## **Supporting Information for**

## Copolymerization of Phenylselenide Substituted Maleimide with Styrene and Its Oxidative Elimination Behavior

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Figure S1. <sup>1</sup>H NMR spectrum of MSM in CDCl<sub>3</sub>



Figure S2. <sup>13</sup>C NMR spectrum of MSM in CDCl<sub>3</sub>



Figure S3. <sup>77</sup>Se NMR spectrum of MSM in CDCl<sub>3</sub>



**Figure S4**. Direct <sup>1</sup>H NMR analysis spectra of monomer conversion in CDCl<sub>3</sub>. The integral of the signal of proton belong to toluene in the <sup>1</sup>H NMR was set to be 1.0, served as internal standard for the decrease of the integral of double bonds belong to each monomer. 0 h and 9 h was toward initial time and the ninth hour of the whole copolymerization. Conv.st=  $(I_{5.26,0h} - I_{5.26,0h}) / I_{5.26,0h} \times 100\%$ ; Conv.MSM=  $(I_{5.86,0h} - I_{5.86,0h}) / I_{5.86,0h} \times 100\%$ . The conversion of monomer at other times is similar to the calculation illustrated above.



Figure S5. Relationships of  $M_n$  and molecular weight distribution (*D*) with conversion of reversible addition-fragmentation chain transfer (RAFT) copolymerization of St and MSM with different molar ratios



**Figure S6.** GPC curves of reversible addition-fragmentation chain transfer (RAFT) copolymerization of St and MSM with different molar ratios, (a)  $[St]_0/[MSM]_0 = 200/6$ ; (b)  $[St]_0/[MSM]_0 = 200/20$ ; (c)  $[St]_0/[MSM]_0 = 200/60$ ; (d)  $[St]_0/[MSM]_0 = 200/200$ ,  $V_{\text{St}} = 1.0 \text{ mL}$ ,  $V_{\text{toluene}} = 50 \text{ }\mu\text{L}$ , temperature 70 °C.



Figure S7. FT-IR spectrum of the RAFT copolymer



**Figure S8.** UV-*vis* spectra of copolymers in THF: (a) before oxidization (b) after oxidization



Figure S9. Excitation spectra of copolymers in THF at 10 mg/mL: (a) before oxidization (b) after oxidization