## Synthesis of Phenylhydrazone of 5-Acetyl-3-(Methylsulfanyl)-1,2,4-Triazine and 3-Methyl-5-(Methylsulfanyl)-1-Phenyl-1H-Pyrazolo[4,3-e][1,2,4]Triazine from Pivotal Intermediate

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As part of ongoing research programme on bicyclic heterocycles [1-4] we have elaborated a new approach to 1 H -pyrazolo[4,3-e][1,2,4]triazine derivative $\mathbf{3}$ and its synthetic precursor $\mathbf{2}$ by reaction of oxime 5-acetyl-3-(methylsulfanyl)-1,2,4-triazine (1) with phenylhydrazine hydrochloride under different reaction conditions.


## Phenylhydrazone of 5-acetyl-3-(methylsulfanyl)-1,2,4-triazine 2

To a solution of the oxime $\mathbf{1}(184 \mathrm{mg}, 1 \mathrm{mmol})$ and phenylhydrazine hydrochloride ( $288 \mathrm{mg}, 2 \mathrm{mmol}$ ) in ethanol $(10 \mathrm{ml}) 37 \% \mathrm{HCl}(0.3 \mathrm{ml})$ was added. The mixture was heated at $40^{\circ} \mathrm{C}$ for 9 hours and then the solvent was evaporated in vacuo. The solid was collected by filtration, washed with water and recrystallized from ethanol/water mixture (1:1) to give $\mathbf{2}$ in $27 \%$ yield.

Melting point: $224^{\circ} \mathrm{C}$.
${ }^{1}{ }^{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta=2.30(\mathrm{~s}, 3 \mathrm{H}) ; 2.69(\mathrm{~s}, 3 \mathrm{H}) ; 6.99-7.08(\mathrm{~m}, 1 \mathrm{H}) ; 7.23-7.28(\mathrm{~m}, 2 \mathrm{H}) ; 7.32-7.41(\mathrm{~m}$, 2H); 8.05 (s, 1H, NH); 9.63 (s, 1H).

IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): $3240(\mathrm{NH}) ; 2980,1600,700$.
EI-MS (70eV, m/z): 259 (7) [ $\left.\mathrm{M}^{+}\right] ; 147$ (45); 129 (100); 112 (54); 70 (90).
Elemental Analysis: Calculated for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{~S}$ : C $55.60 \%$; H 5.02\%; N $27.03 \%$. Found: C $55.53 \%$; H 5.09\%; N 26.99\%.

## 3-Methyl-5-(methylsulfanyl)-1-phenyl-1H-pyrazolo[4,3-e][1,2,4]triazine 3

To a solution of the oxime $\mathbf{1}(184 \mathrm{mg}, 1 \mathrm{mmol})$ and phenylhydrazine hydrochloride ( $216 \mathrm{mg}, 1.5 \mathrm{mmol}$ ) in ethanol $(10 \mathrm{ml})$ was added $37 \% \mathrm{HCl}(0.3 \mathrm{ml})$. The mixture was heated at reflux for 5 hours and then the
solvent was evaporated in vacuo. The solid was collected by filtration, washed with water and recrystallized from ethanol/water mixture (1:1) to give $\mathbf{3}$ in $18 \%$ yield.

Melting point: $105{ }^{\circ} \mathrm{C}$.
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=2.73(\mathrm{~s}, 3 \mathrm{H}) ; 2.77(\mathrm{~s}, 3 \mathrm{H}) ; 7.29-7.40(\mathrm{~m}, 1 \mathrm{H}) ; 7.50-7.61(\mathrm{~m}, 2 \mathrm{H})$; 8.31-8.38 (m, 2H).

IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ): 2920, 1590, 1500, 1390, 760.

EI-MS (70eV, $m / z$ ): 257 (43) $\left[\mathrm{M}^{+}\right] ; 232$ (3); 216 (22); 93 (41); 77 (100).
Elemental Analysis: Calculated for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{~S}$ : C 56.03\%; H 4.28\%; N $27.23 \%$. Found: C 55.67\%; H 4.13\%; N 27.05\%.

## References

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