

Supporting Information

Conversion of Glycerol to Value Added Products in a Semi-Continuous Batch Reactor Using Noble Metals Supported on ZSM-11 Zeolite

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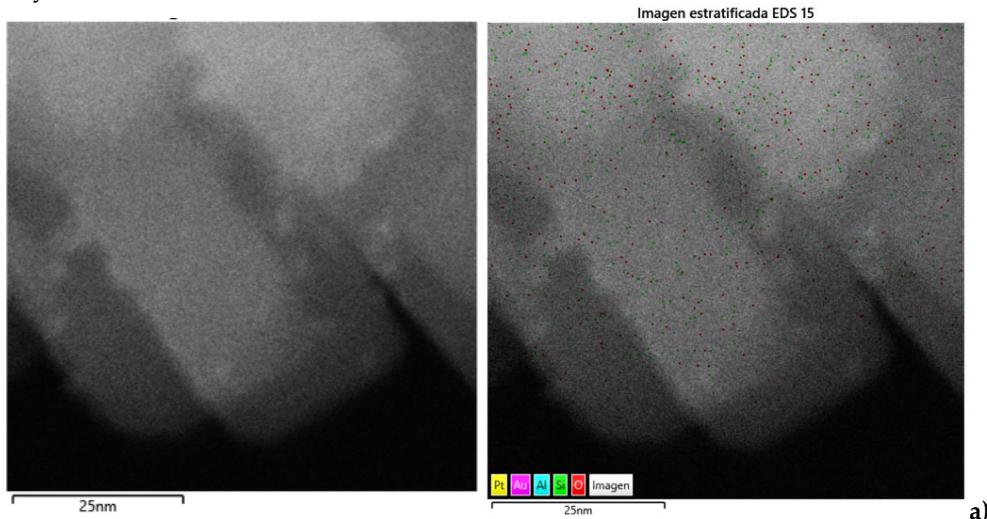
S1. TEM images, particle size distribution (a) and XEDS spectra (b) of the bimetallic catalyst: Au-Pt- ZSM-11.

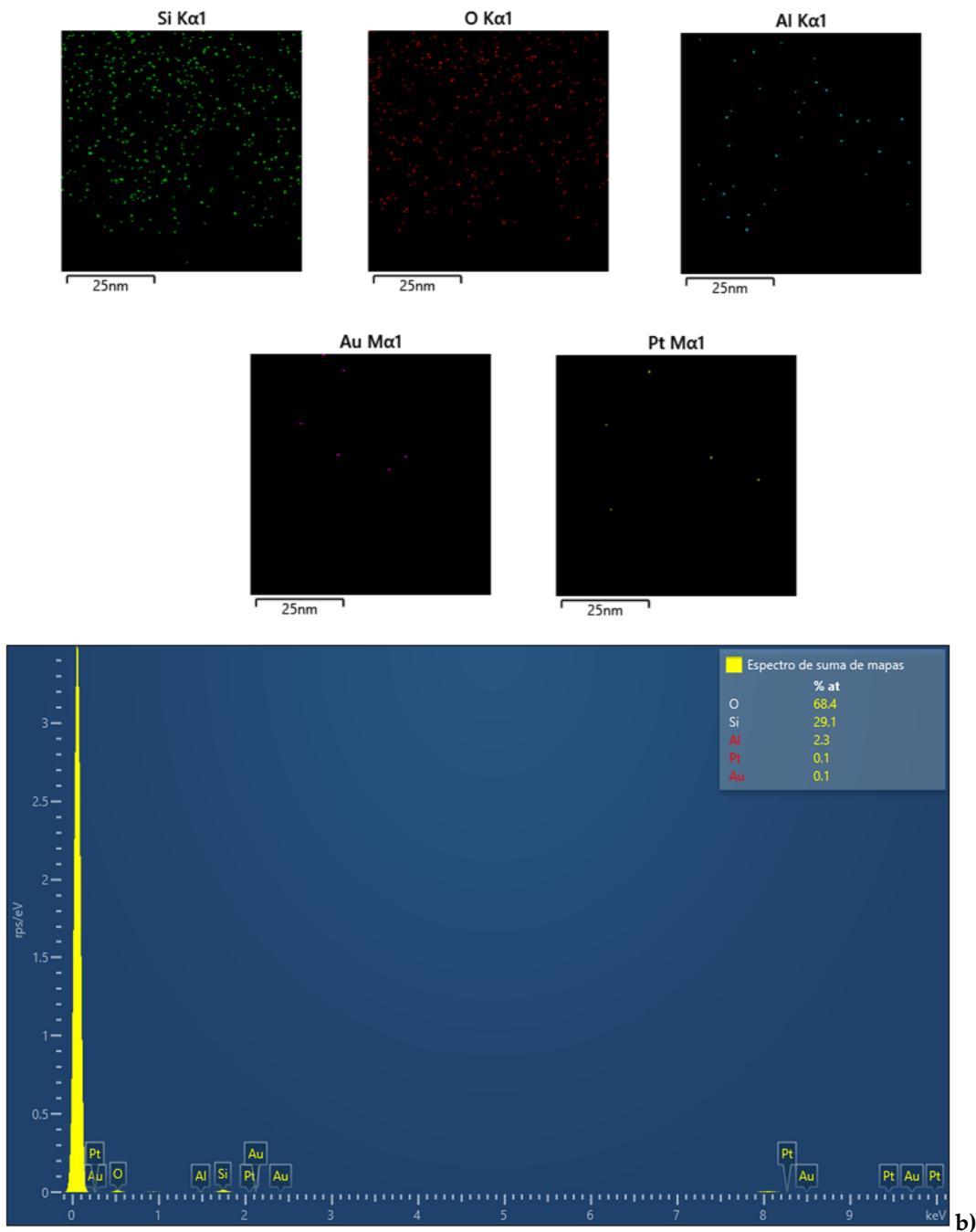
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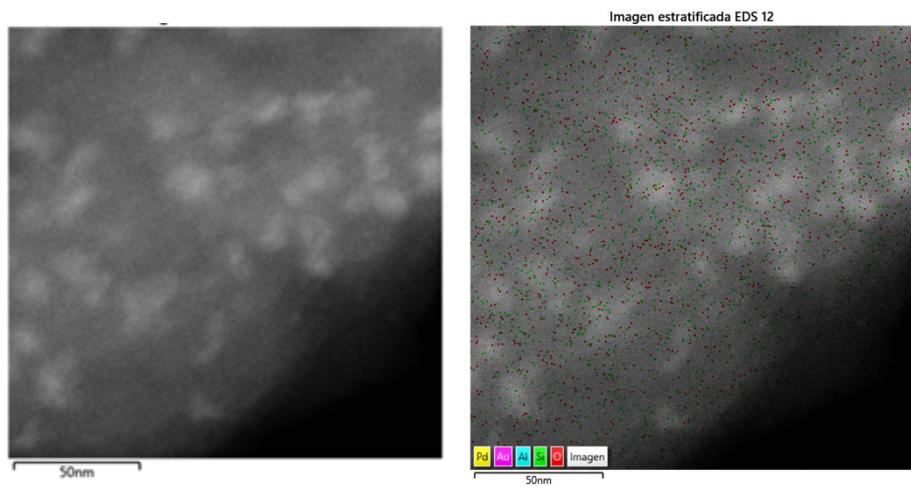
S4. GLY conversion and selectivity (mol%) towards LA, GA and DHA over different Metal/ZSM-11.

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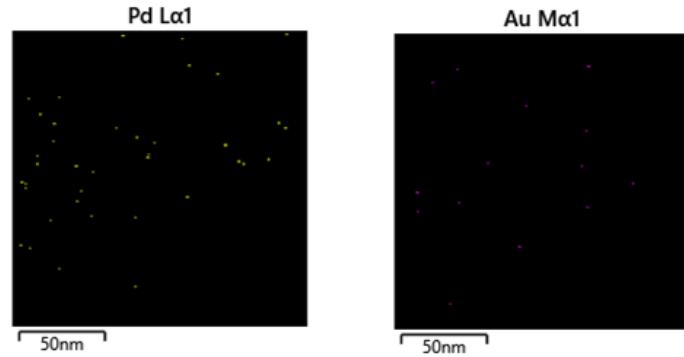
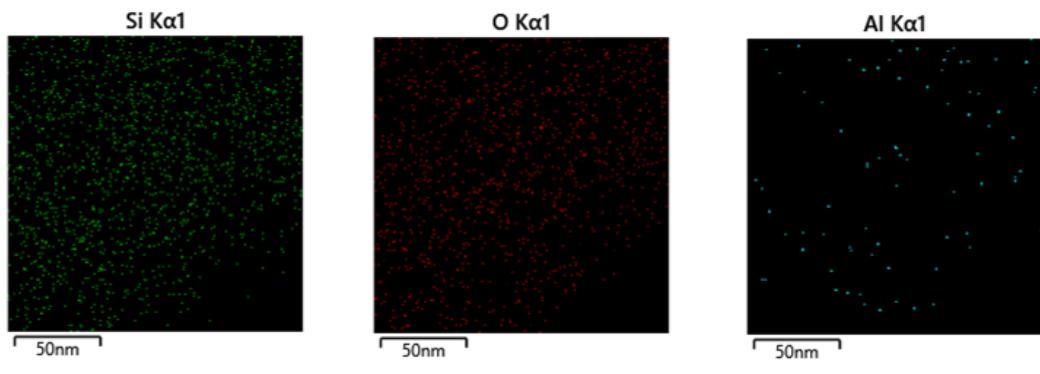


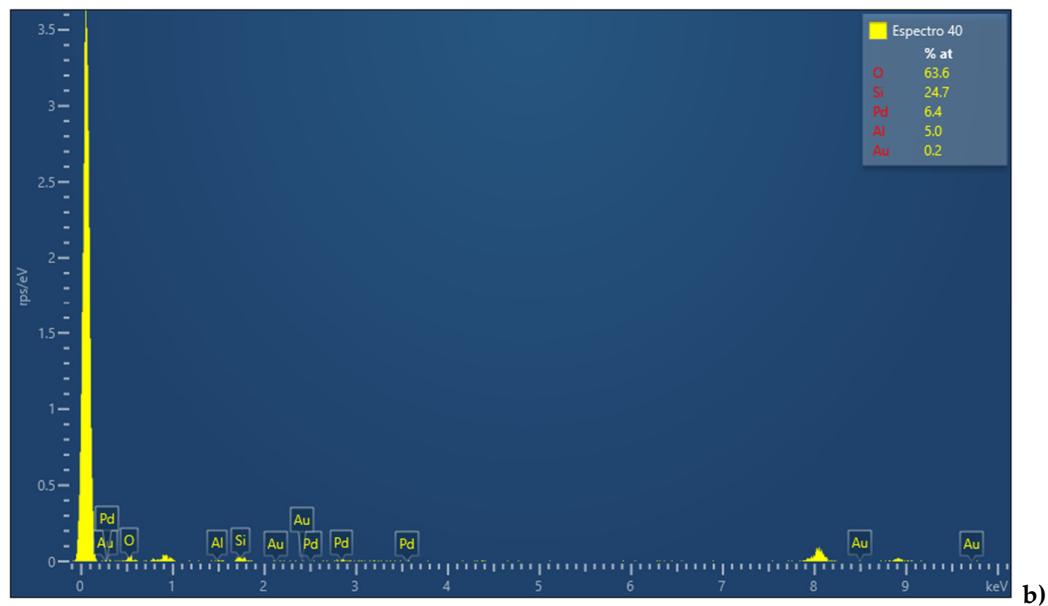


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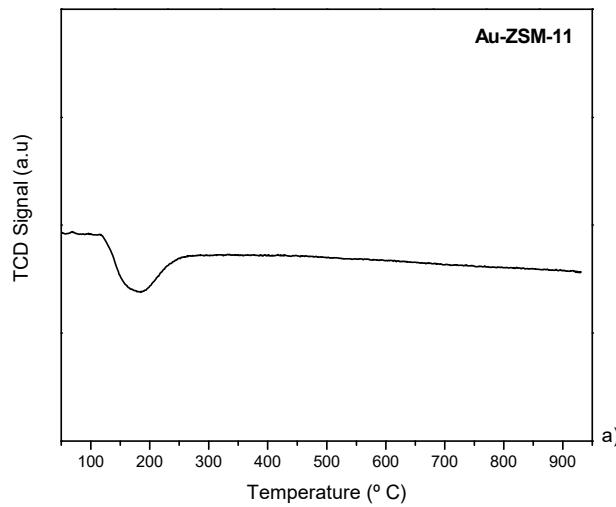


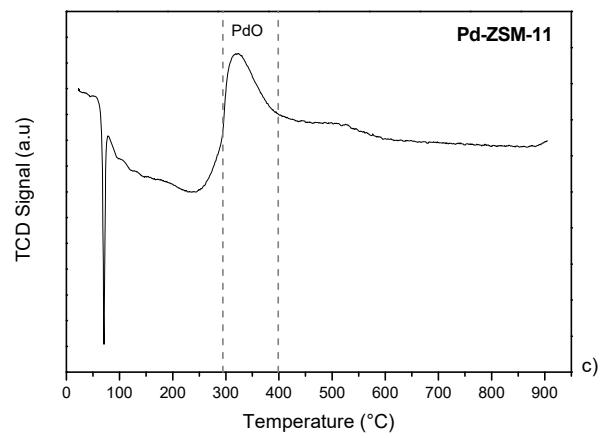
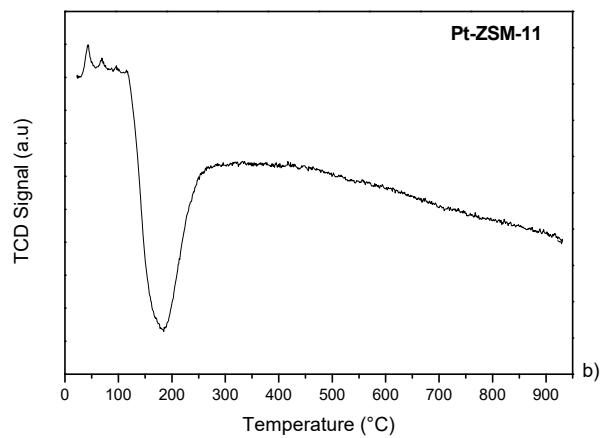
a)





S3. Temperature programmed reduction (TPR analysis) of monometallic catalysts (a) Au-ZSM-11, (b) Pt-ZSM-11, (c) Pd-ZSM-11.





S4. GLY conversion and selectivity (mol.%) towards LA, GA and DHA over different Metal/ZSM-11.

At 60 min of reaction

Catalyst	Reaction Time (min)	GLY Conversion (mol%)	Selectivity (mol%)		
			LA	GA	DHA
Pt -ZSM-11 ^a	60	51.9	16.5	-	-
Pd-ZSM-11 ^a	60	36.9	-	71.9	-
Au-ZSM-11 ^a	60	46.8	-	41.4	-
Au-Pt-ZSM-11 ^a	60	16.8	8.9	20.5	-
Au-Pd-ZSM-11 ^a	60	16.0	7.3	19.5	-
Cu-ZSM-11 ^b	60	27.1	68.0	-	19.3
Cu-ZSM-11 (T) ^b	60	24.1	51.4	-	3.1
Cr-ZSM-11 ^b	60	13.7	-	-	11.4
Cr-ZSM-11 (T) ^b	60	23.8	10.2	-	18.1

At 120 min of reaction

Catalyst	Reaction Time (min)	GLY Conversion (mol%)	Selectivity (mol%)		
			LA	GA	DHA
Pt -ZSM-11 ^a	120	56.6	25.4	-	-
Pd-ZSM-11 ^a	120	32.8	-	63.9	-
Au-ZSM-11 ^a	120	48.4	-	48.5	-
Au-Pt-ZSM-11 ^a	120	21.0	36.9	25.1	-
Au-Pd-ZSM-11 ^a	120	16.0	12.2	22.5	-
Cu-ZSM-11 ^b	120	42.5	49.0	-	25.2
Cu-ZSM-11 (T) ^b	120	31.5	41.5	-	5.0
Cr-ZSM-11 ^b	120	40.6	-	-	28.3
Cr-ZSM-11 (T) ^b	120	42.5	5.3	-	27.5

a- Reaction conditions: 35 mL sol. GLY (0.25 M), NaOH/GLY = 2 (mol/mol), GLY/metal = 400 (g/g), at 70 °C and atmospheric pressure.

b- Reaction conditions: 30 mL sol. GLY (0.5 M), H₂O₂/GLY = 2 (mol/mol), GLY/metal = 230 (g/g), at 60 °C and atmospheric pressure.

S5. GLY conversion and selectivity (mol%) towards majority products over different noble metals supports.

Catalyst	Reaction Condition	X GLY (%)	S GA (%)	S DHA (%)	Ref.
Au/C	Sol. GLY (0.3 M, 30 mL), T= 100 °C, 11 bar O ₂ , NaOH/GLY = 2 (mol/mol), t= 0.5 h	6.8	67	-	[1]
Pd/C		29	83	-	[1]
Pt/C		16	70	-	[1]
Pt/AC	Sol. GLY (0.3M, 50 mL), T= 60 °C, 150 mL/min O ₂ , t = 6 h	50	47.4	-	[2]
Pt/CNTs	Sol. GLY (0.1 M, 50 g), T= 60 °C, 150 mL/min O ₂ , 600 rpm	22	19.5	10.4	[3]
Pd-Au/C	Sol. GLY (5%w/w, 20 g), T= 80 °C, 0.3 Mpa O ₂ , t = 4 h	6.1	-	47.8	[4]
Au-Pt/MgO	Sol. GLY (0.3 M, 15 mL), T = 80 °C, 300 KPa O ₂ , t = 4 h, GLY/metal = 1000 (mol/mol)	45	40	-	[5]
Au-Pt/SiO ₂	Sol. GLY (0.3 M), T= 80 °C, 3 atm O ₂ GLY/metal = 500 mol/mol	30	61	18	[6]

X: conversion of GLY ; S: Selectivity

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