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Article

# Effect of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> Dopants on the Stable Dielectric Properties of BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub>-Based Materials

Sea-Fue Wang <sup>†,\*</sup>, Yung-Fu Hsu, Yu-Wen Hung and Yi-Xin Liu

Department of Materials and Mineral Resources Engineering, National Taipei University of Technology, Taipei 106, Taiwan; E-Mails: yfhsu@ntut.edu.tw (Y.-F.H.); lovemyky@gmail.com (Y.-W.H.); evie19831123@gmail.com (Y.-X.L.)

- <sup>†</sup> Current Address: Department of Materials and Mineral Resources Engineering, National Taipei University of Technology, 1, Sec. 3, Chung-Hsiao E. Rd., Taipei 106, Taiwan.
- \* Author to whom correspondence should be addressed; E-Mail: sfwang@ntut.edu.tw; Tel.: +886-2-2771-2171 (ext. 2735).

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Abstract: In this study, BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of Ta<sub>2</sub>O<sub>5</sub> dopant were investigated for their ability to enhance high-temperature stability to meet X9R specifications. The results were compared to those for ceramics with the common Nb<sub>2</sub>O<sub>5</sub> additive. The best composition appeared to be 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> with 2 mol% Ta<sub>2</sub>O<sub>5</sub> dopant sintered at 1215 °C, which had a dielectric constant of 1386, a tanð value of 1.8%, temperature coefficients of capacitance (TCCs) of –1.3% and 1.2%, and electrical resistivities of  $2.8 \times 10^{12}$  and  $1.5 \times 10^{10} \Omega \cdot \text{cm}$  at 25 °C and 200 °C, respectively. Its microstructure consisted of fine equiaxed grains with a perovskite structure and an average grain size of 0.46 µm and some rod-like grains of second-phase Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> with a size of approximately 3.2 µm. The Ta<sub>2</sub>O<sub>5</sub> dopant caused a reduction in the grain size and a slight increase in trapped pores. The temperature dependence of the dielectric constant flattened and the Curie point was dramatically suppressed with the addition of Ta<sub>2</sub>O<sub>5</sub> dopant, leading to smooth dielectric temperature characteristics over a relatively broad temperature range. The X9R formulations and their dielectric properties were highly repeatable in this study.

Keywords: dielectric properties; X9R; formulation; microstructure

### 1. Introduction

Multilayer ceramic capacitors (MLCCs) are crucial electronic components used in virtually every area of electronics, and the number and variety of their applications have steadily grown. During the past decade, MLCC manufacturers have been driven to accelerate progress in miniaturization, enhancement of volumetric efficiency, reinforcement of stability and reliability against environment, and cost reduction, none of which can be achieved without an improvement in materials formulation and processing technology. The compositions of most ceramic capacitors are based on ferroelectric barium titanate (BaTiO<sub>3</sub>) materials. BaTiO<sub>3</sub> can be either chemically or physically modified to exhibit the required temperature-stable dielectric behavior [1].

MLCCs used in automobile electronic parts, such as anti-lock brake systems, engine electronic control units, crank angle sensors, and programmed fuel injections, are exposed to high-temperature working conditions of greater than approximately 130 °C in the engine room of an automobile. MLCCs with X8R specifications (X8R: -55 °C to 150 °C,  $\Delta C/C_{25 \circ C} \leq \pm 15\%$ ) were developed to satisfy this requirement. Because of the dramatic variation in permittivity of BaTiO<sub>3</sub> ceramics above the Curie temperature ( $T_c \approx 125$  °C), the X8R formulation makes use of the core-shell structure to retain the temperature-stable characteristic by adding rare earth elements with a smaller ionic radius, such as Tm, Yb, Lu, Sc, Er, and Y, to shift the T<sub>c</sub> to higher temperatures [2–4]. Nevertheless, for aerospace, oil drilling, and various other applications, MLCCs in electronic devices must withstand even harsher working environments, typified by temperatures higher than 150 °C. Accordingly, Yuan et al. and a few other researchers have recently started to work on the development of dielectric formulations capable of satisfying X9R specifications (-55 °C to 200 °C,  $\Delta C/C_{25 \circ C} \leq \pm 15\%$ ) [5–11]. The utilization of an effective Curie-point shifter, such as (Bi0.5Na0.5)TiO3 or PbTiO3, to increase the T<sub>c</sub> point along with the formation of the core-shell structure was found to be an effective method for obtaining high-temperature stability [12–16]. Dielectric formulations, including BaTiO<sub>3</sub>-based compositions, such as BaTiO<sub>3</sub>-(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> [5,7,8,14,17-19], BaTiO<sub>3</sub>-Bi(Zn<sub>0.5</sub>Ti<sub>0.5</sub>)O<sub>3</sub> [16,20], and BaTiO<sub>3</sub>–Pb(Ti<sub>0.55</sub>Sn<sub>0.45</sub>)O<sub>3</sub> [10] systems, and (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub>-based compositions, such as (Bi0.5Na0.5)TiO3-BaTiO3-CaTiO3 [6,9,21] systems, have been developed recently to meet the temperature coefficient of capacitance (TCC) requirements of the X9R specifications. Another approach was a multi-phase composite consisting of a high- $T_c$  material and a low- $T_c$  material. For instance, new X9R dielectrics based on the BaTiO<sub>3</sub>-LiTaO<sub>3</sub> compositions reported by Wang et al. can be sintered in a reducing atmosphere and are compatible with the base metal electrode (BME) process [22].

In the BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> system, (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> shows strong ferroelectricity at room temperature, undergoes a phase transition with a broad dielectric maximum approximately close to 320 °C [8], and is considered as an ideal material to shift the Curie temperature of BaTiO<sub>3</sub> above 150 °C [15]. The distortion and deformation of BaTiO<sub>3</sub> structure induced by the substitution of Na<sup>+</sup> and Bi<sup>3+</sup> into Ba sites by (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> addition improves the stable temperature characteristics of the dielectric properties of BaTiO<sub>3</sub>. With the addition of Co<sub>3</sub>O<sub>4</sub> and Nb<sub>2</sub>O<sub>5</sub>, the stable dielectric properties of BaTiO<sub>3</sub> were tailored to meet the X9R specification by adjusting the core–shell structure [14,15]. It was also shown in the literature that the dielectric properties of BaTiO<sub>3</sub> or bismuth-based compounds can be significantly enhanced with the addition of Ta<sub>2</sub>O<sub>5</sub>, through tailoring the microstructure, improving densification, and shifting Curie temperature [23,24]. In this

study, BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> dielectric ceramics incorporating various amounts of Ta<sub>2</sub>O<sub>5</sub> additive were investigated to enhance their high temperature stability to meet the X9R specifications. The effect of the Ta<sub>2</sub>O<sub>5</sub> additive on the densification, crystalline phase, microstructure development, and dielectric properties of BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics compared to those of the common Nb<sub>2</sub>O<sub>5</sub> additive under the same processing conditions were investigated and are discussed below.

### 2. Experimental Procedure

Dielectric powders of  $(1 - x)BaTiO_3-x(Bi_{0.5})TiO_3$  ceramics with various amounts of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> were prepared using the solid-state reaction technique. Highly pure (>99.9% purity) BaTiO<sub>3</sub> (Ferro, reagent grade), Bi<sub>2</sub>O<sub>3</sub> (SHOWA, reagent grade), Na<sub>2</sub>CO<sub>3</sub> (SHOWA, reagent grade), TiO<sub>2</sub> (SHOWA, reagent grade), Ta<sub>2</sub>O<sub>5</sub> (Alfa, reagent grade), and Nb<sub>2</sub>O<sub>5</sub> (Alfa, reagent grade) were used as raw materials. The host material (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> was prepared by mixing appropriate amounts of Bi<sub>2</sub>O<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub>, and TiO<sub>2</sub> and calcining them at 800 °C for 12 h, and the phase was identified using X-ray diffraction (XRD, Rigaku DMX-2200, Rigaku, Tokyo, Japan). Formulations based on compositions of  $(1 - x)BaTiO_3-x(Bio_5Na_{0.5})TiO_3$  (x = 0.05, 0.1, and 0.2) with various amounts of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> (0, 1.0, 1.5, and 2.0 mol%) were mixed and milled in methyl alcohol solution using polyethylene jars and zirconia balls for 24 h and then dried at 80 °C in an oven overnight. The mixed powders were added to 4 wt% of a 15 wt% Polyvinyl Alcohol (PVA) solution and were pressed into disc-shaped compacts under a uniaxial pressure of 120 MPa. The samples were then heat treated at 550 °C for 4 h to eliminate the PVA, followed by sintering at 1200 °C to 1245 °C. The heating and cooling rates were controlled at 5 °C/min for all samples, and the dwell time was set to 2 h.

The apparent density of the sintered ceramics was measured on five samples for each composition using the Archimedes method with de-ionized water. Phase identification on the polished surfaces of the sintered bulk ceramics was performed using X-ray diffraction (XRD, Siemens D5000, Karlsruhe, Germany). The microstructures of the sintered samples were examined with scanning electron microscopy (SEM, JEOL 6500F, Tokyo, Japan) and energy-dispersive spectroscopy (EDS). The average grain sizes of the sintered samples were determined using the linear intercept, which was calculated as the mean value of the measurement diagonals of about 200 grains. The dielectric properties of the sintered samples were measured as a function of temperature using an automatic measurement system incorporating a Delta Design Environmental Chamber with an LCR meter (HP 4284A, Santa Clara, CA, USA) at a frequency of 1 kHz. Before measurement, electrodes were fabricated on the disc samples using silver paste applied on opposing surfaces and fired at 750 °C. The measurements were performed during heating cycles from -55 °C to 130 °C at a rate of 2 °C/min with an accuracy of 0.1 °C. An alternating voltage of 1 V was applied. The electrical resistivity of the bulk ceramics was determined by a high resistance meter (HP 4339B, Santa Clara, CA, USA) at 25 and 200 °C at a DC voltage of 500 V. The measurements of dielectric properties and electrical resistivity were performed on at least three samples in each case, with an accuracy of  $\pm 3\%$ .

#### 3. Results and Discussion

To determine the best composition of the host  $(1 - x)BaTiO_3 - x(Bi_{0.5}Na_{0.5})TiO_3$  ceramic, formulations with an x value of 0.05, 0.1, or 0.2 were prepared and characterized. While the sintering

temperature of BaTiO<sub>3</sub> is typically above 1300 °C, the addition of (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> significantly reduced the sintering temperature to 1230 °C for an x value of 0.05 and to 1215 °C for x values of 0.1 and 0.2, which was caused by the formation of the liquid phase [14,25]. XRD results indicated the presence of the tetragonal BaTiO<sub>3</sub> perovskite phase (tetragonal, P4/mmm; JCPDF 75-0462) with trace amounts of second-phase Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> (monoclinic, C2/c; JCPDF 77-1566), which increased with the (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> content. This observation is in accord with those reported in the literature [25,26]. In addition, the peaks corresponding to BaTiO<sub>3</sub> in the XRD patterns shifted to lower angles due to the incorporation of the smaller ions Na<sup>+</sup> (VI: 1.02 Å) and Bi<sup>3+</sup> (VI: 1.03 Å) in the Ba<sup>2+</sup> (VI: 1.35 Å) sites. The lattice constants calculated from XRD patterns appeared to be a = 3.991 Å and c = 4.025 Å. The dielectric properties of the  $(1 - x)BaTiO_3 - x(Bi_{0.5}Na_{0.5})TiO_3$  at the maximum density, measured at a frequency of 1 KHz, are listed in Table 1. The Curie point of the  $(1 - x)BaTiO_3 - x(Bi_{0.5}Na_{0.5})TiO_3$ increased with the (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> content and was 145 °C, 170 °C, and >200 °C for x values of 0.05, 0.1, and 0.2, respectively. This increase occurred because Bi-O bonds are weaker than Ba-O bonds, the substitution of Bi<sup>3+</sup> into Ba sites leads to strengthened Ti–O bonds, and more energy is needed to maintain the balance of  $Ti^{4+}$  displacement in the TiO<sub>6</sub> octahedrons; thus,  $T_c$  increased [14,15,19]. The dielectric constant of  $(1 - x)BaTiO_3 - x(Bi_{0.5}Na_{0.5})TiO_3$  at 25 °C decreased from 2066 to 677 while the tanδ value increased from 2.9% to 3.5% as the x value increased from 0.05 to 0.2. The TCC at the cold end (-55 °C) decreased slightly from -20.8% to -24.6%, whereas the TCC at the hot end (200 °C) significantly increased from 27.0% to 641.0%, which was simply due to the increase in Tc. Because a high dielectric constant, a low tand value, and a stable TCC areneeded for X9R materials, the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic was selected as the host material to incorporate the Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> additives in this study.

Figure 1a shows the sintered densities of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with 1.0%, 1.5%, and 2.0% Ta<sub>2</sub>O<sub>5</sub> sintered at different temperatures. In all cases, the sintered density increased the sintering temperature, and the maximum densification was reached at 1215 °C. With further increases in sintering temperature, the sintered density gradually declined because of the trapped porosities associated with the fast grain growth and partly the evaporation of the volatile species such as bismuth and sodium. The weight losses of the Ta<sub>2</sub>O<sub>5</sub>-doped 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics during sintering were found to be less than 1.5 wt% through careful monitoring in this study, suggesting that the deviations of the sintered compositions from the nominal composition were only slight. The sintered density of the ceramics decreased slightly with an increase in Ta<sub>2</sub>O<sub>5</sub> content, though Ta has a high molecular weight than the substituted Ti. For instance, the maximum densities of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with 1.0%, 1.5%, and 2.0% Ta<sub>2</sub>O<sub>5</sub> were 5.82, 5.81, and 5.79 g/cm<sup>3</sup>, respectively. The results imply that the addition of  $Ta_2O_5$  inhibits the sintering of the host material. As shown in Figure 1b, the sintered densities of 0.9BaTiO<sub>3</sub>-0.1Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with 1.0%, 1.5%, and 2.0% Nb<sub>2</sub>O<sub>5</sub> dopant showed a trend similar to those with added Ta<sub>2</sub>O<sub>5</sub>. Apparently, both Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> additives slightly obstructed the densification of the 0.9BaTi O<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic.

	Sintering	Sinter I Density (s/see)	<b>T</b> <sub>-</sub> (0 <b>C</b> )	V (25.0C)		TCC		Resistivity * ( $\Omega \cdot cm$ )		
	Starting Composition	Condition (°C/h)	Sintered Density (g/cm <sup>2</sup> )	1e (°C)	K (25 °C)	tano (25 °C)	−55 °C	200 °C	25 °C	200 °C
	$0.95BaTiO_3-0.05(Bi_{0.5}Na_{0.5})TiO_3$	1230/2	5.78	145	2066	2.9%	-20.8	27.0	$4.4  imes 10^{12}$	$9.2  imes 10^{10}$
	$0.9BaTiO_3-0.1(Bi_{0.5}Na_{0.5})TiO_3$	1215/2	5.78	170	1044	3.4%	-26.2	202.0	$7.2  imes 10^{12}$	$1.4  imes 10^{11}$
	0.8BaTiO <sub>3</sub> -0.2(Bi <sub>0.5</sub> Na <sub>0.5</sub> )TiO <sub>3</sub>	1215/2	5.82	>200	677	3.5%	-24.6	641.0	$1.9  imes 10^{13}$	$5.1  imes 10^{10}$

**Table 1.** Dielectric properties of (1 - x)BaTiO<sub>3</sub>-x(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics at their maximum sintered densities.

\* Resistivity of the samples was measured at a DC voltage of 500 V.



Figure 1. Apparent densities of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of (a) Ta<sub>2</sub>O<sub>5</sub> and (b) Nb<sub>2</sub>O<sub>5</sub> sintered at different temperatures for 2 h.

Figure 2a shows the XRD patterns of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of Ta<sub>2</sub>O<sub>5</sub> sintered at 1215 °C. Only a major phase of tetragonal perovskite was found in the XRD patterns. Several small peaks at around 30° and 43° were due to the Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> phase, which was also observed in the pure 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic. The addition of Ta<sub>2</sub>O<sub>5</sub> in the host material did not trigger any additional second phase, implying that the Ta<sup>5+</sup> ions were substituted into the B-sites of the perovskite lattices and the excess Ti<sup>4+</sup> ions caused the formation of the Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> phase. It was also found that the peaks corresponding to the perovskite phase shifted slightly to lower angles because of the ionic radius of Ta5+ (VI; 0.64 Å), which is slightly larger than that of Ti<sup>4+</sup> (VI; 0.605 Å). As revealed in Figure 2b, for the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of added Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C, the XRD patterns are quite similar to those of the sintered ceramics with Ta<sub>2</sub>O<sub>5</sub>. However, another second-phase Ba<sub>6</sub>Ti<sub>14</sub>Nb<sub>2</sub>O<sub>39</sub> (orthorhombic phase, Bm21b; JCPDF 38-1431) was observed as the Nb<sub>2</sub>O<sub>5</sub> content increased to 2 mol%, in addition to the Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> phase. The calculated lattice constants of the sintered ceramics with 2.0 mol% Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> were a = 3.981 Å and c = 4.068 Å and a = 3.981 Å and c = 4.031 Å, respectively. The larger variation in the lattice constants of the Ta<sub>2</sub>O<sub>5</sub>-doped ceramic than of the Nb<sub>2</sub>O<sub>5</sub>-doped ceramic is correlated with the greater solubility of the former dopant in the perovskite structure than the latter. The limited solubility of Nb<sub>2</sub>O<sub>5</sub> led to the formation of Ba<sub>6</sub>Ti<sub>14</sub>Nb<sub>2</sub>O<sub>39</sub> precipitates.

Figure 3 shows the microstructures of the sintered surfaces and fracture surfaces of the pure 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic and those with added Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C. All the microstructures of the sintered surfaces were composed of two types of grains: fine equiaxed grains and some enlarge rod-like grains. EDS analysis identified the former as the perovskite phase BaTiO<sub>3</sub>-(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> and the latter as the Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> phase. The microstructures of the fracture surfaces indicated that all fractures were transgranular mode and revealed the presence of very little porosity. Figure 3a,b show the sintered surface and fracture surface of the pure 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic sintered at 1215 °C, which included fine equiaxed grains with an average grain size of 0.57 µm and rod-like grains with a size of approximately 3.2 µm, similar to those reported in the literature [14]. The microstructures of the sintered surfaces and fracture surfaces of the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with 1.0 and 2.0 mol% Ta<sub>2</sub>O<sub>5</sub> sintered at 1215 °C are shown in Figure 3c,d, and in Figure 3e,f, respectively. Compared with the average grain size of the pure 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic, the average grain size decreased slightly with the addition of Ta<sub>2</sub>O<sub>5</sub>, while the number of intergranular pores appeared to increase. The average grain size of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> with 1.0% and 2.0% Ta<sub>2</sub>O<sub>5</sub> was determined to be 0.55 and 0.46 µm, respectively. The Ta<sub>2</sub>O<sub>5</sub> dopant caused a reduction in the grain boundary mobility, which led to a decrease in grain size and an increase in trapped pores. The microstructures of the sintered surfaces of the 0.9BaTiO<sub>3</sub>-0.1(Bi0.5Na0.5)TiO<sub>3</sub> ceramics with 1.0 mol% and 2.0 mol% Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C are shown in Figure 3g,h, respectively; the equiaxed grains had an average size of 0.51 and 0.50  $\mu$ m, respectively. The Nb<sub>2</sub>O<sub>5</sub>-doped ceramics appeared to have greater porosity than the Ta<sub>2</sub>O<sub>5</sub>-doped ceramics. According to Figure 3i,j, which present the microstructures of the fracture surfaces, many fine second-phase precipitates of approximately 50 nm in size were distributed in the microstructure, which was determined to be Ba<sub>6</sub>Ti<sub>14</sub>Nb<sub>2</sub>O<sub>39</sub> based on the XRD results shown in Figure 2b. The fine second-phase precipitates suppressed the grain growth because of the pinning effect, which dragged the grain boundaries during sintering.

Overall, it can be concluded that the solubility of Ta<sub>2</sub>O<sub>5</sub> dopant is greater than that of Nb<sub>2</sub>O<sub>5</sub> dopant in the 0.9BaTiO<sub>3</sub>--0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic. The addition of Ta<sub>2</sub>O<sub>5</sub> resulted in the lattice expansion of 0.9BaTiO<sub>3</sub>--0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic and reduced the grain boundary mobility and thus slightly inhibited the densification. The average grain size slightly decreased and the number of trapped pores increased with increase of Ta<sub>2</sub>O<sub>5</sub> content. In addition to the pre-existing Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub> phase, the escalation in addition of Nb<sub>2</sub>O<sub>5</sub> in 0.9BaTiO<sub>3</sub>--0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic triggered the formation of Ba<sub>6</sub>Ti<sub>14</sub>Nb<sub>2</sub>O<sub>39</sub> precipitates with increased quantity. The distributed second phase suppressed the sintered density and generated porosity in 0.9BaTiO<sub>3</sub>--0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic.



Figure 2. Cont.



**Figure 2.** XRD patterns of  $0.9BaTiO_3-0.1(Bi_{0.5}Na_{0.5})TiO_3$  ceramics with various amounts of (a) Ta<sub>2</sub>O<sub>5</sub> and (b) Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C for 2 h.



Figure 3. Cont.



**Figure 3.** SEM micrographs of the sintered surfaces (**a**), (**c**), (**e**). (**g**), and (**i**) and the fractured surfaces (**b**), (**d**), (**f**), (**h**), and (**j**) of  $0.9BaTiO_3-0.1$  (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic sintered at 1215 °C for 2 h. (**a**) and (**b**) are pure  $0.9BaTiO_3-0.1(Bi_{0.5}Na_{0.5})TiO_3$  ceramic; (**c**) and (**d**) correspond to the ceramic with 1.0 mol% Ta<sub>2</sub>O<sub>5</sub>; (**e**) and (**f**) correspond to the ceramic with 2.0 mol% Ta<sub>2</sub>O<sub>5</sub>; (**g**) and (**h**) correspond to the ceramic with 1.0 mol% Nb<sub>2</sub>O<sub>5</sub>; and (**i**) and (**j**) correspond to the ceramic with 2.0 mol% Nb<sub>2</sub>O<sub>5</sub>.

Figures 4 and 5 show the TCCs and dielectric losses *versus* temperature, measured at a frequency of 1 kHz, of 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of added Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C, respectively. Some of the dielectric properties, including the Curie point, dielectric constant, and tanδ value at 25 °C and TCCs at both 25 °C and 200 °C, are also listed in Table 2. The temperature dependence of the 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics became flattened with

an increase in the dopant contents of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub>. The TCC curves near the Curie point were dramatically suppressed, and a broad and diffuse phase transition occurred in all samples, leading to smooth dielectric temperature characteristics over a relatively broad temperature range (-55 to 200 °C). The temperature corresponding to the maximum of the diffuse phase transition ( $T_m$ ) increased with the Ta<sub>2</sub>O<sub>5</sub> content, but there was no visible change with an increase in the Nb<sub>2</sub>O<sub>5</sub> content. For instance,  $T_m$  corresponding to the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with 1.0, 1.5, and 2.0 mol% Ta<sub>2</sub>O<sub>5</sub> content appeared to be around 145 °C, 150 °C, and 180 °C, respectively, while those with 1.0, 1.5 and 2.0 mol% Nb<sub>2</sub>O<sub>5</sub> had a similar  $T_m$  of approximately 140 °C. It is apparent that the Ta<sub>2</sub>O<sub>5</sub> content can effectively elevate the  $T_m$  of the sintered ceramics.



**Figure 4.** (a) TCCs and (b) dielectric losses *versus* temperature of  $0.9BaTiO_3-0.1(Bi_{0.5}Na_{0.5})TiO_3$  ceramics with various amounts of Ta<sub>2</sub>O<sub>5</sub> sintered at 1215 °C (measured frequency = 1 kHz; heating rate = 2 °C/min; applied voltage = 1 V).



**Figure 5.** (a) Temperature coefficients of capacitance and (b) dielectric losses *versus* temperature of  $0.9BaTiO_3-0.1(Bi_{0.5}Na_{0.5})TiO_3$  ceramics with addition of various amounts of Nb<sub>2</sub>O<sub>5</sub> sintered at 1215 °C (measured frequency = 1 kHz; heating rate = 2 °C/min; applied voltage = 1 V).

A .1.1°4°		Sintered Density (g/cm <sup>3</sup> )	<i>T</i> <sub>m</sub> (°C)	K (25 °C)	tanð (25 °C) –	TCC		Resistivity * (Ω·cm)	
Additives	Sintering Condition (°C/ n)					−55 °C	200 °C	25 °C	200 °C
1.0 mol% Ta <sub>2</sub> O <sub>5</sub>	1215/2	5.82	145	1788	3.6%	-38.8	2.1	$2.6  imes 10^{12}$	$1.3  imes 10^{11}$
1.5 mol% Ta <sub>2</sub> O <sub>5</sub>	1215/2	5.81	150	1800	2.3%	-16.1	-6.4	$3.5  imes 10^{12}$	$8.2\times10^{10}$
2.0 mol% Ta <sub>2</sub> O <sub>5</sub>	1215/2	5.79	180	1386	1.8%	-1.3	5.2	$2.8  imes 10^{12}$	$1.5  imes 10^{10}$
1.0 mol% Nb <sub>2</sub> O <sub>5</sub>	1215/2	5.79	140	1471	3.9%	-37.7	15.7	$9.0\times10^{12}$	$9.1\times10^{10}$
1.5 mol% Nb <sub>2</sub> O <sub>5</sub>	1215/2	5.77	140	1790	3.1%	-18.4	-16.5	$6.2  imes 10^{12}$	$5.5\times10^{10}$
2.0 mol% Nb <sub>2</sub> O <sub>5</sub>	1215/2	5.74	140	1865	1.4%	-9.4	-19.8	$7.8  imes 10^{12}$	$5.1\times10^{10}$

Table 2. Dielectric properties of 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic with various amounts of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> at their maximum sintered densities.

\* Resistivity of the samples was measured at a DC voltage of 500 V.

The TCCs of the 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic with 1.0 mol% Ta<sub>2</sub>O<sub>5</sub> at -55 and 200 °C were -38.8% and 2.1%, respectively, which failed to meet the EIA X9R specifications because of the large variation of capacitance between the hot and cold ends (See Figures 4a and 5a). As the Ta<sub>2</sub>O<sub>5</sub> dopant content in the sintered ceramic increased, the capacitance variation at both -55 °C and 200 °C became flattened. The TCCs of the sintered ceramics became X9R-qualified with 2.0 mol% Ta<sub>2</sub>O<sub>5</sub>. The TCCs at -55 °C and 200 °C for the sintered ceramics with 1.5 and 2.0 mol% Ta<sub>2</sub>O<sub>5</sub> were, respectively, -16.1% and -6.4% and -1.3% and 5.2%. The dielectric constant of the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic tended to decrease with the Ta<sub>2</sub>O<sub>5</sub> content and appeared to be 1788, 1800, and 1386 for 1.0, 1.5 and 2.0 mol% Ta<sub>2</sub>O<sub>5</sub>, respectively. However, the dielectric constant increased from 1471 to 1790 and 1865 as the Nb<sub>2</sub>O<sub>5</sub> content rose from 1.0 to 1.5 and 2.0 mol%, respectively. Apparently, the dielectric constant increased with the addition of Nb<sub>2</sub>O<sub>5</sub> dopant, even though the differences in their microstructures, such as grain size and porosity, seemed to be trivial.

Capacitors for high-temperature applications must possess a low dielectric loss at both room and high temperatures to prevent overheating and failure. As shown in Figures 4b and 5b, all samples revealed a similar trend in their dielectric-loss curves. The dielectric losses of the sintered 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> continued to drop with the measurement temperature. For instance, the dielectric loss of the sintered ceramic with 2.0 mol% Ta<sub>2</sub>O<sub>5</sub> was 3.3%, 1.8%, and 1.7% at temperatures of -55, 25, and 200 °C, respectively. The dielectric loss did not increase at high temperature, unlike that of most ferroelectric ceramics, where the dielectric loss increases because of the increase in ionic mobility or a ferroelectric-paraelectric transition at high temperatures. The dielectric losses of the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic were found to decrease significantly with the addition of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub>. For instance, it decreased from 3.6% to 1.8% and from 3.9% to 1.4% at 25 °C when the Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> dopant content increased from 1.0 to 2.0 mol%, respectively, which is close to the values for common X7R and X8R ceramics with a similar thickness. The electrical resistivities of the 0.9BaTiO<sub>3</sub>-0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with added Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> obtained at 25 °C and 200 °C are also listed in Table 1. All samples appeared to be good insulators, with resistivities ranging from  $2.6 \times 10^{12} \Omega \cdot cm$  to  $9.0 \times 10^{12} \Omega \cdot cm$  at 25 °C and from  $1.5 \times 10^{10} \Omega$  cm to  $1.3 \times 10^{11} \Omega$  cm at 200 °C. There was a decrease of approximately two orders of magnitude in resistivity as the temperature rose from 25 °C to 200 °C.

In this study, it was found that Ta<sub>2</sub>O<sub>5</sub> dopant was more efficient to increase the temperature stability of dielectric constant of the 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic, compared with Nb<sub>2</sub>O<sub>5</sub> dopant. Particularly, the host ceramic with 2 mol% Ta<sub>2</sub>O<sub>5</sub> dopant hadthe best composition in terms of dielectric properties. The X9R formulations and their dielectric properties were highly repeatable in this study, which was not the case with the lead-based X9R ceramics reported in the literature [10].

### 4. Conclusions

BaTiO<sub>3</sub>–(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramics with various amounts of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> dopants were explored in the study to compare high-temperature stability for meeting X9R specifications. Addition of Ta<sub>2</sub>O<sub>5</sub> and Nb<sub>2</sub>O<sub>5</sub> dopants resulted in areduction in grain size of the sintered ceramics. The microstructure of the former contained fine equiaxed perovskite grains and some rod-like grains of second-phase Ba<sub>6</sub>Ti<sub>17</sub>O<sub>40</sub>, while that of the latter also included Ba<sub>6</sub>Ti<sub>14</sub>Nb<sub>2</sub>O<sub>39</sub> precipitates. The best

dielectric properties of the 0.9BaTiO<sub>3</sub>–0.1(Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> ceramic with 2 mol% Ta<sub>2</sub>O<sub>5</sub> dopant sintered at 1215 °C: a dielectric constant of 1386, a tanð of 1.8%, and TCCs of -1.3% and 1.2% and electrical resistivities of  $2.8 \times 10^{12}$  and  $1.5 \times 10^{10} \Omega \cdot cm$ , respectively, at 25 and 200 °C. The X9R formulations and their dielectric properties were shown to be highly repeatable in this study.

## **Author Contributions**

Sea-Fue Wang and Yung-Fu Hsu conceived and designed the experiments; Yu-Wen Hung and Yi-Xin Liu performed the experiments and analyzed the data; and Sea-Fue Wang wrote the paper.

## **Conflicts of Interest**

The authors declare no conflict of interest.

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