

Supplementary

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An Easy-made, Economical and Efficient Carbon-doped Amorphous TiO₂ Photocatalyst Obtained by Microwave Assisted Synthesis for the Degradation of Rhodamine B

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Received: 12 October 2017; Accepted: 11 December 2017; Published: 20 December 2017



Materials 2017, 10, x; doi: FOR PEER REVIEW

www.mdpi.com/journal/materials



Figure S1. Elemental mapping from the EDS analysis of all the photocatalyst samples prepared in this work.



Figure S2. The relation between energy $(h\nu, eV)$ and $(F(R)h\nu)^2$ for TiO₂-DP25.



Figure S3. UV-vis spectra for the monitoring of the absorption of Rhodamine B in the dark onto the catalyst surface for the CD-aTiO₂ samples prepared.



Figure S4. $-\ln(C/C_0)$ vs t plot in which it is apparent that the Rhodamine B degradation data do not fit for a pseudo first order kinetic equation.



Figure S5. Emission spectrum for the lamp used in the photocatalytic experiments with visible light.



Figure S6. Comparison of the Rhodamine B solution (20 mg/L) degradation using a carbon doped TiO2 photocatalyst (DTiB-04) and another TiO2 material without carbon (DTiB-06).



Figure S7. Rhodamine B solution (5 mg/L) degradation plots for the DTiB-05 and the DTiB-05TT photocatalysts.



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