Supporting Information

Competition in Coordination Assemblies: 1D-Coordination Polymer or 2D-Nets Based on Co(NCS)2 and 4'-(4-methoxyphenyl)-3,2':6',3"terpyridine

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Figure S1. Electrospray mass spectrum of 2.



Figure S2. 126 MHz ${}^{13}C{}^{1}H$ NMR spectrum of **2** (CDCl₃, 298 K). * = CDCl₃. See Scheme 2 for atom labelling.



Figure S3. NOESY spectrum of **2** (500 MHz ¹H NMR, CDCl₃, 298 K).



Figure S4. HMQC spectrum of **2** (500 MHz ¹H, 126 MHz ¹³C{¹H}, CDCl₃, 298 K).



Figure S5. HMBC spectrum of **2** (500 MHz ¹H, 126 MHz ¹³C{¹H}, CDCl₃, 298 K).



Figure S6. Powder XRD pattern (295 K) of the bulk material from experiment I compared to that predicted from the single crystal structure (130 K) of $[Co(2)(NCS)_2(MeOH)_2]_n$.



Figure S7. Powder XRD pattern (295 K) of the bulk material from experiment III compared to that predicted from the single crystal structure (130 K) of $[Co(2)(NCS)_2(MeOH)_2]_n$.



Figure S8. Top: Overlay of the PXRD patterns from the powders obtained from the 1 : 1 and 1 : 2 bulk reactions of Co(NCS)₂ and **2**. Middle and lower figures: Predicted powder patterns from the single crystal structures of $[Co(2)(NCS)_2(MeOH)_2]_n$ and $[\{Co(2)_2(NCS)_2\}\cdot 3MeOH]_n$.



Figure S9. Powder XRD pattern (295 K) of the bulk material from experiment II compared to those predicted from the single crystal structures (130 K) of $[Co(2)(NCS)_2(MeOH)_2]_n$, $[{Co(2)_2(NCS)_2}\cdot 3MeOH]_n$ and $[{Co(2)_2(NCS)_2}\cdot 2.2CHCl_3]_n$.