## Supplementary Materials: Immobilized Palladium Nanoparticles on Zirconium Carboxy-Aminophosphonates Nanosheets as an Efficient Recoverable Heterogeneous Catalyst for Suzuki- Miyaura and Heck Coupling

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Figure S1. IR spectra of ZPGly and Pd@ZPGly-1.

The IR spectrum of ZPGly shows the bands ascribable to the stretching and bending of the O-H bond at 3424 cm<sup>-1</sup> and 1640 cm<sup>-1</sup>, due to the hydration water molecules. The bands at 3002 and 2995 cm<sup>-1</sup> are due to the C-H stretching. The broad band centered at 2600 cm<sup>-1</sup> is characteristic of a R<sub>3</sub>-NH<sup>+</sup> group, proving the presence of protonated amino groups. The band at 1739 cm<sup>-1</sup> is associated with the C=O stretching of the -COOH group, accompanied by the bands at 1427 (O-H bending) and 1244 cm<sup>-1</sup> (C-O stretching). The bands from 950–1200 cm<sup>-1</sup> are related to the P-O stretchings, whereas in the region below 900 cm<sup>-1</sup>, a number of bands are found, which are not straightforwardly assigned (various bending modes, Zr-O stretchings).

In the sample Pd@ZPGly-15, typical bands of the COOH group disappear, while the typical bands of the COO<sup>-</sup> group appear at 1567 cm<sup>-1</sup> and at 1370 cm<sup>-1</sup>, ascribable to asymmetric and symmetric stretching of COO<sup>-</sup>. These findings suggest that the proton was transferred to the propylamine and that both palladium ions and propylammonium ions take part in neutralizing the negative charge of carboxylate groups.



**Figure S2.** TEM images of Pd@ZPGly-1 (**a**), Pd@ZPGly-1R (**b**), Pd@ZPGly-7 (**c**), Pd@ZPGly-7R (**d**) and Pd@ZPGly-15 at a low magnification.



**Figure S3.** STEM (above) and HRTEM (below) images of Pd@ZPGly-15. The size distribution of the smaller population of Pd nanoparticles on Pd@ZPGly-15 is also reported.

## Procedure for hot filtration test in the Suzuki reaction in aq. EtOH azeotrope with Pd@ZPGly-15 catalyst:

In a vial, Pd@ZPGly-15 catalyst (0.47 mg, 2.11 mmol of Pd per g, 0.1 mol%), phenylboronic acid (**2a**, 134 mg, 1.1 mmol), K<sub>2</sub>CO<sub>3</sub> (167 mg, 1.10 mmol), 4-bromotoluene (**1a**, 171 mg, 1 mmol) and ethanol 96% (2.4 mL) were placed. The reaction mixture was stirred at 80 °C for 20 min (conv. 71%), and the hot filtration was performed. K<sub>2</sub>CO<sub>3</sub> (167 mg) was added again and the reaction mixture stirred at 80 °C.

Reaction time 1 h; Conv. 74%

Reaction time 24 h; Conv. 74%

## Procedure for the homogeneous Suzuki reaction in aq. EtOH azeotrope performed with Pd(OAc)<sub>2</sub>:

The amount of Pd(OAc)<sup>2</sup> is comparable to that which leached during the reaction with Pd@ZPGly (11 ppm, Table 3, Entry 3)

In a vial, phenylboronic acid (2a, 134 mg, 1.1 mmol), K2CO3 (167 mg, 1.10 mmol), 4-

bromotoluene (1a, 171 mg, 1 mmol), ethanol 96% (2.4 mL) and Pd(OAc)<sub>2</sub> ( $1.7 \cdot 10^{-5}$  mmol, 3.9  $\mu$ L of a solution of 1 mg/1 mL in acetone) were placed. The reaction mixture was stirred at 80 °C for 24 h. Reaction time 1 h; Conv. 33%

Reaction time 24 h; Conv. 41%