



Editorial Hydrometallurgy

Suresh K. Bhargava ^{1,*}, Mark I. Pownceby ^{2,*} and Rahul Ram ^{1,*}

- ¹ Centre of Advanced Materials & Industrial Chemistry, School of Applied Sciences, RMIT University, GPO Box 2476, Melbourne, VIC 3000, Australia
- ² CSIRO Mineral Resources, Private Bag 10, Clayton South, VIC 3169, Australia
- * Correspondence: suresh.bhargava@rmit.edu.au (S.K.B.); mark.pownceby@csiro.au (M.I.P.); rahul.ram@rmit.edu.au (R.R.)

Received: 18 May 2016; Accepted: 18 May 2016; Published: 23 May 2016

Hydrometallurgy, which involves the use of aqueous solutions for the recovery of metals from ores, concentrates, and recycled or residual material, plays an integral role in the multi-billion dollar minerals processing industry. It involves either the selective separation of various metals in solution on the basis of thermodynamic preferences, or the recovery of metals from solution through electro-chemical reductive processes or through crystallisation of salts. There are numerous hydrometallurgical process technologies used for recovering metals, such as: agglomeration; leaching; solvent extraction/ion exchange; metal recovery; and remediation of tailings/waste. Hydrometallurgical processes are integral across various stages in a typical mining recovery and mineral processing circuits be it in situ leaching (where solution is pumped through rock matrices); heap leaching (of the ROM or crushed ore); tank/autoclave leaching (of the concentrate/matte obtained from floatation); electro-refining (of the blister product from smelting routes); and the treatment of waste tailings/slags from the aforementioned processes. Modern hydrometallurgical routes to extract metals from their ores are faced with a number of issues related to both the chemistry, geology and engineering aspects of the processes involved. These issues include declining ore grade, variations in mineralogy across the deposits and geo-metallurgical locations of the ore site; which would influence the hydrometallurgical route chosen. The development of technologies to improve energy efficiency, water/resources consumption and waste remediation (particularly acid-rock drainage) across the circuit is also an important factor to be considered. Therefore, there is an ongoing development of novel solutions to these existing problems at both fundamental scales and pilot plant scales in order to implement environmentally sustainable practices in the recovery of valuable metals.

The Present Issue

We are delighted to be the Guest Editors for this Special Issue of Hydrometallurgy published in the journal *Metals*. With a total of 22 papers covering both fundamental and applied research, this issue covers all aspects of hydrometallurgy from comprehensive review articles [1,2], theoretical modelling [3] and experimental simulations [4], surface studies of dissolution mechanisms and kinetics [5], pre-treatment by roasting [6] or carbonation [7] to enhance recovery, aqueous carbonation as a means of CO_2 sequestration [8], biological systems [9–11], solvent and liquid-liquid extraction [12–14], nanoparticle preparation [15,16] and the development of novel and/or environmentally sustainable methods for the treatment of wastes and effluents for the recovery of valuable metals and products [17–22]. The number and of quality of submissions makes this Special Issue of *Metals* the most successful to date. As Guest Editors, we would especially like to thank Dr. Jane Zhang, Managing Editor for her support and active role in the publication. We are also extremely grateful to the entire staff of the *Metals* Editorial Office, who productively collaborated

on this endeavour. Furthermore, we would like to thank all of the contributing authors for their excellent work.

References

- 1. Huang, H.-H. The Eh-pH Diagram and Its Advances. *Metals* **2016**, *6*, 23. [CrossRef]
- 2. Rutledge, J.; Anderson, C.G. Tannins in Mineral Processing and Extractive Metallurgy. *Metals* **2015**, *5*, 1520–1542. [CrossRef]
- 3. Wadnerkar, D.; Pareek, V.K.; Utikar, R.P. CFD Modelling of Flow and Solids Distribution in Carbon-in-Leach Tanks. *Metals* **2015**, *5*, 1997–2020. [CrossRef]
- 4. Santini, T.C.; Fey, M.V.; Gilkes, R.J. Experimental Simulation of Long Term Weathering in Alkaline Bauxite Residue Tailings. *Metals* **2015**, *5*, 1241–1261. [CrossRef]
- 5. Li, Y.; Qian, G.; Li, J.; Gerson, A.R. Chalcopyrite Dissolution at 650 mV and 750 mV in the Presence of Pyrite. *Metals* **2015**, *5*, 1566–1579. [CrossRef]
- Yoon, H.-S.; Kim, C.-J.; Chung, K.W.; Jeon, S.J.; Park, I.; Yoo, K.; Jha, K. The Effect of Grinding and Roasting Conditions on the Selective Leaching of Nd and Dy from NdFeB Magnet Scraps. *Metals* 2015, *5*, 1306–1314. [CrossRef]
- 7. Santos, R.M.; Van Audenaerde, A.; Chiang, Y.W.; Iacobescu, R.I.; Knops, P.; Van Gerven, T. Nickel Extraction from Olivine: Effect of Carbonation Pre-Treatment. *Metals* **2015**, *5*, 1620–1644. [CrossRef]
- 8. Jo, H.; Jo, H.Y.; Rha, S.; Lee, P.-K. Direct Aqueous Mineral Carbonation of Waste Slate Using Ammonium Salt Solutions. *Metals* **2015**, *5*, 2413–2427. [CrossRef]
- 9. Fedje, K.K.; Modin, O.; Strömvall, A.-M. Copper Recovery from Polluted Soils Using Acidic Washing and Bioelectrochemical Systems. *Metals* **2015**, *5*, 1328–1348. [CrossRef]
- 10. Sueoka, Y.; Sakakibara, M.; Sera, K. Heavy Metal Behaviour in Lichen-Mine Waste Interactions at an Abandoned Mine Site in Southwest Japan. *Metals* **2015**, *5*, 1591–1608. [CrossRef]
- 11. Castro, L.; Blázquez, M.L.; González, F.; Munoz, J.A.; Ballester, A. Exploring the Possibilities of Biological Fabrication of Gold Nanostructures Using Orange Peel Extract. *Metals* **2015**, *5*, 1609–1619. [CrossRef]
- 12. Paiva, A.P.; Martins, M.E.; Ortet, O. Palladium(II) Recovery from Hydrochloric Acid Solutions by *N*,*N*'-Dimethyl-*N*,*N*'-Dibutylthiodiglycolamide. *Metals* **2015**, *5*, 2303–2315. [CrossRef]
- Lu, D.; Chang, Y.; Wang, W.; Xie, F.; Asselin, E.; Dreisinger, D. Copper and Cyanide Extraction with Emulsion Liquid Membrane with LIX 7950 as the Mobile Carier: Part 1, Emulsion Stability. *Metals* 2015, *5*, 2034–2047. [CrossRef]
- Saito, S.; Ohno, O.; Igarashi, S.; Kato, T.; Yamaguchi, H. Separation and Recycling for Rare Earth Elements by Homogeneous Liquid-Liquid Extraction (HoLLE) Using a pH-Responsive Fluorine-Based Surfactant. *Metals* 2015, 5, 1543–1552. [CrossRef]
- Zeng, X.; Niu, L.; Song, L.; Wang, X.; Shi, X.; Yan, J. Effect of Polymer Addition on the Structure and Hydrogen Evolution Reaction Property of Nanoflower-Like Molybdenum Disulfide. *Metals* 2015, *5*, 1829–1844. [CrossRef]
- 16. King, S.R.; Massicot, J.; McDonagh, A.M. A Straightforward Route to Tetrachlorauric Acid from Gold Metal and Molecular Chlorine for Nanoparticle Synthesis. *Metals* **2015**, *5*, 1454–1461. [CrossRef]
- 17. Inoue, K.; Gurung, M.; Xiong, Y.; Kawakita, H.; Ohto, K.; Alam, S. Hydrometallurgical Recovery of Precious Metals and Removal of Hazardous Metals Using Persimmon Tannin and Persimmon Wastes. *Metals* **2015**, *5*, 1921–1956. [CrossRef]
- 18. Park, K.; Jung, W.; Park, J. Decontamination of Uranium-Contaminated Sand and Soil Using Supercritical CO₂ with a TBP-HNO₃ Complex. *Metals* **2015**, *5*, 1788–1798. [CrossRef]
- 19. Slimi, R.; Girard, C. "High-Throughput" Evaluation of Polymer-Supported Triazolic Appendages for Metallic Cations Extraction. *Metals* **2015**, *5*, 418–427. [CrossRef]
- 20. Lee, S.-H.; Kwon, O.; Yoo, K.; Alorro, R.D. Removal of Zn from Contaminated Sediment by FeCl₃ in HCl Solution. *Metals* **2015**, *5*, 1812–1820. [CrossRef]

- 21. Wei, Y.-L.; Wang, Y.-S.; Liu, C.-H. Preparation of Potassium Ferrate from Spent Steel Pickling Liquid. *Metals* **2015**, *5*, 1770–1787. [CrossRef]
- 22. Siciliano, A. Use of Nanoscale Zero-Valent Iron (NZVI) particles for Chemical Dentrification under Different Operating Conditions. *Metals* 2015, *5*, 1507–1519. [CrossRef]



© 2016 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC-BY) license (http://creativecommons.org/licenses/by/4.0/).