Systematic study of the behavior of different metal and metalcontaining particles under the microwave irradiation and transformation of nanoscale and microscale morphology

Evgeniy O. Pentsak^a, Vera A. Cherepanova^a, Mikhail A. Sinayskiy^b, Andrey V. Samokhin^b and Valentine P. Ananikov^a*

 ^a Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, Leninsky prospekt 47, Moscow, 119991, Russia;
 ^b Baikov Institute of Metallurgy and Material Science (IMET), Russian Academy of Sciences, Leninsky prospekt 49, Moscow 119334, Russia. E-mail: val@ioc.ac.ru; http://AnanikovLab.ru

Table of content

Table S1. Initial samples and their characteristics 5
Table S2. Effect of microwave irradiation on powders of carbides, metals and their compounds7
Table S3. Effect of microwave irradiation on mixture of graphite powder with metal-contained samples 10
Table S4. Measurement of the specific surface of powders by the BET method13
SEM and EDX investigation of changes in metal compounds under microwave treatment conditions14
SEM images of initial sample of Cu powder15
SEM images of Cu powder sample after MW treatment16
SEM images of initial sample of Cu/C powder17
SEM images of Cu/C powder sample after MW treatment18
SEM images of initial sample of Pt powder19
EDX data of initial sample of Pt powder20
SEM images of Pt powder sample after MW treatment21
EDX data of Pt powder sample after MW treatment23
SEM images of initial sample of Ag powder24
EDX data of initial sample of Ag powder25
SEM images of Ag powder sample after MW treatment26
EDX data of Ag powder sample after MW treatment28
SEM images of initial sample of Re powder29
EDX data of initial sample of Re powder
SEM images of Re powder sample after MW treatment31
EDX data of Re powder sample after MW treatment32
SEM images of initial sample of Fe/C powder33

EDX data of initial sample of Fe/C powder	34
SEM images of Fe/C powder sample after MW treatment	35
EDX data of Fe/C powder sample after MW treatment	37
SEM images of initial sample of Mo-Fe-C powder	38
EDX data of initial sample of Mo-Fe-C powder	39
SEM images of Mo-Fe-C powder sample after MW treatment	40
EDX data of Mo-Fe-C powder sample after MW treatment	42
SEM images of initial sample of Mo/C powder	43
EDX data of initial sample of Mo/C powder	44
SEM images of Mo/C powder sample after MW treatment	45
EDX data of Mo/C powder sample after MW treatment	47
SEM images of initial sample of MoS_2 powder	48
EDX data of initial sample of MoS_2 powder	49
SEM images of MoS_2 powder sample after MW treatment	50
EDX data of MoS_2 powder sample after MW treatment	52
SEM images of initial sample of WC powder	53
EDX data of initial sample of WC powder	54
SEM images of WC powder sample after MW treatment	55
EDX data of WC powder sample after MW treatment	56
SEM images of initial sample of TiC powder	57
EDX data of initial sample of TiC powder	58
SEM images of TiC powder sample after MW treatment	59
EDX data of TiC powder sample after MW treatment	60
SEM images of initial sample of W-C powder	61
EDX data of initial sample of W-C powder	62
SEM images of W-C powder sample after MW treatment	63
EDX data of W-C powder sample after MW treatment	64
SEM images of initial sample of V-C powder	65
SEM images of V-C powder sample after MW treatment	66
SEM images of initial sample of Cr-C powder	67
EDX data of initial sample of Cr-C powder	68
SEM images of Cr-C powder sample after MW treatment	69
EDX data of Cr-C powder sample after MW treatment	70
SEM images of initial sample of W-V-C powder	71
EDX data of initial sample of W-V-C powder	72
SEM images of W-V-C powder sample after MW treatment	73

EDX data of W-V-C powder sample after MW treatment.....77 Images of changes in the morphology of graphite in the presence of metal compounds under microwave SEM images of morphology of pure graphite after MW treatment78 SEM images of changes in graphite morphology in the presence of Pt after MW treatment79 SEM images of changes in graphite morphology in the presence of Re after MW treatment.......80 SEM images of changes in graphite morphology in the presence of Co after MW treatment.......82 SEM images of changes in graphite morphology in the presence of Mo-Fe-C after MW treatment90 SEM images of changes in graphite morphology in the presence of TiN after MW treatment97 EDX data of graphite powder with TiC sample after MW treatment100 SEM images of changes in graphite morphology in the presence of SiC №2 after MW treatment102 SEM images of changes in graphite morphology in the presence of MoS₂ after MW treatment103 EDX data of graphite powder with MoS₂ sample after MW treatment......104 SEM images of changes in graphite morphology in the presence of Al-B after MW treatment105 SEM images of changes in graphite morphology in the presence of W-C after MW treatment106 SEM images of changes in graphite morphology in the presence of V-C after MW treatment107 SEM images of changes in graphite morphology in the presence of Cr-C after MW treatment108 EDX data of graphite powder with Cr-C sample after MW treatment......110 SEM images of changes in graphite morphology in the presence of W-V-C after MW treatment111 SEM images of changes in graphite morphology in the presence of Al₂O₃ after MW treatment112

	SEM images of changes in graphite morphology in the presence of SiO_2 after MW treatment	114
	SEM images of changes in graphite morphology in the presence of WO_3 after MW treatment	115
	SEM images of changes in graphite morphology in the presence of ZnO after MW treatment	117
	SEM images of changes in graphite morphology in the presence of ZrO ₂ after MW treatment	119
	SEM images of changes in graphite morphology in the presence of TiO_2 after MW treatment	120
	SEM images of changes in graphite morphology in the presence of CoO after MW treatment	121
	SEM images of changes in graphite morphology in the presence of Fe_2O_3 after MW treatment	122
	SEM images of changes in graphite morphology in the presence of SnO_2 after MW treatment	123
	SEM images of changes in graphite morphology in the presence of CuO after MW treatment	124
	SEM images of changes in graphite morphology in the presence of Y_2O_3 after MW treatment	125
	SEM images of changes in graphite morphology in the presence of MgO after MW treatment	126
	SEM images of changes in graphite morphology in the presence of Cr_2O_3 after MW treatment	127
	SEM images of changes in graphite morphology in the presence of ZrO_2 -SiO ₂ after MW treatment	128
SI	EM images and the macro photographs of samples before and after MW treatment	129
Х	RD data of samples before and after MW processing	133
	XRD diffraction pattern of Ag powder sample after MW treatment	133
	XRD diffraction pattern of initial Re powder	134
	XRD diffraction pattern of Re powder sample after MW treatment	135
	XRD diffraction pattern of initial WC powder	136
	XRD diffraction pattern of WC powder sample after MW treatment	137
	XRD diffraction pattern of MoS_2 powder sample after MW treatment	138
	XRD diffraction pattern of initial W-C powder	139
	XRD diffraction pattern of W-C powder sample after MW treatment	140
	XRD diffraction pattern of W-V-C powder sample after MW treatment	141

	S _{sp} , m²/g D	_	_	Element contains, w%		
Sample		D _{BET,} nm	Raw material	[C]	[0]	[N]
				[•]	[0]	[]
Pt	5	56	$Pt(NH_3)_2(NO_2)_2$			
Re	6.1	47	NH_4ReO_4			
Ag	3.8	150	Ag			
Со	12.9	52	Co(OH) ₂			
Fe/C	5.5	139	$Fe_2O_3 + C_3H_8$	0.37		
Ni	3.8	177	NiO			
Cu	15.4	50	Cu			
Cu/C	17.8	43	Cu(CH ₃ COO) ₂	1.74		
W-Ni-Fe (90-7-3%)	4.1	80	WO ₃ , NiO, Fe ₂ O ₃			
Mo/C	18.3	32	MoO ₃	0.22		
Mo-Fe-C	12.9	52	MoO ₃ :Fe ₂ O ₃ = 1:1	2.7		
Cu-W (9:1)	6.5	98	CuO:WO ₃ =9:1		2.63	
W-Cu (1:1)	6.5	67	CuO:WO ₃ =1:1		3.44	
WC	7.5	51	WO ₃			
AIN	66.5	28	Al		14.2	25.6
AION	30	56	Al		37	7.9
TiN	24.7	45	TiH ₂		3.4	19.6
TiCN	21.5	-	TiCl ₄	9.4	5.5	9
TiC	20	61	TiCl ₄	15.2	1.7	0.13
SiC (1)	25.6	73	SiCl ₄	27.8		
SiC (2)	82.6	23	SiCl ₄	34.3		
MoS ₂	45.1	27	MoO ₃ , S	1.42		
Al-B	19.3	109	Al:B=1:2			
W-C	46	8	WO ₃	7		
V-C	89	12	V ₂ O ₅	16		
Cr-C	21.8	50	Cr	20.2		
W-V-C	21.4	19	WO ₃ :V ₂ O ₅ =10:1	5.7		
Al ₂ O ₃	69.4	22	Al			
SiO ₂	282	8	Schist			
WO ₃	4	205	WO ₃			
ZnO	20	53	ZnO			
ZrO ₂	14.9	68	Zr			
 TiO ₂	81	18	Ti			
CoO	7.5	150	Co(OH) ₂			
Fe ₂ O ₃	-	-	Fe			
SnO ₂	61.1	14	SnO ₂			
CuO	-	-	Cu			
Y ₂ O ₃	8	150	Y(CH ₃ COO) ₃			
MgO	60.5	28	Mg			

Table S1. Initial samples and their characteristics

Cr ₂ O ₃	15.5	74	Cr
ZrO ₂ -SiO ₂	67	19	ZrSiO ₄

Table S2. Effect of microwave irradiation on powders of carbides, metals and their compounds

Nia	Desitates	T	\ <i>P</i> = -1	National and the
Nº	Particles	Type of	visuai	Microscopy study
		process	changes	
1.	Pt	Type 2	Fusion	Part of the particles fused into a
				teardrop particle with a diameter of
	Micrometer particles of			500 μ m. The domain structure is
	irregular shape (1-100			visible on the particle surface, in
	microns) consisting of			some places there is a wavy
	spherical particles with a			morphology with a period of about
	diameter of 10-500 nm			20 nm
	(d -25 nm)			201111.
	(d _{av} =35 mm).			
2.	Ag	Type 2	Fusion	Fusion of the powder with
	0	- /		formation a particle of 1-2 mm in
	Micrometer particles of			size with the domain structure
	irregular shape (1-100			occurred locally. The surface is
	microns) consisting of			relatively smooth covered with
	rounded particles and			nononarticles (d =167 nm) and with
	irregular change particles			tranoparticles (u _{av} =107 mill) and with
	(100 500 and 157 and			traces of crystallization of sliver.
	$(100-500 \text{ hm}, \text{d}_{av}=157 \text{ hm})$			
	connected in short chains.			
3.	Cu	Type 2	No	No morphological changes were
0.		.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	changes	observed. d _{av} =55 nm.
	Rough micrometer particles,		0	,
	consisting of nano-sized			
	subunits $(d = 54 \text{ nm})$ of			
	irregular shape			
	inegular shape.			
4.	Re	Type 3	Color	There are no significant changes in
		<i>"</i>	change	morphology. Traces of fusion are
	A granular mass consisting of			noticeable, oblong particles,
	spherical particles with a			irregularly shaped particles.
	diameter of 2-600 nm			polyhedral particles have appeared.
	(d _w =43 nm).			At the same time the particle
				diameter remained the same
				(d -42 nm)
				$(u_{av}=43 \text{ IIII}).$
5.	Fe/C	Type 4	Color	Polycrystalline structures consisting
	-	/1 -	change	of angular polyhedral particles
	A granular mass consisting of		and	(d _{av} =427 nm). Also rounded large
	spherical particles with a		fusion	particles (ca. 500 µm) with a rough
	diameter of 5-300 nm			surface, formed as a result of fusion
	(d=127 nm)			and with traces of polyhodral
	(Sav 127 1111).			structures are present
				structures, are present.

6.	Cu/C	Type 2	No	No morphological changes were
	Rough micrometer particles consisting of irregularly shaped subunits with diameters up to 300 nm (d _{av} =60 nm).		changes	
7.	WC Rough micrometer particles consisting of angular polyhedral particles with a diameter of 0.2-1 μm, between which irregular nanoparticles are located (d _{av} =28 nm).	Type 4	Color change	Micrometer particles with a rough surface. Average diameter of subunits is 58 nm. The polyhedral particles disappeared.
8.	TiC A granular mass consisting of nano-sized irregular shaped subunits (d _{av} =63 nm).	Type 4	Color change	No morphological changes were observed. Average diameter of subunits is 61 nm.
9.	W-C Granular mass, consisting of nanoscale subunits (up to 50 nm, d _{av} =9 nm).	Туре 3	Color change	No morphological changes were observed. Average diameter of subunits is 13 nm.
10.	V-C A granular mass consisting of nano-sized irregular shaped subunits (up to 50 nm, d _{av} =25 nm).	Type 3	Compacti on	No morphological changes were observed. Average diameter of subunits is 29 nm.
11.	Cr-C A granular mass consisting of nano-sized irregular shaped subunits (d_{av} =42 nm). Also, spherical particles of regular shape with a diameter of 20-30 µm are observed.	Type 3	Color change	No morphological changes were observed. Average diameter of subunits is 40 nm.

12.	W-V-C Micrometer granular particles consisting of an irregular shaped nanoparticle (up to 20 nm, d _{av} =17 nm). Layered inclusions up to 1 micrometer in diameter and	Type 4	Color change	Micrometer particles with a diverse structure: needle crystals (width ca. 1 μ m), oblong microparticles (width ca. 300 nm), spherical microparticles (d _{av} =2.5 μ m), consisting of rounded particles up to 400 nm in diameter (d _{av} =214 nm), polyhedral particles.
	needle particles (rarely) were observed.		Color	Hilly irregularly changed particles up
13.	MO-Fe-C Granular mass consisting of rounded irregularly shaped subunits (d _{av} =50 nm).	Type 4	change and fusion	to 100 microns in diameter with melting traces. Also large millimeter particles with traces of crystallization, polyhedral structures embedded in a smooth surface are present. Agglomerates of nanoparticles are present on the surface of large particles.
14.	Mo/C Granular mass consisting of rounded particles and particles of irregular shape (d _{av} =36 nm) connected in short chains.	Туре4	Crystals growth	The crystals with several millimeters long and width 300 microns. The surface of the crystals is plate-like. Also, lamellar microcrystals are present
15.	MoS₂ A granular mass consisting of nano-sized irregular formed subunits (d _{av} =18 nm).	Type 4	Crystals growth	The crystals with several millimeters long and 300 microns wide. The surface of the crystals is plate-like. Also, lamellar microcrystals are present.

Types of process.

Type 1) very weak MW-absorption, heating up to 200 °C for Al_2O_3 , SiO_2 , WO_3 , ZnO, ZrO_2 , TiO_2 , CoO, Fe_2O_3 , SnO_2 , CuO, Y_2O_3 , MgO, Cr_2O_3 , ZrO_2 -SiO₂, Al-B, SiC, AlN, AlON, TiN, TiCN, Cu-W (9:1), W-Cu (1:1), W-Ni-Fe (90-7-3%), Ni, Co.

Type 2) weak MW-absorption or reflection of microwaves: single spark discharge.

Type 3) middle MW-absorption: red heat and/or red sparks

Type 4) intensive MW-absorption: spark discharges, glow of plasma, flame appearance with red heat.

Table S3. Effect of microwave irradiation on mixture of graphite powder with metal-contained samples

N⁰	Particles	Type of process	Microscopy study
1.	Pt	The appearance of a flame	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
2.	Re	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
3.	Ag	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
4.	Со	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
5.	Fe/C	Red heat of graphite	A slight etching of the surface of graphite. Single particles of compound on the surface of graphite.
6.	Ni	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
7.	Cu	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
8.	Cu/C	The appearance of a flame	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
9.	W-Ni-Fe	The appearance of a flame	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded and polyhedral particles.
10.	Mo/C	Red heat of graphite	Single particles of the compound and agglomerates on the graphite surface.
11.	Mo-Fe-C	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
12.	Cu-W	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
13.	W-Cu	The appearance of a flame	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
14.	WC	Red heat of graphite, rare spark discharges	Etching of the graphite surface. The graphite surface is covered with rounded and needle particles of \$10

15.	AIN	Red heat of graphite	Single particles of compound on the surface of graphite.
16.	AION	Red heat of graphite	Single particles of compound on the surface of graphite.
17.	TiN	Red heat of graphite	Etching of the graphite surface. Single particles of compound on the surface of graphite.
18.	TiCN	Red heat of graphite	Single particles of compound on the surface of graphite.
19.	TiC	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles and their agglomerates.
20.	SiC (1)	Red heat of graphite	The surface of graphite is unevenly covered agglomerates.
21.	SiC (2)	Red heat of graphite	The surface of graphite is unevenly covered agglomerates.
22.	MoS ₂	The appearance of a flame	Plate crystals on the surface of graphite.
23.	Al-B	Red heat of graphite	Single particles of compound on the surface of graphite.
24.	W-C	The appearance of a flame	Slight etching in rare areas. The surface of graphite is unevenly covered with rounded particles and their agglomerates.
25.	V-C	The appearance of a flame	Slight etching in rare areas. The surface of graphite is unevenly covered with rounded particles and their agglomerates.
26.	Cr-C	The appearance of a flame	Etching of the graphite surface. The surface of graphite is coated with particles of 10-200 nm in size.
27.	W-V-C	The appearance of a flame	Etching of the graphite surface. The graphite surface is covered with rounded and needle particles of nanometer and micrometer size.
28.	Al ₂ O ₃	Red heat of graphite	Single particles of compound on the surface of graphite.

nanometer and micrometer size.

29.	SiO ₂	Red heat of graphite	Single particles of the compound and agglomerates on the graphite surface. Part of substance is sublimated on quartz vial.
30.	WO ₃	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with rounded particles.
31.	ZnO	Red heat of graphite	Etching of the graphite surface. The graphite surface is covered with rounded and plate crystals.
32.	ZrO ₂	Red heat of graphite	Single particles of compound on the surface of graphite.
33.	TiO ₂	Red heat of graphite	The surface of graphite is unevenly covered with rounded particles and their agglomerates.
34.	CoO	Red heat of graphite	Single particles of compound on the surface of graphite.
35.	Fe ₂ O ₃	The appearance of a flame, spark discharges	Etching on individual areas of the graphite surface. Single particles of compound on the surface of graphite.
36.	SnO ₂	Red heat of graphite	The surface of graphite is unevenly covered with rounded particles.
37.	CuO	Red heat of graphite	Etching of the graphite surface. The surface of graphite is unevenly covered with polyhedral particles.
38.	Y ₂ O ₃	Red heat of graphite	Etching of the graphite surface. The surface of graphite in rare areas is unevenly covered with filament particles.
39.	MgO	Red heat of graphite	Single particles of compound on the surface of graphite.
40.	Cr ₂ O ₃	Red heat of graphite	Single particles of compound and agglomerates on the surface of graphite.
41.	ZrO ₂ -SiO ₂	Red heat of graphite	The surface of graphite is covered with rounded particles and their agglomerates.

Table S4. Measurement of the specific surface of powders by the BET method

Destidae	Specific surface area, S _{sp} , m ² /g			
Particles	Initial sample	Sample after MW treatment		
Ag	3.8	~0		
Re	6.1	0.19		
W-V-C	21.4	1.74		
MoS ₂	45.1	0.44		
W-C	46	4.77		
wc	7.5	2.31		

SEM and EDX investigation of changes in metal compounds under microwave treatment conditions

A target-oriented approach was utilized for optimization of the analytic measurements. Before measurements, the samples were placed on a 25 mm aluminum specimen stub and fixed by conductive graphite adhesive tape. Sample morphology was studied under native conditions to exclude the metal coating surface effects. The observations were carried out using a Hitachi SU8000 field-emission scanning electron microscope (FE-SEM). The images were acquired in a secondary electron mode at a 2-30 kV accelerating voltage and at working distances of 8-10 mm.

EDX studies were carried out using an Oxford Instruments X-max EDX system. The increased content of carbon and oxygen in the EDX analysis can be resulted from their presence in the carbon adhesive tape and specimen stub which used to fix samples for SEM study.

SEM images of initial sample of Cu powder



Figure S1. SEM image of initial sample of Cu powder.



Figure S2. SEM image of initial sample of Cu powder.

SEM images of Cu powder sample after MW treatment



Figure S3. SEM image of Cu powder sample after MW treatment.



Figure S4. SEM image of Cu powder sample after MW treatment.

SEM images of initial sample of Cu/C powder



Figure S5. SEM image of initial sample of Cu/C powder.



Figure S6. SEM image of initial sample of Cu/C powder.

SEM images of Cu/C powder sample after MW treatment



Figure S7. SEM image of Cu/C powder sample after MW treatment.



Figure S8. SEM image of Cu/C powder sample after MW treatment.

SEM images of initial sample of Pt powder



Figure S9. SEM image of initial sample of Pt powder.



Figure S10. SEM image of initial sample of Pt powder.

EDX data of initial sample of Pt powder



1µm





Figure S11. EDX study of initial Pt sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of platinum (d), carbon (e), and oxygen (f) distributions. The original nanopowder does not contain carbon. The presence of carbon and the increased oxygen content in the EDX analysis result by their presence in the carbon tape and stub which fixed samples for SEM.

1μm

SEM images of Pt powder sample after MW treatment



Figure S12. SEM image of Pt powder sample after MW treatment.



Figure S13. SEM image of Pt powder sample after MW treatment.



Figure S14. SEM image of Pt powder sample after MW treatment.



Figure S15. SEM image of Pt powder sample after MW treatment.

EDX data of Pt powder sample after MW treatment



25µm

25µm



Figure S16. EDX study of Pt powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of platinum (d), carbon (e), and oxygen (f) distributions.

25µm

25µm

SEM images of initial sample of Ag powder



Figure S17. SEM image of initial sample of Ag powder.



Figure S18. SEM image of initial sample of Ag powder.

EDX data of initial sample of Ag powder



Figure S19. EDX study of initial Ag sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of silver (d), carbon (e), and oxygen (f) distributions. The original nanopowder does not contain carbon. The presence of carbon and the increased oxygen content in the EDX analysis result by their presence in the carbon tape and stub which fixed samples for SEM.

SEM images of Ag powder sample after MW treatment



Figure S20. SEM image of Ag powder sample after MW treatment.



Figure S21. SEM image of Ag powder sample after MW treatment.



Figure S22. SEM image of Ag powder sample after MW treatment.



Figure S23. SEM image of Ag powder sample after MW treatment.

EDX data of Ag powder sample after MW treatment



Figure S24. EDX study of Ag powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of silver (d), carbon (e), and oxygen (f) distributions. The presence of carbon and oxygen in the EDX analysis result by their presence in the carbon tape and stub which fixed samples for SEM.

SEM images of initial sample of Re powder



Figure S25. SEM image of initial sample of Re powder.



Figure S26. SEM image of initial sample of Re powder.

EDX data of initial sample of Re powder



Figure S27. EDX study of initial Re sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of rhenium (d), carbon (e), and oxygen (f) distributions. The original nanopowder does not contain carbon. The presence of carbon and the increased oxygen content in the EDX analysis result by their presence in the carbon tape and stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

SEM images of Re powder sample after MW treatment



Figure S28. SEM image of Re powder sample after MW treatment.



Figure S29. SEM image of Re powder sample after MW treatment.

EDX data of Re powder sample after MW treatment



Re Map Sum Spectrum Element Wt% At% Re 87.0 33.7 cps/eV С 4.9 29.7 0.5 0 8.1 36.6 Re Re b Total: 100 100 С 2 6 8 keV 4 C Kα1_2 Ο Κα1 Re Ma

2.5µm

.

2.5µm

e

Figure S30. EDX study of Re powder sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of rhenium (d), carbon (e), and oxygen (f) distributions.

SEM images of initial sample of Fe/C powder



Figure S31. SEM image of initial sample of Fe/C powder.



Figure S32. SEM image of initial sample of Fe/C powder.

EDX data of initial sample of Fe/C powder

.

2.5µm





Figure S33. EDX study of initial Fe/C sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of iron (d), carbon (e), and oxygen (f) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

2.5µm

2.5µm

SEM images of Fe/C powder sample after MW treatment



Figure S34. SEM image of Fe/C powder sample after MW treatment (fusion area).



Figure S35. SEM image of Fe/C powder sample after MW treatment (fusion area).



Figure S36. SEM image of Fe/C powder sample after MW treatment (fusion area).



Figure S37. SEM image of Fe/C powder sample after MW treatment (powdered area).
EDX data of Fe/C powder sample after MW treatment



10µm





Figure S38. EDX study of Fe/C powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of iron (d), carbon (e), and oxygen (f) distributions.

SEM images of initial sample of Mo-Fe-C powder



Figure S39. SEM image of initial sample of Mo-Fe-C powder.



Figure S40. SEM image of initial sample of Mo-Fe-C powder.

EDX data of initial sample of Mo-Fe-C powder



Mo Map Sum Spectrum Element Wt% At% 2 47.9 Mo 18.7 cps/eV Fe 31.0 20.8 С 43.2 13.8 0 7.3 17.2 b⁰ С Total: 100 100 . keV 6 8 2 4 0

Mo Lα1





Fe La1_2





Figure S41. EDX study of initial Mo-Fe-C sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), iron (e), oxygen (f), and carbon (g) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

2.5µm

SEM images of Mo-Fe-C powder sample after MW treatment



Figure S42. SEM image of Mo-Fe-C powder sample after MW treatment (fusion area).



Figure S43. SEM image of Mo-Fe-C powder sample after MW treatment (fusion area).



Figure S44. SEM image of Mo-Fe-C powder sample after MW treatment (fusion area).



Figure S45. SEM image of Mo-Fe-C powder sample after MW treatment (powdered area).

EDX data of Mo-Fe-C powder sample after MW treatment







<u>О Ка</u>1

10µm

Figure S46. EDX study of Mo-Fe-C powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), iron (e), carbon (f), and oxygen (g) distributions.

SEM images of initial sample of Mo/C powder



Figure S47. SEM image of initial sample of Mo/C powder.



Figure S48. SEM image of initial sample of Mo/C powder.

EDX data of initial sample of Mo/C powder



5µm

5µm



Figure S49. EDX study of initial Mo/C sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), carbon (e), oxygen (f) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

5µm

5µm

SEM images of Mo/C powder sample after MW treatment



Figure S50. SEM image of Mo/C powder sample after MW treatment.



Figure S51. SEM image of Mo/C powder sample after MW treatment.



Figure S52. SEM image of Mo/C powder sample after MW treatment.



Figure S53. SEM image of Mo/C powder sample after MW treatment.

EDX data of Mo/C powder sample after MW treatment



Figure S54. EDX study of Mo/C powder sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), carbon (e), oxygen (f) distributions.

SEM images of initial sample of MoS_2 powder



Figure S55. SEM image of initial sample of MoS₂ powder.



Figure S56. SEM image of initial sample of MoS₂ powder.

EDX data of initial sample of MoS₂ powder





Figure S57. EDX study of initial MoS_2 sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), sulfur (e), carbon (f), and oxygen (g) distributions. The original nanopowder does not contain carbon. The presence of carbon and the increased oxygen content in the EDX analysis result by their presence in the carbon tape and stub which fixed samples for SEM.

2.5µm

SEM images of MoS_2 powder sample after MW treatment



Figure S58. SEM image of MoS₂ powder sample after MW treatment.



Figure S59. SEM image of MoS₂ powder sample after MW treatment.



Figure S60. SEM image of MoS₂ powder sample after MW treatment.



Figure S61. SEM image of MoS₂ powder sample after MW treatment.

EDX data of MoS₂ powder sample after MW treatment



2.5µm

2.5µm



Figure S62. EDX study of MoS₂ powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of molybdenum (d), carbon (e) and oxygen (f) distributions.

2.5µm

2.5µm

SEM images of initial sample of WC powder



Figure S63. SEM image of initial sample of WC powder.



Figure S64. SEM image of initial sample of WC powder.

EDX data of initial sample of WC powder



W Map Sum Spectrum Element Wt% At% 1 W 83.0 25.8 Va/eV 0.5 С 11.5 54.6 0 9.5 19.6 W W W 0 Total: 100 100 b 111 С 0 2 4 6 8 keV W Mα1 C Kα1_2 Ο Κα1

Figure S65. EDX study of initial WC sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), carbon (e), and oxygen (f) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

2.5µm

2.5µm

2.5µm

SEM images of WC powder sample after MW treatment



Figure S66. SEM image of WC powder sample after MW treatment.



Figure S67. SEM image of WC powder sample after MW treatment.

EDX data of WC powder sample after MW treatment



2.5µm



2.5µm

C

Figure S68. EDX study of WC powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), carbon (e), and oxygen (f) distributions.

2.5µm

e

2.5µm

SEM images of initial sample of TiC powder



Figure S69. SEM image of initial sample of TiC powder.



Figure S70. SEM image of initial sample of TiC powder.

EDX data of initial sample of TiC powder



Figure S71. EDX study of initial TiC sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of titanium (d), carbon (e), and oxygen (f) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

SEM images of TiC powder sample after MW treatment



Figure S72. SEM image of TiC powder sample after MW treatment.



Figure S73. SEM image of TiC powder sample after MW treatment.

EDX data of TiC powder sample after MW treatment





Figure S74. EDX study of TiC powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of titanium (d), carbon (e), and oxygen (f) distributions.

SEM images of initial sample of W-C powder



Figure S75. SEM image of initial sample of W-C powder.



Figure S76. SEM image of initial sample of W-C powder.

EDX data of initial sample of W-C powder





Figure S77. EDX study of initial W-C sample: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), carbon (e), and oxygen (f) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

SEM images of W-C powder sample after MW treatment



Figure S78. SEM image of W-C powder sample after MW treatment.



Figure S79. SEM image of W-C powder sample after MW treatment.

EDX data of W-C powder sample after MW treatment



1µm





Figure S80. EDX study of W-C powder after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), carbon (e), and oxygen (f) distributions.

SEM images of initial sample of V-C powder



Figure S81. SEM image of initial sample of V-C powder.



Figure S82. SEM image of initial sample of V-C powder.

SEM images of V-C powder sample after MW treatment



Figure S83. SEM image of V-C powder sample after MW treatment.



Figure S84. SEM image of V-C powder sample after MW treatment.

SEM images of initial sample of Cr-C powder



Figure S85. SEM image of initial sample of Cr-C powder.



Figure S86. SEM image of initial sample of Cr-C powder.

EDX data of initial sample of Cr-C powder



Figure S87. EDX study of initial sample of Cr-C powder: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of chromium (d) and carbon (e) distributions.

SEM images of Cr-C powder sample after MW treatment



Figure S88. SEM image of Cr-C powder sample after MW treatment.



Figure S89. SEM image of Cr-C powder sample after MW treatment.

EDX data of Cr-C powder sample after MW treatment



Figure S90. EDX study of Cr-C powder sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of chromium (d), carbon (e) and oxygen (f) distributions.

SEM images of initial sample of W-V-C powder



Figure S91. SEM image of initial sample of W-V-C powder.



Figure S92. SEM image of initial sample of W-V-C powder.

EDX data of initial sample of W-V-C powder



2.5µm



W Mα1





C Kα1_2 2.5µm

At%

28.1

5.2

49.5

17.2

100

2.5µm



2.5µm

Figure S93. EDX study of initial sample of W-V-C powder: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), vanadium (e), carbon (f) and oxygen (g) distributions. The increased oxygen content in the EDX analysis result by its presence in aluminum stub which fixed samples for SEM. In addition, an oxide film on the surface of the sample can be a source of the oxygen signal.

S72
SEM images of W-V-C powder sample after MW treatment



Figure S94. SEM image of W-V-C powder sample after MW treatment.



Figure S95. SEM image of W-V-C powder sample after MW treatment.



Figure S96. SEM image of W-V-C powder sample after MW treatment.



Figure S97. SEM image of W-V-C powder sample after MW treatment.



Figure S98. SEM image of W-V-C powder sample after MW treatment.



Figure S99. SEM image of W-V-C powder sample after MW treatment.



Figure S100. SEM image of W-V-C powder sample after MW treatment.



Figure S101. SEM image of W-V-C powder sample after MW treatment.

EDX data of W-V-C powder sample after MW treatment



V Kα1

W Map Sum Spectrum 0.5 oks/e/ VV W w W **b**° Т т Ō keV 8 2 4 6

Element	Wt%	At%
W	64.3	14.5
V	9.4	7.5
С	11.3	39.1
0	15.0	38.9
Total:	100	100

C Kα1_2

W Mα1



25µm

f

25µm



Figure S102. EDX study of W-V-C powder sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of tungsten (d), vanadium (e), carbon (f) and oxygen (g) distributions.

С

25µm

Images of changes in the morphology of graphite in the presence of metal compounds under microwave treatment conditions

SEM images of morphology of pure graphite after MW treatment



Figure S103. SEM image of pure graphite powder after MW treatment.



Figure S104. SEM image of pure graphite powder after MW treatment.

SEM images of changes in graphite morphology in the presence of Pt after MW treatment



Figure S105. SEM image of graphite powder with Pt sample after MW treatment.



Figure S106. SEM image of graphite powder with Pt sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Re after MW treatment



Figure S107. SEM image of graphite powder with Re sample after MW treatment.



Figure S108. SEM image of graphite powder with Re sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Ag after MW treatment



Figure S109. SEM image of graphite powder with Ag sample after MW treatment.



Figure S110. SEM image of graphite powder with Ag sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Co after MW treatment



Figure S111. SEM image of graphite powder with Co sample after MW treatment.



Figure S112. SEM image of graphite powder with Co sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Fe/C after MW treatment



Figure S113. SEM image of graphite powder with Fe/C sample after MW treatment.



Figure S114. SEM image of graphite powder with Fe/C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Ni after MW treatment



Figure S115. SEM image of graphite powder with Ni sample after MW treatment.



Figure S116. SEM image of graphite powder with Ni sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Cu after MW treatment



Figure S117. SEM image of graphite powder with Cu sample after MW treatment.



Figure S118. SEM image of graphite powder with Cu sample after MW treatment.



Figure S119. SEM image of graphite powder with Cu sample after MW treatment.



Figure S120. SEM image of graphite powder with Cu sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Cu/C after MW treatment



Figure S121. SEM image of graphite powder with Cu/C sample after MW treatment.



Figure S122. SEM image of graphite powder with Cu/C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of W-Ni-Fe after MW treatment



Figure S123. SEM image of graphite powder with W-Ni-Fe sample after MW treatment.



Figure S124. SEM image of graphite powder with W-Ni-Fe sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Mo/C after MW treatment



Figure S125. SEM image of graphite powder with Mo/C sample after MW treatment.



Figure S126. SEM image of graphite powder with Mo/C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Mo-Fe-C after MW treatment



Figure S127. SEM image of graphite powder with Mo-Fe-C sample after MW treatment.



Figure S128. SEM image of graphite powder with Mo-Fe-C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Cu-W after MW treatment



Figure S129. SEM image of graphite powder with Cu-W sample after MW treatment.



Figure S130. SEM image of graphite powder with Cu-W sample after MW treatment.

SEM images of changes in graphite morphology in the presence of W-Cu after MW treatment



Figure S131. SEM image of graphite powder with W-Cu sample after MW treatment.



Figure S132. SEM image of graphite powder with W-Cu sample after MW treatment.

SEM images of changes in graphite morphology in the presence of WC after MW treatment



Figure S133. SEM image of graphite powder with WC sample after MW treatment.



Figure S134. SEM image of graphite powder with WC sample after MW treatment.

EDX data of graphite powder with WC sample after MW treatment



5µm

5µm



Figure S135. EDX study of graphite powder with WC sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of carbon (d), tungsten (e) and oxygen (f) distributions.

5µm

5µm

SEM images of changes in graphite morphology in the presence of AlN after MW treatment



Figure S136. SEM image of graphite powder with AIN sample after MW treatment.



Figure S137. SEM image of graphite powder with AIN sample after MW treatment.

SEM images of changes in graphite morphology in the presence of AlON after MW treatment



Figure S138. SEM image of graphite powder with AION sample after MW treatment.



Figure S139. SEM image of graphite powder with AION sample after MW treatment.

SEM images of changes in graphite morphology in the presence of TiN after MW treatment



Figure S140. SEM image of graphite powder with TiN sample after MW treatment.



Figure S141. SEM image of graphite powder with TiN sample after MW treatment.

SEM images of changes in graphite morphology in the presence of TiCN after MW treatment



Figure S142. SEM image of graphite powder with TiCN sample after MW treatment.



Figure S143. SEM image of graphite powder with TiCN sample after MW treatment.

SEM images of changes in graphite morphology in the presence of TiC after MW treatment



Figure S144. SEM image of graphite powder with TiC sample after MW treatment.



Figure S145. SEM image of graphite powder with TiC sample after MW treatment.

EDX data of graphite powder with TiC sample after MW treatment



2.5µm

Figure S146. EDX study of graphite powder with TiC sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of carbon (d), titanium (e) and oxygen (f) distributions.

SEM images of changes in graphite morphology in the presence of SiC Nº1 after MW treatment



Figure S147. SEM image of graphite powder with SiC №1 sample after MW treatment.



Figure S148. SEM image of graphite powder with SiC №1 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of SiC Nº2 after MW treatment



Figure S149. SEM image of graphite powder with SiC №2 sample after MW treatment.



Figure S150. SEM image of graphite powder with SiC №2 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $\ensuremath{\mathsf{MoS}_2}$ after MW treatment



Figure S151. SEM image of graphite powder with MoS₂ sample after MW treatment.



Figure S152. SEM image of graphite powder with MoS₂ sample after MW treatment.

EDX data of graphite powder with MoS₂ sample after MW treatment



25µm



25µm

Figure S153. EDX study of graphite powder with MoS₂ sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of carbon (d), molybdenum (e) and oxygen (f) distributions.

SEM images of changes in graphite morphology in the presence of Al-B after MW treatment



Figure S154. SEM image of graphite powder with Al-B sample after MW treatment.



Figure S155. SEM image of graphite powder with Al-B sample after MW treatment.

SEM images of changes in graphite morphology in the presence of W-C after MW treatment



Figure S156. SEM image of graphite powder with W-C sample after MW treatment.



Figure S157. SEM image of graphite powder with W-C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of V-C after MW treatment



Figure S158. SEM image of graphite powder with V-C sample after MW treatment.



Figure S159. SEM image of graphite powder with V-C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Cr-C after MW treatment



Figure S160. SEM image of graphite powder with Cr-C sample after MW treatment.



Figure S161. SEM image of graphite powder with Cr-C sample after MW treatment.


Figure S162. SEM image of graphite powder with Cr-C sample after MW treatment.



Figure S163. SEM image of graphite powder with Cr-C sample after MW treatment.

EDX data of graphite powder with Cr-C sample after MW treatment



Figure S164. EDX study of graphite powder with Cr-C sample after MW treatment: SEM image (a); EDX spectrum of this area (b); element composition (c) and maps of carbon (d), chromium (e) and oxygen (f) distributions.

SEM images of changes in graphite morphology in the presence of W-V-C after MW treatment



Figure S165. SEM image of graphite powder with W-V-C sample after MW treatment.



Figure S166. SEM image of graphite powder with W-V-C sample after MW treatment.

SEM images of changes in graphite morphology in the presence of Al_2O_3 after MW treatment



Figure S167. SEM image of graphite powder with Al_2O_3 sample after MW treatment.



Figure S168. SEM image of graphite powder with Al_2O_3 sample after MW treatment.



Figure S169. SEM image of graphite powder with AI_2O_3 sample after MW treatment.



Figure S170. SEM image of graphite powder with Al_2O_3 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $\ensuremath{\text{SiO}_2}$ after MW treatment



Figure S171. SEM image of graphite powder with SiO₂ sample after MW treatment.



Figure S172. SEM image of graphite powder with SiO_2 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of WO_3 after MW treatment



Figure S173. SEM image of graphite powder with WO₃ sample after MW treatment.



Figure S174. SEM image of graphite powder with WO₃ sample after MW treatment.



Figure S175. SEM image of graphite powder with WO₃ sample after MW treatment.



Figure S176. SEM image of graphite powder with WO₃ sample after MW treatment.

SEM images of changes in graphite morphology in the presence of ZnO after MW treatment



Figure S177. SEM image of graphite powder with ZnO sample after MW treatment.



Figure S178. SEM image of graphite powder with ZnO sample after MW treatment.



Figure S179. SEM image of graphite powder with ZnO sample after MW treatment.



Figure S180. SEM image of graphite powder with ZnO sample after MW treatment (sublimated particles).

SEM images of changes in graphite morphology in the presence of \mbox{ZrO}_2 after MW treatment



Figure S181. SEM image of graphite powder with ZrO₂ sample after MW treatment.



Figure S182. SEM image of graphite powder with ZrO₂ sample after MW treatment.

SEM images of changes in graphite morphology in the presence of TiO_2 after MW treatment



Figure S183. SEM image of graphite powder with TiO₂ sample after MW treatment.



Figure S184. SEM image of graphite powder with TiO₂ sample after MW treatment.

SEM images of changes in graphite morphology in the presence of CoO after MW treatment



Figure S185. SEM image of graphite powder with CoO sample after MW treatment.



Figure S186. SEM image of graphite powder with CoO sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $\ensuremath{\mathsf{Fe}}_2\ensuremath{\mathsf{O}}_3$ after MW treatment



Figure S187. SEM image of graphite powder with Fe_2O_3 sample after MW treatment.



Figure S188. SEM image of graphite powder with Fe_2O_3 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of \mbox{SnO}_2 after MW treatment



Figure S189. SEM image of graphite powder with SnO₂ sample after MW treatment.



Figure S190. SEM image of graphite powder with SnO₂ sample after MW treatment.

SEM images of changes in graphite morphology in the presence of CuO after MW treatment



Figure S191. SEM image of graphite powder with CuO sample after MW treatment.



Figure S192. SEM image of graphite powder with CuO sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $Y_2 O_3$ after MW treatment



Figure S193. SEM image of graphite powder with Y_2O_3 sample after MW treatment.



Figure S194. SEM image of graphite powder with Y_2O_3 sample after MW treatment.

SEM images of changes in graphite morphology in the presence of MgO after MW treatment



Figure S195. SEM image of graphite powder with MgO sample after MW treatment.



Figure S196. SEM image of graphite powder with MgO sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $\mbox{Cr}_2\mbox{O}_3$ after MW treatment



Figure S197. SEM image of graphite powder with Cr_2O_3 sample after MW treatment.



Figure S198. SEM image of graphite powder with Cr₂O₃ sample after MW treatment.

SEM images of changes in graphite morphology in the presence of $\text{ZrO}_2\text{-}\text{SiO}_2$ after MW treatment



Figure S199. SEM image of graphite powder with ZrO_2 -SiO₂ sample after MW treatment.



Figure S200. SEM image of graphite powder with ZrO₂-SiO₂ sample after MW treatment.

SEM images and the macro photographs of samples before and after MW treatment



Figure S201. Initial Cu powder: SEM image (a) and macrophotography (b); Cu powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S202. Initial Pt powder: SEM image (a) and macrophotography (b); Pt powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S203. Initial Ag powder: SEM image (a) and macrophotography (b); Ag powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S204. Initial Re powder: SEM image (a) and macrophotography (b); Re powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S205. Initial Fe/C powder: SEM image (a) and macrophotography (b); Fe/C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S206. Initial Mo-Fe-C powder: SEM image (a) and macrophotography (b); Mo-Fe-C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S207. Initial Mo/C powder: SEM image (a) and macrophotography (b); Mo/C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S208. Initial MoS_2 powder: SEM image (a) and macrophotography (b); MoS_2 powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S209. Initial WC powder: SEM image (a) and macrophotography (b); WC powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S210. Initial TiC powder: SEM image (a) and macrophotography (b); TiC powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S211. Initial W-C powder: SEM image (a) and macrophotography (b); W-C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S212. Initial V-C powder: SEM image (a) and macrophotography (b); V-C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S213. Initial Cr-C powder: SEM image (a) and macrophotography (b); Cr-C powder after MW-treatment: SEM image (c) and macrophotography (d).



Figure S214. Initial W-V-C powder: SEM image (a) and macrophotography (b); W-V-C powder after MW-treatment: SEM image (c) and macrophotography (d).

XRD data of samples before and after MW processing

XRD diffraction pattern of Ag powder sample after MW treatment



Figure S215. XRD diffraction pattern of Ag powder sample after MW treatment.

XRD diffraction pattern of initial Re powder



Figure S216. XRD diffraction pattern of initial Re powder.

XRD diffraction pattern of Re powder sample after MW treatment



Figure S217. XRD diffraction pattern of Re powder sample after MW treatment.

XRD diffraction pattern of initial WC powder



Qualitative analysis results

Figure S218. XRD diffraction pattern of initial WC powder.

XRD diffraction pattern of WC powder sample after MW treatment

Qualitative analysis results			
Phase name	Formula	Figure of merit	Phase reg. detail
Tungsten Oxide	WO ₃	0.3933252532370825	10830950 (ICDD)
Unnamed mineral, syn (NR)	WC	0.7970322278003963	510939 (ICDD)
Wolfram	W	0.9117691763570458	40806 (ICDD)



Figure S219. XRD diffraction pattern of WC powder sample after MW treatment with the reference diffraction patterns of WO_3 (PDF 01-083-0950) and other.

S137

XRD diffraction pattern of MoS₂ powder sample after MW treatment

Qualitative analysis results			
Phase name	Formula	Figure of merit	Phase reg. detail
Tugarinovite, syn	MoO ₂	0.5947432249246876	10781069 (ICDD)
Molybdite, syn	MoO ₃	0.763158262495746	10747911 (ICDD)
chi-Mo ₄ O ₁₁	Mo ₄ O ₁₁	1.148356889559762	50337 (ICDD)
Molybdenum	Мо	0.9209986404861873	40809 (ICDD)
Molybdenum Sulfide	Mo_2S_3	1.20307476716514	30656963 (ICDD)



Figure S220. XRD patterns of the of MoS_2 powder sample after MW treatment, with the reference diffraction patterns of MoO_2 (PDF 01-078-1069), MoO_3 (PDF 01-074-7911), Mo_4O_{11} (PDF 00-005-0337) and other.

XRD diffraction pattern of initial W-C powder

Qualitative analysis results

Phase name	Formula	Figure of merit	Phase reg. detail
Tungsten Carbide	WC _{1-x}	0.8545924599599198	201316 (ICDD)
Tungsten Carbide	W ₂ C	1.388907051635667	30653896 (ICDD)
Unnamed mineral, syn (NR)	WC	1.315216631957023	251047 (ICDD)
Wolfram	W	1.667238816853466	40806 (ICDD)



Figure S221. XRD diffraction pattern of initial W-C powder.

XRD diffraction pattern of W-C powder sample after MW treatment

Qualitative analysis results			
Phase name	Formula	Figure of merit	Phase reg. detail
Tungsten Oxide	WO ₃	0.4869597054349091	10830950 (ICDD)
Tungsten	W	1.047058038640831	10892767 (ICDD)
beta-WO ₃	WO ₃	1.034093539710211	10894480 (ICDD)



Figure S222. XRD patterns of the of W-C powder sample after MW treatment, with the reference diffraction patterns of WO_3 (PDF 01-083-0950), and other.

XRD diffraction pattern of W-V-C powder sample after MW treatment

/|||

20

20

0

100

50

100

50

100

50

0

h

40

Qualita

Quan	lalive all	aiyəiə i cə	uita										
Phase	name					Formula		Figure of	of merit		Phase	reg. de	etail
Tungst	ten Oxide					WO ₃		0.41269	0977837	8652	10830	950 (IC	DD)
Tungst	ten Oxide					WO ₃		0.72448	8063971	2449	10891	287 (IC	DD)
Tungst	ten					W		1.14343	3753013	287	10892	767 (IC	DD)
Phas	e data pa	ttern											
	65000												
	60000												
	55000												
	50000		h										
	45000												
	40000												
	35000												
	30000												
()	25000												
(cb;	20000												
sity	15000												
ten	10000		M .		M								
-	5000			Λ		14	 h wh						

Mr MM MM

~~~

80

Fungsten Oxide

80

Tungsten Oxide, W O3, 01-083-095

W O3, 01-089-128

Tungsten, W, 01-089-2767

60

60

Figure S223. XRD patterns of the of W-V-C powder sample after MW treatment, with the reference diffraction patterns of WO<sub>3</sub> (PDF 01-083-0950), WO<sub>3</sub> (PDF 01-089-1287) and W (PDF 01-089-2767).

2-theta (deg)

40

#### S141