

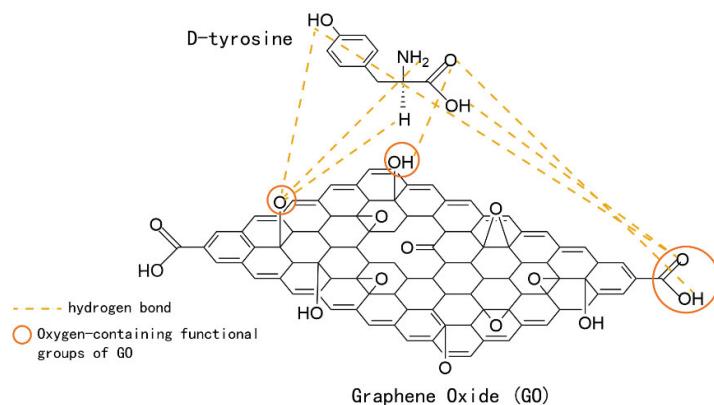
# Supplementary Information: A membrane with Strong Resistance to Organic and Biological Fouling Using Graphene Oxide and D-Tyrosine as Modifiers

Jiarui Guo <sup>1,2</sup>, Yan Zhang <sup>1,2,\*</sup>, Fenghua Chen <sup>2</sup> and Yuman Chai <sup>2</sup>

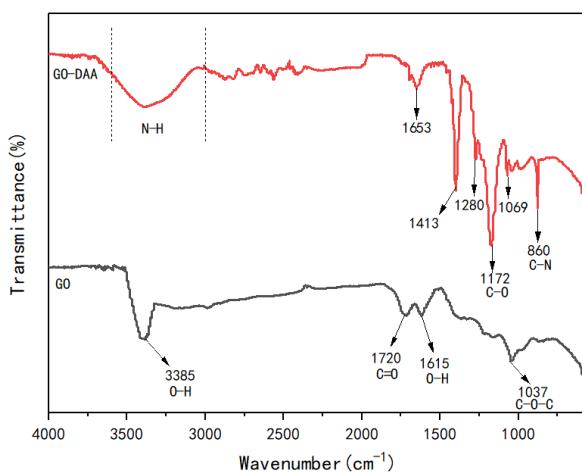
<sup>1</sup> Key Laboratory of Water Quality Science and Water Environment Recovery Engineering, Beijing University of Technology, Beijing 100124, China

<sup>2</sup> Faculty of Architecture, Civil and Transportation Engineering, Beijing University of Technology, Beijing 100124, China; guojiarui@emails.bjut.edu.cn (J.G.); chenfh@cnpe.cc (F.C.); chaiyuman@163.com (Y.C.)

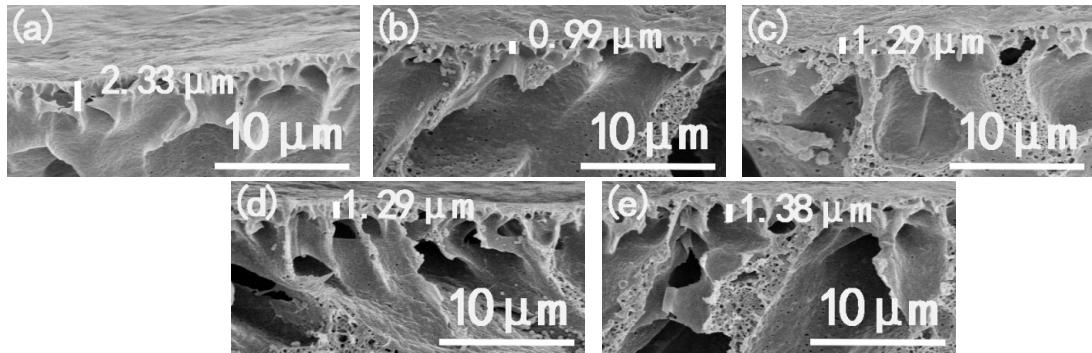
\* Correspondence: yzhang@bjut.edu.cn



**Figure S1.** Schematic diagram of the Intermolecular hydrogen bonding between GO and D-tyrosine in the membrane matrix.



**Figure S2.** FTIR spectra of GO and GO-DAA.



**Figure S3.** The surface layer thickness of the membranes. (a) PVDF; (b) P-GO; (c) P-GO-50; (d) P-GO-100; (e) P-GO-150.

**Table S1.** Components in the polymer casting solutions.

Membrane	Substance and concentration				
	DMAc (wt %)	PVP (wt %)	PVDF (wt %)	GO (wt %)	D-tyrosine (mg/L)
PVDF	80	2	18	/	/
P-GO	79.8	2	18	0.2	/
P-GO-50DAA	79.8	2	18	0.2	50
P-GO-100DAA	79.8	2	18	0.2	100
P-GO-150DAA	79.8	2	18	0.2	150

**Table S2.** Elemental composition of three membranes' surface.

Membrane	Atomic percentage (%)		
	C 1s	O 1s	N 1s
PVDF	83.05	11.11	5.84
P-GO	86.54	8.57	4.89
P-GO-DAA	84.69	9.03	6.28

**Table S3.** Carbon chemical bond composition of three membranes' surface.

Membrane	Carbon chemical bond (%)			
	C-C	C-F	C=C	C-O/C-N
PVDF	45.04	41.11	8	5.85
P-GO	51.06	30.18	11.87	6.89
P-GO-DAA	41.51	30.99	16.52	10.97

**Table S4.** Average pore size and porosity of fouled membranes.

Membrane	Average pore size (nm)	Porosity (%)
PVDF	2.9 ( $\pm 1.6$ )	16.5 ( $\pm 1.8$ )
P-GO	7.1 ( $\pm 1.8$ )	40.1 ( $\pm 2.6$ )
P-GO-50DAA	9.9 ( $\pm 2.1$ )	58.8 ( $\pm 1.8$ )
P-GO-100DAA	13.2 ( $\pm 1.7$ )	72.0 ( $\pm 2.3$ )
P-GO-150DAA	12.3 ( $\pm 1.9$ )	69.6 ( $\pm 2.1$ )

**Table S5.** Infrared absorption peaks with corresponding chemical bonds.

Absorption peaks (cm <sup>-1</sup> )	chemical bonds
860	C-N
1037	C-O-C
1320	O-H (in phenol group)
1413	N-H
1615	O-H
1720	C=O
3385	O-H (in carboxyl acid group)
3300~3500	N-H