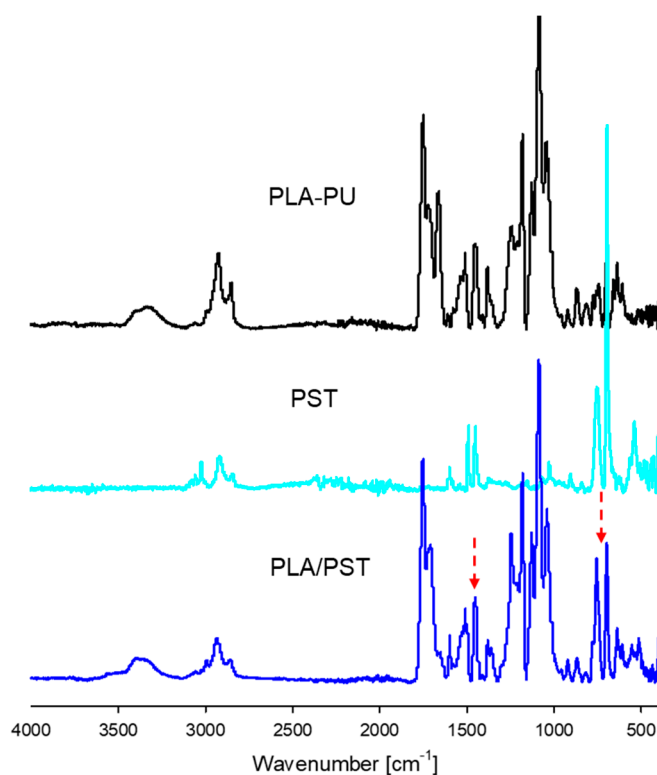


Figure S3. FTIR spectra of PLA-PU, additionally synthesized PST, and PLA/PST copolymer. Red arrows indicate the signals corresponding to aromatic ring (698 cm^{-1} and 1492 cm^{-1}) for PST [38].

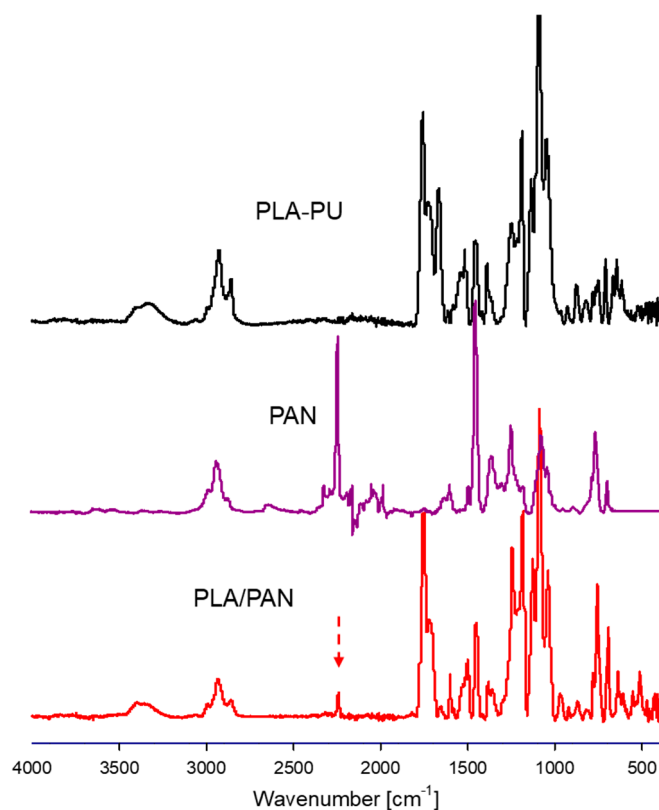


Figure S4. FTIR spectra of PLA-PU, additionally synthesized PAN, and PLA/PAN copolymer. Red arrows indicate the signal corresponding to nitrile group in PAN (2240 cm⁻¹) [39].

Homopolymers - PST and PAN for comparison in FTIR analyses were obtained via radical polymerizations of ST and AN correspondingly, initiated with AIBN.

5. The stability of PLA- based PU during heating

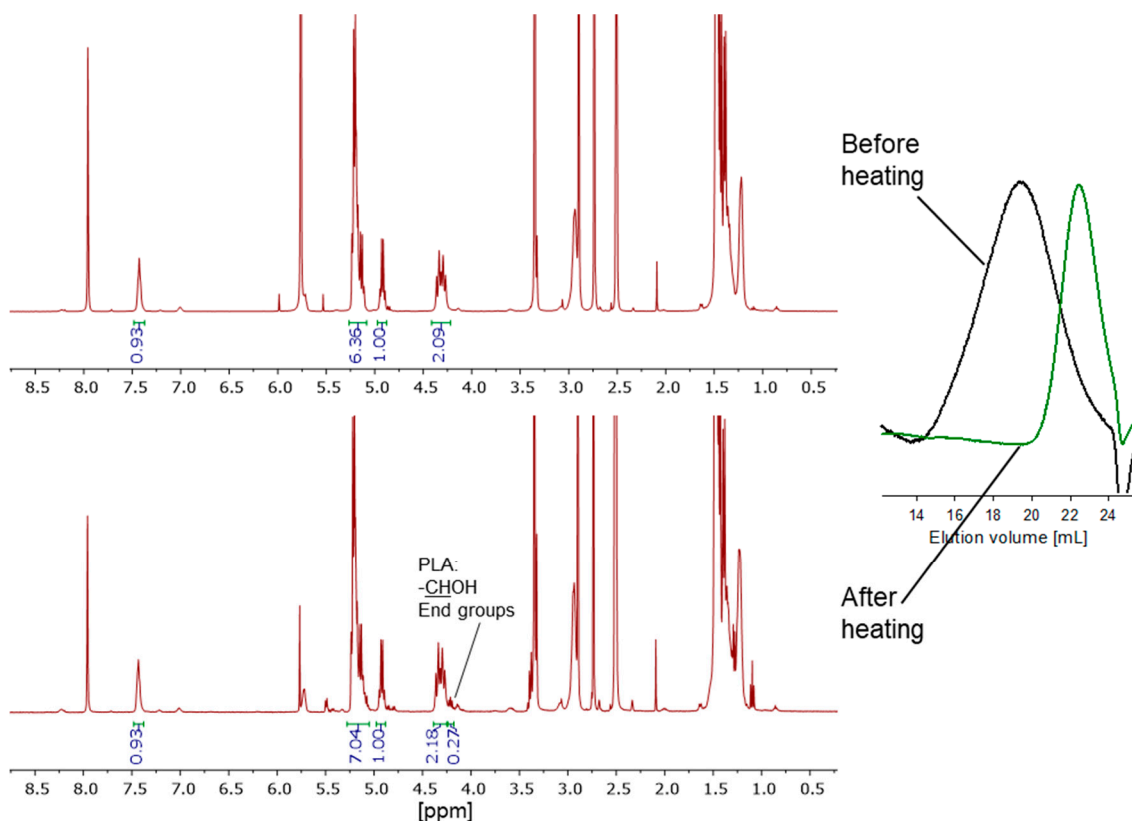


Figure S5 . ^1H NMR spectra of PLA-PU prepared without TPE-diET before and after heating in DMF at 85 °C for 24h and SEC curves for PU before and after heating (DMF as an eluent).

6. Diffusion coefficients of PLA-PU/PVM copolymers and corresponding homopolymers on the basis of DOSY NMR measurements

Table S1. The diffusion coefficients for groups in PLA/PVM copolymers and homopolymers corresponding to particular signals in ^1H NMR spectra shown in Fig. 5 and Fig. 6 in the main text.

PLA/PAN		PLA/PST	
Signal [ppm]	Diffusion coefficient $\times 10^7$, [cm^2/s]	Signal [ppm]	Diffusion coefficient $\times 10^7$, [cm^2/s]
PLA-PU		PLA-PU	
7.33-7.24	1.27	7.33-7.24	1.41
7.24-7.16	1.29	7.24-7.16	1.42

7.06-6.96	1.27	7.06-6.96	1.49
6.68-6.57	1.28	6.68-6.57	1.41
5.22-5.13	1.26	5.22-5.13	1.46
4.93-4.87	1.26	4.93-4.87	1.41
4.22-4.16	1.27	4.22-4.16	1.43
4.07-3.98	1.26	4.07-3.98	1.46
1.56-1.34	1.26	2.93-2.89	1.45
		1.56-1.34	1.40
PAN		PST	
3.16-3.09	0.78	7.25-6.88	7.44
2.14-2.06	0.78	6.88-6.31	7.47
		1.90-1.27	7.44
PLA-PU/PAN copol.		PLA-PU/PST copol	
7.76-7.02	2.80	7.75-7.62	3.25
6.89	2.83	7.56-7.41	3.20
6.64	2.82	7.23-6.30	3.21
5.22-5.13	2.81	5.22-5.13	3.21
4.93-4.87	2.78	4.93-4.87	3.18
4.38-4.16	2.80	4.22-4.16	3.27
3.22-3.05	2.85	2.95-2.89	3.25
2.20-1.90	2.86	1.56-1.34	3.26
1.51-1.41	2.76	1.21-1.19	3.25