

Supporting Information

Density Functional Theory Optimization of Cobalt- and Nitrogen-Doped Graphene Catalysts for Enhanced Oxygen Evolution Reaction

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Supplementary Figures:

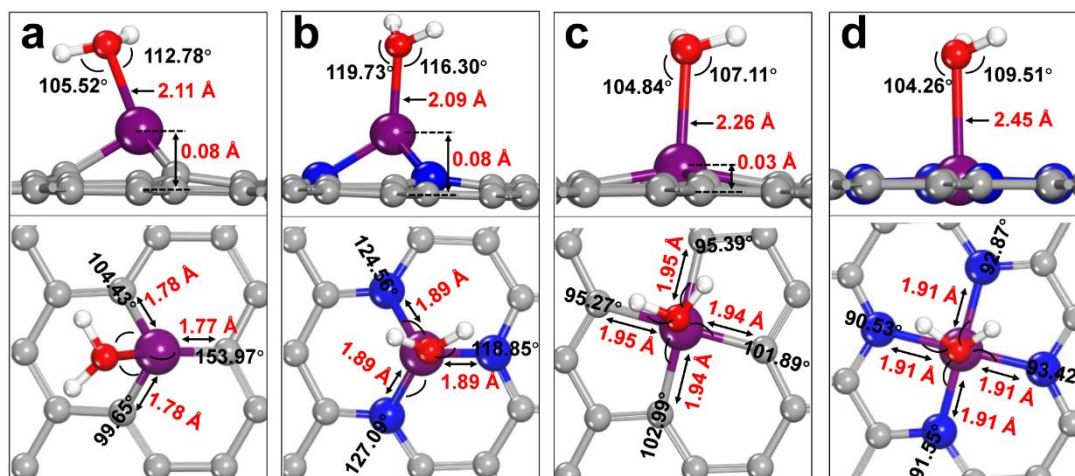


Figure S1. Bond lengths and bond angles of the stable Co-N-g catalyst of (a) SV-GN, (b) SV-N3, (c) DV-GN, and (d) DV-N4.

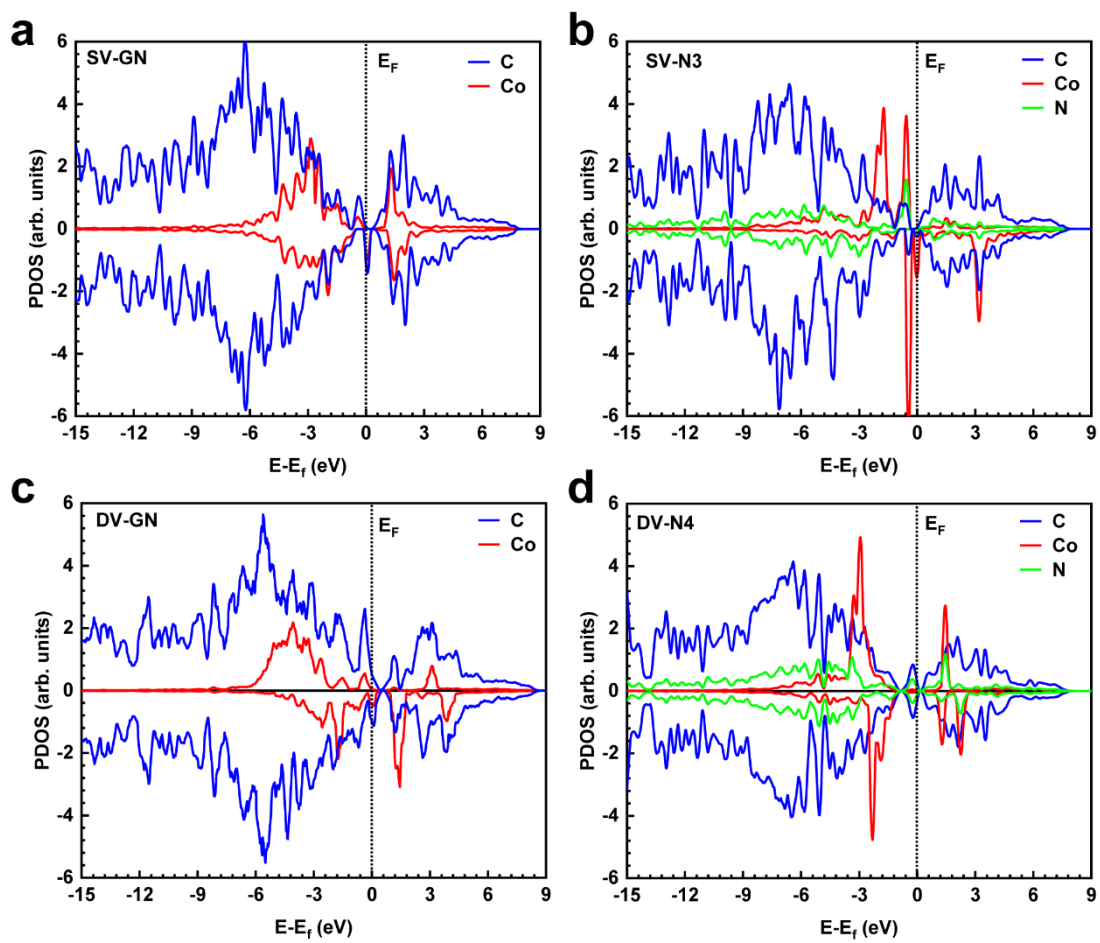


Figure S2. PDOS of specified elements C (blue), Co (red) and N (green), for Co-N-g catalyst of (a) SV-GN, (b) SV-N3, (c) DV-GN, and (d) DV-N4.

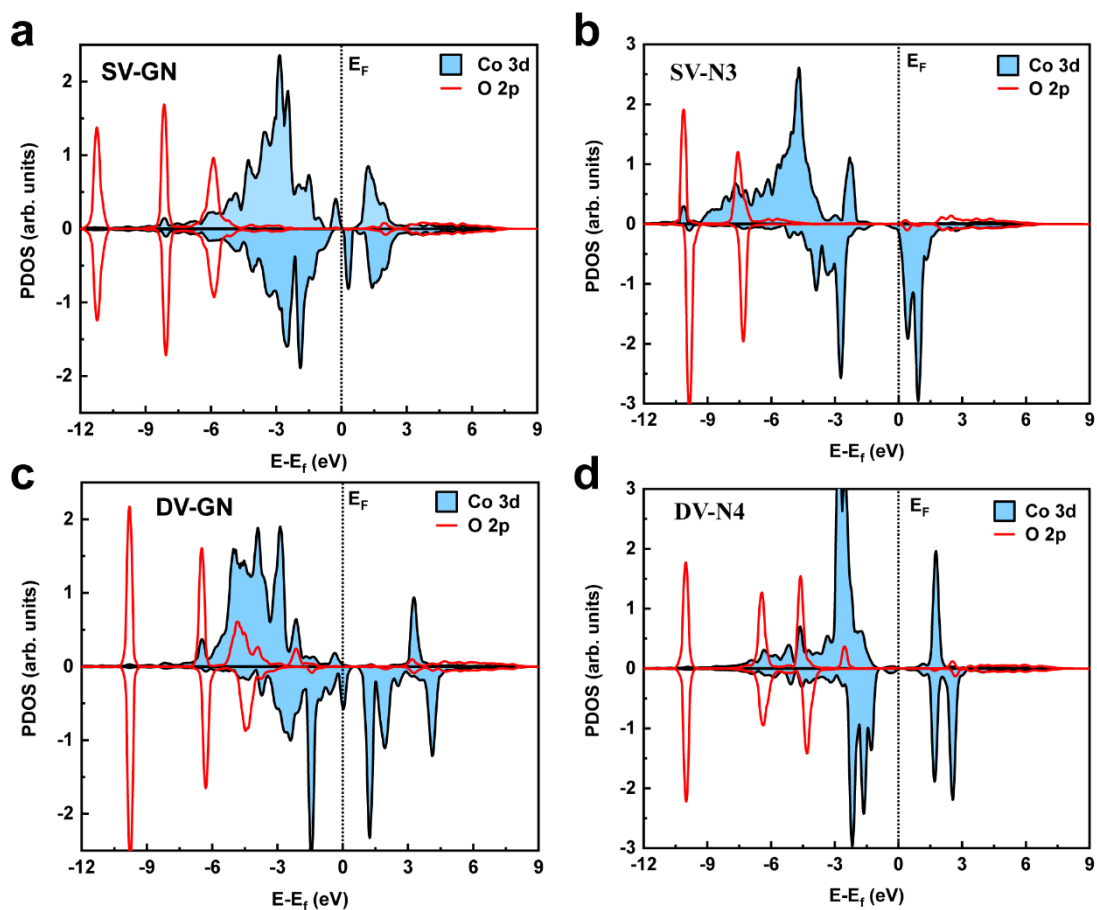


Figure S3. PDOS of Co-3d (blue) and O-2p states (red) after H₂O adsorption on the catalyst surface.

Table S1. Bader charge and d-band center values of SV-GN, SV-N3, DV-GN, and DV-N4 catalysts with H₂O adsorbed.

Catalysts	Bader charge (eV)		d-band center (eV)	
	Co of catalysts	Co of H ₂ O adsorbed on catalysts	O of H ₂ O adsorbed on catalysts	Co of H ₂ O adsorbed on catalysts
SV-GN	0.50	0.63	-1.19	-2.45
SV-N3	0.73	1.06	-1.24	-5.22
DV-GN	0.84	0.92	-1.29	-3.39
DV-N4	0.95	0.98	-1.25	-2.68

Table S2. HOMO, LUMO, and HOMO-LUMO gap^a of α electrons and β electrons for SV-GN, SV-N3, DV-GN, and DV-N4 catalysts with H₂O adsorbed. The unit is eV.

	SV-GN		SV-N3		DV-GN		DV-N4	
	α electron	β electron	α electron	β electron	α electron	β electron	α electron	β electron
HOMO	-7.99	-8.79	-6.41		-8.95	-8.29	-8.46	-7.32
LUMO	-0.10	-0.13	-3.27		-0.79	-0.99	-0.13	-0.04
HOMO-LUMO gap	7.89	8.66	3.14		8.16	7.30	8.33	7.28

^a HOMO-LUMO gap is the difference between LUMO and HOMO energy levels.