

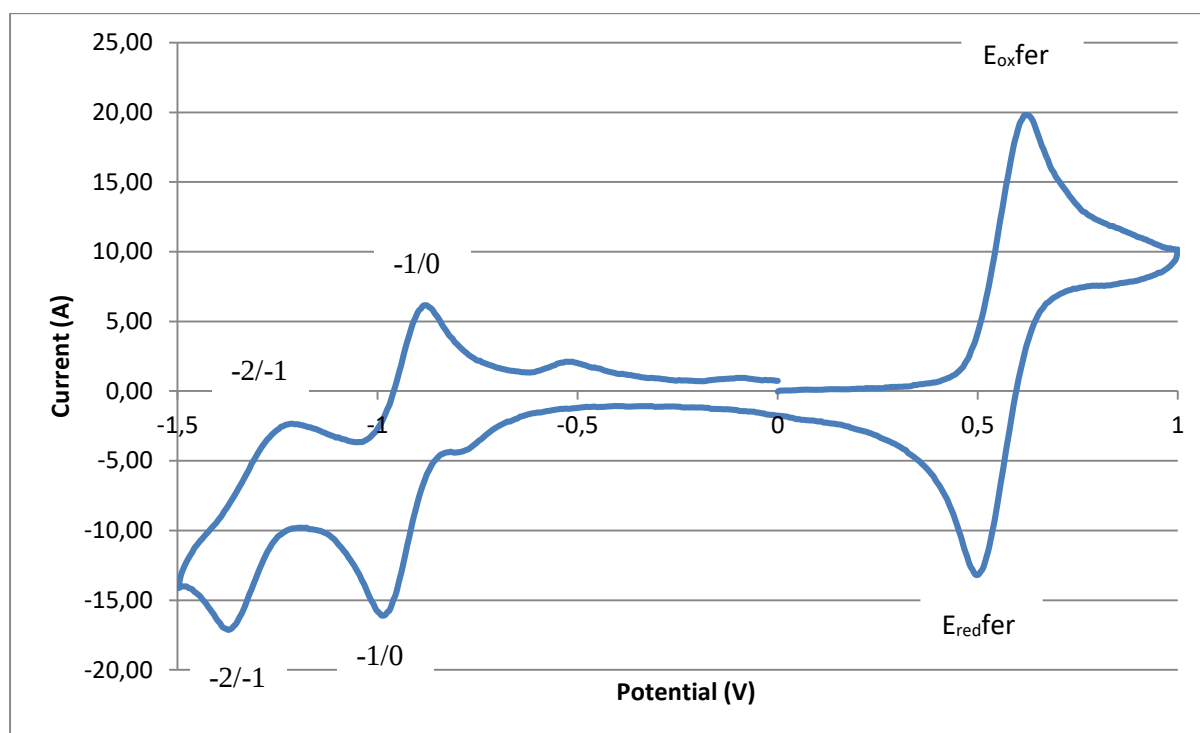
## Supporting Information

**3,5-Bis[5-(thiazol-2-yl)thien-2-yl]-4*H*-1,2,6-thiadiazin-4-one**

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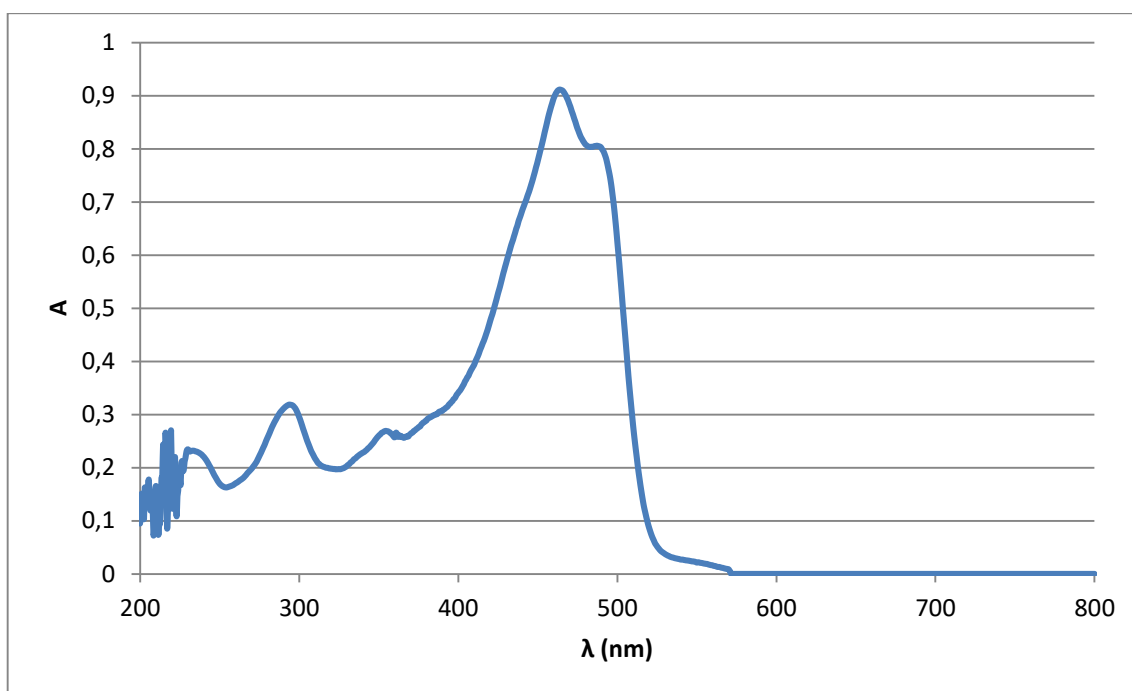
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## S1. Cyclic Voltammetry



**Figure S1.** Cyclic voltammogram of pentamer **9**. The voltammogram was run in a 1.0 mM solution of compound **9** in dry (over  $\text{CaH}_2$ ), HPLC grade DCM containing  $\text{TBAPF}_6$  (0.1 M) as an electrolyte. A three-electrode electrochemical cell was used with a glassy carbon disk as working electrode ( $\varnothing$  3 mm), Pt wire as counter electron and Ag/AgCl (1.0 M KCl) as reference electrode. Scan rate  $50 \text{ mV} \cdot \text{s}^{-1}$ . Temperature =  $20^\circ \text{C}$ .  $\text{Fc}/\text{Fc}^+$  ( $E_{\text{Fc}/\text{Fc}^+} = 0.475 \text{ V vs. SCE}$ ) [1] was used as an internal reference.

## S2. UV-vis Spectrum of pentamer **9**



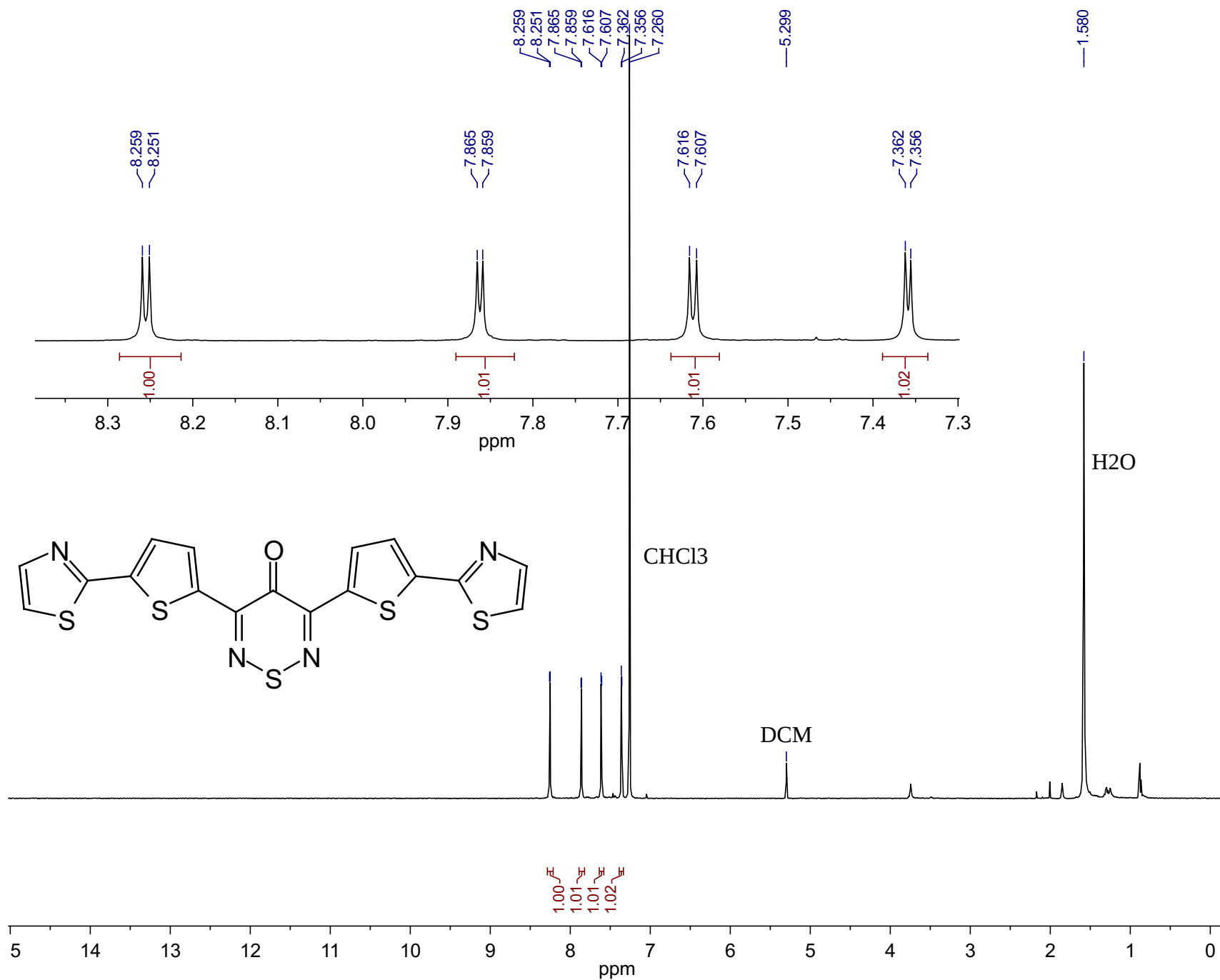
**Figure S2.** UV-vis absorption spectrum of pentamer **9** in CH<sub>2</sub>Cl<sub>2</sub> at 0.02 mM. Peaks: 300 (log  $\epsilon$  4.21), 354 (4.17), 470 (4.69), 489 inf (4.65).

## S3. References

1. Aranzaes, J. R.; Daniel, M.-C.; Astruc, D. Metallocenes as references for the determination of redox potentials by cyclic voltammetry - Permethylated iron and cobalt sandwich complexes, inhibition by polyamine dendrimers, and the role of hydroxy-containing ferrocenes. *Can. J. Chem.* **2006**, *84*, 288-299. DOI: 10.1139/v05-262.

**S4.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Spectra of pentamer 9**

<sup>1</sup>H NMR of 3,5-bis[5-(thiazol-2-yl)thien-2-yl]-4*H*-1,2,6-thiadiazin-4-one (9)



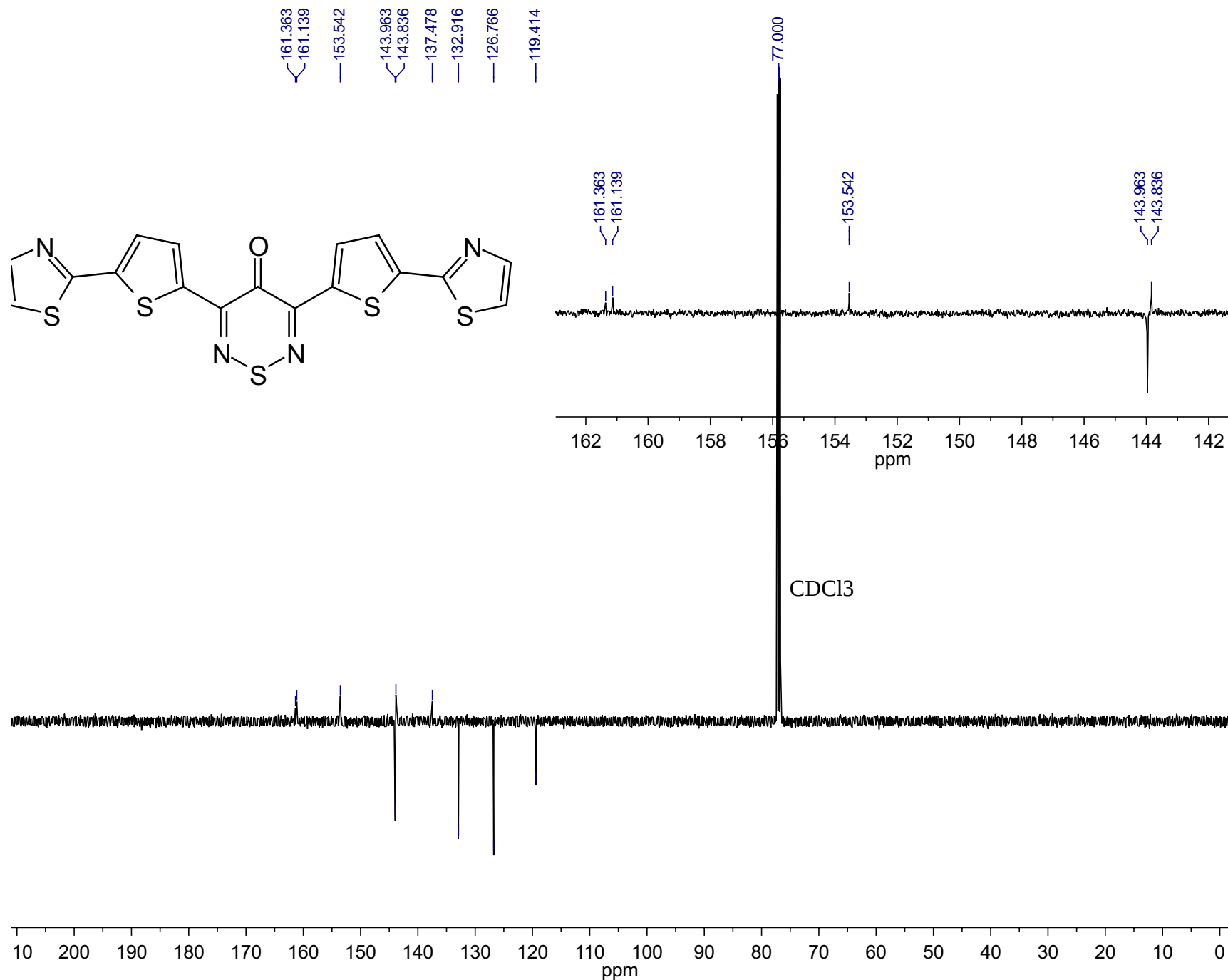
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<sup>13</sup>C NMR of 3,5-bis[5-(thiazol-2-yl)thien-2-yl]-4*H*-1,2,6-thiadiazin-4-one (**9**)



Current Data Parameters

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