



Supplementary Materials:

TiO₂ Nanotube Layers Decorated with Al₂O₃/MoS₂/Al₂O₃ as Anode for Li-ion Microbatteries with Enhanced Cycling Stability

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Received: date; Accepted: date; Published: date

Keywords: TiO₂ nanotube; MoS₂; Al₂O₃; Atomic Layer Deposition; Li-ion microbatteries

Calculation to determine the porosity of TNTs

The porosity calculation is based on the amount of TiO₂ nanotubes per cm² on obtained SEM images. In average, there are 46.83 nanotubes per μm². The nanotube diameters were measured using our proprietary software “nanomeasure”. In average, the inner and outer diameter of the nanotubes was 120.95 nm and 150.54 nm, respectively. Area of the TNTs was calculated by assuming the nanotubes have a shape of a circle at the top.

$$\text{Area}_{\text{TNT}} = \pi \cdot (R^2 - r^2)$$

Where,

Area_{TNT} = Area of a single TiO₂ nanotubes

R and r = outer and inner diameter of TiO₂ nanotubes

$$A_{\text{TNT}} = 6309.41 \text{ nm}^2$$

So, the area occupied by TNTs on planar area of 1 cm² is,

$$\begin{aligned} \text{Area}_{\text{TNT}/\text{cm}^2} &= \text{Number of nanotubes per cm}^{-2} \cdot \text{Area}_{\text{TNT}} \\ &= 0.295 \end{aligned}$$

Therefore, the free area is 70.5 %.

Figure S1 shows the EDX spectrum of Al₂O₃/MoS₂/Al₂O₃-TNTs before and after 100 cycles at The presence of Mo, Al, S, Ti, and O peaks confirms the Al₂O₃/MoS₂/Al₂O₃ surface coating and the underlying self-supporting TNTs. The additional C and Cu peaks are attributed to the membrane

and the TEM sample grid. Fe, Co, and Zr peaks are from the TEM column and Si is ascribed to the SDD EDX detectors.

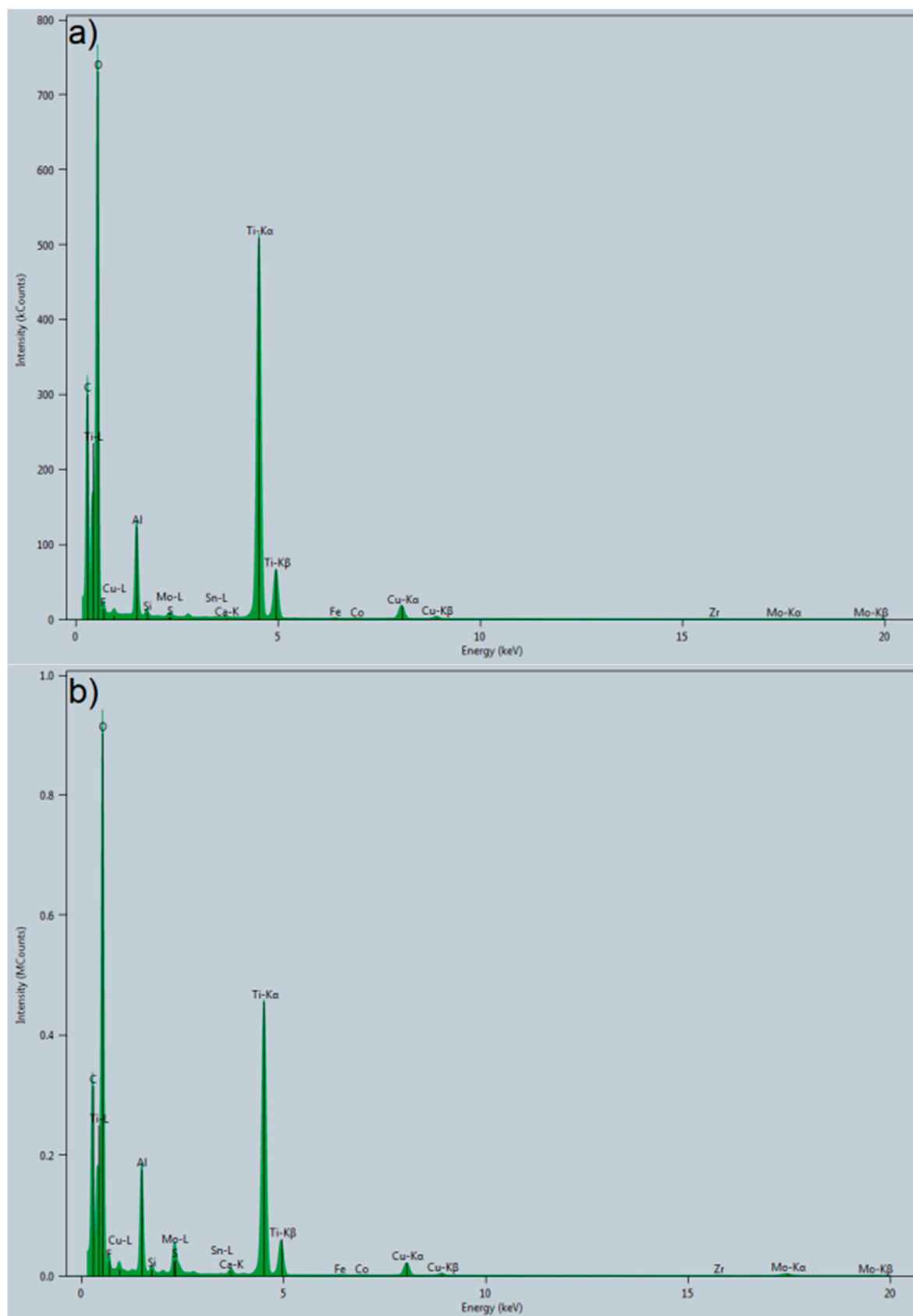


Figure S1. EDX spectrum of Al₂O₃/MoS₂/Al₂O₃-TNTs (a) before and (b) after 100 galvanostatic cycles.