

## Article

# Seasonal Variations and Sources of Airborne Polycyclic Aromatic Hydrocarbons (PAHs) in Chengdu, China

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**Abstract:** The concentrations of polycyclic aromatic hydrocarbons (PAHs) in the air of Chengdu, a southwest city of China, were determined from March 2015 to February 2016. Here, two diagnostic ratios (DR) were determined and a principal component analysis/multiple linear regression (PCA/MLR) analysis was performed to identify the sources of PAHs during the four seasons. The gaseous and particle phase samples were analyzed separately. The sampled air had a gas-to-particle ratio of 4.21, and between 18.7% and 31.3% of the total detected PAHs were found in the particulate phase. The total concentration of all 16-PAHs combined (gas + particles) varied from 176.94 in summer to 458.95 ng·m<sup>−3</sup> in winter, with a mean of 300.35 ± 176.6 ng·m<sup>−3</sup>. In the gas phase, phenanthrene(Phe) was found at the highest concentrations in all four seasons, while benzo[b]fluoranthene(BbF) and (in winter) chrysene(Chr) were the highest in the particle phase. The DR of Fluroanthene (Flua)/(Flua + Pyrene (Pyr)) was higher in the gas phase than in the particle phase, while the Indeno[1,2,3-cd]pyrene(IcdP)/(IcdP + Benzo[ghi]perylene (BghiP)) ratio was more variable in the gas than that in the particle phase. The main sources for both phases were a mixture of liquid fossil fuel combustion and the burning of biomass and coal, with clear seasonal variation. Principal Component Analysis/Multiple Linear Regression (PCA/MLR) analysis identified the main PAH sources as coal burning (52%) with motor vehicle exhaust and coke (48%) in spring; coal (52%), coke (21%), and motor vehicle exhaust (27%) in summer; coal (47%), vehicle exhaust (34%), and coke (19%) in autumn; and coal (58%) and vehicle exhaust (42%) in winter.

**Keywords:** PAHs; source apportionment; diagnostic ratios; PCA/MLR

## 1. Introduction

Polycyclic aromatic hydrocarbons (PAHs) are toxic pollutants that can be distributed widely by air pollution. They are semi-volatile and persistent with different distribution characteristics according to their physicochemical properties. In polluted air, PAHs can be present both in the gas phase and bound to particles. PAHs with lower molecular weight are more volatile and, together with those that are water-soluble, are typically found in the gas phase, while PAHs with higher molecular weight and higher octanol-water partition coefficients are typically more lipophilic and are easier absorbed by particles [1]. The origin of air-borne PAHs can be natural causes (e.g., wild fires, volcanic activity), but human activity is mostly responsible [2,3]. These activities include the burning of wood and fossil fuels for industry, transport, heating, and electricity production. For control of air quality and to reduce human exposure to these toxic chemicals, the sources of PAHs need to be investigated.

For such investigations, determination of diagnostic ratios and analysis by multivariate statistics is very useful [1,4]. Multivariate statistical analysis can be performed by principal

component analysis (PCA) or principal component combined with multiple regression (PCA/MLR). These receptor-oriented models are independent on the composition and degradation of PAHs and are widely used [5]. In China, where air pollution can be severe, a limited number of studies have been performed to identify PAHs sources by the use of receptor-oriented models. However, a comprehensive survey for emission sources of PAHs that differentiates between air-borne gas and particle phases is lacking for southwest China, where local topographic, climate conditions, and human use of energetic sources differ from those in better-studied northern areas of China.

The city of Chengdu, the capital city of Sichuan province, is typical for cities in the southwest of China. It is located in the Chengdu plain at 540 m elevation and is surrounded by mountains. This topography limits the migration and diffusion of pollutants from the city to surrounding areas, causing severe local air pollution. In the current study, the atmospheric concentrations of PAHs in the gas and particle phase were measured on a weekly basis during a complete year (March 2015–February 2016). Diagnostic ratios were analyzed by PCA/MLR to investigate individual PAHs in both the gas and particle phases in air in the city.

## 2. Materials and Methods

### 2.1. Sample Collection

The sampling site (E104.148937, N30.682027) was located at the university campus, approximately 15 m above ground level in the east of Chengdu. Samples were taken once a week, resulting in a total of 98 samples (52 gas phase samples and 52 particle phase samples).

Air samples were collected using a high-volume air sampler with an air flow rate of  $0.8 \text{ m}^3 \cdot \text{min}^{-1}$ , an acquisition time of 24 h, and a sampling volume of approximately  $1150 \text{ m}^3$ . During sampling, the air was drawn through a glass fiber filter (GFF) ( $20 \text{ cm} \times 25 \text{ cm}$ ) to collect the particles, after which it passed a cartridge containing two layers of polyurethane foam plugs (PUF) with a total length of 10 cm (each PUF had a length of 5.0 cm, and a diameter of 9.5 cm) to collect gas-phase compounds. After sampling, the filters were wrapped and sealed with pre-baked aluminum foil, while the plugs were placed in acetone-rinsed glass flasks, to be stored at  $-35^\circ \text{C}$  until required for analysis.

### 2.2. Experimental Procedure

The filter and plug samples were Soxhlet-extracted with dichloromethane and a mixture of acetone: hexane (1:1 *v/v*) for 24 h, respectively. Extract volumes were reduced to 5 mL using a rotary evaporator. For the filters, 15 mL hexane was first added to the solvent before the volume was reduced. The extracts were then passed through a cleaning column containing anhydrous sodium sulfate and silica gel (Merck, Darmstadt, Germany) and eluted with a mixture of dichloromethane and hexane (1:1 *v/v*). The solution was rotary-evaporated again, solvent-exchanged into isooctane, and finally reduced to a volume of 1 mL under a gentle nitrogen gas flow. All sample solutions were kept in gas chromatography (GC) vials for analysis.

A series of 16 PAHs (Supelco Co., Bellefonte, PA, USA) have been identified by the US Environmental Protection Agency (EPA) as priority pollutants [1,6,7]. These were analyzed by gas chromatography/mass spectrometry (GC/MS) using an Agilent 6890 N GC coupled with an Agilent 5973 mass spectrometer detector. A  $60 \text{ m} \times 0.25 \text{ mm} \times 0.25 \mu\text{m}$  HP-5MS capillary column (Agilent Co., Santa Clara, CA, USA) in selected ion monitoring (SIM) mode was used. A  $2.0\text{-}\mu\text{L}$  volume of sample was injected in the splitless mode. The column temperature was programmed as follows: the temperature was kept constant at  $90^\circ \text{C}$  for 1 min, then raised to  $180^\circ \text{C}$  with  $10^\circ \text{C} \cdot \text{min}^{-1}$  increments, kept at  $180^\circ \text{C}$  for 1 min, then increased to from 180 to  $280^\circ \text{C}$  with  $3^\circ \text{C} \cdot \text{min}^{-1}$ , and kept at  $280^\circ \text{C}$  for 20 min. Five mixed standard solutions containing the 16 PAHs at different standard concentrations were used to spike the PUF and GFF, and recovery was tested (see below). These data were used for the calculation of PAH concentrations in the true PUF and GFF samples. The total PAHs ( $\text{R}_{16}\text{PAHs}$ ) was defined as the sum of the 16 individual compounds.

### 2.3. Quality Control

According to the quality assurance and control method recommended by the EPA in the process of the experiment, the blank experiment and the standard recovery test were set up during the analysis of the sample. The filed blanks and the laboratory blanks were able to be detected by naphthalene (Nap), and the remaining 15 were all below the method detection limit. Therefore, blank correction was carried out in the analysis of the results.

Most PAHs were not detectable in the filed blanks and laboratory blanks, except naphthalene, which was detected at very low concentrations in some blanks. Therefore, blank correction was carried out in the analysis of the results. The average recoveries of the four surrogates were 74–104% for GFF and 63–112% for PUF. As defined by the EPA, a pre-cleaned PUF and GFF spiked with PAHs were used to determine the method detection limits (MDLs), which were calculated as the Student's *t*-value by multiplying the standard deviation (SD) of five replicate samples. With a 1151-m<sup>3</sup> air sample, the MDLs ranged from 5 to 12 pg·m<sup>-3</sup>. On a monthly basis, the second, downstream PUF plug (5.0 cm long) was divided into a 3 cm upstream and a 2 cm downstream fragment, and these fragments were analyzed separately to check any possible leakage of gaseous PAHs. Overall, the average proportion of PAHs collected by the 2.0 cm downstream fragment of the second plug was  $9 \pm 4\%$  of the total gaseous PAHs collected, meaning that the two plugs combined (10 cm in total) have sufficient capacity to adsorb the gaseous PAHs during the 24 h sampling period.

### 2.4. Diagnostic Ratios

Different fuels and different combustion or burn conditions will release different types of PAHs and in different relative ratios. Thus, the ratio of the different marker compounds depends on the emission source. Based on previous research, we calculated the following ratios: Flua/(Flua + Pyr) and IcdP/(IcdP + BghiP). These ratios were used to identify the possible sources of the detected PAHs.

The PAHs Flua, Pyr, Icdp, and BghiP have similar physicochemical properties so these were chosen to give a characteristic ratio, as they are relatively stable in the atmospheric environment [8,9]. When isotopic analysis methods were compared with the diagnostic ratios of Flua/(Flua + Pyr) and IcdP/(IcdP + BghiP), they pointed to similar sources [10,11]. Therefore, in this study we chose the ratios of Flua/(Flua + Pyr) and IcdP/(IcdP + BghiP) to analyze the sources of PAHs in the samples.

### 2.5. Principal Component Analysis/Multiple Linear Regression (PCA/MLR)

Multivariate statistical analysis was performed using the SPSS statistical software package (SPSS 13.0) with varimax rotation, and principal components (factors) with eigenvalues >0.8 were used to extract the possible sources.

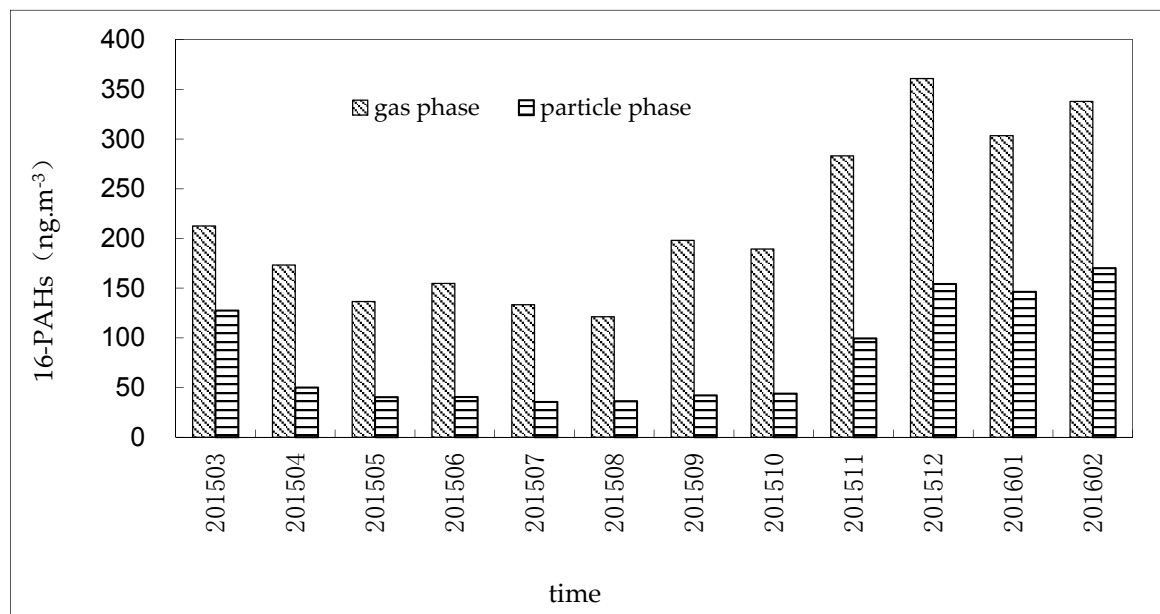
## 3. Results and Discussion

### 3.1. Concentration of Total PAHs

Figure 1 shows the concentration variation characteristic of PAHs in the atmospheric gas phase and particle phase within a year of the sampling period. During the sampling period, atmospheric PAHs concentration in the gas phase and the particle phase of the study area shows a similar variation, with higher PAHs concentration in winter and spring and lower PAHs concentration in summer and autumn. Peak values appeared in December, January, and February.

A summary of the total PAH concentrations (the sum of 16 individual PAHs) determined in the gas and particulate phases of air collected per seasonal average are presented in Table 1. During all seasons, the total PAH content was higher in the gas than in the particulate phase, with between 18.7% and 31.3% of the total detected PAHs (gaseous plus particulate) in the latter. The average total (gaseous plus particulate) PAH concentrations varied from 176.94 in summer to 458.95 ng·m<sup>-3</sup> in winter, with a yearly mean of  $300.35 \pm 176.6$  ng·m<sup>-3</sup>. The concentration range of the total detected PAHs in the gas phase varied between 70.48 and 645.27 ng·m<sup>-3</sup> for individual samples, with a yearly average of

$218.35 \pm 117.2 \text{ ng}\cdot\text{m}^{-3}$ ; for the particulate phase the yearly average was  $82.00 \pm 64.8 \text{ ng}\cdot\text{m}^{-3}$ . Thus, the majority of PAHs in the air are found in the gas phase, and the content of particle-bound PAHs is relatively low; the highest concentrations are found in winter and spring, while lower concentrations occur during summer and autumn.



**Figure 1.** Atmospheric PAHs gas-particle concentration distribution in Chengdu.

**Table 1.** Mean concentration ( $\text{ng}\cdot\text{m}^{-3}$ ) and standard deviation for total, gaseous, and particulate PAHs during four different seasons of one year in air samples from Chengdu.

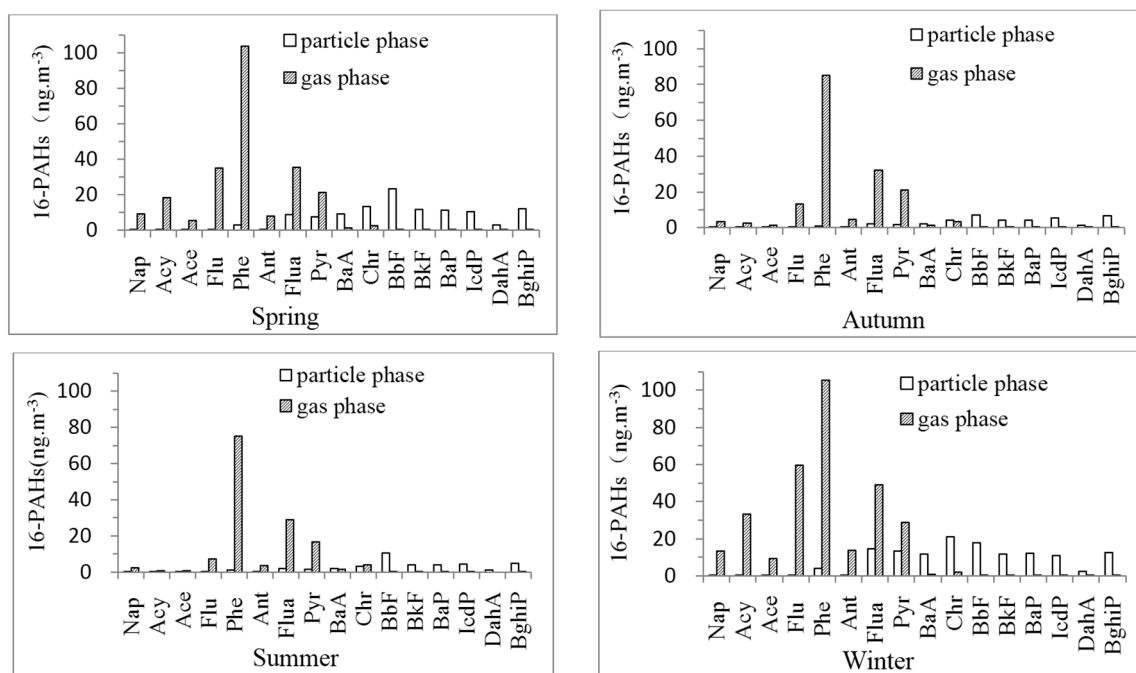
Season	Gas	Particle	Total PAHs
Spring	$228.44 \pm 90.3$	$106.94 \pm 69.0$	$335.38 \pm 154.5$
Summer	$136.21 \pm 40.5$	$38.53 \pm 12.7$	$176.94 \pm 48.2$
Autumn	$172.26 \pm 52.8$	$41.58 \pm 17.5$	$213.84 \pm 61.9$
Winter	$322.76 \pm 146.6$	$136.19 \pm 65.2$	$458.95 \pm 207.6$
Annual average	$217.09 \pm 115.9$	$82.00 \pm 64.8$	$300.35 \pm 176.6$

The PAHs concentration ratio of gas to particles detected in Chengdu was 4.21, which is higher than those reported from a few other regions in China and abroad. For instance, a gas to particle ratio of 1.2 was reported for Harbin in China [12], and a ratio of 2.3 was detected in Seoul, Korea [13]. However, our ratio is lower than that obtained from Turkey (9.36) [14], Birmingham (5–6) [7], Guangzhou (13.2) [15] or Nanjing (8.9) in China, or Athens, Greece (5.2).

### 3.2. Concentration of Individual PAHs

The averages of individual PAH levels detected in the gas and particle phase during the four seasons are shown in Figure 2. The highest concentrations detected in the gas phase were for Phe, Fluorene (Flu), Flua, and Pyr. In particular, Phe could make up to 46.3% of total PAHs in the gas phase. Figure 2 further shows that the concentration of PAHs in the gas phase and particle phase followed similar seasonal trends with concentrations during winter > spring > autumn > summer. PAHs with a low molecular weight, typically containing two or three organic ring structures only, dominated in the gas phase, while PAHs with higher molecular weight and more organic rings (five or six) were present in the particle phase. In the latter, BbF could make up to 16.1% of the total. Four organic ring PAHs are more likely to be enriched in the particulate phase at low temperatures, while PAHs with five or six

organic rings are more likely to be adsorbed in particulates at high temperatures. Studies have shown that PAHs with four to six ring components mainly originate from the emissions of high temperature combustion. It is gradually distributed from the atmospheric particulates to the gas phase [16,17].

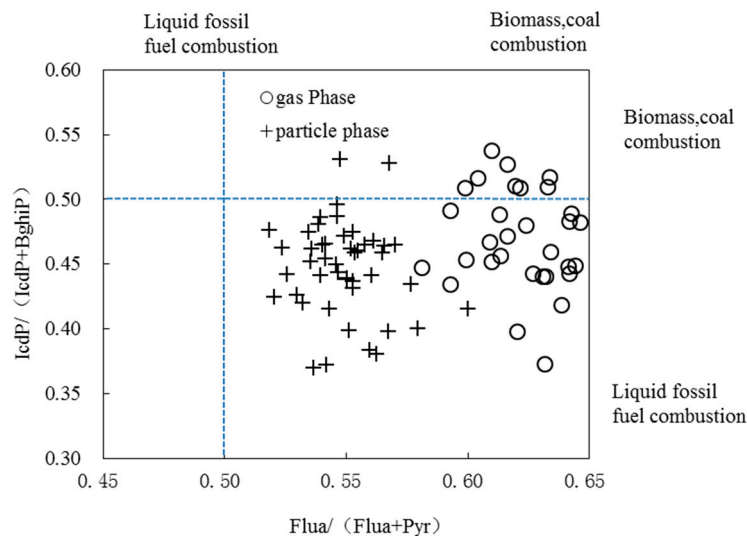


**Figure 2.** The concentration ( $\text{ng}\cdot\text{m}^{-3}$ ) of the 16 PAHs in the gas phase and particle phase during the different seasons. The PAHs are shown in order of decreasing molecular weight. (Note: Naphthalene (Nap) Acenaphthylene(Acy) Acenaphthene(Ace) Anthracene(Ant) Benzo[a]anthracene(BaA) Benzo[k]fluoranthene(BkF) Benzo[a]pyrene(BaP) Dibenz[a,h]anthracene(DahA)).

### 3.3. Sources of PAHs

#### 3.3.1. Diagnostic Ratios

The possible sources of PAHs in air can be identified by diagnostic ratios of individual PAH compounds, due to their different distributions in different energy sources [18]. We determined the diagnostic ratios of  $\text{Flua}/(\text{Flua} + \text{Pyr})$  and  $\text{IcdP}/(\text{IcdP} + \text{BghiP})$  in the gas and particle air samples. The burning of biomass and coal typically results in values  $>0.5$  for both of these ratios, while the combustion of liquid fossil fuels produce values  $<0.5$  [1,8]. In Figure 3, the obtained ratios are plotted against each other, showing that gas phase samples produced significantly higher values of  $\text{Flua}/(\text{Flua} + \text{Pyr})$  than particle samples, while the  $\text{IcdP}/(\text{IcdP} + \text{BghiP})$  ratio was more variable for gas than for particle samples. These two ratios clearly separated gas from particle phase samples. In the particle phase, the  $\text{Flua}/(\text{Flua} + \text{Pyr})$  ratio ranged between 0.38 and 0.58 (mean  $0.54 \pm 0.03$ ) and the  $\text{IcdP}/(\text{IcdP} + \text{BghiP})$  ratio ranged between 0.29 and 0.53 (mean  $0.44 \pm 0.04$ ). In the gas phase, these values were 0.39–0.68,  $0.62 \pm 0.04$  and 0.40–0.54,  $0.46 \pm 0.04$ , respectively. These results suggest that the PAHs were derived from mixed sources of vehicle exhaust and coal/biomass combustion.



**Figure 3.** The diagnostic ratio of  $\text{Flua}/(\text{Flua} + \text{Pyr})$  plotted against  $\text{IcdP}/(\text{IcdP} + \text{BghiP})$  obtained for gas and particle samples. The division between completely liquid fossil fuel combustion sources and biomass/coal burning are indicated by dotted lines.

### 3.3.2. Principal Component Analysis

We next used PCA to identify source characteristics of the PAH concentrations. By means of factor analysis, the principal components (PCs) displaying loading values  $>0.8$  were extracted with total accumulated loading accounting for more than 90% of the variance. As presented in Table 2, two or three PCs were extracted for each season. For spring, PC 1 plus PC 2 explained 92.6% of the total variance. For summer and autumn, the three retained PCs accounted for 93.6% and 95.3%, respectively, of the total variance in the dataset, while the two PCs for winter contributed 94.7%.

**Table 2.** Factor patterns of principle components (PC) 1 and 2, for individual PAHs (gas plus particle phase) in air from Chengdu, China.

PAHs	Spring		Summer			Autumn			Winter	
	PC1	PC2	PC1	PC2	PC3	PC1	PC2	PC3	PC1	PC2
Nap	0.227	<b>0.914</b>	−0.149	<b>0.853</b>	0.359	0.414	<b>0.817</b>	−0.127	0.747	0.489
Acy	0.469	<b>0.858</b>	0.174	<b>0.817</b>	−0.168	0.403	<b>0.883</b>	−0.165	<b>0.894</b>	0.283
Ace	0.296	<b>0.853</b>	0.032	<b>0.935</b>	−0.126	0.448	0.771	−0.216	<b>0.834</b>	0.464
Flu	0.211	<b>0.874</b>	−0.133	<b>0.946</b>	0.201	0.467	<b>0.866</b>	−0.081	<b>0.865</b>	0.404
Phe	<b>0.857</b>	0.039	0.250	0.462	0.752	0.556	0.458	0.626	0.288	<b>0.925</b>
Ant	0.650	0.543	0.114	−0.348	0.578	0.728	0.456	0.306	<b>0.914</b>	0.323
Flua	<b>0.924</b>	0.414	0.442	0.129	<b>0.848</b>	0.493	−0.107	<b>0.831</b>	0.741	0.589
Pyr	<b>0.887</b>	0.450	0.472	0.046	<b>0.840</b>	0.745	−0.357	0.498	<b>0.833</b>	0.526
BaA	0.696	0.576	0.686	−0.374	0.352	<b>0.928</b>	−0.175	0.273	<b>0.867</b>	0.475
Chr	<b>0.895</b>	0.494	<b>0.818</b>	0.060	0.569	<b>0.911</b>	−0.246	0.285	0.741	0.632
BbF	<b>0.820</b>	0.389	<b>0.933</b>	0.049	0.178	<b>0.920</b>	−0.236	−0.245	0.741	0.612
BkF	<b>0.862</b>	0.512	<b>0.956</b>	0.000	0.158	<b>0.851</b>	−0.461	−0.276	<b>0.825</b>	0.527
BaP	0.685	0.685	<b>0.876</b>	0.237	0.181	<b>0.945</b>	−0.137	−0.285	<b>0.872</b>	0.455
IcdP	<b>0.823</b>	0.632	<b>0.969</b>	−0.109	0.153	<b>0.864</b>	−0.214	−0.374	<b>0.912</b>	0.411
DahA	0.779	0.613	<b>0.921</b>	−0.055	0.309	<b>0.946</b>	−0.199	−0.335	<b>0.859</b>	0.441
Bghip	<b>0.838</b>	0.522	<b>0.964</b>	−0.071	0.199	<b>0.853</b>	−0.238	−0.379	<b>0.927</b>	0.379
Explained variance (%)	92.6%		93.6%			95.3%			94.7%	

PAH values mostly contributing to the PCs are shown in bold.

As shown in the table, in spring the cumulative variance contribution rate was mostly due to (in decreasing order of importance) Flua, Chr, Pyr, BkF, Phe, IcdP, Bghip, and BbF for PC1; and Nap, Flu, Acy, and Ace for PC2. Although BaA and DahA did not contribute in spring, they did so in the other

seasons; Ant only contributed to the variance contribution rate in winter. As summarized by others, Flua, Pyr, and Chr are considered as tracers of coal combustion, but Flu and Acy are more typical tracers of coke emissions, and BkF, IcdP, and Bghip are tracers of vehicle exhaust [1,7]. We conclude that PC1 and PC2 of spring can be considered as representative of pyrogenic sources, including coal burning, vehicle emissions, and coke.

Of the PC1 extracted from the data obtained in summer, the strongest contributors (IcdP, Bhghip, and BkF) point to traffic-related sources. There were over  $4.28 \times 10^6$  vehicles in Chengdu in 2015, so automobile exhaust is a big source of PAHs in this area. In summer, when other sources related to heating are low, the relative contribution of traffic will increase, which can explain the results. A similar result was found in Dalian, China [19]. Indeed, heating-related sources such as coke and coal were only apparent in PC2 (Flu and Acy, tracers of coke oven sources) and PC3 (Flua and Pyr, tracers of coal burning). This is because these energy sources are not only used for heating but also for local industrial production.

In autumn, again three PCs were extracted, and the PAHs mainly contributing to PC1 were BaP, BaA, BbF, DahA, and Bhgip. This composition is typical for a PAH source of vehicle (gasoline and diesel) exhaust emissions. PC2 is highly loaded by Flu and Acy (coke), and PC3 is highly loaded by Flua and Phe (coal).

In winter, two PCs were extracted. The individual contributors of PC1 are weaker than those for the first principal components defined for the other seasons. This illustrates a more mixed source of the PAHs in winter with, amongst others, Bghip, Ant, and IcdP as the main tracers.

### 3.3.3. Principal Component Analysis Combined with Multiple Linear Regression

Next, PCA/MLR analysis was conducted on the data of the four seasons. Two or three principle component factors were extracted, and a multiple regression equation was calculated to determine the rate of contribution of the energy sources the PAHs represent, for the different seasons. The correlations obtained were strong, with Pearson's correlation coefficients  $R^2 > 0.990$  (Table 3). For spring, factor  $x_1$  explains 52% of the total measured PAHs, and identifies coal as the average source, with strong similarities between the two profiles [4,20]. This profile is predominantly loaded by Flua, Pyr, and Phe; these are typically used as markers for coal [7]. Factor  $x_2$  is characterized by high BkF, IcdP, Bghip, Flu, and Acy, and is responsible for 48% of the measured PAHs. Flu and Acy are tracers of coke, whereas BkF, IcdP, and Bghip result from vehicle exhaust. The obtained profile is similar to the PAHs produced by vehicle exhaust and coke ovens in Dalian, a city in the northeast of China [21,22].

**Table 3.** Multivariate linear regression model identifying the main factors of the different seasons.

Relevant Parameter	Multiple Regression Equation	Rate of Contribution (%)	<i>p</i>	<i>R</i> <sup>2</sup>
Spring	$y = 0.650 x_1 + 0.590 x_2$	$x_1$ (coal): 52% $x_2$ (coke + motor vehicle exhaust): 48%	0.000	0.999
Summer	$y = 0.301 x_1 + 0.232 x_2 + 0.579 x_3$	$x_1$ (motor vehicle exhaust): 27% $x_2$ (coke): 21% $x_3$ (coal): 52%	0.001 0.001 0.000	0.991
Autumn	$y = 0.406 x_1 + 0.340 x_2 + 0.707 x_3$	$x_1$ (motor vehicle exhaust): 34% $x_2$ (coke): 19% $x_3$ (coal): 47%	0.000	0.990
Winter	$y = 0.724 x_1 + 0.514 x_2$	$x_1$ (coal): 58% $x_2$ (motor vehicle exhaust): 42%	0.001 0.000	0.998

For the summer season of Chengdu, factor  $x_1$ , explaining 27% of the total measured PAHs, mainly identifies motor vehicle exhaust sources, while factor  $x_2$  points to coke (rate of contribution 21%) and, with a contribution to factor 3 of 52%, coal burning is also identified.

During autumn, factor  $x_1$  shows that motor vehicle exhaust is a source of PAHs, contributing 34%. Exhibiting a profile similar to diesel engine emissions [21], factor  $x_2$  points to coke (with 19%),

while the most likely source for factor  $x_3$  is coal (47%). Finally, during winter, as expected, coal burning is the most important PAH source with 58% contributing to factor  $x_1$ , which is higher than the contribution of motor vehicle exhaust (42%) during this cold season. In the study area, atmospheric PAHs mainly originate from coal and biomass burning, followed by motor vehicle exhaust and coke. The contribution of coal combustion is different in different seasons, and the contribution of coal combustion in winter and spring to PAHs is higher than that in summer and autumn.

The PCA/MLR method is widely used to analyze the source of PAHs. For instance, Harrison et al. used this technique to identify the source of PAHs in atmosphere in Birmingham and showed that there the largest contributions were dust and motor vehicle exhaust (about 50%), with a contribution by BaP of about 88% [7]. Likewise, it was shown that the contribution to PAHs in Chicago were due to coal (48%), natural gas (26%), motor vehicle exhaust (9%), and coke (14%) [1]. Our data show the effect of season on the observed distributions of PAHs, where traffic is a stronger source in summer, and coal is a stronger source in winter.

#### 4. Conclusions

The diagnostic ratios of Flua/(Flua + Pyr) and IcdP/(IcdP + BghiP) were used to analyze the source of PAHs in the gas and particle air samples. The results suggest that the PAHs were derived from mixed sources of vehicle exhaust and coal/biomass combustion. Seasonal variation of sources resulting in PAHs in air collected in Chengdu, China, was detected by the application of diagnostic ratios and PCA/MLR models. Both models identified similar trends of the sources responsible for individual PAHs. The PCA/MLR model was used to estimate the contribution of each source, and clear differences were observed between seasons for relative source contributions. Coal burning was the major source (52%) in spring, followed by traffic emissions (48%), while in summer the main sources were coal combustion (52%), coke ovens (21%), and traffic emissions (27%). During autumn coal combustion was also the most important PAH source (47%), followed by traffic emissions (34%), and coke ovens (19%). Finally, during winter coal was the major source (58%), followed by traffic emissions (42%). These research results can provide suggestions and references for the prevention of local and nationwide air pollution.

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**Author Contributions:** Wenlai Xu conceived and designed the experiments; Ju Yang performed the experiments; Wenlai Xu analyzed the data; Huiyu Cheng contributed reagents and analysis tools; Wenlai Xu wrote the paper.

**Conflicts of Interest:** The authors declare no conflict of interest.

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