

Supplementary materials for

Bimetallic Pt-IrO_x/g-C₃N₄ photocatalysts for the highly efficient overall water splitting under visible light

Nikolay D. Sidorenko ¹, Polina A. Topchiyan ^{1,2}, Andrey A. Saraev ¹, Evgeny Yu. Gerasimov ¹, Angelina V. Zhurenok ¹, Danila B. Vasilchenko ^{1,2} and Ekaterina A. Kozlova ^{1,*}

¹ Federal Research Center Boreskov Institute of Catalysis 630090 Novosibirsk, Russia; n.sidorenko@g.nsu.ru (N.D.S.), asaraev@catalysis.ru (A.A.S.), gerasimov@catalysis.ru (E.Y.G.), angelinazhurenok@gmail.com (A.V.Z).

² Nikolaev Institute of Inorganic Chemistry, Siberian Branch of the Russian Academy of Science, 630090 Novosibirsk, Russian Federation; topchiyan@niic.nsc.ru (P.A.T), vasilchenko@niic.nsc.ru (D.B.V).

* Correspondence: kozlova@catalysis.ru

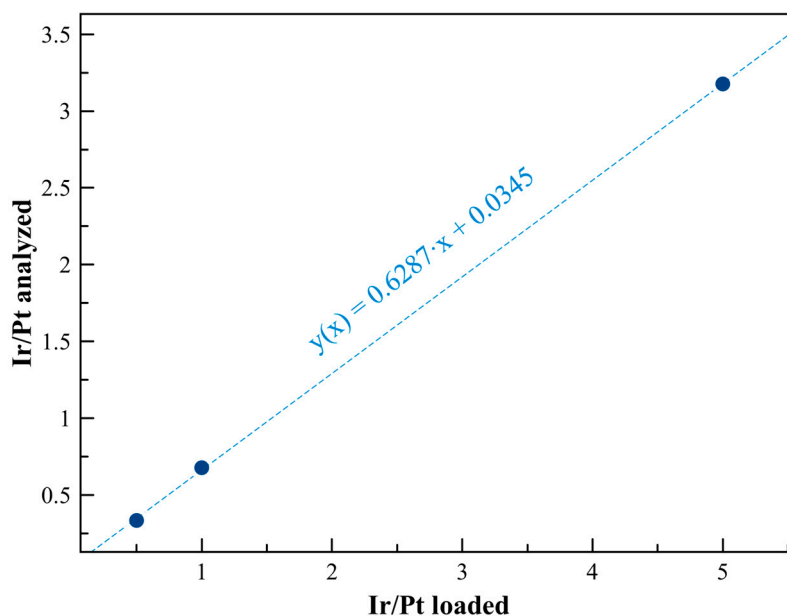


Figure S1. Correlation between loaded Ir:Pt ratios and the same value determined with ICP AES chemical analysis in the as prepared Ir_xPt_{0.1}/g-C₃N₄(1) catalysts (x = 0.5, 0.1, 0.01).

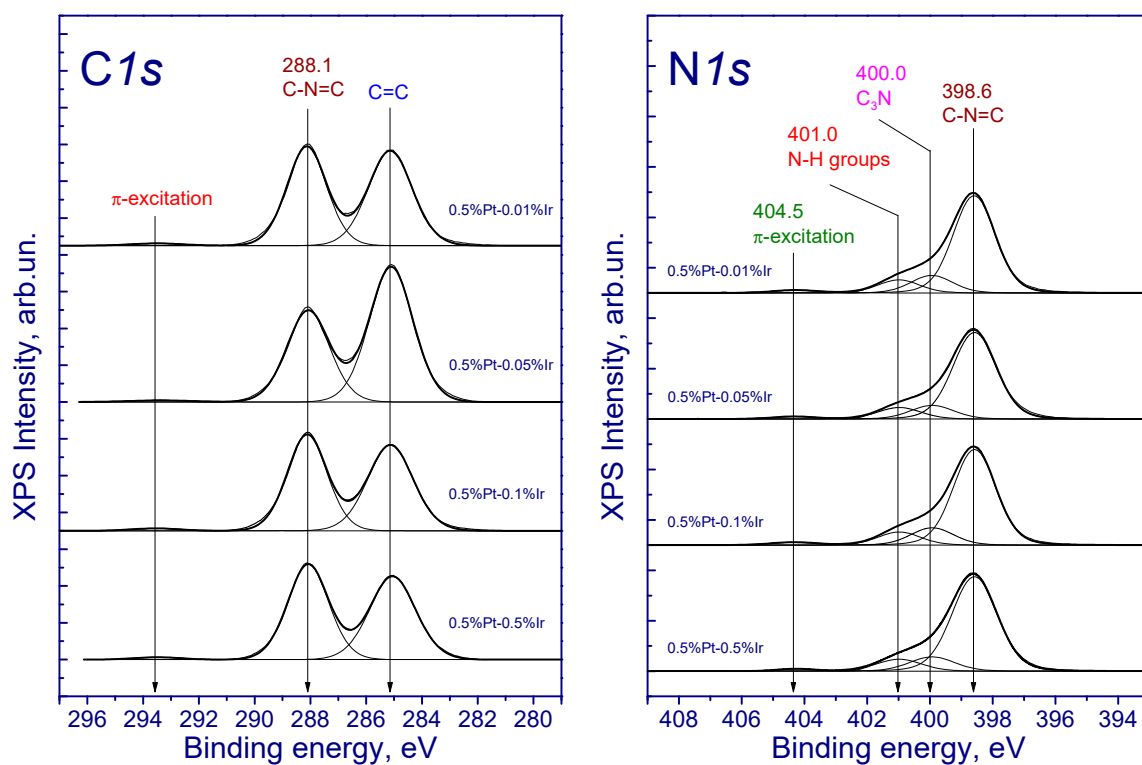


Figure S2. C1s and N1s core-level spectra of photocatalysts. The N1s spectra are normalized to the integral intensity of the C1s peaks corresponding to the spectrum of g-C₃N₄.

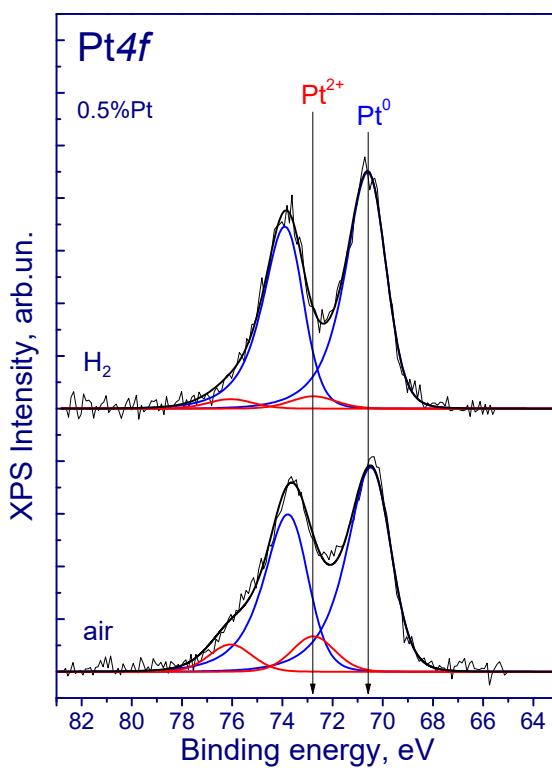


Figure S3. XPS spectra of Pt4f of the photocatalysts Pt_{0.5}/g-C₃N₄ after treatment and hydrogen and consecutive treatment in air. Spectra are normalized to the integral peak intensity C1s corresponding to the carrier spectrum (g-C₃N₄).

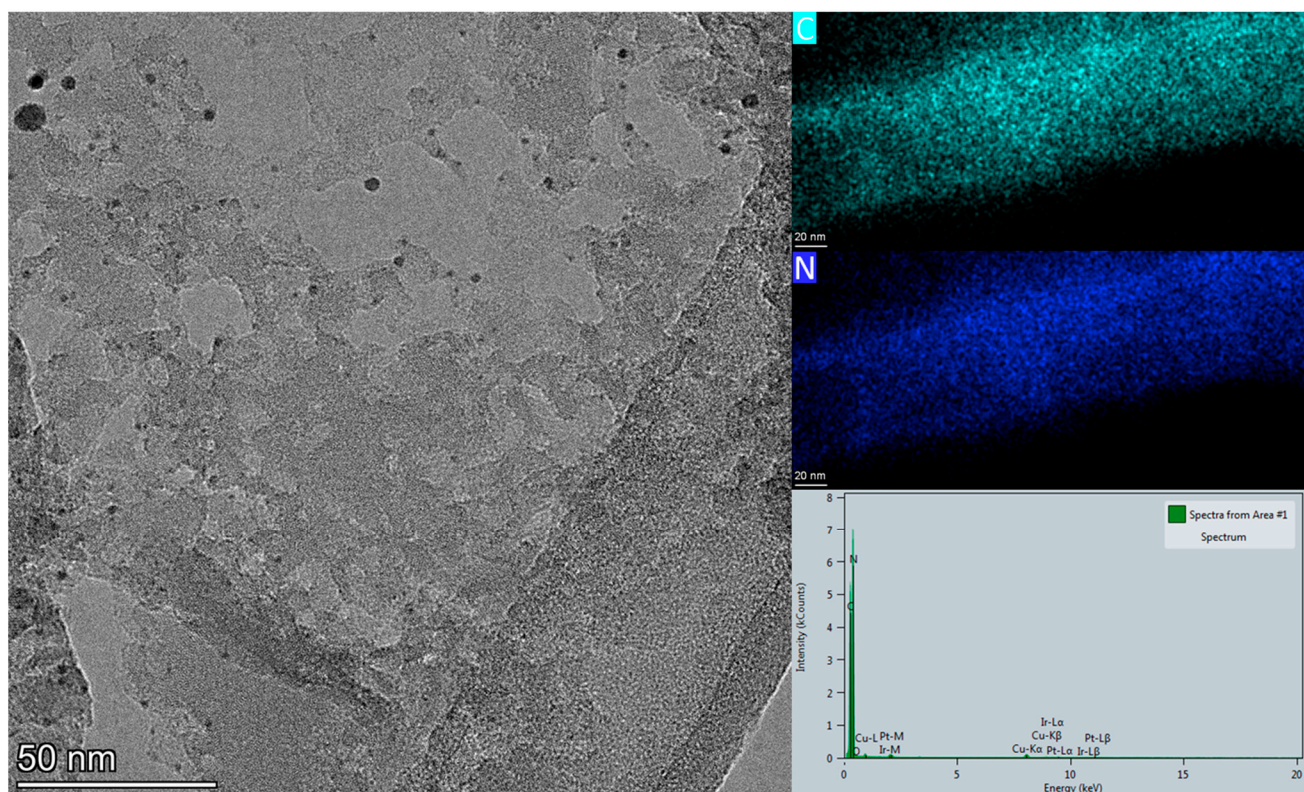


Figure S4. HR-TEM of the $\text{Ir}_{0.5}\text{Pt}_{0.5}/\text{g-C}_3\text{N}_4(1)$ photocatalyst, EDS mapping of the C and N elements in the $\text{Ir}_{0.5}\text{Pt}_{0.5}/\text{g-C}_3\text{N}_4(1)$ photocatalyst and spectrum from EDS mapping.

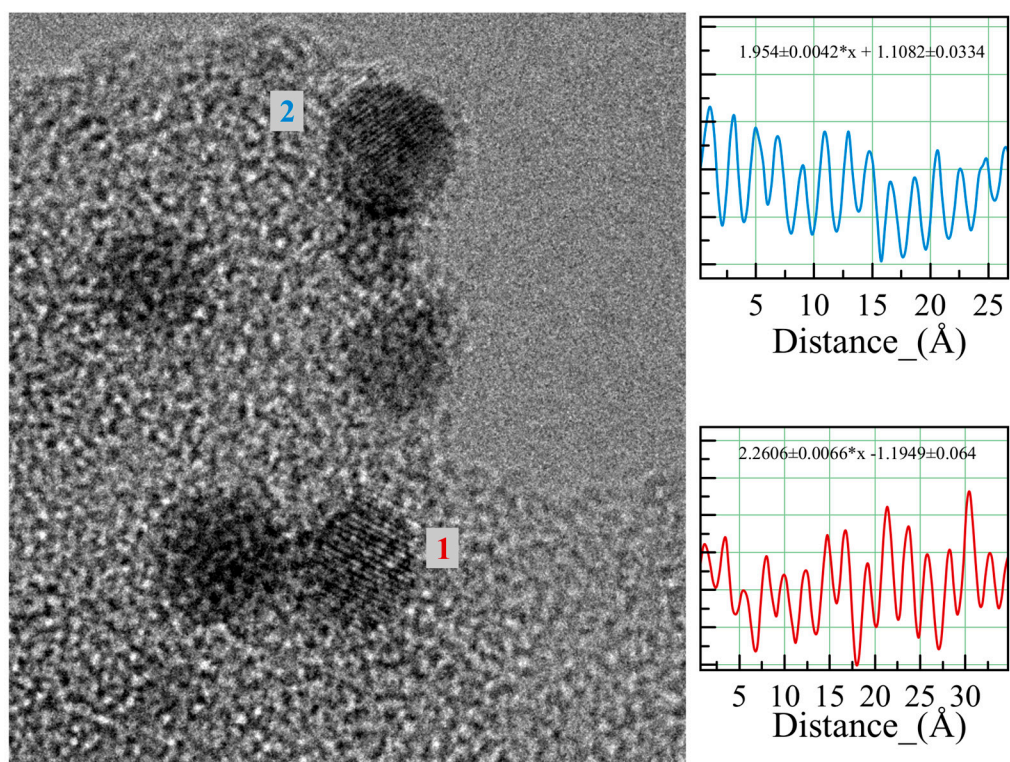


Figure S5. Analysis of the interplanar distances observed in HR-TEM micrographs of Pt nanoparticles in the $\text{Ir}_{0.5}\text{Pt}_{0.5}/\text{g-C}_3\text{N}_4(1)$ photocatalyst.

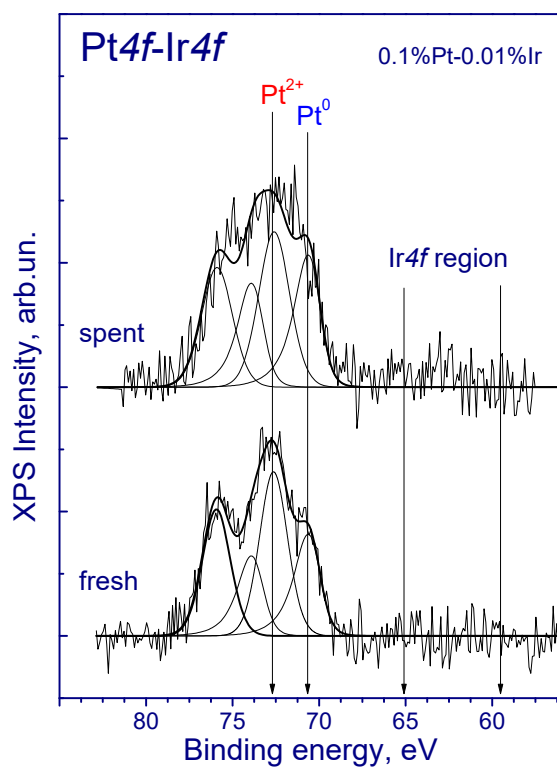


Figure S6. XPS spectra of Pt4f and Ir4f of the spent and fresh Ir_{0.01}Pt_{0.1}/g-C₃N₄(1) photocatalyst. Spectra are normalized to the integral peak intensity C1s corresponding to the carrier spectrum (g-C₃N₄).

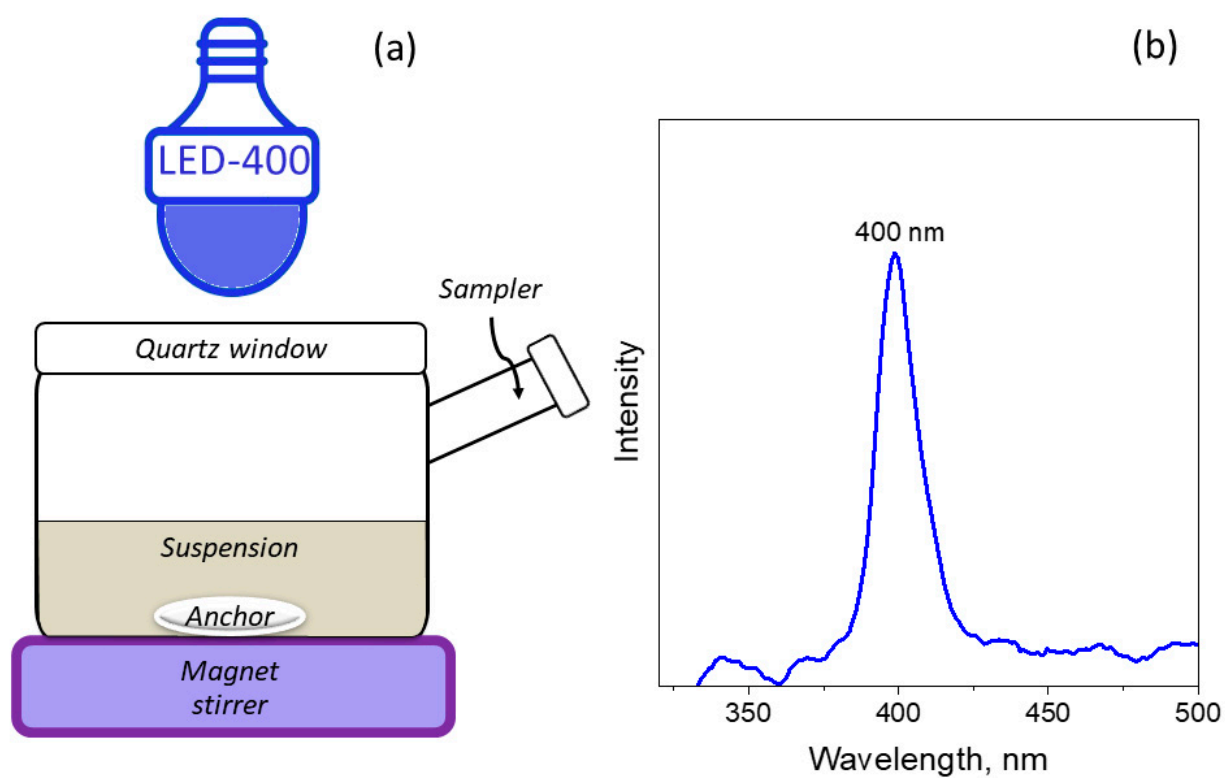


Figure S7. (a) The scheme of photocatalytic reactor utilized in this work and (b) the spectrum of 400 nm LED used for irradiation of catalysts suspensions.

Table S1. The percentage composition of all elements calculated from spectrum from EDS mapping.

Element	Mass fraction, %	Mass error, %
C	47.87	4.91
N	51.06	11.48
O	0.00	0.02
Ir	0.13	0.14
Pt	0.94	0.14

Table S2. Data on the photocatalytic hydrogen and oxygen evolution from water without the addition of electron donors.

Photocatalyst	Experimental conditions	Light source	Activity, $\mu\text{mol H}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$	Activity, $\mu\text{mol O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$	Ref.
3%Pt/0.5%CoP/g-C ₃ N ₄	deionized H ₂ O 100 ml pH = 3 photocatalyst 28 mg	300 W Xe lamp with cut-off filter $\lambda > 420 \text{ nm}$	14.7	7.4	[8]
CdS/1.3% Ni ₂ P/g-C ₃ N ₄	ultrapure H ₂ O 50 ml photocatalyst 50 mg	300 W Xe lamp with cut-off filter $\lambda > 420 \text{ nm}$	11.2	5.5	[13]
3%Pt/10%MnO ₂ /g-C ₃ N ₄	pure H ₂ O 100 ml photocatalyst 20 mg	300 W Xe lamp with cut-off filter $\lambda > 400 \text{ nm}$	60.6	28.9	[3]
3%Pt/BiVO ₄ /g-C ₃ N ₄	pure H ₂ O 100 ml photocatalyst 100 mg	300 W Xe lamp with cut-off filter $\lambda > 420 \text{ nm}$	156	73	[44]
1%Pt/g-C ₃ N ₄ /ITO/Co-BiVO ₄	pure H ₂ O 100 ml photocatalyst 30 mg	300 W Xe lamp	95.4	40.2	[45]
0.5%Pt/0.13%Co/g-C ₃ N ₄	pure H ₂ O 50 ml photocatalyst 20 mg	300 W Xe lamp with cut-off filter $\lambda > 400 \text{ nm}$	85.7	21.3	[46]
7%Co _{0.6} Ni _{1.4} P/CdS/g-C ₃ N ₄	pure H ₂ O 100 ml photocatalyst 30 mg	300 W Xe lamp with cut-off filter $\lambda \geq 420 \text{ nm}$	122.0	57.5	[47]
1%Pt/3%IrO ₂ /g-C ₃ N ₄	distilled H ₂ O 100 ml photocatalyst 50 mg	300 W Xe lamp with cut-off filter $\lambda \geq 420 \text{ nm}$	101	49.1	[6]

Table S3. Data on the photocatalytic hydrogen and peroxide evolution from pure water presented in the literature.

Photocatalyst	Experimental conditions	Light source	Activity, $\mu\text{mol H}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$	Activity, $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$	Ref.
64 wt.% CdS@ZnIn ₂ S ₄	100 ml pure water 15 mg photocatalyst	Xe lamp 300 W with filter $\lambda > 400$ nm	540.3	604.8	[54]
8.3 wt.% carbon dots/CoP	20 ml ultrapure water 10 mg photocatalyst	White LED (420 nm $\leq \lambda < 700$ nm)	239.0	466.0	[52]
0.5 wt.% CoO/TiO ₂ /SiO ₂	25 ml ultrapure water 40 mg photocatalyst	White LED (420 nm $\leq \lambda < 700$ nm, 5 W)	1460.0 $\mu\text{mol H}_2 \text{ g}_{\text{CoO}}^{-1} \text{ h}^{-1}$ (7.30 $\mu\text{mol H}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	1390.0 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{CoO}}^{-1} \text{ h}^{-1}$ (6.95 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	[18]
1 wt.% Pt/1.45 wt.% Na ⁺ -doped-g-C ₃ N ₄	200 ml deionized water 200 mg photocatalyst	Sodium lamp high-pressure 250 W (400 $< \lambda < 800$ nm)	900.0	880.0 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1}$ (146.7 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	[24]
1 wt.% Pt/2.3 wt.% K ⁺ -doped-g-C ₃ N ₄	200 ml deionized water 200 mg photocatalyst	Sodium lamp high-pressure 250 W (400 $< \lambda < 800$ nm)	550.0	620.0 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1}$ (103.3 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	[20]
Co _x P/P-doped-g-C ₃ N ₄ (x=1 or 2)	80 ml pure water 50 mg photocatalyst	300 W Xe lamp (incident light density = 450 mW/cm ²) with cut off filter $\lambda \geq 420$ nm	75.0 $\mu\text{mol H}_2$ (300.0 $\mu\text{mol H}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	70.0 $\mu\text{mol H}_2\text{O}_2$ (280.0 $\mu\text{mol H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{ h}^{-1}$)	[48]
Au nanoparticles/Oxygen vacancies BiOBr/TiO ₂ (P25)	100 ml deionized water 50 mg photocatalyst	Xe lamp 300 W	384.0	200.0	[55]
Ni ₂ P/carbon dots-x (x=20 mg)	20 ml ultrapure water 10 mg photocatalyst	$\lambda > 420$ nm	256.6	1281.4	[49]
3% Co-doped-mesoporous g-C ₃ N ₄	25 ml ultrapure water 10 mg photocatalyst	White LED	1.82 $\mu\text{mol H}_2 \text{ h}^{-1}$	1.65 $\mu\text{mol H}_2\text{O}_2 \text{ h}^{-1}$	[53]

		(420 nm \leq $\lambda < 700$ nm) Xe lamp 300 W with cut off filter λ > 325 nm	(182.0 μmol $\text{H}_2 \text{ g}_{\text{cat}}^{-1} \text{h}^{-1}$)	(165.0 μmol $\text{H}_2\text{O}_2 \text{ g}_{\text{cat}}^{-1} \text{h}^{-1}$)	
5% black P nanodots/ $\text{Ti}_3\text{C}_2\text{T}_x@ \text{TiO}_2$	100 ml pure water 10 mg photocatalyst		564.8	400.0	[50]
1 wt.% Ni/3 wt.% CoP/ P-doped-g- C_3N_4	60 ml water 50 mg photocatalyst	Xe lamp	124.0	447.0	[51]