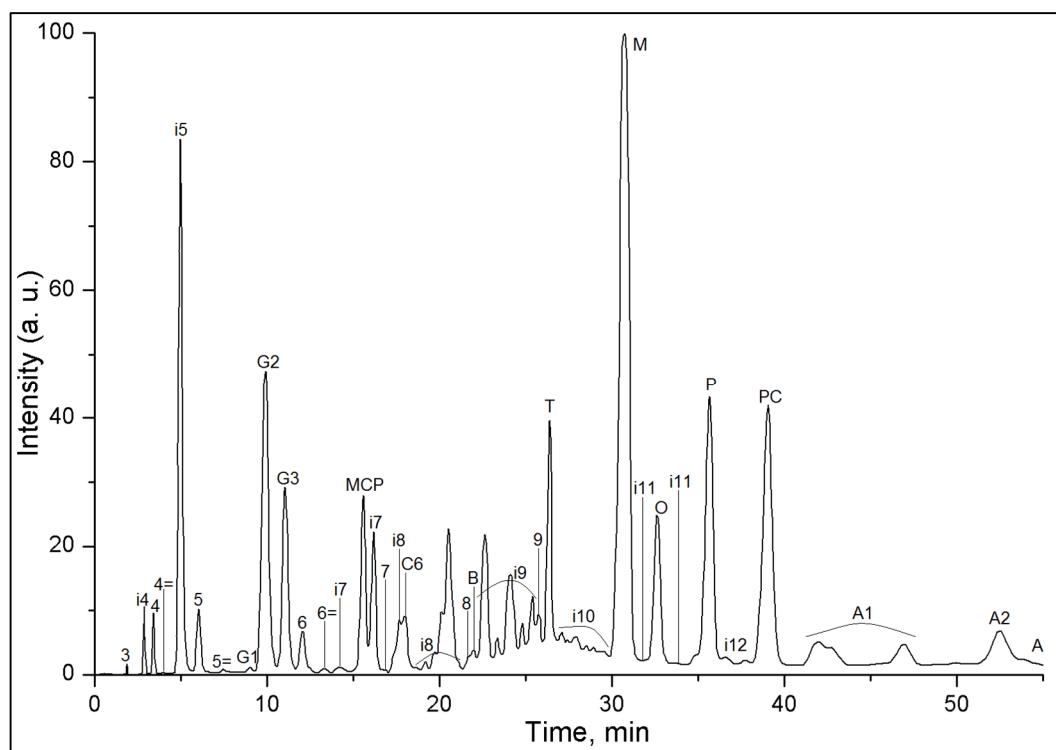


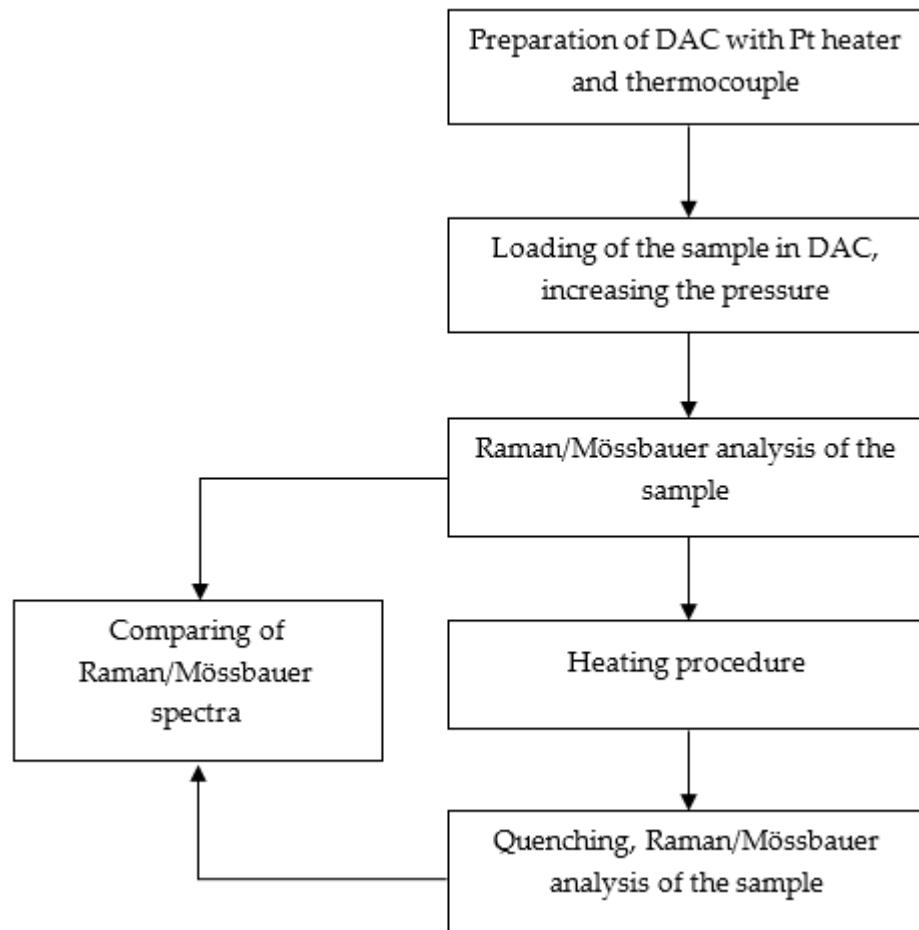
## Supplementary materials



**Figure S1.** Pt-Pt/Rh thermocouple, touching the surface of the diamond.



**Figure S2.** Chromatogram of the synthetic hydrocarbon system (a.u.—arbitrary units).



**Figure S3.** Summary of the completed experimental procedure.

**Table S1.** Characteristics of the crude oil from the Korchaginskoe deposit (Astrakhan region).

Number	Property	Units	Value
1	Pour point	°C	-24
2	Density at 20 °C	kg/m <sup>3</sup>	817
3	Molecular weight	kg/kmol	211
4	Viscosity at 20 °C	mm <sup>2</sup> /s	4.9
5	Asphaltene content	% weight	0.2
6	Silica gel pitch content	% weight	10.1
7	Petroleum wax content	% weight	5.6
8	Congelation point of paraffin	°C	+60
9	Sulphur content	% weight	0.543
10	Average diameter of dispersed particles	nm	290
11	Dispersed particles phase concentration	* 10 <sup>-12</sup> cm <sup>-3</sup>	1.082

**Table S2.** Composition of the synthetic hydrocarbon system.

Symbol	Compound	Conc. <sup>1</sup> , %	Symbol	Compound	Conc., %
3	propane C <sub>3</sub> H <sub>8</sub>	0.05	CG	cyclogexane, C <sub>6</sub> H <sub>12</sub>	1.33
i4	i-C <sub>4</sub> , i-butane, CH(CH <sub>3</sub> ) <sub>3</sub>	0.38	8	n-C <sub>8</sub> , n-octane, C <sub>8</sub> H <sub>18</sub>	0.17
4	n-C <sub>4</sub> , n-butane, C <sub>4</sub> H <sub>10</sub>	0.48	B	benzene, C <sub>6</sub> H <sub>6</sub>	0.40
4=	butylenes, C <sub>4</sub> H <sub>8</sub>	0.05	i9	i-C <sub>9</sub> , i-nonanes	7.14
i5	i-C <sub>5</sub> , i-pentane (2-methylbutane)	7.44	9	n-C <sub>9</sub> , n-nonane, C <sub>9</sub> H <sub>20</sub>	0.54
5	n-C <sub>5</sub> , n-pentane, C <sub>5</sub> H <sub>12</sub>	1.08	T	toluene, C <sub>6</sub> H <sub>5</sub> -CH <sub>3</sub>	3.44
5=	pentenes, C <sub>5</sub> H <sub>10</sub>	0.13	i10	i-C <sub>10</sub> , i-decanes ethylbenzene+meta-xylene + para-xylene, C <sub>6</sub> H <sub>5</sub> -C <sub>2</sub> H <sub>5</sub> + meta-C <sub>6</sub> H <sub>5</sub> -(CH <sub>3</sub> ) <sub>2</sub> + para-C <sub>6</sub> H <sub>5</sub> - (CH <sub>3</sub> ) <sub>2</sub>	0.71 22.30
G1	2,3-dimethylbutane	0.15	M	i-C <sub>11</sub> , i-undecanes ortho-xylene, ortho-C <sub>6</sub> H <sub>5</sub> - (CH <sub>3</sub> ) <sub>2</sub>	0.37 3.64
G2	2-methylpentane	7.45	i11	i-C <sub>12</sub> , i-dodecanes	0.30
G3	3-methylpentane	4.06	O	pseudocumene, 1,2,4- trimethylbenzene	0.18
6	n-C <sub>6</sub> , n-hexane, C <sub>6</sub> H <sub>14</sub>	0.94	P	ethyltoluens	7.74
6=	hexenes, C <sub>6</sub> H <sub>12</sub>	0.23	i12	i-C <sub>13</sub> , i-dodecanes	0.30
i7	i-C <sub>7</sub> , i-heptanes	2.84	PC	ethylylenes	13.53
MCP	methylcyclopentane	3.28	A1	durol, 1,2,4,5- thetramethylbenzene	2.60
7	n-C <sub>7</sub> , n-heptane, C <sub>7</sub> H <sub>16</sub>	0.16	A2	aromatic hydrocarbons C <sub>11+</sub>	0.78
i8	i-C <sub>8</sub> , i-octanes	6.13	A		

<sup>1</sup> Molar concentration.

---

<sup>1</sup> Molar concentration.