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Characterizations of Gelatin from the Skin of American Bullfrog (*Rana catesbeiana*) as Affected by Extraction Temperature

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Abstract: We investigated the effect of extraction temperature on the gel properties of gelatin from the skin of the American bullfrog (*Rana catesbeiana*) and the mechanisms. The textural and rheological properties of bullfrog gelatin extracted at 45 °C (G45), 55 °C (G55), and 65 °C (G65) were measured. The molecular weight distributions, microstructures, and amino acid compositions of the bullfrog gelatins were also determined. G45, G55, and G65 had gel strengths of 272.1, 225.6, and 205.8 g and hardness values of 28.1, 24.0, and 22.5 N, respectively. The gelling temperatures ranged from 19.3 to 23.9 °C, and the melting temperatures ranged from 28.9 to 31.5 °C. All the results were compared with those of commercial porcine gelatin. We propose that the higher gel strength of G45 with a higher band intensity of α 2-chains compared with G55 and G65 was more likely to form ordered and strong cross-links. The gelatin extracted at a lower temperature (G45) had a finer gel structure, suggesting that it would be more difficult to disrupt by applied force. Gelatin extracted at a lower temperature demonstrated better properties with α 2-chains and a fine gel structure. These results provide basic information on the extraction of American bullfrog skin gelatin for industrial applications.

Keywords: American bullfrog; textural properties; gel structures

1. Introduction

Gelatin is a fibrous protein hydrolyzed from a collagen that is mainly derived from the skin, connective tissues, and bones of animals [1]. Gelatin has been widely used as a stabilization, gelation, and emulsion agent in food and non-food industries [2,3]. The world usage of gelatin is about 200,000 metric tons yearly [4], which is predominantly manufactured from the skin and bones of pigs and cows. However, the outbreaks of foot-and-mouth ailments and bovine spongiform encephalopathy (BSE) have caused panic among customers [5].

Recently, chicken skin [6], chicken deboner residue [7], duck feet [8], camel skins [9], goat skin [10], cod fish skin [11,12], salmon fish skin [13], and shark byproducts [14] as additional sources have been tapped for gelatin extraction in attempts to increase safety. American bullfrogs (*Rana catesbeiana*) may be a new and safe source of gelatin with no threat of BSE. In China, the bullfrog annual production currently exceeds 100,000 tons, which is processed into products with no skin [15]. Gelatin can be extracted from different species of frogs in the genus *Rana*, such as *Rana tigerina* [2], *Rana nigromaculata* [16], and *Rana esculanta* [17].

The American bullfrog (*Rana catesbeiana*) is an important economic amphibian and one of the largest frogs in the genus *Rana* [15]. As bullfrog meat has developed in international gastronomy in various styles [18], the hatcheries of American bullfrogs have increased all around the world [19], generating a great amount of skin as by-products. Extracting gelatin from American bullfrog skin can reduce waste in the bullfrog industry and increase the



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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). revenue for the processor and farmer. However, there are no reports on the production of gelatin from American bullfrogs and details on the gel properties.

Gel properties (gel strength) are critical indices for evaluating the quality of gelatin [17]. The raw material, pretreatment, and extraction conditions (the pH, time, and particularly temperature) can influence the properties of gelatin [20,21]. Changing the extraction temperatures can cause different degrees of gelatin degradation, which leads to different molecular weight distributions [22]. The molecular weight distributions and amino acid composition were reported as important factors that affected the gel properties of gelatin [9]. However, to the best of our knowledge, the effects of extraction temperature on the characteristics of American bullfrog gelatin have not yet been evaluated.

American bullfrog skin may represent a good source for gelatin extraction due to its abundance, low cost, and safety. In this study, we aimed to characterize the gelatin extracted from American bullfrog skin at various extraction temperatures to provide basic information regarding American bullfrog skin gelatin for potential industrial applications.

2. Materials and Methods

2.1. Materials

American bullfrog skin was obtained from restaurant scrap (Beijing, China) and stored at -20 °C for no longer than 3 months until further use. High molecular weight markers (5–245 kDa) were bought from Beijing Solarbio Science & Technology Co., Ltd. (Beijing, China). Porcine skin gelatin (type A) was obtained from Sigma Aldrich Inc., (St. Louis, MO, USA) (V900863). All other chemicals used were of analytical grade.

2.2. Extraction of Gelatin from Bullfrog Skin

Bullfrog skin gelatin was extracted using the procedure described by Hafsteinsson et al. [23] with some modifications. The bullfrog skins were first cleaned to eliminate all residues of muscle then cut into small pieces (2 × 3 cm). The skin was soaked in 0.1 mol·L⁻¹ NaOH solution 1:10 (w/v) below 10 °C for 3 h, followed by washing with water until the wash water was almost neutral. The skin was soaked in 0.05 mol·L⁻¹ CH₃COOH solution 1:10 (w/v) under gentle stirring below 10 °C for 3 h. This was followed by washing with water until the wash water was almost neutral. Extractions with water 1:6 (w/v) were stirred for 6 h at 45, 55, and 65 °C The gelatin samples were filtered by cheesecloth and then centrifuged at 1589× g for 10 min. The supernatant was freeze-dried.

2.3. Determination of the Yield and Proximate Composition

2.3.1. Yield Determination

The yield of gelatin was calculated based on the following equation:

yield (%) =
$$\frac{\text{Weight of freeze dried gelatin } (g)}{\text{Weight of initial skin } (g)} \times 100$$
 (1)

2.3.2. Proximate Analysis

The moisture content was calculated by the weight lost during drying. The freezedried gelatin samples were first weighed and then placed in an oven at 105 °C After 2 h of drying, the gelatin was re-weighed, then put back in the oven at 105 °C for 1 h, and then re-weighed again. The above operation was repeated until the difference between the two masses was less than 2 mg. The protein content of the gelatin samples was determined using the Kjeldahl method. The nitrogen conversion factor used for the gelatin was 5.55.

2.4. Determination of Gel Strength

The method of Fernandez-Daz et al. [24] was used to determine the gel strength. Gelatin solutions (6.67%, w/v) were prepared at 60 °C in distilled water and kept at 10 °C for 16–18 h before analysis. The gel strength of the samples was determined by a Brookfield CT3 Texture Analyzer (Leatherhead Food Research Association Texture Analyzer Brookfield,

USA) with a 12.7 mm diameter probe. The speed of the plunger was 0.5 mm/s. The maximum force was recorded when the penetration distance reached 4 mm.

2.5. Rheological Behavior

The rheological behaviors of the gelatin solutions (6.67%, *w/v*) were analyzed using a rheometer (AR1500ex, TA Instruments, New Castle, DE, USA) using a 40 mm parallel plate. Temperature sweeps were performed from 50 to 10 °C and 10 to 50 °C with cooling/heating rates of 1.0 °C/min. The frequency and strain amplitude were set at 1 Hz and 0.1%. The cross-over point of the storage modulus (G') and loss modulus (G'') was considered as the gelling or melting temperature of each gelatin gel. The angular frequency sweep of the gelatin solutions with a range of 0.1-100 rad/s were measured with a stress value of 1 Pa. All the analyses were performed in triplicate.

2.6. Texture Profile Analysis (TPA)

The texture profiles of gelatin gels (6.67%, *w*/*v*) were analyzed using the previously published method of Huang et al. [25]. A TMS-Pro Texture Analyzer (Food Technology Corporation, Sterling, VA, USA) with a 50-mm diameter aluminum cylindrical probe (P/50) was used to measure the hardness, springiness, cohesiveness, and chewiness. Each sample was poured into a mold and kept at 10 °C for 16–18 h. The gelatin samples (d: 3 cm × h: 2.5 cm) were compressed to 50% of the original height for two cycles at a speed of 60 mm/min.

2.7. Electrophoretic Analysis

The gelatin samples were measured as described by Laemmli et al. [26]. A gelatin solution (1 mg protein/mL water) was mixed in a 1:4 (v/v) ratio with loading buffer (6% 1 M Tris-HCl, 50% glycerol, 10% [w/v] SDS, 1% [w/v] bromophenol blue, and 0.5% 2-mercaptoethanol, pH 6.8). The mixture was heated in a boiling water bath for 5 min. The stacking and resolving gel from the PAGE Gel Fast Preparation Kit (Shanghai epizyme Biotech Co., Ltd. Shanghai, China) was subjected to electrophoresis.

The electrophoresis was run at a constant voltage of 80 V/gel for the stacking gel, then run at 120 V/gel for 60 min until the resolving gel reached the bottom of the gel. This was followed by staining with Coomassie brilliant blue R-250 (0.25% w/v). A rainbow protein maker (5–245 kDa) was used to estimate the molecular weight distributions. The gels were scanned with an Imager 600 (Amersham, UK) gel-imaging system.

2.8. Microstructure Analysis of Gelatin

The microstructure of the gelatin (6.67%, w/v) was elucidated using S-3000N cryoscanning electron microscopy (cryo-SEM, Hitachi Co., Tokyo, Japan). The samples were deposited in the slots of a stub with rivets and then frozen by plunging them into slush nitrogen. After being fractured, the free water of the gels was sublimated at -85 °C for 30 min, and the gel was sputter-coated with gold (Model PP3000T, Quorum Technologies, East Grinstead, UK).

2.9. Amino Acid Composition Analysis

The amino acid composition was measured according to GB 5009.124-2016 [27].

2.10. Statistical Analysis

The experiments were performed in triplicate (except for the amino acid composition data). All data underwent analysis of variance, and significant differences (p < 0.05) between the means were determined using Tukey's test using SPSS 17.0, SPSS Inc. (Chicago, IL, USA).

3. Results and Discussion

3.1. Yield and Proximate Composition

The gelatin yields extracted at the three temperatures from bullfrog skin (45, 55, and 65 °C) are shown in Table 1. The yields of G45, G55, and G65 were 9.6%, 11.7%, and 12.3% (on a wet weight basis), respectively. Generally, an increase in extraction temperature was associated with a higher yield. The results were consistent with those of Nagarajan et al. [22] and Kittiphattanabawon et al. [28], who reported that gelatin yield increased as the extraction temperature increased. Increasing the extract temperature provides more energy for the disruption of the stabilizing collagen structures by breaking hydrogen bonds and peptide bonds [29]. As the extraction temperature increases, gelatin undergoes more helix-to-coil transitions and becomes easier to extract into the water, leading to a higher yield [30].

Table 1. The extraction yield, proximate composition, and gelling and melting temperatures (°C) of gelatin from bullfrog skin extracted at different temperatures.

Properties	PG	G45	G55	G65	Bullfrog Skin
Moisture (%)	$10.75\pm0.67\mathrm{b}$	$6.12\pm0.51~\mathrm{a}$	7.26 ± 0.85 a	$6.23\pm1.01~\mathrm{a}$	67.61 ± 0.15
Protein (%)	$87.34\pm0.09~\mathrm{a}$	87.21 ± 0.62 a	87.75 ± 0.24 a	87.83 ± 0.22 a	19.94 ± 0.77
Yield (%)	-	$9.63\pm0.28~\mathrm{a}$	$11.69\pm0.25\mathrm{b}$	$12.24\pm0.36\mathrm{b}$	-
Gelling temperature (°C)	$23.9\pm1.7~\mathrm{a}$	$23.4\pm0.4~\text{a}$	$22.8\pm1.4~\mathrm{a}$	$19.3\pm1.4~\mathrm{a}$	-
Melting temperature (°C)	$31.5\pm1.0~\mathrm{a}$	$32.3\pm0.3~\mathrm{a}$	$31.7\pm1.6~\mathrm{a}$	$28.9\pm0.6~\mathrm{a}$	-

Mean \pm SD (n = 3). PG: porcine gelatin. Within each row, different lowercase letters mean significant differences between different groups (p < 0.05).

The moisture and protein of gelatin extracted at different temperatures (45, 55, and 65 °C from the bullfrog skin were 6.1–7.3% and 87.2–87.8%, respectively. The moisture contents of all bullfrog skin gelatin samples were below the prescribed limit (15%), which was lower than commercial porcine gelatin (PG) (p > 0.05). There were no significant differences in the moisture or protein contents among G45, G55, and G65. All bullfrog skin gelatin samples showed low levels of moisture content and high levels of proteins. The bullfrog skin moisture as a major component was 67.6%. Zhang et al. [16] showed a similar result that the moisture content of bullfrog (*Rana nigromaculata*) skin was 74.0%.

3.2. Gel Strength of Gelatin

The gel strength of the gelatins from bullfrog skin at different extraction temperatures are shown in Figure 1. G45 showed the highest gel strength (**b**: 272.1 g) compared with G55 (**c**: 225.6 g) and G65 (**d**: 205.8 g) (p < 0.05), which were lower than PG (**a**: 474.3 g). Therefore, the extraction temperatures directly affected the gel strength of the bullfrog gelatin. This agrees with the studies by Liu et al. [31] and Sinthusamran et al. [21], who reported that high extraction temperatures significantly decreased the gel strength of gelatin.

Gelatins with different molecular weight distributions and amino acid compositions have different gel strength. Higher extraction temperatures might cause more hydrolysis, the more hydrolysis likely leads to shorter chains. The shorter chains cannot align properly, and the junction zone cannot form to a higher degree [20]. The amino acid composition has been reported to be one of the most important factors affecting the gel strength of gelatin [22].



Figure 1. Gel strength of the bullfrog skin gelatin extracted at different temperatures.

3.3. Rheological Properties

3.3.1. Temperature Sweep

Figure 2 shows the temperature development of G' upon cooling and heating. G' was higher than of G'' (not shown) at lower temperatures indicated that gelatin exhibited solid behavior, demonstrating gelatin molecules in a triple-helix arrangement. As shown in Figure 2, the G' of all gelatin samples increased sharply as the extraction temperature decreased, the cross-over point of G' and G'' (not shown) indicated that the formation of the gel was considered to be the gelling temperature of the gelatin gels.



Figure 2. The storage modulus (G') of gelatin (6.67%, w/v) from bullfrog skin extracted at different temperatures upon (**A**) cooling from 50 to 10 °C and subsequent (**B**) heating from 10 to 50 °C.

The gelation process for gelatin is the transition from single strands to a triple helix via hydrogen bonding, ionic interaction, hydrophobic association, van der Waals forces, and self-assembly [32]. The maximum values of the G' values of G45, G55, and G65 were 2814, 2470, and 2266 Pa, respectively, showing that higher temperature extraction reduced the crosslinking between gelatin molecules. This result agrees with the gel strength (Figure 1).

As shown in Table 1, the gelling and melting temperatures of G45, G55, and G65 were 19.3–23.4 °C and 28.9–32.3 °C, respectively, and there was no marked difference among all bullfrog skin gelatins and PG (p > 0.05). The gelling and melting temperatures in this study were higher than those of gelatins from the skins of camel (15.2–11.1 and 18.4–21.6 °C, respectively) [9], croaker fish (17.4 °C and 23.8 °C, respectively) [33] and similar with goat skin gelatin (21.2–25.2 °C and 30.7–34.1 °C, respectively) [20]. The difference in the gelling and melting properties of bullfrog gelatin compared with other gelatins can be attributed to the difference in extraction conditions, amino acid compositions, and protein chain length [29].

3.3.2. Frequency Sweep

Measuring the angular frequency of the modulus can be used to evaluate the strength of the gel network [34]. The cross-linking behavior of bullfrog skin gelatin was characterized by performing a dynamic rheological test on a 6.67% (w/v) gelatin sample at a constant temperature (10 °C) (Figure 3). The G' values were higher than the G" values (not shown) during the angular frequency range studied, confirming that gelatins were capable of forming a network and possessed a solid-like gel structure at 10 °C [25]. G45 showed numerically higher G' values compared with G55 and G65 (Figure 3). This observation agrees with the results of a study by Abedinia et al. [8]. As explained above, the stronger inter-molecular interactions in G45 compared with those in G55 and G65 may result in higher G' values [34].



Figure 3. The storage modulus (G') of gelatin (6.67%, w/v) from bullfrog skin extracted at different temperatures during angular frequency sweep at 10 °C.

3.4. Texture Profile Analysis

Texture profile analysis (TPA) is closely related to the sensory evaluation of gels [25]. The TPA results of the gelatins in this study are presented in Table 2. The hardness of G45, G55, and G65 were 28.1, 24.0, and 22.5 N, respectively. The higher extraction temperature could result in the lower hardness of gelatin gels. The hardness of all bullfrog skin gelatin was lower than porcine skin gelatin. This result is consistent with the gel strength findings (Figure 1). There were no significant differences in the springiness, cohesiveness, or chewiness among all gelatin samples (p > 0.05). A study reported that the textural properties of gelatin gels can be influenced by the amino acid compositions and molecular weight distributions [25].

Table 2. The texture profile analysis (TPA) parameters (hardness, springiness, cohesiveness, and chewiness) of gelatin (6.67%, w/v) from bullfrog skin extracted at different temperatures.

Sample	Hardness (N)	Springiness (mm)	Cohesiveness	Chewiness (mJ)
PG	$44.57\pm2.23~\mathrm{a}$	$4.29\pm0.51~\mathrm{a}$	$0.92\pm0.02~\mathrm{a}$	174.92 ± 13.67 a
G45	$28.07\pm1.61~\mathrm{b}$	$5.91\pm0.15\mathrm{b}$	$0.92\pm0.03~\mathrm{a}$	151.77 ± 2.28 a
G55	$24.03\pm0.56bc$	$6.40\pm0.47\mathrm{b}$	$0.88\pm0.04~\mathrm{a}$	135.27 ± 5.24 a
G65	$22.47\pm2.24~\mathrm{c}$	$6.38\pm0.40b$	$0.94\pm0.02~\mathrm{a}$	135.46 ± 19.69 a

Mean \pm SD (n = 3). Within each column, different lowercase letters indicate significant differences between different groups (p < 0.05).

3.5. SDS–Polyacrylamide Gel Electrophoresis (SDS-PAGE)

The protein patterns of the gelatin samples are shown in Figure 4. All the gelatin samples contained α - and β -chains with molecular weights of approximately 100 and 200 kDa, as the major components. This indicates that the α - and β -chains of the mother collagen were retained with rare degradation [29]. Among all the gelatin samples, G65 possessed the lowest α - and β -chain band intensity (as observed visually), while G45 showed a higher band intensity of α 2-chains over G55 or G65. Gomez-Guillen et al. [35] reported that gelatins with higher α -chain contents possessed better functional properties.





We noted that the bands at around 70 kDa were more intense in G65, suggesting that more drastic degradation occurred during the extraction process, which is consistent with the findings of Pang et al. [36]. It is likely that more degradation occurred in G65 as a result of the higher extraction temperature. These results suggest that the intensities of the α - and β -chains bands of bullfrog skin gelatin were influenced by the extraction temperature. The result was partly in accordance with Tan et al. [37], who found a high extraction temperature (75 °C) resulted in a decrease in the major protein components (a- and β -chains) of black tilapia (*Oreochromis mossambicus*) gelatin.

3.6. Microstructures of Gelatin Gels

The gel strength of gelatin was affected by the generally conformation and association of the proteins in the gel matrix [38]. Figure 5 shows the microstructures of the gelatin gels. The structures of all gelatins were sponge or coral-like. Among all the bullfrog gelatins, G45 showed the finest gel network with very small voids. Gelatin extracted at lower temperatures with fine gel structures is consistent with a higher gel strength (Figure 1).

As observed, the gel network of G65 was found to be coarse and heterogenous. This result was partially in agreement with the findings of Sinthusamran et al. [29] who described gelatin extracted at a lower temperature for less time as having a finer gel structure. The microstructure of the gel is known to be closely related to its physical properties [36], and a heterogenous network may be more easily disrupted by applied force [28].



Figure 5. Microstructures of gelatin (6.67%, w/v) from bullfrog skin extracted at different temperatures. Magnification: 3000 times.

3.7. Amino Acid Composition

The amino acid composition of the extracted bullfrog gelatin and PG is reported in Table 3. There were five dominant amino acids in bullfrog gelatin, including glycine (27.1–27.2%), proline (13.4–13.6%), glutamic acid (12.5–12.7%), alanine (10.6–10.7%), and arginine (8.2–8.4%). Low contents of tyrosine (0.4–0.5%), methionine (0.5–0.8%), and histidine (0.9%) were observed in all the bullfrog gelatin samples. All the bullfrog gelatins showed similar amino acid compositions. Glycine, in all the gelatin samples, represented approximately a third of the total amino acids [39]. Based on the glycine content as the major component in gelatin, the bullfrog gelatin appeared similar to porcine skin gelatin.

Table 3. The approximate amino acid composition (mg/100 mg protein) of gelatins from bullfrog skin extracted at different temperatures.

Amino Acids (g/100 g Protein)	PG	G45	G55	G65
Aspartic acid	5.68 ± 0.12	6.07 ± 0.11	6.14 ± 0.03	6.03 ± 0.01
Threonine	1.10 ± 0.02	1.55 ± 0.09	1.62 ± 0.00	1.52 ± 0.01
Serine	2.99 ± 0.05	4.45 ± 0.05	4.70 ± 0.35	4.40 ± 0.01
Glutamic acid	12.36 ± 0.22	12.71 ± 0.02	12.62 ± 0.09	12.53 ± 0.00
Glycine	27.46 ± 0.62	27.23 ± 0.11	27.21 ± 0.17	27.06 ± 0.00
Alanine	10.85 ± 0.23	10.65 ± 0.05	10.63 ± 0.10	10.59 ± 0.01
Valine	2.26 ± 0.02	1.42 ± 0.01	1.42 ± 0.02	1.40 ± 0.01
Methionine	0.69 ± 0.01	0.83 ± 0.01	0.80 ± 0.01	0.54 ± 0.03
Isoleucine	1.11 ± 0.02	0.98 ± 0.01	0.97 ± 0.03	0.99 ± 0.02
Leucine	3.32 ± 0.09	2.62 ± 0.01	2.64 ± 0.02	2.61 ± 0.01
Tyrosine	0.37 ± 0.00	0.47 ± 0.02	0.45 ± 0.00	0.44 ± 0.01
Phenylalanine	1.83 ± 0.04	1.81 ± 0.00	1.80 ± 0.00	1.79 ± 0.04
Lysine	3.90 ± 0.06	3.89 ± 0.02	3.86 ± 0.00	3.84 ± 0.00
Histidine	0.68 ± 0.02	0.90 ± 0.00	0.91 ± 0.01	0.89 ± 0.01
Arginine	8.24 ± 0.02	8.24 ± 0.12	8.40 ± 0.18	8.29 ± 0.10
Proline	14.62 ± 0.38	13.61 ± 0.21	13.43 ± 0.33	13.56 ± 0.01
Hydrophobic acid	20.06 ± 0.34	18.32 ± 0.09	18.25 ± 0.12	17.93 ± 0.02

The proline content of G45 (13.6%) was lower than in PG (14.6%) and was higher than shark extracts (9.0%) [14]. This result was consistent with G45 showing a lower gel strength compared with PG. Proline can form hydrogen bonds between hydroxyl groups in hydroxyproline and water molecules, and these bonds contribute to the strength of the gelatin gel [29]. The hydrophobic amino acid content of G45 (18.3%) was lower than that of PG (20.1%), and these amino acids can form hydrophobic associations, which also contribute to the gel strength [40]. The differences in the amino acid content might be due to the different sources and manufacturing processes of the gelatin [41].

4. Conclusions

The present study investigated the influence of the extraction temperature on the gel properties of bullfrog skin gelatin. Generally, the gel strength and hardness of gelatin ranged from low to moderate as the extraction temperature increased. These phenomena could be explained by the variations in the molecular weight distributions and microstructures of gelatin gels. Gelatin with a low extraction temperature showed a high α_2 -chain band intensity and fine gel structure, which can improve the gel properties. These results showed that American bullfrog skin has the potential to be a new source of gelatin for use in food and non-food products. A large amount of by-product from bullfrog skin could be effectively used.

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