



## **Supplementary Materials**

Experimental section

HOOC- $\bigvee$ -NO<sub>2</sub>  $\xrightarrow{\text{Glucose/NaOH}}$  HOOC- $\bigvee$ -N $\searrow$ -COOH Azoa Figure S1. Synthesis route of the 4,4-azodibenzoic acid.

4,4-azodibenzoic acid (Azoa) was easy to synthetic According to our previous work. The chemical structures and synthetic procedures for the Azoa are shown in scheme S1. The characterization data of the monomer were confirmed by <sup>1</sup>H-NMR and Mass Spectrometry (MS). The characterization data of the monomer are as follows: Azoa (Fig. S1), <sup>1</sup>H NMR ( $\delta$ , ppm, DMSO-d $\delta$ ): 12.23 (a, 1H, -COOH), 8.47-7.71 (b, 4H, Ar-H), 6.71 (c, 4H, Ar-H). ppm = 3.33 and 2.67, which belong to the solvent of DMSO-d $\delta$ . Mass spectrometry (MS) (m/z) [M] calcd for C14H10N2O4, 270.24; found, 270 + 1.

Supplementary tables and figures



Figure S2. <sup>1</sup>H-NMR spectra of Azoba in DMSO-d<sub>6</sub>.



Figure S3. SEM images of (a) P1, (b) P2, (c) P3, and (d) P4, respectively.



Figure S4. Typical AFM images of (a) P1, and (b) P4, respectively.

Sample	Td1 (°C) <sup>a</sup>	Td2 (°C)a	Tm(°C)b	Tg(°C) <sup>b</sup>			
P1	204.3	382.8	25.1	104.3			
P2	212.6	397.6	30.3	103.5			
Р3	221.0	410.9	34.2	103.1			
P4	234.2	410.4	33.6	102.9			

 Table S1
 Thermal properties of the A-SMPUs

a.  $T_{d1}$  is the peak temperature on the first stage;  $T_{d2}$  is the peak decomposition temperature on the second stages, measured by DTG.

b. Evaluated by DSC during the second heating process at a rate of 10 °C min<sup>-1</sup> under nitrogen atmosphere.

Table S2 Shape recovery	y ratio and shape f	ixity ratio of A-SMP	'Us in the triple sh	ape memory cycle
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	First shape	Second shape	First strain	Second strain	Total strain
Samples	fixity ratio	fixity ratio	recovery ratio	recovery ratio	recovery ratio
	ratio (%)	ratio (%)	(%)	(%)	(%)
P1	65.9	96.1	98.0	80.6	95.2
P2	49.4	87.7	94.8	70.0	94.4
P3	38.7	86.3	95.9	96.3	96.1
P4	57.9	98.3	93.3	67.2	92.2