## **Supplementary Materials**

## Characterization and applications of robustly grafted kaolinite using ionic liquid with naphthyl functionality

Gustave Kenne Dedzo<sup>1,2\*</sup> and, Christian Detellier<sup>1</sup>

- <sup>1</sup> Center for Catalysis Research and Innovation and Department of Chemistry and Biomolecular Sciences, University of Ottawa, Ottawa, Ontario K1N 6N5, Canada.
- <sup>2</sup> Laboratory of Analytical Chemistry, Faculty of Science, University of Yaounde I, B.P. 812, Yaoundé, Cameroon.
- \* Correspondence: gkennede@uottawa.ca



Figure S1. Powdered XRD patterns of (a) kaolinite, (b) K-NI and K-NI sonicated for 30 minutes in (c) water and (d) toluene.



Figure S2.TGA curves of K-NI (a) under  $N_2$ , (b) under air.



Figure S3. FTIR Spectra of (a) kaolinite and (b) K-NI.



Figure S4. Effect of the scanning rate on the signal recorded on GCE/K-NI. Voltammograms recorded in 1 mM solution of  $K_3Fe(CN)_6$  and 0.2 M KCl as supporting electrolyte at varying scanning rate. Inset variation of peak currents as a function of the square root of the scan rate.



Figure S5: UV-Vis spectra of (A)  $10^{-4}$  M PNP aqueous solution: (a) before adsorption, after 10 hours contact time with (b) kaolinite 2g L<sup>-1</sup>and (c) K-NI 2g L<sup>-1</sup>; (B) 5x10<sup>-5</sup> M PNP ethanol solution, (a) before adsorption and (b) after 10 hours contact time with K-NI 1g L<sup>-1</sup>.