

Article

Creep Behaviour and Microstructural Characterization of VAT 36 and VAT 32 Superalloys

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Received: 2 October 2018; Accepted: 23 October 2018; Published: 27 October 2018



Abstract: Superalloys are used primarily for the aerospace, automotive, and petrochemical industries. These applications require materials with high creep resistance. In this work, evaluation of creep resistance and microstructural characterization were carried out at two new nickel intermediate content alloys for application in aerospace industry and in high performance valves for automotive applications (alloys VAT 32 and VAT 36). The alloys are based on a high nickel chromium austenitic matrix with dispersion of intermetallic L₁₂ and phases containing different (Nb,Ti)C carbides. Creep tests were performed at constant load, in the temperature range of 675–750 °C and stress range of 500–600 MPa. Microstructural characterization and failure analysis of fractured surfaces of crept samples were carried out with optical and scanning electron microscopy with EDS. Phases were identified by Rietveld refinement. The results showed that the superalloy VAT 32 has higher creep resistance than the VAT 36. The superior creep resistance of the alloy VAT 32 is related to its higher fraction of carbides (Nb,Ti)C and intermetallic L₁₂ provided by the amount of carbon, titanium, and niobium in its chemical composition and subsequent heat treatment. During creep deformation these precipitates produce anchoring effect of grain boundaries, hindering relative slide between grains and therefore inhibiting crack formation. These volume defects act also as obstacles to dislocation slip and climb, decreasing the creep rate. Failure analysis of surface fractures of crept samples showed intergranular failure mechanism at crack origin for both alloys VAT 36 and VAT 32. Intergranular fracture involves nucleation, growth, and subsequent binding of voids. The final fractured portion showed transgranular ductile failure, with dimples of different shapes, generated by the formation and coalescence of microcavities with dissimilar shape and sizes. The occurrence of a given creep mechanism depends on the test conditions. At creep tests of VAT 32 and VAT 36, for lower stresses and higher temperature, possible dislocation climb over carbides and precipitates would prevail. For higher stresses and intermediate temperatures shear mechanisms involving stacking faults presumably occur over a wide range of experimental conditions.

Keywords: creep; superalloy VAT 36; superalloy VAT 32; high temperature

1. Introduction

Creep is the slow and continuous deformation of a solid with time, under the influence of mechanical stresses [1–4]. Superalloys play a role in the development of jet engine technology [5–8]. Over the past 20 years, the thrust of jet engines has increased by more than 60% whereas the fuel

consumption has fallen by 15–20%, and these improvements are, in part, the result of improvements in the high-temperature properties of superalloys [9].

A variety of high-performance materials is used in modern jet engines. Aluminum and carbon-fibers composites are used in the coolest sections of engines (operating at temperatures under 150 °C), such as the fan and inlet casing, to reduce weight. Titanium ($\alpha + \beta$ and β) alloys are used in engine components with operating temperatures around 550 °C, which includes parts in the fan and compressor sections. Superalloys are used for components that operate above 550 °C, such as the blades, discs, vanes, and other parts found in the combustion chamber and other high-temperature engine sections [10].

Materials used in the hottest engine components, such as high-pressure turbine blades and discs, must have high strength, fatigue life, fracture toughness, creep resistance, hot-corrosion resistance, and low thermal expansion properties. Nickel-based superalloys are the appropriate material for these engine components bearing in mind their capability to operate at high temperatures for long periods of time [10].

Nickel-based superalloys used in jet engines have high concentration of alloying elements (up to about 50% by weight) to provide strength, creep resistance, fatigue endurance, and corrosion resistance at high temperature. The types and concentration of alloying elements determines whether the superalloy is a solid solution-hardened or precipitation-hardened material. Precipitation-hardened superalloys are used in the hottest engine components, with their high-temperature strength and creep resistance improved by presence of L_{12} and other precipitates that have high thermal stability [10,11].

Iron-nickel superalloys are used in jet engines for their high-temperature properties and low thermal expansion. These superalloys, which contain 15–60% iron and 25–45% nickel, are used in blades, discs, and engine casings that require low thermal expansion properties. Other alloying elements such as Nb, Mo, W, Ti, Al, and Cr are also added deliberately to the formation of new phases with ability to support exposure to oxidizing environments at high temperatures for reasonable periods of time [11,12].

However, there is a growing interest in new superalloys with good creep resistance and low cost. For this purpose two new superalloys have been developed by Villares Metals, the VAT 36 and VAT 32, both with intermediate nickel content but different amounts of carbon, iron, titanium, and niobium. The VAT 32 and VAT 36 poly-grained alloys can replace superalloys in turbines but its main application is for high temperature automotive valves. The VATs alloys intend to replace UNS N07751 and UNS N07080 (Nimonic 80A). These alloys are commonly applied in high temperature components such as those for automotive valves production for high performance internal combustion engines, considering its high hot strength. However, the high cost is still a problem due to the high nickel contents. The new developed alloys (VAT 32 and VAT 36) present economic advantages caused by its lower nickel content and are based on a high nickel-chromium austenitic matrix with dispersion of L_{12} phases and different carbides. In this work, these alloys were subjected to creep tests at constant load, at a temperature range of 675–750 °C and a stress range of 500–600 MPa. The creep tests were conducted in accordance with ASTM E139. The study is complemented by microstructural characterization with optical microscopy (OM), Scanning Electron Microscopy (SEM), Energy Dispersive Spectroscopy (EDS), and Rietveld refinement. The VAT 32 alloy showed higher creep resistance and these results were correlated to formation of higher amount of carbides and intermetallic phases caused by higher mass percentage of carbon and different amounts of alloying elements.

2. Materials and Methods

Cylindrical bars of superalloys VAT 36 and VAT 32 were provided by Villares Metals Company (Sumaré, Brazil). Specimens for creep tests were prepared with shapes and dimensions depicted in Figure 1.

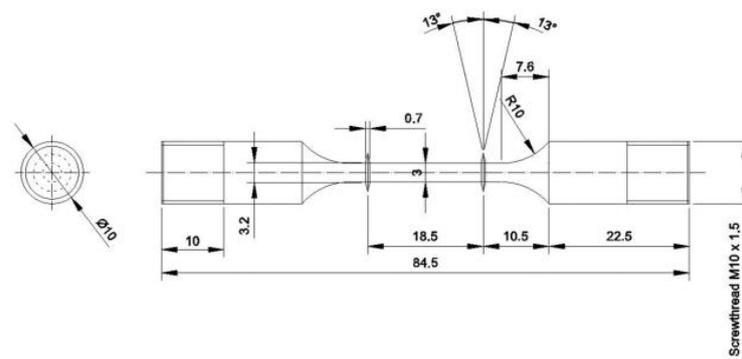


Figure 1. Dimensions of the specimen for creep tests in mm.

The specimens were subjected to heat treatment comprising two steps: (a) solution, held at 1050 °C during 30 min with a heating rate of 20 °C/min, followed by air cooling; (b) aging, carried out at 750 °C during 4 h, with a heating rate of 10 °C/min followed by air cooling. The creep tests were accomplished according to ASTM E-139-11 [13]. Electrical systems and controllers (Denison Mayes Group, Leeds, UK) were adapted in the furnace (Denison Mayes Group, Leeds, UK). A linear variable differential displacement transducer (LVDT, Denison Mayes Group, Leeds, UK) was used to obtain elongation measurements, and a Cromel-Alumel thermocouple was used for temperature control as detailed in [14]. The creep tests were performed at temperatures of 675, 700 and 750 °C and in constant load mode at stresses of 500, 550, and 600 MPa. A diffractometer Philips model PW 3710 (Philips, Best, The Netherlands) was used for Rietveld refinement, operated with $\text{CuK}\alpha$ radiation ($\lambda = 1.54184 \text{ \AA}$). The specimens were scanned at 2θ in a range of $10^\circ \leq 2\theta \leq 125^\circ$ and step increment of 0.02° .

The samples preparation for analysis via Optical Microscopy (Zeiss, Jena, Germany) and Scanning Electron Microscopy (SEM) (Jeol, Tokyo, Japan) followed the standard grinding procedure using silicon carbide sandpapers. Polishing was accomplished with diamond paste Codemaq with 0.3 nm. Etching was carried out with the kalling's reagent, with the following composition: 33 mL HCl, 33 mL ethanol, 33 mL water, and 1.5 g of CuCl_2 . The scanning electron microscope was used for microstructural and energy dispersive spectroscopy (EDS) analyses. The same microscope was operated aiming to evaluate fracture surfaces of crept samples and to identify the dominant fracture mechanism in each case.

3. Results and Discussion

3.1. Chemical Composition

The chemical compositions of VAT 36 and VAT 32 alloys are shown in Table 1.

Table 1. Hemical compositions: VAT 36 and VAT 32. (Mass.%).

Alloys/Elem.	Ni	Fe	Cr	C	Ti	Nb	Al
VAT 36	35.8	40.21	18.6	0.05	1.14	2.0	1.90
VAT 32	32.0	44.14	15.5	0.26	2.0	3.90	1.90

The VAT 32 alloy has about five times more carbon and approximately two times more titanium and niobium, when compared to VAT 36. Titanium and niobium preferably combine with carbon to form carbides. The excess of titanium and niobium unreacted with carbon combines with nickel giving rise to intermetallic phases L_{12} and increases the creep resistance.

3.2. Microstructural Evaluation after HT

Images obtained with optical and scanning electron microscopy showed both alloys (VAT 32 and VAT 36) with heterogeneous microstructures, with twinned areas, and grains of average size of 71.83 μm to 32 VAT and 70.75 μm to 36 VAT.

3.2.1. Microstructural Evaluation of Superalloy VAT 36

The microstructures for VAT 36 alloy are depicted in Figures 2 and 3.

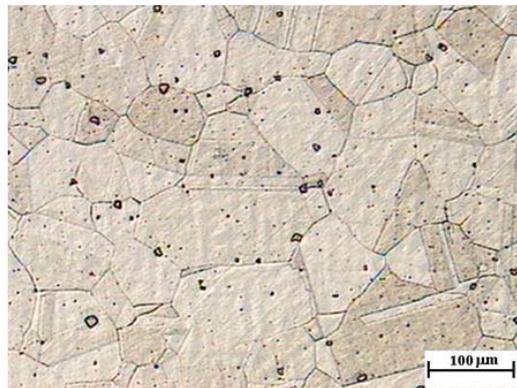


Figure 2. Optical micrograph of heat treated superalloy VAT 36. Magnification: 100 \times . Heterogeneous grain sizes and small amount of precipitates. Small twinned areas are observed.

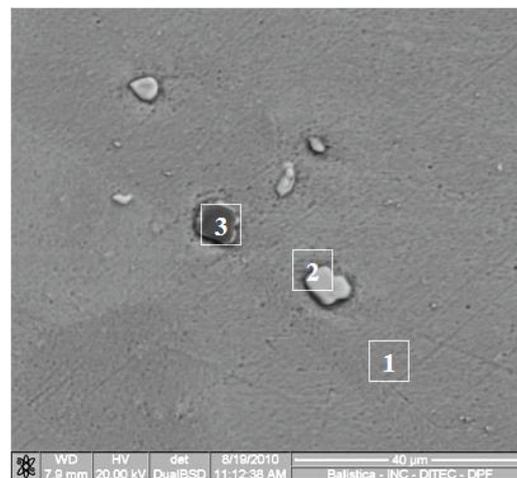


Figure 3. Scanning Electron Microscopy (SEM) micrograph of heat treated superalloy VAT 36 in backscattered electron image. Magnification: 3000 \times . Matrix (1) and two different precipitates (2 and 3).

Matrix chemical composition (1) was already presented in Table 1 for this alloy. Precipitates highlighted (2 and 3) were evaluated via EDS with semi quantitative analysis depicted in Table 2.

Table 2. Semi quantitative analysis of precipitates 2 and 3 of VAT 36, showed in Figure 3.

ZAF—Method Standardless Quantitative Analysis		
Element	Mass. % (Precipitate 2)	Mass. % (Precipitate 3)
C K	8.66	5.67
Fe K	2.28	4.65
Ti K	21.72	44.41
Nb K	67.24	18.69
Cr K	0.04	23.22
Ni K	0.06	3.36
TOTAL	100	100

From the values observed in Table 2 we can infer a possible formation of (Nb,Ti)C carbides in precipitate 2 and (Nb,Ti,Cr) carbides in precipitate 3.

3.2.2. Microstructural Evaluation of Superalloy VAT 32

Figures 4 and 5 depict microstructures for VAT 32 alloy. Precipitates were observed distributed throughout the microstructures, with higher amount of these precipitates for VAT 32 alloy (Figure 4) when compared with VAT 36 (Figure 2).

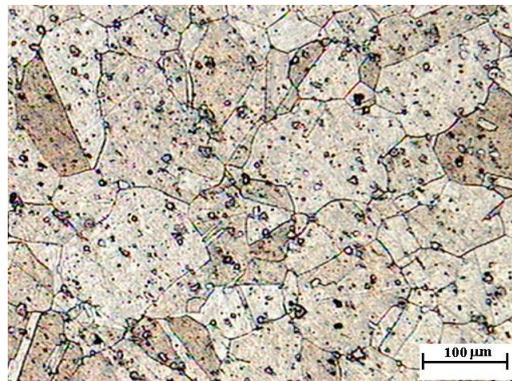


Figure 4. Optical micrograph of heat treated superalloy VAT 32. Magnification: 100×. Higher amount of precipitates is observed, in comparison with VAT 36 microstructure. Negligible twinned areas are observed.

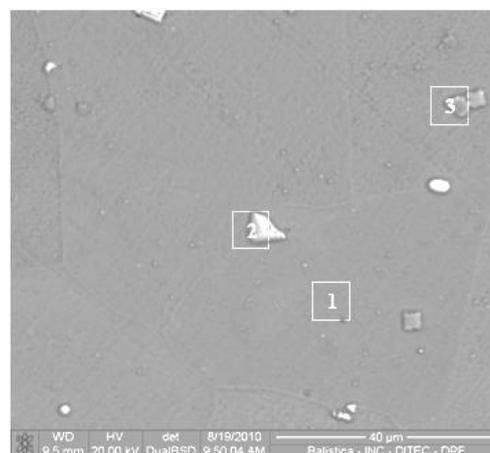


Figure 5. SEM micrograph of heat treated superalloy VAT 32 in backscattered electron mode in which matrix (1) and precipitates (2 and 3) are highlighted. Magnification: 3000×.

Matrix chemical composition of VAT 32 (1) is available in Table 1 for this alloy. Precipitates highlighted (2 and 3) were evaluated via EDS with compositions depicted in Table 3 as follows.

Table 3. Semi quantitative analysis of precipitates 2 and 3 observed in VAT 32, highlighted in Figure 5.

ZAF—Method Standardless Quantitative Analysis		
Element	Mass % (Precipitate 2)	Mass % (Precipitate 3)
C K	5.23	2.23
Fe K	2.64	1.78
Ti K	23.51	31.30
Nb K	65.34	62.39
Cr K	1.32	1.34
Ni K	1.96	0.96
Total	100	100

Both precipitates are quite similar for this alloy, possibly (Nb,Ti)C carbides.

The chemical analysis of both alloys VAT 36 and VAT 32 (Table 1) showed high mass percentage of iron, nickel, and chromium, as expected for Fe-Ni-Cr based superalloys. Other elements such as carbon, aluminum, titanium, and niobium are also present, however at lower concentrations. These elements form carbides and intermetallic compounds and, as a result, cause the alloy precipitation hardening as described in [15]. Evaluation of EDS semi quantitative analysis, obtained from precipitates observed in Figures 3 and 5 showed considerably higher concentration of niobium and titanium at precipitates for VAT 32 alloy, indicative of (Nb,Ti)C carbides formation and possibly intermetallic L1₂. For VAT 36, EDS analysis showed possibly formation of (Nb,Ti,Cr) carbides at precipitate 3. The weight percentage of precipitates is about 1.2% to VAT 36 and 5.5% to the VAT 32, as calculated in GSAS software in Rietveld refinement. The largest amount of precipitates for the alloy VAT 32 is due to higher concentration of carbon, titanium, and niobium in its chemical composition.

3.3. Rietveld Refinement

The peaks of the intermetallic L1₂ phases [Ni₃Al, Ni₃(Al_{0.5}Ti_{0.5})] are superimposed with the highest peak of γ austenitic matrix. A refinement of the crystal structure was accomplished by the Rietveld method (Figures 6 and 7), due to the difficulty of distinguishing the phase γ and other precipitates with similar lattice parameters as described in [16,17]. Table 4 contains the lattice parameters and calculated weight percentages of the phases for the alloys VAT 36 and VAT 32.

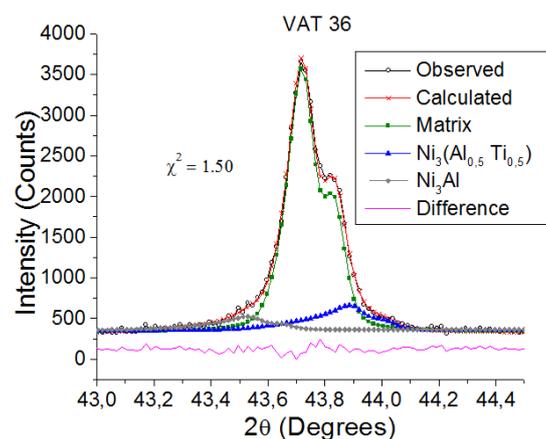


Figure 6. Rietveld refinement carried out at $2\theta = 44^\circ$ for the alloy VAT 36.

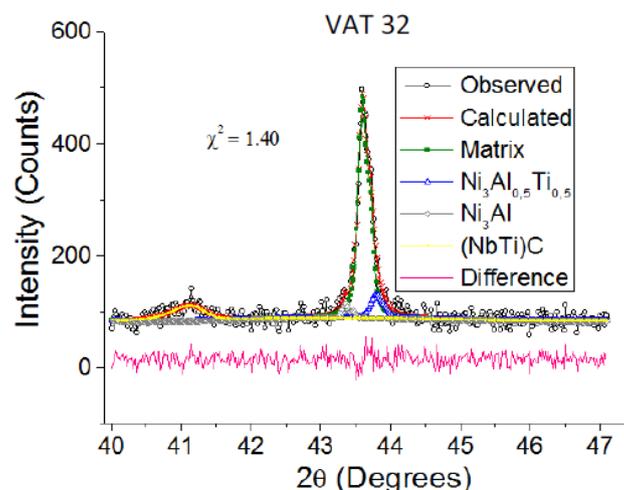


Figure 7. Rietveld refinement carried out in $40^\circ \leq 2\theta \leq 47^\circ$ (VAT 32).

Table 4. Lattice parameters and percentages of the phases for the alloys VAT 36 and VAT 32.

VAT 36		
Phases	wt.% Calculated	Lattice Parameters (Å)
Matrix (γ)	92	$a = b = c = 3.581$
L1 ₂ Phase	5	$a = b = c = 3.596$
L1 ₂ Phase	3	$a = b = c = 3.568$
VAT 32		
Phases	wt.% Calculated	Lattice Parameters (Å)
Matrix (γ)	64	$a = b = c = 3.590$
L1 ₂ Phase	9	$a = b = c = 3.611$
L1 ₂ Phase	5	$a = b = c = 3.581$
(NbTi)C	22	$a = b = c = 4.384$

3.4. Creep Tests

Figures 8–10 show the creep curves corresponding to deformation ϵ as a function of the time t for VAT 36 and VAT 32 alloys in creep tests carried out for a temperature range of 675–750 °C and stress range of 500–600 MPa.

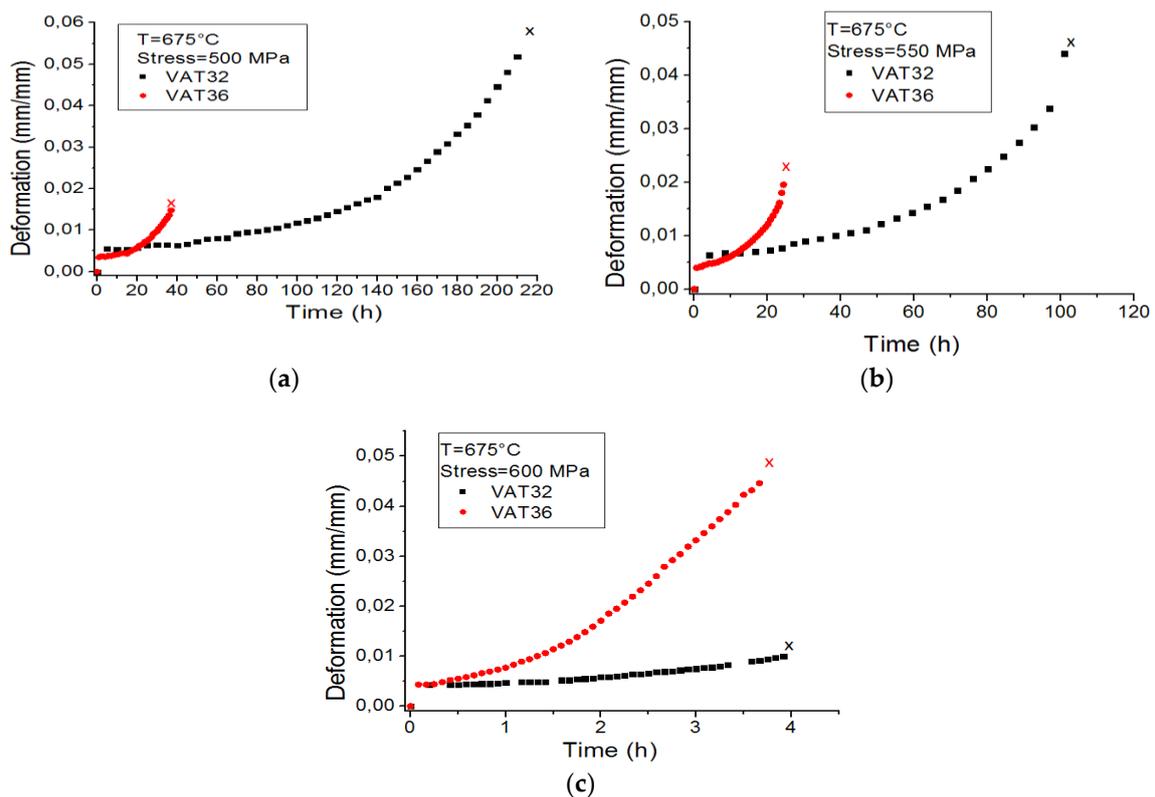


Figure 8. Creep curves of the alloy VAT 36 and VAT 32 at 675 °C at the stress of (a) 500. (b) 550 and (c) 600 MPa.

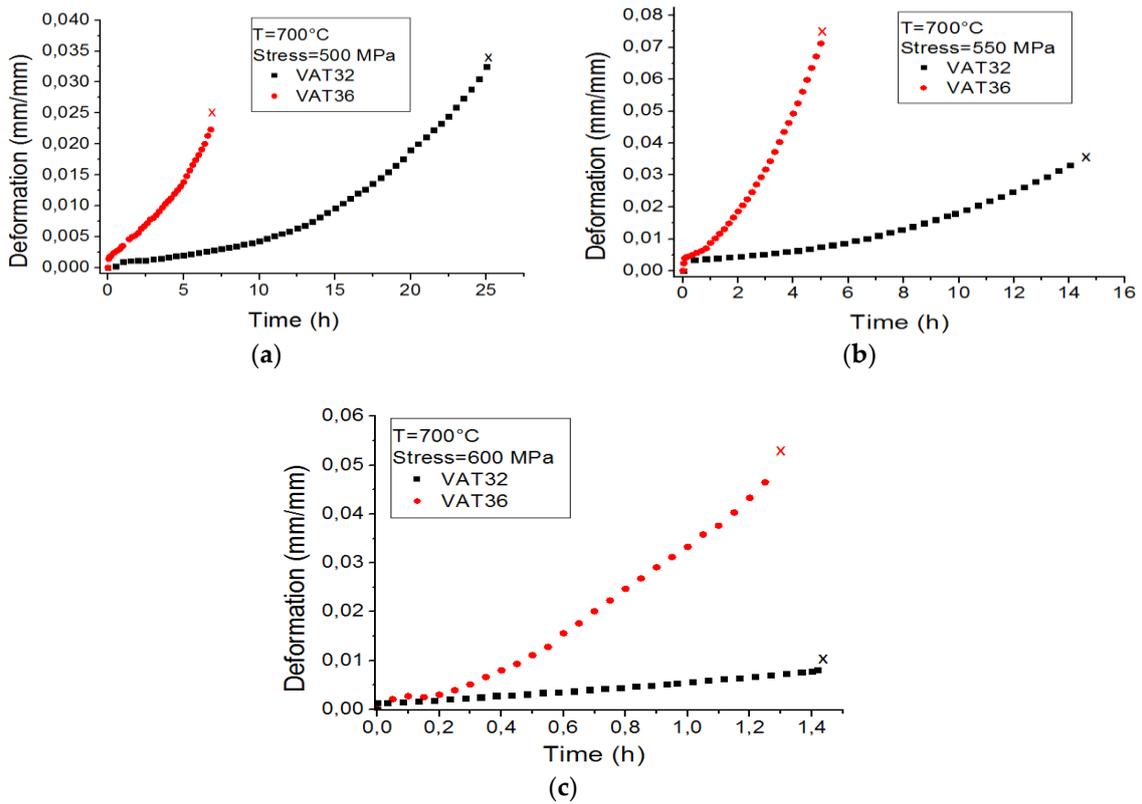


Figure 9. Creep curves of the alloys VAT 36 and VAT 32 at 700 °C at the stress of: (a) 500. (b) 550 and (c) 600 MPa.

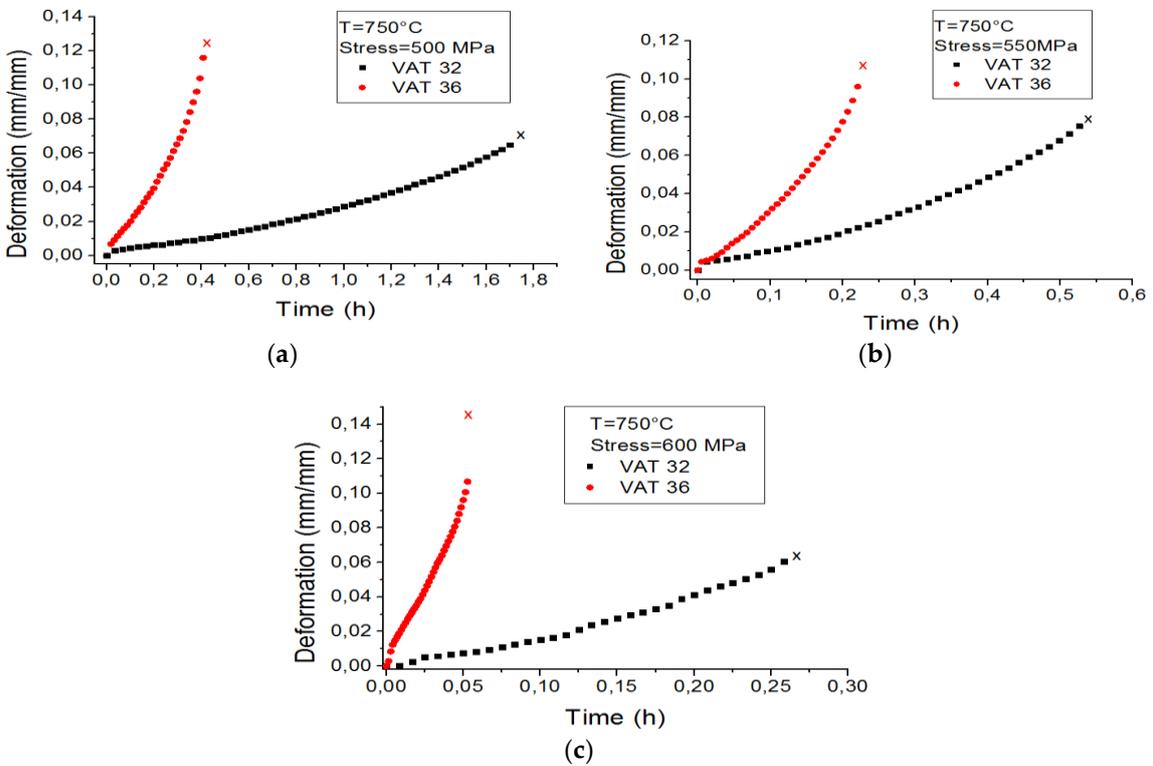


Figure 10. Creep curves of the alloys VAT 36 and VAT 32 the 750 °C at the stress of: (a) 500. (b) 550 and (c) 600 MPa.

Table 5 shows the creep data of all the creep curves.

Table 5. Creep tests data of the alloys VAT 36 and VAT 32 at temperatures (T) of 675, 700, and 750 °C and stresses (σ) of 500, 550, and 600 MPa, showing creep rate ($\dot{\epsilon}_s$), time of failure (t_f) and reduction of area (RA).

T (°C)	Material	σ (MPa)	$\dot{\epsilon}_s$ (1/h)	t_f (h)	RA (%)
675	VAT 36	500	3.25×10^{-4}	37.25	3.31
		550	9.08×10^{-4}	25.27	3.30
		600	4.01×10^{-3}	3.76	5.91
	VAT 32	500	5.08×10^{-5}	214.58	3.30
		550	1.75×10^{-4}	102.33	2.00
		600	7.56×10^{-4}	3.96	1.32
700	VAT 36	500	2.36×10^{-3}	6.95	14.12
		550	7.89×10^{-3}	5.12	11.64
		600	2.58×10^{-2}	1.30	12.26
	VAT 32	500	4.4×10^{-4}	25.16	4.61
		550	1.55×10^{-3}	14.57	3.96
		600	5.93×10^{-3}	1.43	3.26
750	VAT 36	500	1.71×10^{-1}	0.41	19.29
		550	3.87×10^{-1}	0.23	16.45
		600	14.65×10^{-1}	0.054	17.40
	VAT 32	500	2.15×10^{-2}	1.73	5.91
		550	8.43×10^{-2}	0.53	5.26
		600	2.38×10^{-1}	0.26	4.61

Table 5 shows smaller reduction of area after creep tests for VAT 32 alloy. Therefore this alloy has lower ductility than the VAT 36. The creep rate is also lower for VAT 32 for all temperatures and stresses by, at least, one order of magnitude. This higher creep resistance for this alloy (VAT 32) is attributed to a larger amount of MC carbides and intermetallic compound L1₂. The VAT 32 showed also higher times to failure at lower stresses.

At 675 °C, when the stress is increased from 500 to 600 MPa, the time to rupture decreases 54.2 times for the VAT 32 and to 9.9 times for the VAT 36. At 700 °C, this reduction is of 17.6 times for the VAT 32 and 5.3 times for the VAT 36. At a temperature of 750 °C, t_f decreases 6.6 times for the VAT 32 and 7.6 times for the VAT 36. Increase of creep tests temperatures results in higher ductility and the material becomes less sensitive to stress, causing lower differences between the times of rupture for higher stresses for both compositions. In general, both alloys showed typical behavior in creep tests, with decrease in the creep rate and increase of the rupture time with reduction of the stress or temperature. This indicates that the alloys exhibits creep rate sensitivity to both temperature and applied stress. For metals and alloys, the greater the temperature and higher strain rate, the earlier the stage of cavities and cracks formation will be [18].

The higher creep resistance of the alloy VAT 32 in comparison to VAT 36 is caused by the largest fraction of carbides (Nb,Ti)C and intermetallic L1₂ in VAT 32 alloys. These precipitates originated from the higher amount of carbon, titanium, and niobium in the VAT 32 chemical composition. These precipitates are stable and have a low rate of coalescence. As a result dislocations sliding are hindered by these obstacles during creep deformation [19]. Carbides also produce an anchoring effect at grain boundaries, hence delaying relative sliding between grains, affecting the ease with which vacancies can be generated in these regions [20,21]. Dependence of the steady state creep rate with stress and temperature could be expressed as an Arrhenius type relationship, and represented by the Equation (1) [21].

$$\dot{\epsilon}_s = B_0 \sigma^n \exp\left(\frac{-Q_C}{RT}\right) \quad (1)$$

where B_0 is a factor dependent on the stress and the structure of the material, σ is the stress applied, R is the gas constant, and T is the absolute temperature. The concept of a mechanism responsible for creep phenomenon can be characterized by different values of stress exponent n combined with the activation energy Q_c . Therefore, values of n and Q_c were determined in this work, based on Equation (1) and the data in Table 3, as can be seen in Figures 11 and 12.

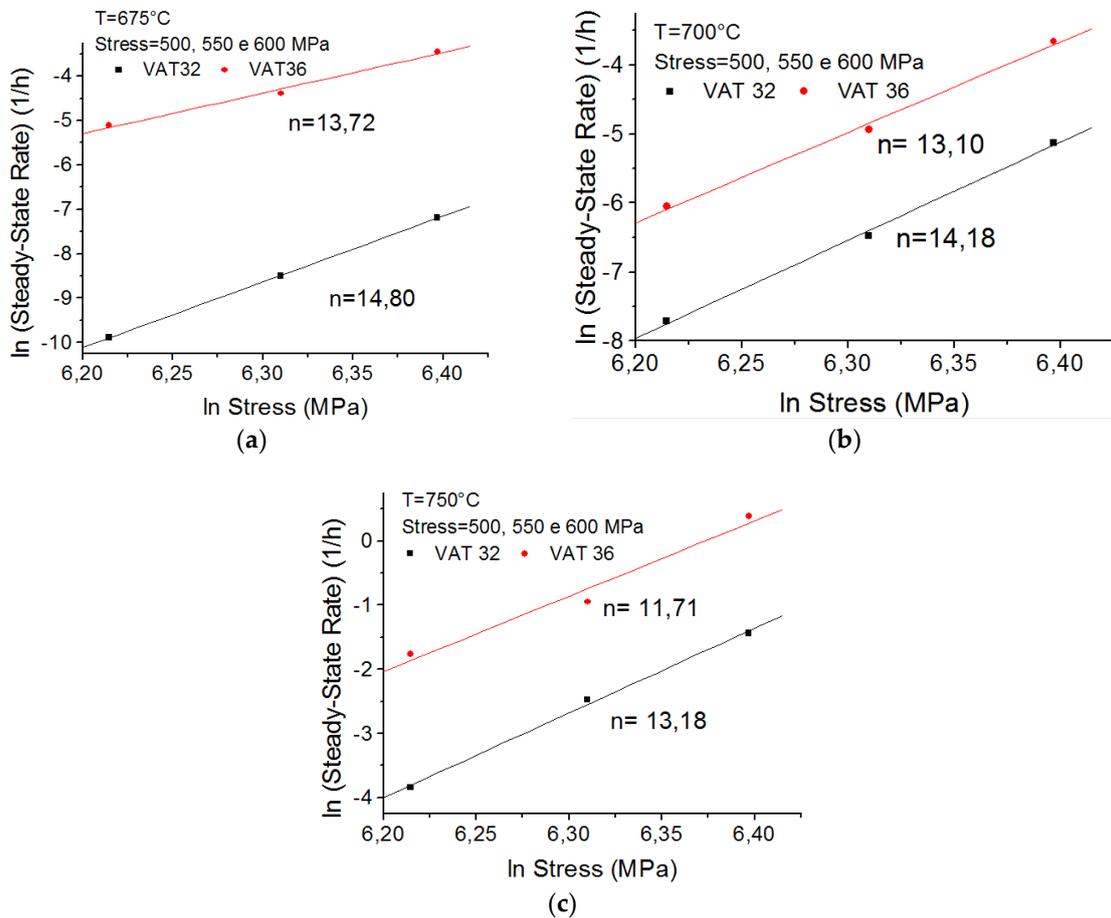


Figure 11. Dependence of the steady-state creep rate with the applied stress to the alloys VAT 32 and VAT 36 at: (a) 675 °C. (b) 700 °C and (c) 750 °C.

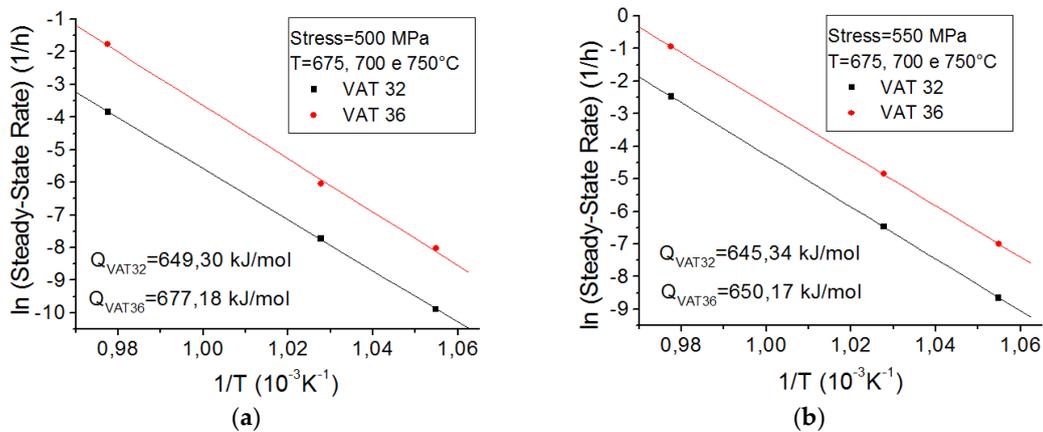


Figure 12. Cont.

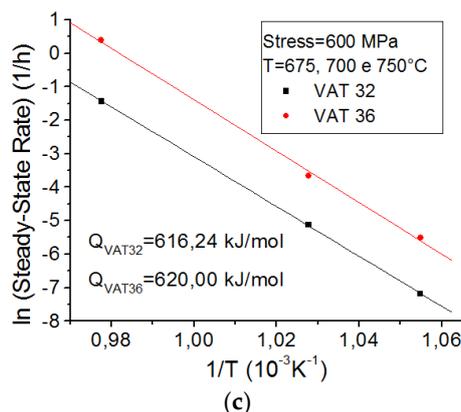


Figure 12. Steady-state creep rate dependence with the temperature for the alloys VAT 32 and VAT 36 at: (a) 500 MPa. (b) 550 MPa and (c) 600 MPa.

At 675 °C a stress exponent of 14.80 was obtained for the VAT 32 and 13.72 for the VAT 36. At 700 °C, n values reached 14.18 for the VAT 32 and 13.10 for the VAT 36. At 750 °C the stress exponents were 13.18 for the VAT 32 and 11.71 for the VAT 36.

Activation energies values at 600 MPa were near 616 kJ/mol for the VAT 32 and 620 kJ/mol for the VAT 36. At 550 MPa these values reached 645 kJ/mol for the VAT 32 and 650 kJ/mol for the VAT 36. At 500 MPa activation energies were 649.30 kJ/mol for the VAT 32 to 677 kJ/mol for the VAT 36.

The occurrence of a given creep mechanism depends on the test conditions. For lower stresses and high temperatures, possibly dislocation climb over carbides and precipitates would prevail. For higher stresses and lower temperatures shear mechanisms involving stacking fault occurs over a wide range of experimental conditions [22,23]. Previous research showed the dependence of velocity of dislocation movement with carbides volumetric fraction [24,25].

Dislocations movement is hindered by carbide particles. Reduced creep rate of VAT 32 is strongly affected by higher amounts of carbides (Nb,Ti)C and intermetallic L1₂, when compared to VAT 36. Those particles behave as obstacles to dislocations movement, with the creep rate dependent on their size and distribution.

3.5. Failure Analysis

Fracture surfaces of crept samples are depicted in fractographs showed in Figures 13–15. Figure 13 shows the fractured surface of sample VAT 36 after creep test. The crack starts with intergranular mode (region A). After reaching critical size, crack propagation occurs at transgranular ductile mode (region B), with dimples.

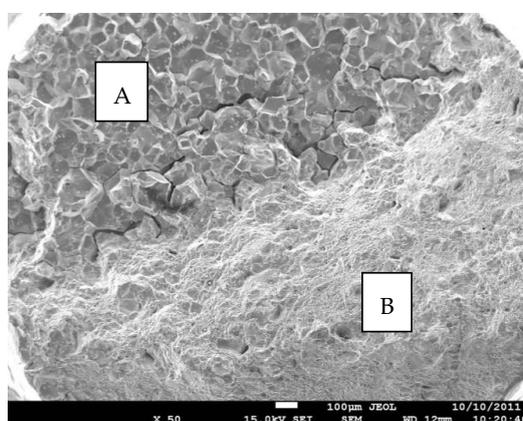


Figure 13. Fracture surface of crept VAT 36 sample, in creep test at 700 °C and stress of 600 MPa, with intergranular (A) and transgranular ductile (B) regions.

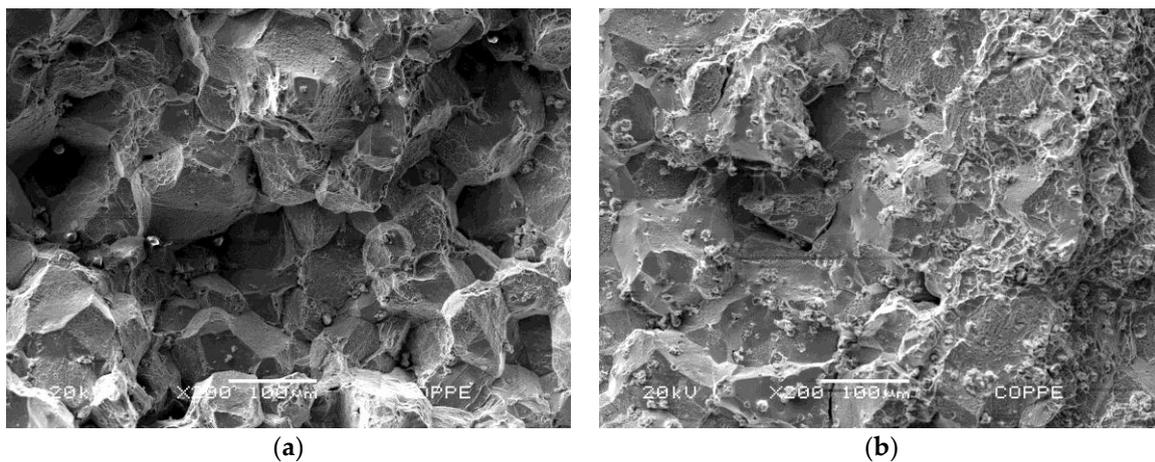


Figure 14. Surface of fracture initiation sites of VAT 36 (a) and VAT 32 (b), crept at 700 °C and 500 MPa, with an intergranular mode for both cases. Considerable amount of precipitated carbides can be seen at the VAT 32 surface.

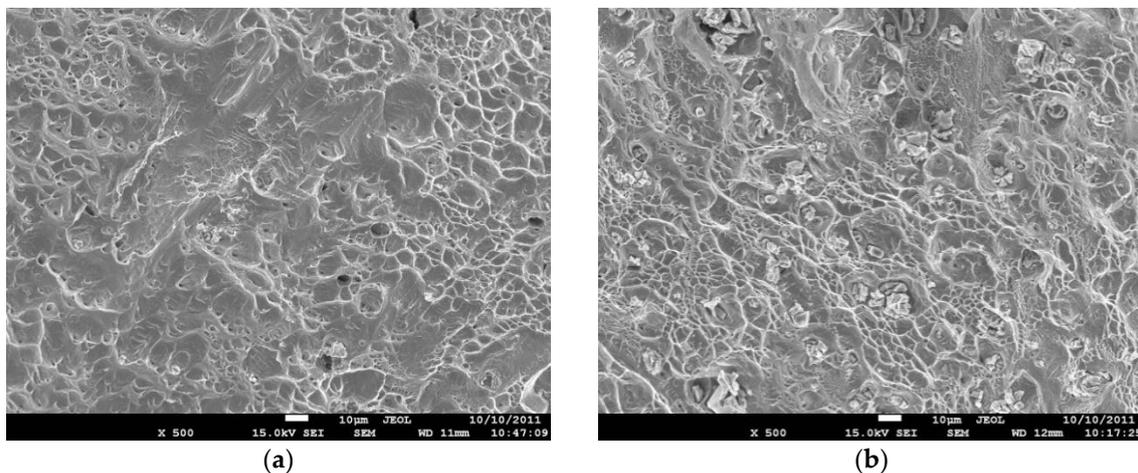


Figure 15. Final portions of fractured surfaces of VAT 36 (a) and VAT 32 (b) alloys, subjected to creep tests at 750 °C showing dimples of ductile fracture.

Detailed examination of the crack initiation of both compositions showed intergranular mode at these regions, as depicted in Figure 14. Higher amount of precipitated carbides can be observed for VAT 32 composition.

At the final region of fractured surfaces, ductile fractures prevailed, with a large amount of dimples. Particulate carbides remained inside dimples for VAT 32 alloy, as depicted in Figure 15b.

The fractographs showed intergranular fracture at creep initial stages for both alloys. Steps of creep intergranular fracture are nucleation, growth, and subsequent binding of voids at triple points. The formation of cavities is controlled by diffusion processes. The stacking of dislocations in regions near the grain boundaries during creep can increase the stress concentration favoring the initiation and propagation of intergranular microcracks [22].

At the final surface cracks, ductile fracture prevails (transgranular ductile fracture) with formation and coalescence of microcavities of variable shapes and sizes. The size and shape of the dimples are governed by the number and distribution of nucleated microcavities and by the level of internal stresses present in the material [26].

The alloy VAT 36 showed more areas of transgranular ductile fracture, considering its higher ductility caused by lower weight percentage of carbides and intermetallics.

4. Conclusions

1. Rietveld refinement allowed qualitative and quantitative evaluation of carbides and intermetallic precipitates in both alloys. The VAT 36 has approximately 8 wt.% of intermetallic L₁₂ and VAT 32, 14 wt.%. The precipitates hindered dislocation slip and grain boundaries slip, mainly in VAT 32 alloy, increasing its creep resistance.
2. The VAT 32 alloy showed smaller reduction in area after creep tests in relation to VAT 36. This result is caused by lower ductility of VAT 32 originated from higher mass percentage of carbon in its composition, favoring formation of phases based on carbides.
3. The higher creep resistance of the alloy VAT 32 is related to its substantial fraction of carbides (Nb,Ti)C and intermetallic L₁₂, provided by its larger carbon content. The excess of titanium and niobium unreacted with carbon combines with nickel giving rise to intermetallic phases L₁₂. These precipitates are stable and have low rate of coalescence. As a result, during creep deformation these precipitates produce anchoring effect of grain boundaries hindering relative slide between grains and therefore causes crack formation delay. These volume defects act also as obstacles to dislocation slip and climb, decreasing the creep rate.
4. Failure analysis of surface fractures of crept samples showed intergranular failure mechanism at crack origin for both alloys VAT 36 and VAT 32. Intergranular fracture involves nucleation, growth, and subsequent binding of voids. The final fractured portion showed transgranular ductile failure, with dimples of different shapes and sizes, typical of ductile failure. Transgranular ductile fracture involves the formation and coalescence of microcavities with dissimilar shape and sizes. The VAT 32 showed smaller areas of intragranular failure mechanism (dimples). This behavior is caused by its lower ductility.
5. Stress exponents obtained in this work were in a range of 14.80–11.71 and activation energies between 677–616 kJ/mol. The occurrence of a given creep mechanism depends on the test conditions. At creep tests of VAT 32 and VAT 36, for lower stresses and high temperatures, dislocation climb over carbides and precipitates possibly prevail. For higher stresses and intermediate temperatures shear mechanisms involving stacking fault can occur over a wide range of experimental conditions

Author Contributions: Conceptualization: C.R.M.d.S., V.J.G., S.J.G., D.A.P.R., J.L.d.A.F., J.A.A.; Validation: C.R.M.S., V.J.G., S.J.G., D.A.P.R., J.L.d.A.F., J.A.A.; Formal Analysis: C.R.M.d.S., V.J.G.; Investigation: C.R.M.d.S., D.A.P.R., V.J.G.; Writing-Review and Editing: C.R.M.d.S., V.J.G.; Supervision: C.R.M.d.S.

Funding: This research received no external funding.

Acknowledgments: We would like to show our gratitude to Villares Metals who provided the samples for creep tests.

Conflicts of Interest: The authors declare no conflict of interest.

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