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Synthesis of 6-chloro-2-oxo-1,2-dihydroquinoline-4-carbohydrazide

R. Bouhfid, E.M. Essassi*

Laboratoire de Chimie Organique Hétérocyclique, Université Mohammed V-Agdal, BP: 1014 Avenue Ibn Batouta, Rabat, Maroc.

E-mail: emessassi@yahoo.fr

*Author to whom correspondence should be addressed

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In previous works we have shown that hydrazides are highly useful starting materials and intermediates in the synthesis of several heterocyclic compounds¹⁻³ of potential biological activities.

The aim of this work is to describe the preparation of a novel compound entitled 6-chloro-2-oxo-1,2-dihydroquinoline-4-carbohydrazide.

$$\begin{array}{c} \text{CO}_2\text{Et} \\ \text{CI} \\ \text{N} \\ \text{O} \end{array} + \text{NH}_2\text{NH}_2, \text{H}_2\text{O} \xrightarrow{\underline{\text{Ethanol}}} \begin{array}{c} \text{CONHNH}_2 \\ \\ \underline{\text{N}} \\ \text{O} \end{array}$$

To a solution of ethyl 2-oxo-1,2-dihydroquinoline-4-carboxylate $\underline{\mathbf{1}}$ (1g, 3.9 mmol) in ethanol, was added hydrazine hydrate 80 % (0.22 mL, 4.6 mmol). The mixture was refluxed for 24h and ice-water was added. The precipate was filtered and recrystallised from ethanol to afford 0.73 g (70 % yields) of product $\underline{\mathbf{2}}$.

Melting point: > 250 °C.

¹H-NMR (300 MHz, DMSO): δ = 4.67 (NH₂); 6.54 (s, 1H, =CH); 7.35-7.78 (m, 3H, H_{Ar}); 9.93 (NH).

¹³C-NMR (300 MHz, DMSO): δ = 121.8 (=CH); 118.0, 125.4, 131.2 (CH_{Ar}); 114.8, 126.3, 139.3, 139.9 (Cq); 161.3 (C=O); 165.0 (CON₂H₃).

MS (EI, m/z): 237.

Elemental analysis: Calculated for $C_{10}H_8ClN_3O_2$: C, 50.54 %; H, 3.39 %; N, 17.68 %; Found: C, 50.60 %; H, 3.42 %; N, 17.74 %;

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