

Article

# Prospects of Biodiesel Production from Macadamia Oil as an Alternative Fuel for Diesel Engines

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Academic Editor: Maurizio Sasso

Received: 18 February 2016; Accepted: 11 May 2016; Published: 25 May 2016

**Abstract:** This paper investigated the prospects of biodiesel production from macadamia oil as an alternative fuel for diesel engine. The biodiesel was produced using conventional transesterification process using the base catalyst (KOH). A multi-cylinder diesel engine was used to evaluate the performance and emission of 5% (B5) and 20% (B20) macadamia biodiesel fuel at different engine speeds and full load condition. It was found that the characteristics of biodiesel are within the limit of specified standards American Society for Testing and Materials (ASTM D6751) and comparable to diesel fuel. This study also found that the blending of macadamia biodiesel–diesel fuel significantly improves the fuel properties including viscosity, density ( $D$ ), heating value and oxidation stability ( $OS$ ). Engine performance results indicated that macadamia biodiesel fuel sample reduces brake power ( $BP$ ) and increases brake-specific fuel consumption ( $BSFC$ ) while emission results indicated that it reduces the average carbon monoxide ( $CO$ ), hydrocarbons ( $HC$ ) and particulate matter ( $PM$ ) emissions except nitrogen oxides ( $NO_x$ ) than diesel fuel. Finally, it can be concluded that macadamia oil can be a possible source for biodiesel production and up to 20% macadamia biodiesel can be used as a fuel in diesel engines without modifications.

**Keywords:** alternative fuel; macadamia oil; properties; blending; emission

## 1. Introduction

The price hiking of petroleum derived fossil fuel and the depletion of the reserve of those fuels have attributed to the necessity of alternative fuel research [1]. Biodiesel is one of the alternative fuels that is produced from vegetable oils, animal fats and waste cooking oil through transesterification process [2–4]. Biodiesel is renewable [5], biodegradable [6] and non-toxic, which have the potential to reduce environment pollution and global warming significantly [7,8]. From the last decades, researchers have been trying to find out the biodiesel sources and already there are more than 350 oil-bearing crops that have been introduced to produce biodiesel [9,10]. The conventional biodiesel sources are palm, jatropha, coconut, sunflower, soybean, rapeseed, jojoba, neem, karanja, calophyllum, moringa, cotton, castor oil, and microalgae [11–15]. The feedstocks of biodiesel should be chosen from the sources that are locally available, easily accessible, and economically feasible and technically viable [16].

Australia is increasingly reliant on imported transportation fuel. The increasing demand and dependency on foreign fossil fuel and environmental concerns prompt to the need to explore opportunities to locally produced alternative sources such as biodiesel [17]. The Federal Government and State Governments have developed relevant policies to promote sustainable biodiesel industry to ensure Australian's long-term energy security. In this context, the research on *Macadamia Integrifolia* (*M. Integrifolia*) oil is paramount as this oil has native distribution as well as is available in Australia. macadamia belongs to Proteaceous family and is widely available in New South Wales and central

Queensland, which could be considered as a potential alternative source to produce biodiesel [18,19]. The fruits of macadamia are very hard and woody with a pointed apex containing one or two seeds. The seeds are brown in colour, and it is the only part that contains oil. The oil contents of seed are around 70% with golden yellow in colour [20]. Currently, the area under macadamias is almost 18,666 ha and Production for 2014 was estimated at 11,400 t kernel. Nearly 400,000 trees were established in the last five years. There are currently around six million macadamia trees under cultivation; about a third of these are yet to reach full production. By 2020, about 20,000 ha are will be planted to produce more than 16,000 t kernel and it is expected that the export value will be more than \$165 million. There are 35% of total production consumes in the local market in which 90% is sold as a kernel. Through the Australian Government's National Residue Survey, the Australian macadamia industry can demonstrate 15 years of 100% compliance with all relevant standards [21].

Many researchers [22] around the world studied the production of biodiesel from different sources, but a very few researcher [20] studied the production of macadamia biodiesel. Yunus *et al.* [22] studied the potential of biodiesel production from *Ceiba pentandra* (*C. pentandra*) and *Nigella sativa* (*N. sativa*) oil. They reported that the fuel properties of the *C. pentandra* biodiesel showed better fuel properties namely flash point (FP), calorific value and viscosity while *N. sativa* exhibited excellent cold flow properties and oxidation stability (OS). Silitonga *et al.* [1] studied the potential of biodiesel production from *Schleichera oleosa* oil. They reported that biodiesel from *Schleichera oleosa* oil could be used in diesel engine due to its properties. Knothe [20] studied the production of biodiesel from macadamia oil but the details data on macadamia biodiesel–diesel blend is not presented in his study. Similarly, there are many studies [23–27] found on the evaluation of different edible and non-edible oil biodiesel in diesel engines, but there is a lack of research on the on the assessment of macadamia biodiesel performance in diesel engines. Tesfa *et al.* [27] studied the emission behaviour of different biodiesel from various sources in a multi-cylinder diesel engine. They found that the engine running with biodiesel blends gave up to 20% higher nitrogen oxides (NO<sub>x</sub>) emission than diesel fuel. Other emissions result using neat biodiesel (B100) indicates that pure biodiesel reduces up to 15% carbon monoxide (CO), 40% CO<sub>2</sub> and 30% total hydrocarbon (THC) emissions, respectively than diesel fuel. Yoon *et al.* [28] studied the performance, emission, and combustion of canola biodiesel in a common rail diesel engine. Their studies indicate that over the entire range of speed, the brake-specific fuel consumption (BSFC) of canola biodiesel was increased while the CO and particulate matter (PM) emissions were significantly decreased. Other emissions like NO<sub>x</sub> emission increased slightly. The blending ratio had an influence on the engine emissions.

The objective of this study is to investigate the potential of macadamia oil as an alternative source along with the evaluation of the 5% and 20% macadamia biodiesel–diesel blends in a diesel engine. The fuel properties of produced biodiesel and its blend with diesel fuel were also analysed according to ASTM standards. The effect of macadamia biodiesel–diesel blending on fuel properties has been reported too.

## 2. Materials and Method

### 2.1. Materials

Crude macadamia oil was purchased from Coles, North Rockhampton, Australia. All other reagents, methanol, filter paper 150 mm were available in our chemical laboratory, Central Queensland University (North Rockhampton, Australia).

### 2.2. Biodiesel Fuel Production Procedure

In this study, a small-scale three-neck laboratory reactor 1 L in size equipped with reflux condenser, thermometer, and magnetic stirrer was used to produce biodiesel from crude macadamia oil. The free fatty acid of crude oil was found around two, which indicate that esterification is not necessary to produce biodiesel. For this reason, biodiesel was produced through only the transesterification process.

Figure 1 shows the production process of macadamia biodiesel. In this process, 1 L of preheated crude macadamia oil was reacted with 6:1 molar ratio of methanol to oil in the presence of 1% (*w/w*) of KOH catalyst. The reaction was maintained at 60 °C for 2 h at 800 rpm. After completion of the reaction, the mixture was poured into a separation funnel for 14 h to be cooled, settled and separated glycerol from biodiesel. The upper part of the funnel contains biodiesel, and the bottom was glycerine, which contains excess methanol and impurities. The biodiesel was collected, and the glycerine was drawn off. The produced biodiesel was then heated at 65 °C to remove any remaining methanol. Then, the biodiesel was washed using warm distilled water to remove all impurities. Finally, the washed biodiesel was dried using Na<sub>2</sub>SO<sub>4</sub> and filtered through a filter paper. Then the final product was collected and stored for characterization. The biodiesel yield was more than 90%, which was calculated using following Equation:

$$\text{Yield\%} = \left( \frac{\text{amount of methyl ester produced}}{\text{amount of oil taken}} \right) \times 100$$

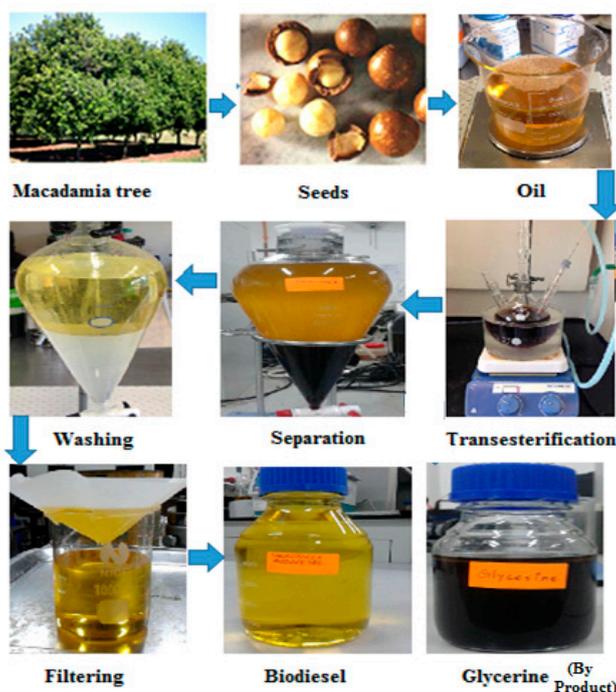


Figure 1. Production of macadamia biodiesel.

### 2.3. Biodiesel–Diesel Blending

The produced biodiesel was then blended with diesel fuel in a beaker glass through magnetic stirrer with maintaining speed at 2000 rpm for 30 min. In this study, biodiesel was blended with diesel fuel at a ratio of 10%, 20%, 30%, 40%, 50%, 60%, 70%, 80% and 90% by volume to examine the blending effect on fuel properties.

### 2.4. Characterization of Biodiesel Fuel

The physico-chemical properties of the crude oil, biodiesel and biodiesel–diesel blend were tested according to the ASTM D6751 standards. Table 1 shows the list of the equipment used in this study to characterise the biodiesel. Fatty acid composition was tested using gas chromatography (Agilent 6890 model, Wilmington, DE 19808-1610 USA). Fourier transform infrared (FT-IR) was done using a Perkin Elmer biodiesel FAME analyser equipped with the MIR TGS detector in the range of 4000–400 cm<sup>-1</sup>.

Cetane number (*CN*), iodine value (*IV*), saponification value (*SV*), degree of unsaturation (*DU*) and long chain saturated factor (*LCSF*) was determined using the following Equations [29]:

$$CN = 46.3 + (5458/SV) - (0.225 \times IV) \quad (1)$$

$$SV = \sum(560 \times A_i)/M_{wi} \quad (2)$$

$$IV = \sum(254 \times A_i \times D)/M_{wi} \quad (3)$$

$$LCSF = 0.1 \times (C16 : 0, wt \%) + 0.5 \times (C18 : 0 wt \%) + 1 \times (C20 : 0 wt \%) + 1.5 \times (C22 : 0 wt \%) + 2.0 \times (C24 : 0 wt \%) \quad (4)$$

$$DU = \sum(MUFA + 2 \times PUFA) \quad (5)$$

where  $A_i$  is the percentage of each component,  $D$  is the number of double bond and  $M_{wi}$  is the molecular mass of each component.

**Table 1.** List of equipment used in this study. *OS*: oxidation stability; *KV*: kinematic viscosity; *D*: density; *FP*: flash point.

Property	Equipment	Standard Method	Accuracy
<i>KV</i>	NVB classic (Norma lab, Valliquerville, France)	ASTM D445	$\pm 0.01 \text{ mm}^2/\text{s}$
<i>D</i>	DM40 LiquiPhysics™ density meter (Mettler Toledo, Columbia, MD, USA)	ASTM D127	$\pm 0.1 \text{ kg}/\text{m}^3$
<i>FP</i>	NPM 440 Pensky-martens FP tester (Norma lab)	ASTM D93	$\pm 0.1 \text{ }^\circ\text{C}$
Cloud and Pour Point	NTE 450 Cloud and pour point tester (Norma lab)	ASTM D2500	$\pm 0.1 \text{ }^\circ\text{C}$
Higher Heating Value	6100EF Semi auto bomb calorimeter (Parr, Moline, IL, USA)	ASTM D240	$\pm 0.001 \text{ MJ}/\text{kg}$
Acid Number	Automation titration rondo 20 (Mettler Toledo)	ASTM D664 and EN 14111	$\pm 0.001 \text{ mg} \cdot \text{KOH}/\text{g}$
<i>OS</i> , 110 °C	873 Rancimat (Metrohm, Herisau, Switzerland)	EN 14112	$\pm 0.01 \text{ h}$

## 2.5. Engine Test

A multi-cylinder diesel engine (model V3300, Kubota Tractor Australia PTY Ltd., Victoria, Australia) was used to perform the performance and emission test. In the performance and emission study, B5 and B20 fuels have been used based on the suggestion from the literature [30,31] that up to 20% biodiesel can be used in a diesel engine with no modifications. Engine performance data were collected at full load condition and at different speeds ranging from 1200 rpm to 2400 rpm at an interval of 100 rpm, whereas emission data were collected only at idle speed (800 rpm) and the speed at which maximum torque was found (1400 rpm). First, the engine was run using diesel fuel for a few minutes to warm up before switching to the biodiesel blend. Furthermore, the engine was run with diesel fuel before it was shut down. Figure 2 shows the engine test bed and Figure 3 shows the schematic diagram of the engine test bed. Table 2 shows the specifications of the engine used in this study. A CODA 5 exhaust gas analyser (Coda Products Pty Ltd., Hamilton, Australia) was used to measure the  $\text{NO}_x$ , hydrocarbons (HC), and CO emissions from the engine and A MPM-4M particulate monitor (Pacific Data System, Queensland, Australia) was used to measure the PM emission.



Figure 2. Engine test bed used in this study.

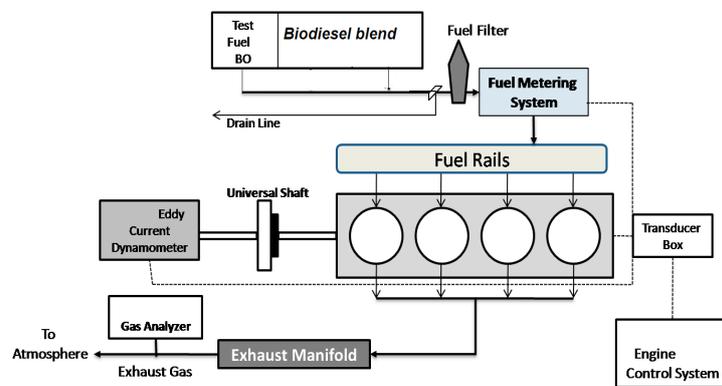


Figure 3. Schematic diagram of the engine test bed.

Table 2. The specifications of the engine used in this study.

Serial No.	Model	Kubota V3300
1	Type	Vertical, 4 cycle liquid cooled diesel
2	No. of cylinder	4
3	Total displacement (L)	3.318
4	Bore × Stroke (mm)	98 × 110
5	Combustion system	E-TVCS
6	Intake system	Natural aspired
7	Output: Gross intermittent (kW/rpm); Net intermittent (Rated power output) (kW/rpm); Net continuous (kW/rpm)	54.5/2600; 50.7/2600; 44.1/2600
8	Rated Torque (N-m/rpm)	230/1400
9	Compression ratio	22.6
10	No load high idling speed (rpm)	2800
11	No load low idling speed (rpm)	700–750
12	Direction of rotation	Counter clockwise (viewed from flywheel side)
13	Governing	Centrifugal fly weight high speed governor
14	Fuel	Diesel fuel No-2-D (ASTM D975)
15	Starter capacity (V-kW)	12–2.5
16	Alternator capacity (V-A)	12–60

### 3. Results and Discussion

#### 3.1. Crude Oil Properties

The crude oil from macadamia was characterised by viscosity, density, FP, acid value, and higher heating value (HHV). The properties of macadamia oil are presented in Table 3. The viscosity of crude oil was found 39.22 mm<sup>2</sup>/s, which is 11–12 times greater than conventional diesel fuel. The FP, pour point, and cold filter plugging point was 167.5 °C, 8 °C, and 9 °C, respectively. The acid value was determined as 4 mg·KOH/g, which is similar to other conventional biodiesel feedstocks.

**Table 3.** Properties of crude macadamia oil.

Serial No.	Properties	Macadamia Oil
1	Dynamic Viscosity (mPa.s)	35.23
2	KV (mm <sup>2</sup> /s) at 40 °C	39.22
3	D (kg/m <sup>3</sup> ) at 15 °C	898.60
4	Higher Heating Value (MJ/kg)	38.20
5	Acid Value (mg·KOH/g)	4
6	FP (°C)	167.5
7	Pour Point (°C)	8
8	Cold Filter Plugging Point (°C)	9

#### 3.2. Characteristics of Macadamia Biodiesel Fuel

The fuel properties of macadamia biodiesel were analysed and compared with diesel and ASTM D6751 standards. Table 4 shows the fuel properties of macadamia biodiesel. It was found that the kinematic viscosity (KV) of macadamia is 4.46 mm<sup>2</sup>/s and the acid value lowered to 0.07 mg·KOH/g. However, all these results are within the specified limit ASTM D6751 standards (1.9–6 mm<sup>2</sup>/s). The FP was found 178.5 °C, which is much higher than diesel fuel (68.5 °C) that indicates macadamia biodiesel fuel is safer to handle and storage. The OS and HHV of macadamia biodiesel were found 3.35 h and 39.90 MJ/kg, respectively.

**Table 4.** Properties of macadamia biodiesel compared to other fuels.

Properties	Unit	<i>M. Integrifolia</i> Biodiesel	Diesel	ASTM D6751 [32]
KV at 40 °C	mm <sup>2</sup> /s	4.46	3.23	1.9-6
D at 15 °C	kg/m <sup>3</sup>	859.2	827.2	-
Higher Heating Value	MJ/kg	39.90	45.30	-
OS	h	3.35	-	3 min
Acid Value	mg·KOH/g	0.07	-	0.05 max
FP	°C	178.5	68.5	130 min
Pour Point	°C	0	0	-
Cloud Point	°C	8	8	report
CFPP	°C	8	5	-
Cetane Number	-	56	48	47 min
Iodine Number	-	77.85	-	-
Saponification Value	-	199	-	-

##### 3.2.1. Fatty Acid Composition

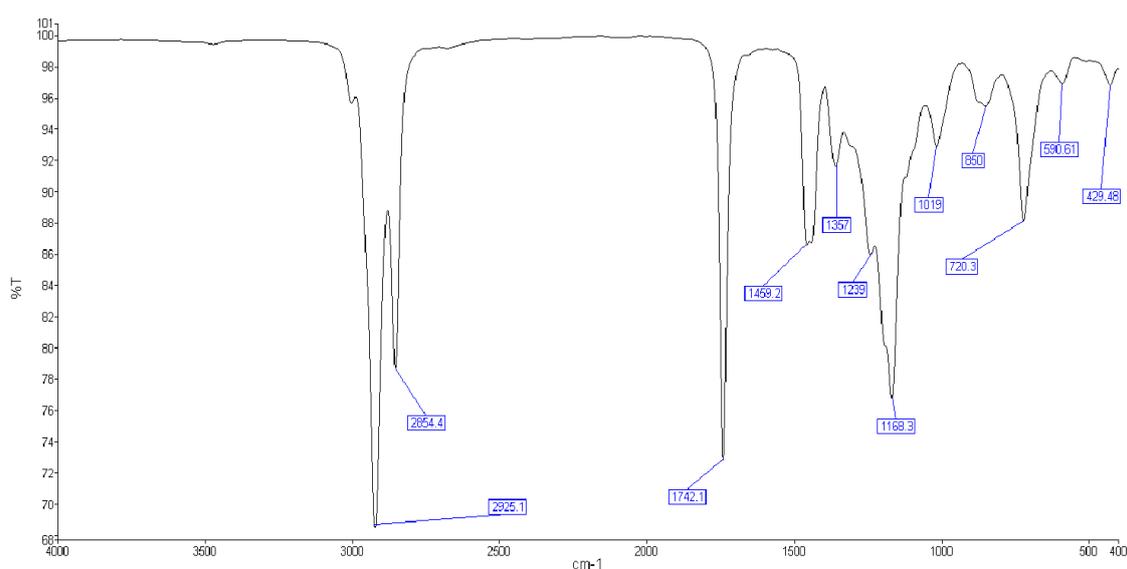
Fatty acids are categorised into saturated and unsaturated fatty acid. A fatty acid that does not contain double bond is known as saturated fatty acid, and that contains double bond is known as unsaturated fatty acid. Table 5 shows the fatty acid composition of macadamia oil. It can be seen that macadamia biodiesel has 15.80% saturated and 82.60% unsaturated fatty acids. Oleic acid (18:1) was the predominant fatty acid (61.3%) in macadamia biodiesel sample. The degree of unsaturation and long chain saturated factor was found 84.50 and 7.24, respectively.

**Table 5.** Fatty acid composition of macadamia biodiesel. DU: degree of unsaturation; LCSF: long chain saturated factor.

Fatty Acids	Molecular Weight	Structure	Formula	<i>M. Integrifolia</i> Biodiesel (wt%)
Lauric	200	12:0	C <sub>12</sub> H <sub>24</sub> O <sub>2</sub>	0.1
Myristic acid	228	14:0	C <sub>14</sub> H <sub>28</sub> O <sub>2</sub>	0.6
Palmitic	256	16:0	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	7.9
Palmitoleic	254	16:1	C <sub>16</sub> H <sub>30</sub> O <sub>2</sub>	16.2
Stearic	284	18:0	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	3.2
Oleic	282	18:1	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	61.3
Linoleic	280	18:2	C <sub>18</sub> H <sub>32</sub> O <sub>2</sub>	2.1
Linolenic	278	18:3	C <sub>18</sub> H <sub>30</sub> O <sub>2</sub>	0.1
Arachidic	312	20:0	C <sub>20</sub> H <sub>40</sub> O <sub>2</sub>	2.7
Eicosenoic	310	20:1	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	2.6
Behenic	340	22:0	C <sub>22</sub> H <sub>44</sub> O <sub>2</sub>	0.9
Erucic	338	22:1	C <sub>22</sub> H <sub>42</sub> O <sub>2</sub>	0.3
Lignoceric	368	24:0	C <sub>24</sub> H <sub>48</sub> O <sub>2</sub>	0.4
Total saturated fatty acid				15.80
Total monounsaturated fatty acid (MUFA)				80.40
Total polyunsaturated fatty acid (PUFA)				2.20
Others				1.6
DU				84.50
LCSF				7.24

### 3.2.2. Fourier Transform Infrared Analysis

Macadamia biodiesel was also characterised using FT-IR to identify the characteristics peaks. The resolution was 4 cm<sup>-1</sup> and eight scans. Figure 4 shows the FT-IR spectrum of the macadamia biodiesel. FT-IR spectra shows the characteristics peaks of biodiesel at 2925–2854 cm<sup>-1</sup> due to C–H stretching vibration and CH<sub>2</sub> asymmetric and symmetric vibration, respectively; 1742 cm<sup>-1</sup> equivalent to C=O stretching vibration; 1459 cm<sup>-1</sup> equivalent to CH<sub>2</sub> shear vibration; 1357 cm<sup>-1</sup> equal to CH<sub>3</sub> bending vibration; 1168 cm<sup>-1</sup> equivalent to C–O–C symmetric stretching vibration; and 1019 cm<sup>-1</sup> equivalent to C–O–C antisymmetric stretching vibration. This result reflects the conversion of triglycerides to the methyl ester.

**Figure 4.** Fourier transform infrared (FT-IR) Spectrum of macadamia biodiesel.

### 3.2.3. Properties of Biodiesel–Diesel Blending

The blending of fossil diesel fuel with biodiesel is an important idea to improve the properties of biodiesel fuel as it is reported that diesel fuel contains zero oxygen [33]. Macadamia has poor

OS, which could be improved by blending with diesel fuel. In Table 6, it can be seen that stability performance of all the blends increased with the addition of diesel fuel with biodiesel and all values are within the specified limits of both ASTM D6751 (3 h) and EN14112 (6 h). The properties result of macadamia biodiesel–diesel blends is given in Table 6. All results indicate that by blending biodiesel with diesel fuel offers a significant improvement of biodiesel fuel properties. All the results were found within the ASTM D6751 standards.

**Table 6.** Properties of macadamia biodiesel–diesel blends. *M. Integrifolia: Macadamia Integrifolia*.

Biodiesel–Diesel Blending	Viscosity (mm <sup>2</sup> /s) at 40 °C	D (kg/m <sup>3</sup> ) at 15 °C	Higher Heating Value (MJ/kg)	OS (h)
<i>M. Integrifolia</i> 10%	3.31	825.30	43.64	43.8
<i>M. Integrifolia</i> 20%	3.40	828.90	42.99	43.0
<i>M. Integrifolia</i> 30%	3.53	832.63	42.47	41.3
<i>M. Integrifolia</i> 40%	3.63	836.20	41.93	38.8
<i>M. Integrifolia</i> 50%	3.78	840.15	41.29	35.3
<i>M. Integrifolia</i> 60%	3.86	843.80	40.79	31.0
<i>M. Integrifolia</i> 70%	4.03	847.67	40.12	25.7
<i>M. Integrifolia</i> 80%	4.15	851.30	39.67	19.6
<i>M. Integrifolia</i> 90%	4.28	855.19	38.94	12.6

The following empirical equations have been developed from Figure 5a–d to predict the *KV*, *D*, *HHV* and *OS* of any macadamia–diesel blend (where *x* denotes % of macadamia biodiesel in the blend).

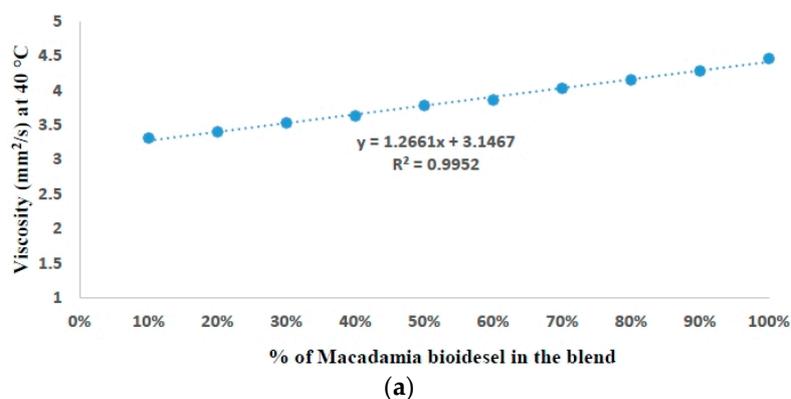
$$KV = 1.2661x + 3.1467 \quad 0 \leq x \leq 100 \quad R^2 = 0.9952 \quad (6)$$

$$D = 37.608x + 821.35 \quad 0 \leq x \leq 100 \quad R^2 = 0.9999 \quad (7)$$

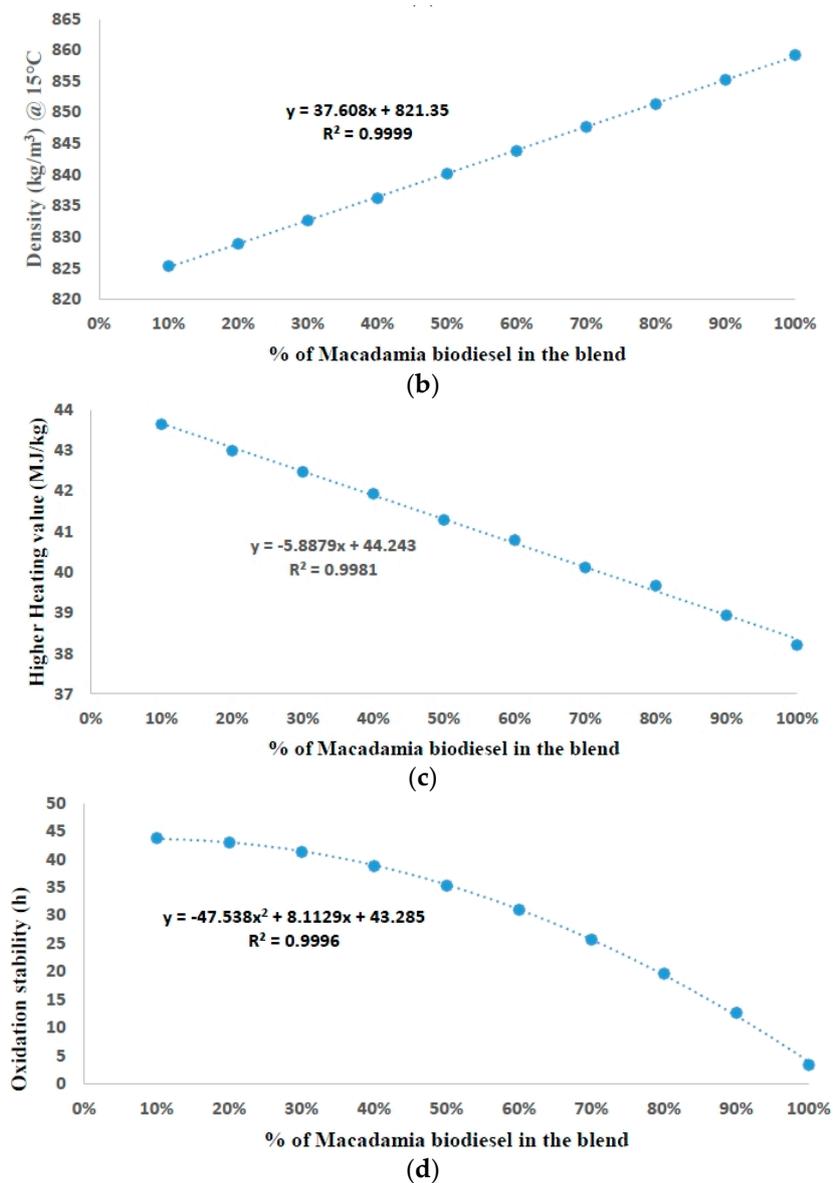
$$HHV = -5.8879x + 44.273 \quad 0 \leq x \leq 100 \quad R^2 = 0.9999 \quad (8)$$

$$OS = -47.538x^2 + 8.1129x + 43.285 \quad 0 \leq x \leq 100 \quad R^2 = 0.9999 \quad (9)$$

It is seen that the blending has a significant effect on both *KV*, *D*, *HHV*, and *OS*. For example, the addition of 10% diesel with biodiesel reduced viscosity from 4.45 mm<sup>2</sup>/s to 4.28 mm<sup>2</sup>/s. The value of *D* and *KV* increases as the percentages of biodiesel in the blends increases, whereas *HHV* and *OS* decrease as the percentages of biodiesel increases in the blends as expected. The *KV*, *D* and *HHV* of blends showed a linear relationship with the significant regression (*R*<sup>2</sup>) value 0.9952, 0.9999 and 0.9981, respectively, whereas the only *OS* of blends showed a polynomial relationship with the *R*<sup>2</sup> value of 0.9996. It is evident that macadamia biodiesel–diesel blend significantly improves the *KV*, *D*, *HHV* and *OS* of biodiesel.



**Figure 5.** Cont.



**Figure 5.** Effect of macadamia biodiesel–diesel blending on (a) *KV*; (b) *D*; (c) higher heating value (*HHV*); and (d) *OS*.

### 3.3. Brake Power

Figure 6 shows the engine brake power (BP) output of B5 (5% macadamia biodiesel and 95% diesel fuel), B20 (20% macadamia biodiesel and 80% diesel fuel) and B0 (pure diesel fuel) at different engine speeds. Figure 6 clearly shows that brake power for diesel fuel is higher than biodiesel blended fuel, which is also supported by other researchers [23,34]. On average, the highest brake power was found for diesel fuel followed by B5 and B20, 37.7, 37 and 36 kW, respectively. At all test speeds, B5 and B20 fuels reduce the brake power by 2% and 4.56%, respectively. The reduction of BP for biodiesel–diesel blended fuel can be attributed to the *HHV* of biodiesel fuel [35]. The *HHV* of macadamia biodiesel is lower than diesel fuel (Table 4).

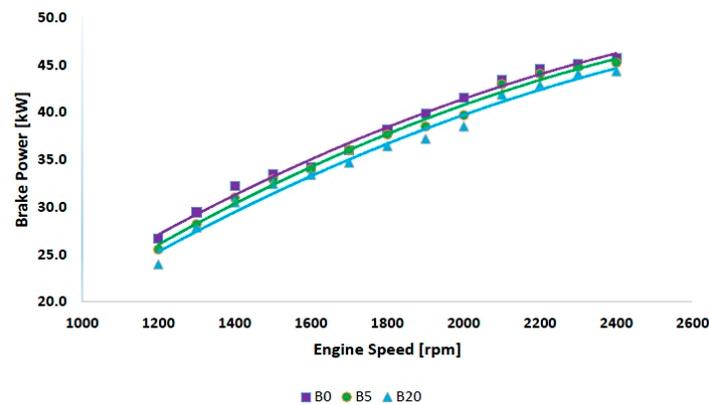


Figure 6. Variation of brake power with speeds.

### 3.4. Brake Specific Fuel Consumption

The properties of fuel such as the calorific value, density, and viscosity play a primary role in engine fuel consumption [25]. Figure 7 shows the BSFC of B5, B20, and B0 at different engine speeds. Figure 7 clearly demonstrates that over the entire range of speed, the macadamia biodiesel–diesel blended fuel gives a higher BSFC than diesel fuel, which is supported by the literature [36]. On average, BSFC of B0, B5 and B20 fuel are 203 g/kWh, 216 g/kWh and 285.2 g/kWh, respectively, and BSFC of B5 and B20 fuels are, respectively, 6% and 28.8% higher than diesel fuel. Higher fuel consumption is the result of higher density and lower calorific value of biodiesel, which causes the higher mass injection for the same volume [37]. The density of macadamia biodiesel is higher and the heating value is lower than the diesel fuel.

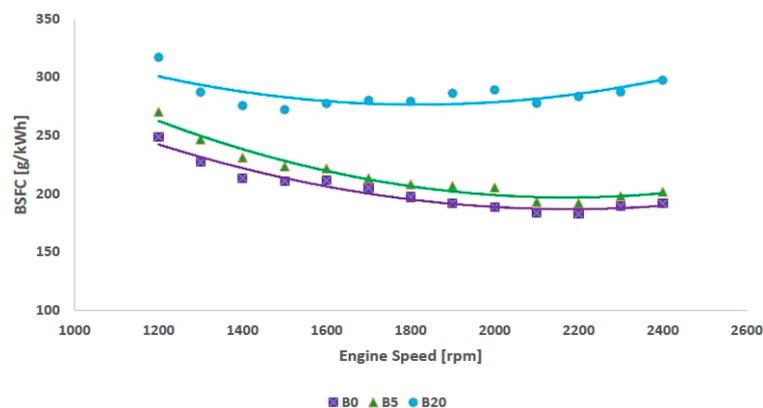


Figure 7. Variation of brake specific fuel consumption with speeds.

### 3.5. Carbon Monoxide Emission

Figure 8 compares the CO emission of B5, B20, and B0 at different engine speeds. Figure 8 clearly shows that, over the entire range of speed, the macadamia biodiesel fuels provide lower CO emission than diesel fuel, which is supported by other studies [38]. Fuel sample B0 produces 1100 ppm and 2200 ppm CO emissions while B5 produces 1010 ppm and 1950 ppm and B20 produces 600 ppm and 1000 ppm at 800 rpm and 1400 rpm, respectively. At 800 rpm, B5 and B20 reduce CO emission by 8.2% and 45.45%, respectively, whereas they reduce CO emission by 11.3% and 54.54%, respectively, at 1400 rpm compared with diesel fuel. The reason of lowering CO emission for biodiesel–diesel blended fuel can be attributed to the higher oxygen contents and higher cetane number of biodiesel fuel. Biodiesel fuel contains 12% higher oxygen, which allows more carbon molecules to burn and combustion becomes completed [39].

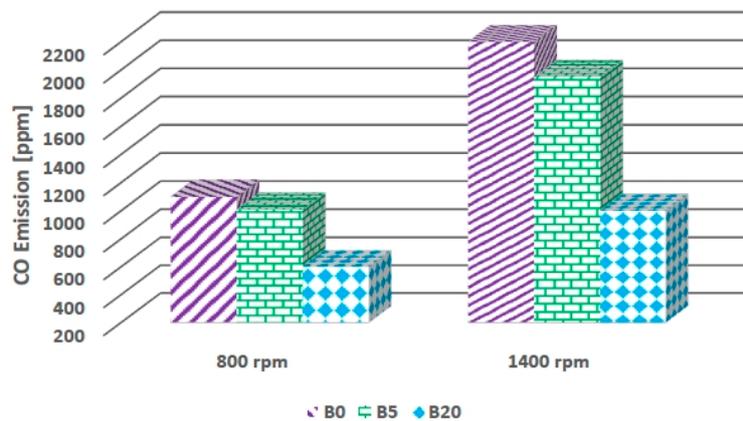


Figure 8. Variation of carbon monoxides emission with speeds.

### 3.6. Hydrocarbon Emission

Figure 9 compares the HC emission of B5, B20, and B0 at different engine speeds. Figure 9 clearly shows that, at all engine speeds, the macadamia biodiesel fuel reduces HC emission significantly than diesel fuel, and this result is in agreement with the results reported by [40]. Fuel sample B0 produces 10 ppm and 14 ppm HC emissions while B5 produces 6 ppm and 12 ppm and B20 produces 5 ppm and 9 ppm at 800 rpm and 1400 rpm respectively. At 800 rpm, B5 and B20 reduce HC emission by 40% and 50%, respectively, whereas they reduce HC emission by 14% and 36%, respectively, at higher engine speed compared with diesel fuel. Lowering HC emission for biodiesel fuel can be explained by the higher oxygen contents and higher cetane number of biodiesel fuel as explained for CO emission [35].

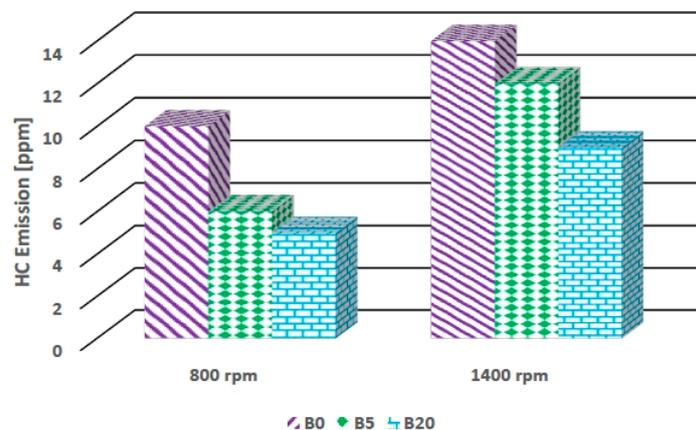
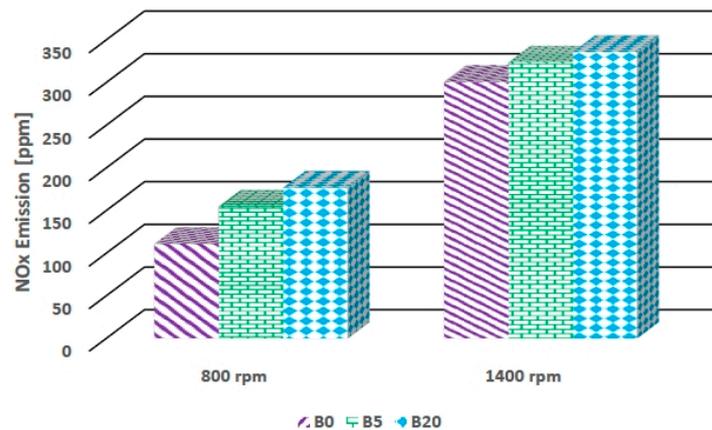


Figure 9. Variation of hydrocarbon emission with speeds.

### 3.7. Nitrogen Oxides Emission

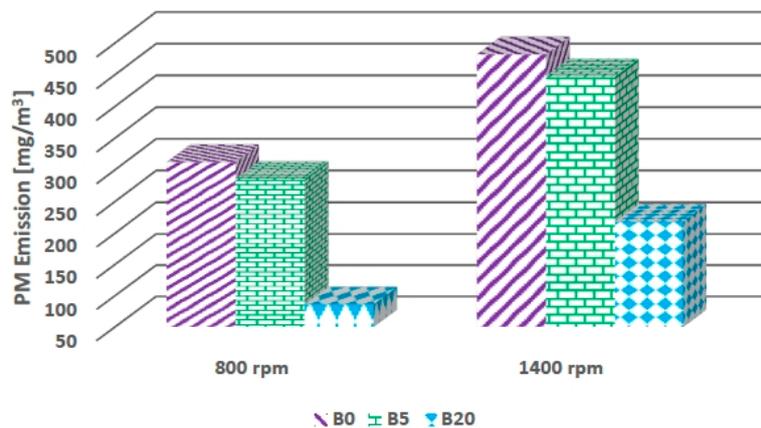
Figure 10 compares the  $\text{NO}_x$  emission of B5, B20, and B0 at different engine speeds. It is seen that, over both speed conditions, biodiesel blended fuels increase  $\text{NO}_x$  emission compared to diesel fuel. Özçelik *et al.* [41] also reported a similar result for  $\text{NO}_x$  emission when they tested camellia biodiesel blend in a diesel engine. Fuel sample B0 produces 111 ppm and 301 ppm  $\text{NO}_x$  emissions while B5 produces 155 ppm and 324 ppm and B20 provides 177 ppm and 336 ppm at 800 rpm and 1400 rpm, respectively. At 800 rpm, B5 and B20 increase  $\text{NO}_x$  emission by 28% and 37%, respectively, whereas they also increase  $\text{NO}_x$  emission by 7% and 10%, respectively, at higher engine speed compared with diesel fuel. Increase in  $\text{NO}_x$  emission for biodiesel fuel can be explained by the advance in combustion and higher oxygen contents of biodiesel fuel [42]. Biodiesel fuel has higher cetane number, which shortens the ignition delay, thereby improving combustion [35].



**Figure 10.** Variation of nitrogen oxides emission with speeds.

### 3.8. Particulate Matter Emission

Figure 11 compares the PM emission of B5, B20, and B0 at different engine speeds. It clearly shows that, at both 800 rpm and 1400 rpm speed, the macadamia biodiesel fuel reduces PM emission significantly compared to diesel fuel, and this result is in agreement with the results reported by Qi *et al.* [43]. Fuel sample B0 produces 312.2 mg/m<sup>3</sup> and 482.1 mg/m<sup>3</sup> PM emissions, while B5 produces 285.4 mg/m<sup>3</sup> and 447.5 mg/m<sup>3</sup> and B20 provides 86.1 mg/m<sup>3</sup> and 218 mg/m<sup>3</sup> at 800 rpm and 1400 rpm, respectively. At 800 rpm, B5 and B20 reduce PM emission by 8.6% and 72.4%, respectively, whereas they reduce PM emission by 7.2% and 54.8%, respectively, at 1400 rpm compared with diesel fuel. Lowering PM emission for biodiesel fuel can be explained by the higher oxygen contents which help to complete combustion [35] and higher cetane number of biodiesel fuel [44].



**Figure 11.** Variation of particulate matter emission with speeds.

## 4. Error Analysis

Experimental error and uncertainty can be arisen from different ways such as instrument selection, condition, calibration, environment, observation, reading, and test planning. Therefore, uncertainty analysis is required to prove the accuracy of the experiments. Table 7 presents the summary of measurement accuracy and the relative uncertainty of the different measured parameters.

**Table 7.** Relative uncertainty for parameters brake power (BP), brake-specific fuel consumption (BSFC), carbon monoxide (CO), hydrocarbons (HC), nitrogen oxides (NO<sub>x</sub>) and particulate matter (PM) emission.

Measurements	Accuracy	Relative Uncertainty	Average Reading for Diesel
BP	±0.07 kW	(±0.07/37.7) = 0.002	37.7
BSFC	±5 g/kWh	(±5/203) = 0.024	203
CO	±10 ppm	(±10/1650) = 0.006	1650
HC	±1 ppm	(±1/12) = 0.083	12
NO <sub>x</sub>	±1 ppm	(±1/206) = 0.005	206
PM	±0.1 mg/m <sup>3</sup>	(±0.1/397) = 0.0003	397

## 5. Conclusions

In this study, macadamia biodiesel was obtained through transesterification process using the base catalyst (KOH) with the superb conversion efficiency (more than 90% of biodiesel was obtained). It was found that all the physical and chemical properties meet the ASTM D6751 standards. Moreover, the blending of macadamia biodiesel with diesel fuel improves the fuel properties significantly, in particular, the OS. Engine performance study shows that the average brake power for macadamia biodiesel blended fuels are up to 4.6% lower and BSFC values are 28.8% higher than diesel fuel due to the lower calorific value of macadamia biodiesel. Blended fuel also reduces CO emissions up to 54.54%, HC emissions up to 50% and PM emissions by 72.4%, but increases NO<sub>x</sub> emission (10%) at higher engine speed compared to diesel fuel. The reason could be attributed to the higher oxygen contents and higher cetane number of biodiesel fuel.

Finally, it can be suggested that macadamia oil can be a potential source for biodiesel production and up to 20% macadamia biodiesel can be used as a fuel for diesel engines as a mixture of diesel fuel without any modification.

**Acknowledgments:** This work was conducted under the International Postgraduate Research Award (IPRA) Scholarship funded by the Central Queensland University, Australia.

**Author Contributions:** The contributions of each author are as follows: Md Mofijur Rahman produced biodiesel, collected experimental data, analysed the numerical results, drafted and revised the manuscript; Mohammad Rasul contributed to the experimental design and thoroughly revised the paper; Nur Md Sayeed Hassan helped to revise the paper according to the reviewer comments; and Justin Hyde checked the revised manuscript.

**Conflicts of Interest:** The authors declare no conflict of interest.

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