

A Theoretical Survey of the UV-Visible of Axially and Peripherally Substituted Boron Subphthalocyanines.

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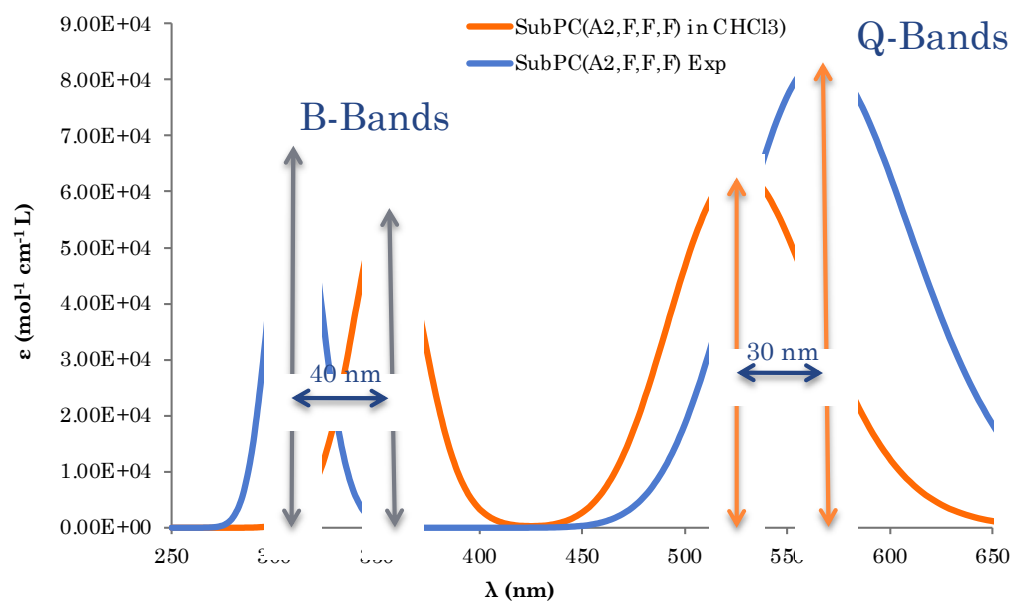


Figure S1. Comparison between the calculated (red curve) and the experimental (blue curve) UV-visible spectra for SubPc (A2,F,F,F) dissolved in CHCl₃.

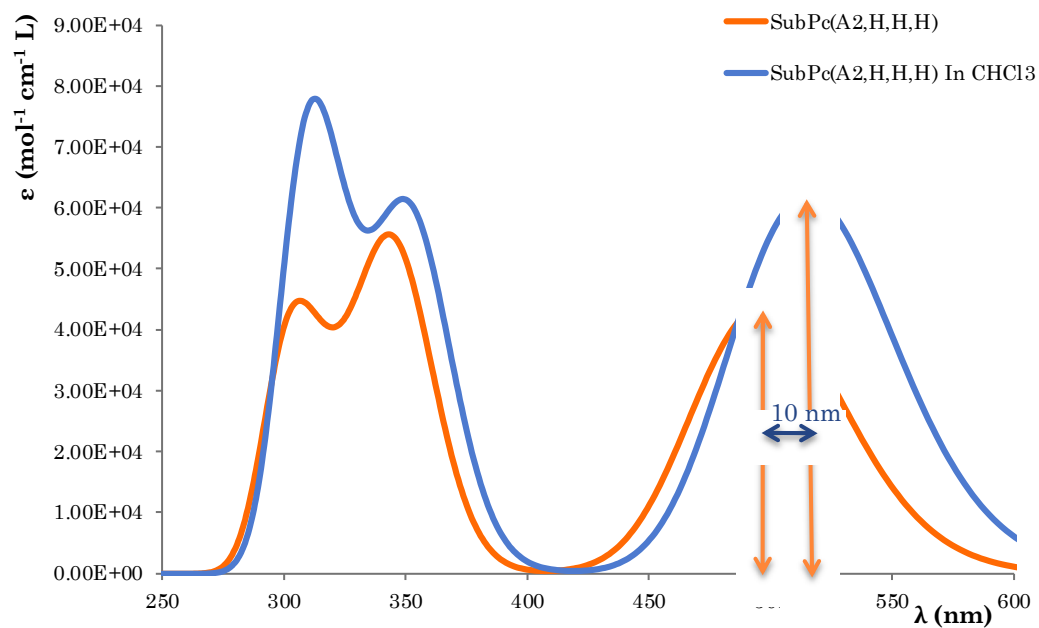


Figure S2. Comparison between the UV-visible spectrum of SubPc(A2,H,H,H) in the gas phase (red curve) and in CHCl_3 solution (blue curve).

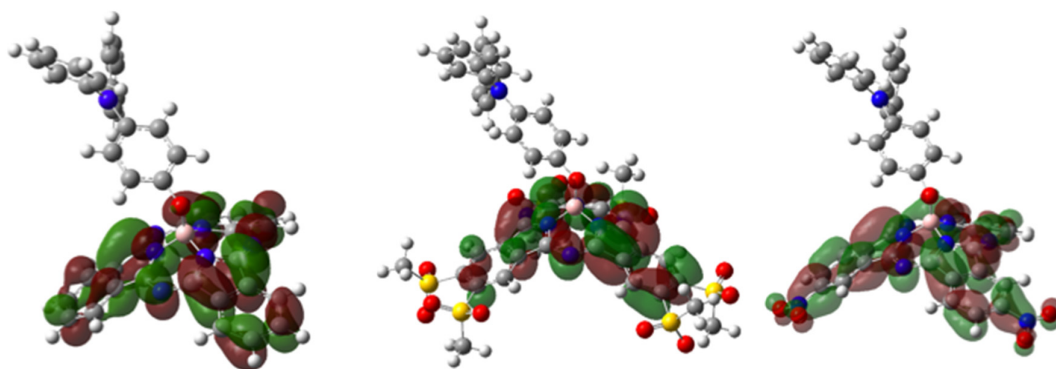
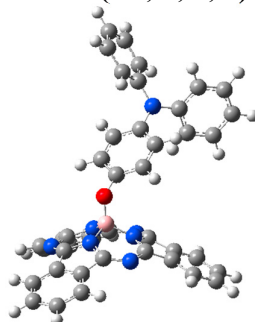


Figure S3: Lowest Unoccupied Molecular Orbital of SubPc(A1,H,H,H), SubPc(A1,SO₂CH₃,SO₂CH₃H,H) and SubPc(A1,NO₂,H,H).

Table S1. B3LYP/6-31G(d) optimized geometries (cartesian coordinates in Å) for the different SubPcs included in this study.

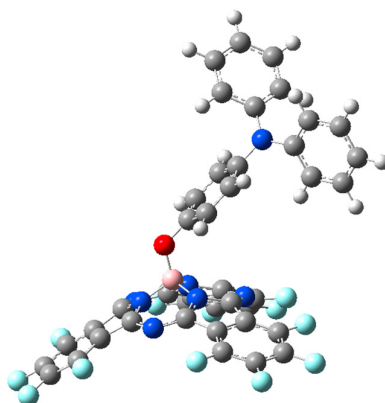
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 C -2.21146100 -3.31209400 4.70445900
 C -2.62324700 -3.34956700 3.37492400
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 N -0.86586400 0.97956700 1.93745100
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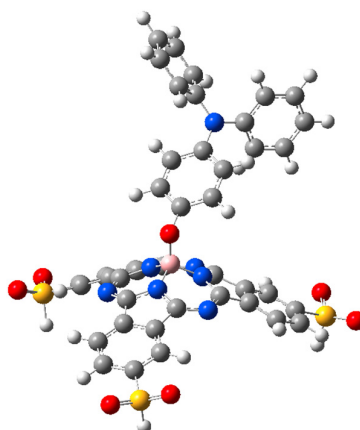
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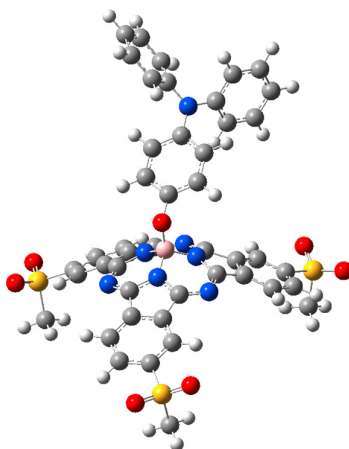
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H 8.92072 -1.51833 -1.53978
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O 2.2371 6.98949 2.92132
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SubPc(A1,SO₂CH₃,H,H)

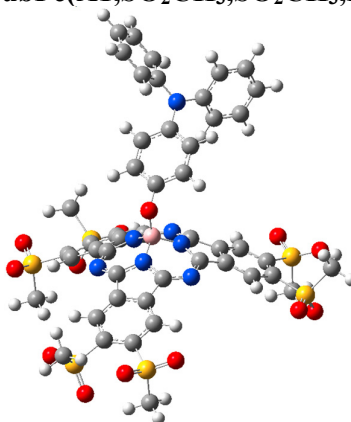


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C 1.50062 2.30333 0.09593
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O 2.08728 7.08406 2.57366
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H 0.38517 -4.41029 5.50192

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H 2.41377 5.19464 5.58544
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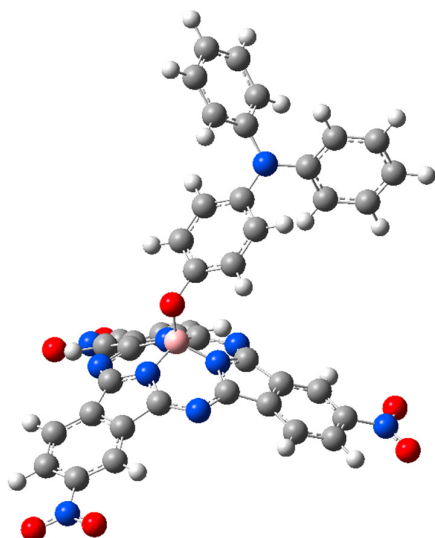


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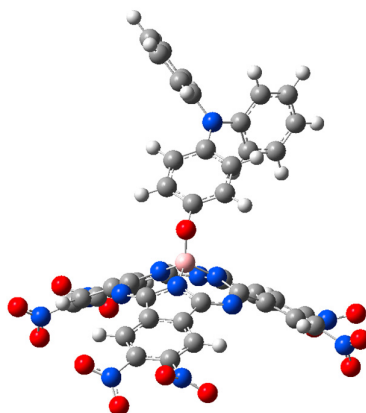
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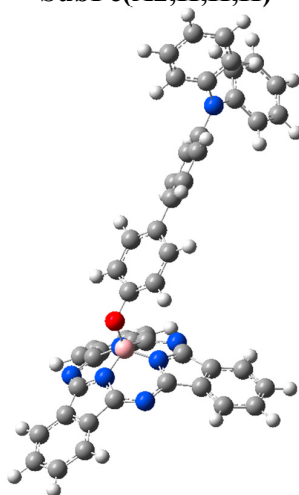


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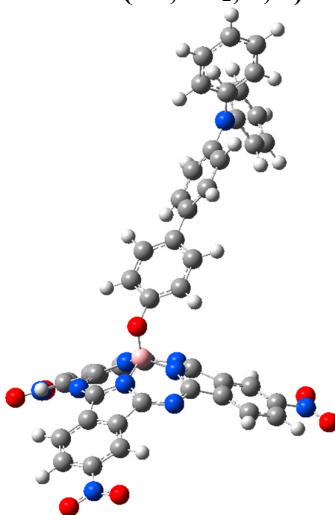


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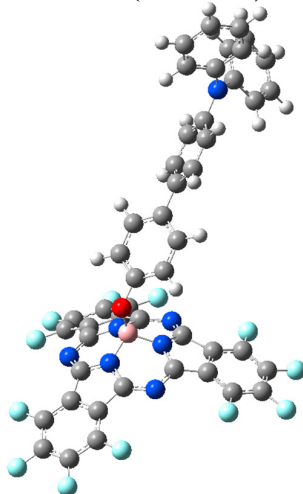


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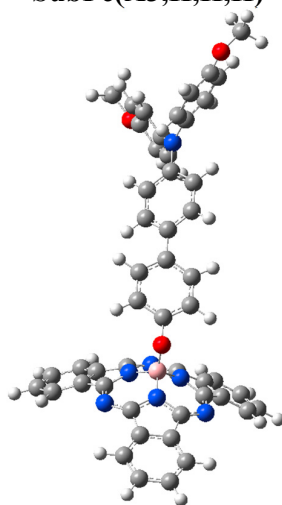


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SubPc(A3,H,H,H)

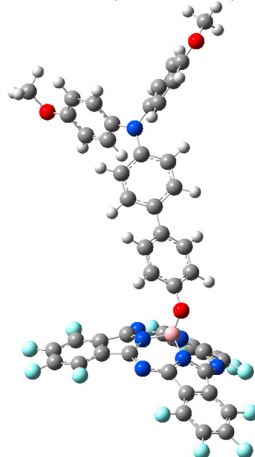


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SubPc(A3,F,F,F)



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Table S2. Q-Band (λ_{\max}) excitation energies in vacuum, oscillator strengths f and dominant electronic transitions, for the SubPcs included in this study

An	R ₁	R ₂	R ₃	λ_{\max} (eV)	nm	f	Main Configuration
A1	H	H	H	2.49	498.9	0.2856	H-1->L+0(+85%)
				2.50	496.70	0.3125	H-1->L+1(+85%)
A1	F	F	F	2.42	512.9	0.2997	H-1->L+0(+85%)
				2.43	510.4	0.3276	H-1->L+1(+85%)
A1	SO ₂ H	H	H	2.46	504.4	0.3401	H-1->L+0(+96%)
				2.47	502.0	0.3726	H-1->L+1(+95%)
A1	SO ₂ CH ₃	SO ₂ CH ₃	H	2.42	511.8	0.4257	H-1->L+0(+96%)
				2.43	509.2	0.4623	H-1->L+1(+95%)
A1	SO ₂ CH ₃	H	H	2.46	504.0	0.3456	H-1->L+0(+96%)
				2.47	501.5	0.3774	H-1->L+1(+96%)
A1	NO ₂	H	H	2.31	536.6	0.3316	H-1->L+0(+96%)
				2.32	534.1	0.3567	H-1->L+1(+96%)
A1	NO ₂	NO ₂	H	2.31	536.9	0.3058	H-1->L+0(+82%)
				2.33	532.4	0.3875	H-1->L+1(+91%)
A2	H	H	H	2.49	498.6	0.2886	H-1->L+0(+96%)
				2.5	496.8	0.3186	H-1->L+1(+95%)
A2	NO ₂	H	H	2.31	535.8	0.3331	H-1->L+0(+95%)
				2.32	534.0	0.3598	H-1->L+1(+95%)
A2	F	F	F	2.42	511.9	0.293	H-1->L+0(+92%)
				2.43	509.5	0.3247	H-1->L+1(+92%)
A3	H	H	H	2.49	498.4	0.2872	H-1->L+0(+95%)
				2.5	496.8	0.3199	H-1->L+1(+95%)
A3	F	F	F	2.42	512.1	0.3025	H-2->L+0(+95%)
				2.43	509.9	0.3378	H-2->L+1(+95%)