

Short Note

7-(3-Chlorophenylamino)-1-cyclopropyl-6-fluoro-8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylic Acid

Yusuf Mohammad Al-Hiari ^{1,*}, Amjad M. Qandil ², Rufaida M. Al-Zoubi ¹, Muhammed H. Alzweiri ¹, Rula M. Darwish ¹, Ghassan F. Shattat ³ and Tariq M. Al-Qirim ³

- ¹ Faculty of Pharmacy, University of Jordan, Amman-11942, Jordan
- ² Faculty of Pharmacy, Jordan University of Science and Technology, Irbid, Jordan
- ³ Faculty of Pharmacy, Al-Zaytoonah Private University of Jordan, Amman-11733, Jordan
- * Author to whom correspondence should be addressed; E-Mail: hiary@ju.edu.jo.

Received: 8 March 2010 / Accepted: 22 March 2010 / Published: 24 March 2010

Abstract: 7-(3-Chlorophenylamino)-1-cyclopropyl-6-fluoro-8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (2) was prepared and fully characterized by NMR, IR, and MS. Compound 2 exhibited good antibacterial activity against gram-positive standard and resistant strains.

Keywords: 7-anilinofluoroquinolone; fluoroquinolone; antibacterial

1. Introduction

To date, fluoroquinolones represent successful synthetic antibacterial agents [1-5]. However, increased prescribing has led to the recent emergence of fluoroquinolone-resistant bacteria which has necessitated the search for newer drugs with efficacy against resistant strains [6,7]. The present work aims at the synthesis of a new fluoroquinolone derivative (2) and screening of its activity.

2. Results and Discussion

An earlier study carried out by our team [4] revealed that substitution of lipophilic groups such as aniline at C-7 of 8-nitrofluoroquinolone (1) produced noticeable increase in gram-positive activity, especially against resistant strains with significant loss of gram-negative activity [4]. This research aims at further investigation of halogenated aniline derivatives exemplified by compound 2. Synthon 1 involved introducing an electron-withdrawing nitro group at C-8 to facilitate coupling of the chloro-

aniline [4]. Although compound **2** was prepared by direct coupling of 7-chloro-6-fluoro-8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**1b**) with a 2 molar excess of 3-chloroaniline in DMF/pyridine (7:3 V/V), the yield was very low after chromatographic separation from many side products. Alternatively, the chloroaniline was reacted with the correspondent ester (**1a**) to produce the target acid **2** in satisfactory yield upon hydrolysis of the ester intermediate (Scheme 1).



Scheme 1. Synthesis of target compound 2.

The *in vitro* antibacterial activity of **2** was evaluated against standard and resistant isolates of grampositive *Staphylococcus aureus* and gram-negative *Escherichia coli* bacteria, using broth dilution method. Compound **2** revealed excellent antimicrobial activity against standard *S. aureus* strains (ATCC6538) with minimum inhibitory concentrations (MIC) of 0.88 μ g/mL. Interestingly, it showed good activity against resistant isolates of *S. aureus* gram-positive bacteria with MIC value of 7.0 μ g/mL. These data are comparable to ciprofloxacin reference which showed MIC values of 2.9 μ g/mL and 1.4 μ g/mL, respectively, against both gram-positive strains.

These findings are in correlation with literature findings that more lipophilic quinolones can better penetrate the lipophilic cell membrane of gram-positive bacteria [8,9].

3. Experimental

Two molar equivalents of 3-chloroaniline (0.36 g) were gradually added to a solution of ethyl 7-chloro-1-cyclopropyl-6-fluoro-8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylate (**1a**) (0.5 g) in 10 mL of DMSO and a few drops of pyridine. The mixture was heated at 70–80 °C under anhydrous conditions. The reaction mixture was left to crystallize, then filtered and the product was left to dry in a dark place. The resulting solid was recrystallized from methanol/chloroform (3x) to give fairly pure ester intermediate (0.15 g, 24%). The intermediate was then hydrolyzed upon dissolving the solid in ethanol and adding conc. HCl (10 mL). The mixture was heated at 70–80 °C for 5 h, then compound **2** was collected by filtration and dried to give a yellow solid (0.13 g, 93%).

M.p.: 240–243 °C (decomp).

IR (KBr, cm⁻¹): v 3434, 3390, 2942, 1735, 1640, 1535, 1467, 1365, 1210, 1225.

¹H-NMR (300 MHz, DMSO-*d*₆): δ 0.95, 1.02 (2m, 4H, H₂-2'/H₂-3'), 3.71 (m, 1H, H-1'), 7.16 (dd, J = 1.0, 7.8 Hz, 1H, Ar-H), 7.39 (dd, J = 7.8, 8.1 Hz, 1H, Ar-H), 7.49 (d, J = 8.4 Hz, 1H, Ar-H), 7.98

Molbank 2010

(m, $J_{H5-F} = 11.2$ Hz, 2H, Ar-H and H-5), 8.50 (br s, 1H, NH, D₂O exchangeable), 8.76 (s, 1H, H-2), 12.95 (br s, 1H, CO₂H).

MS (CI/ESI-ve): *m/z* (% rel. int.): calcd. for C₁₉H₁₃ClFN₃O₅ (417.8): 419 (5), 418 (34), 417 (21), 416 (78), 383 (4), 360 (5), 359 (35), 339 (7), 325 (6), 311 (8), 293 (12), 265 (100), 191 (6), 161 (6).

Elemental Analysis: Calcd. for C₁₉H₁₃ClFN₃O₅, C, 54.62; H, 3.14; N, 10.06. Found: C, 54.61; H, 3.20; N, 9.97.

Acknowledgements

We wish to thank the University of Jordan represented by the Deanship of Academic Research and Hamdi Mango Research Center.

References and Notes

- Wise, R.; Andrews, J.M.; Edwards, L.J. *In vitro* activity of Bay 09867, a new quinoline derivative, compared with those of other antimicrobial agents. *Antimicrob. Agents Chemother.* 1983, 23, 559–564.
- Zhanel, G.G.; Ennis, K.; Vercaigne, L.; Walkty, A.; Gin, A.S.; Embil, J.; Smith, H.; Hoban, D.J. A critical review of the fluoroquinolones: Focus on respiratory tract infections. *Drugs* 2002, 62, 13–59.
- 3. Da Silva, A.D.; De Almeida, M.V.; De Souza, M.V.N.; Couri, M.R.C. Biological activity and synthetic metodologies for the preparation of fluoroquinolones, a class of potent antibacterial agents. *Curr. Med. Chem.* **2003**, *10*, 21–39.
- 4. Al-Hiari, Y.; Al-Mazari, I.; Shakya, A.; Darwish, R.; Abu-Dahab, R. Synthesis and antibacterial properties of new 8-nitrofluoroquinolone derivatives. *Molecules* **2007**, *12*, 1240–1258.
- Baker, W.R.; Cai, S.; Dimitroff, M.; Fang, L.; Huh, K.K.; Ryckman, D.R.; Shang, X.; Shawar, R.M.; Therrien, J.H. A prodrug approach towards the development of water soluble fluoroquinolones and structure-activity relationships of quinoline-3-carboxylic acids. *J. Med. Chem.* 2004, 47, 4693–4709.
- Ball, P. Quinolone generations: Natural history or natural selection. J. Antimicrobial. Chemother. 2000, 46 (Topic T1), 17–24.
- Chen, F.J.; Lo, H.J. Molecular mechanisms of fluoroquinolone resistance. J. Microbiol. Immunol. Infect. 2003, 36, 1–9.
- 8. Renau, T.E.; Sanchez, J.P.; Gage, J.W.; Dever, J.A.; Shapiro, M.A.; Grackeck, S.J.; Domagala, J.M. Structure-activity relationships of the quinolone antibacterials against mycobacteria: Effect of structural changes at N-1 and C-7. *J. Med. Chem.* **1996**, *39*, 729–735.
- 9. Khalil, O.M.; Roshdy, S.M.A.; Shaaban, M.A.; Hasanein, M.K. New 7-substituted fluoroquinolones. *Bull. Fac. Pharm.* **2002**, *40*, 89–96.

© 2010 by the authors; licensee Molecular Diversity Preservation International, Basel, Switzerland. This article is an open-access article distributed under the terms and conditions of the Creative Commons Attribution license (http://creativecommons.org/licenses/by/3.0/).