

A Highly Flexible Supercapacitor Based on MnO₂/RGO Nanosheets and Bacterial Cellulose-Filled Gel Electrolyte

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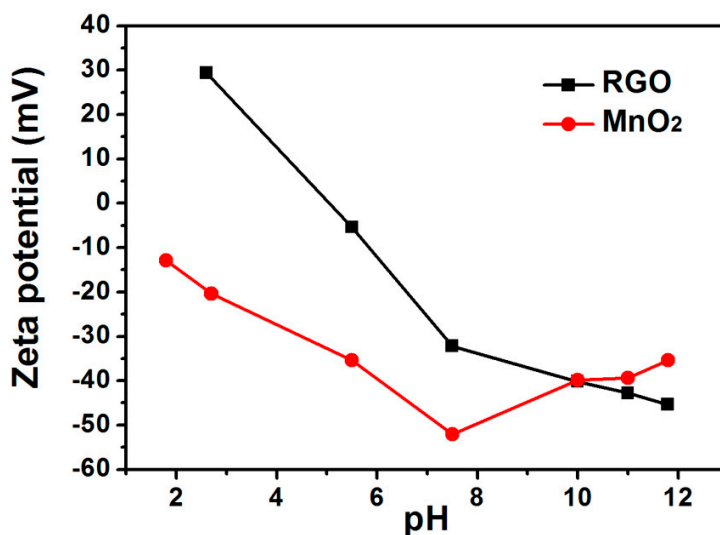


Figure S1. Zeta potential of RGO and MnO₂ as a function of pH, in aqueous dispersion, adjusted by HCl and NH₃·H₂O

Table S1. Zeta potentials (Z) of RGO and MnO₂ at various pH, respectively.

pH	1.8	2.6	2.7	5.5	7.5	10	11	11.8
Z of RGO (mV)	–	29.4	–	-5.4	-32.2	-40.2	-42.8	-45.4
Z of MnO ₂ (mV)	-12.9	–	-20.4	-35.4	-52.1	-39.8	-39.4	-35.4

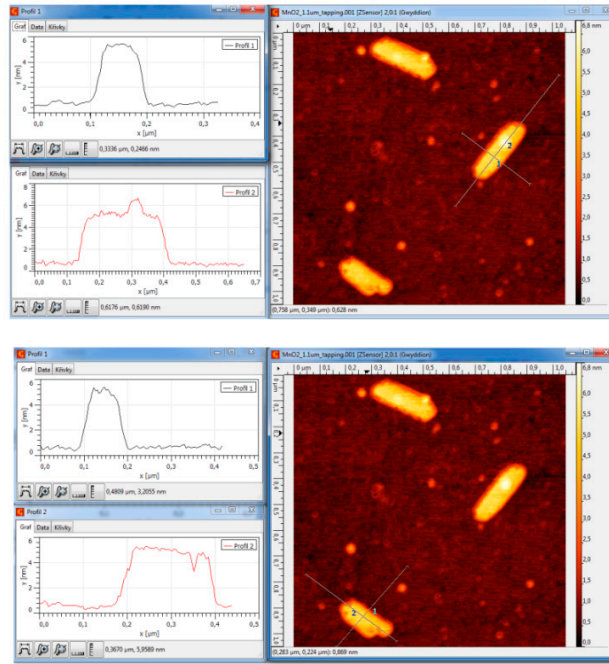


Figure S2. AFM images of MnO₂ show a uniform thickness of ~ 4.5 nm.

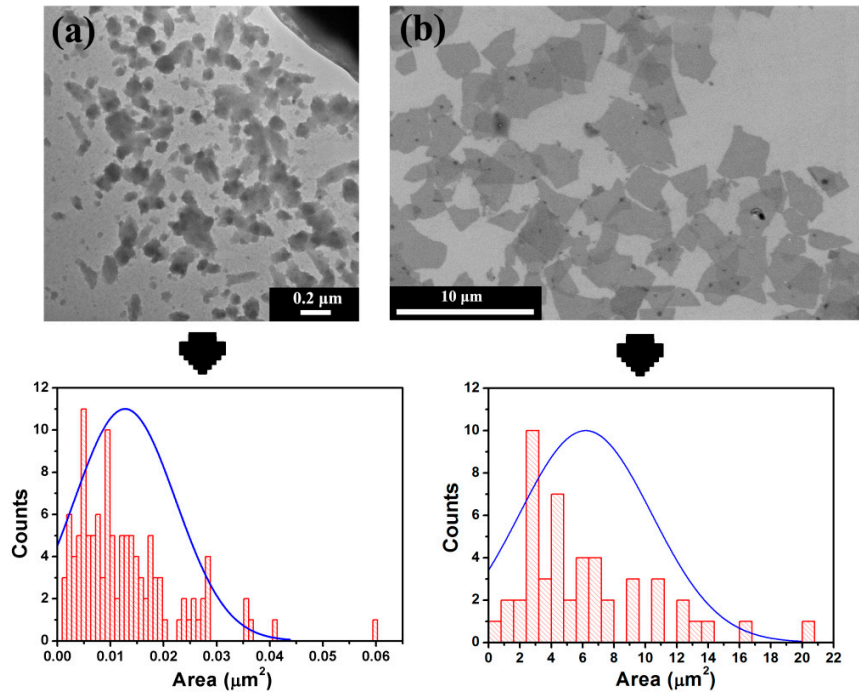


Figure S3. (a) TEM image of MnO₂ presents a lateral size of 100 \sim 300 nm; (b) SEM image of RGO on silicon wafer shows a lateral size of 2 \sim 5 μm and their corresponding area distribution.

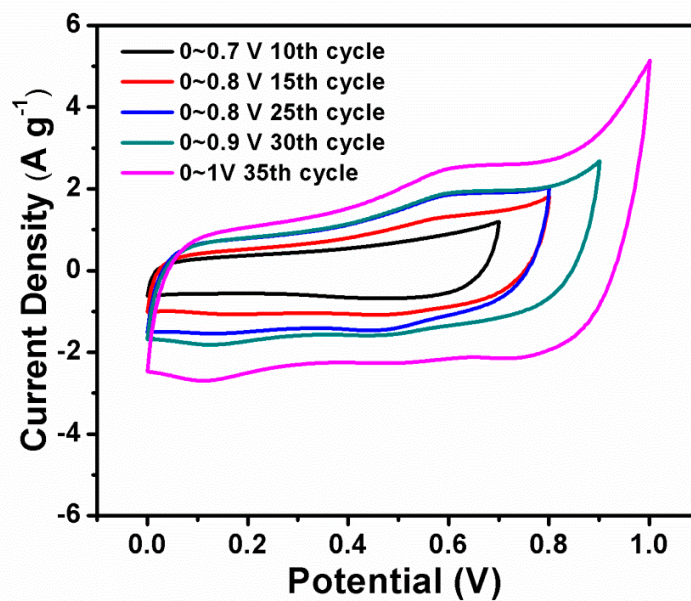


Figure S4. CV curves of MnO_2/RGO at a scan rate of 10 mV s^{-1} with various cycles collected in a three-electrode system with an Ag/AgCl reference in $1 \text{ M Na}_2\text{SO}_4$. The redox peaks appear when the potential window extended and increases after cycling.