

Electronic Supporting Material (ESM)

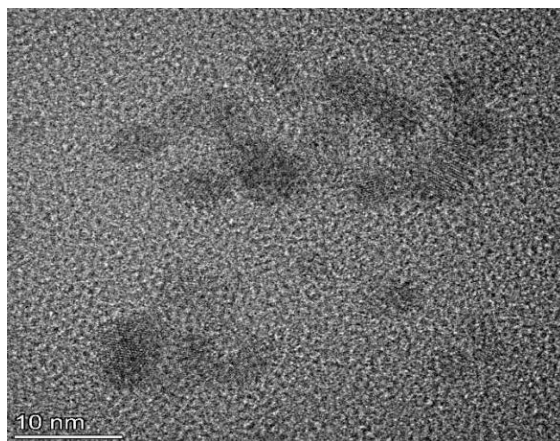


Figure. S1 (A) TEM image,

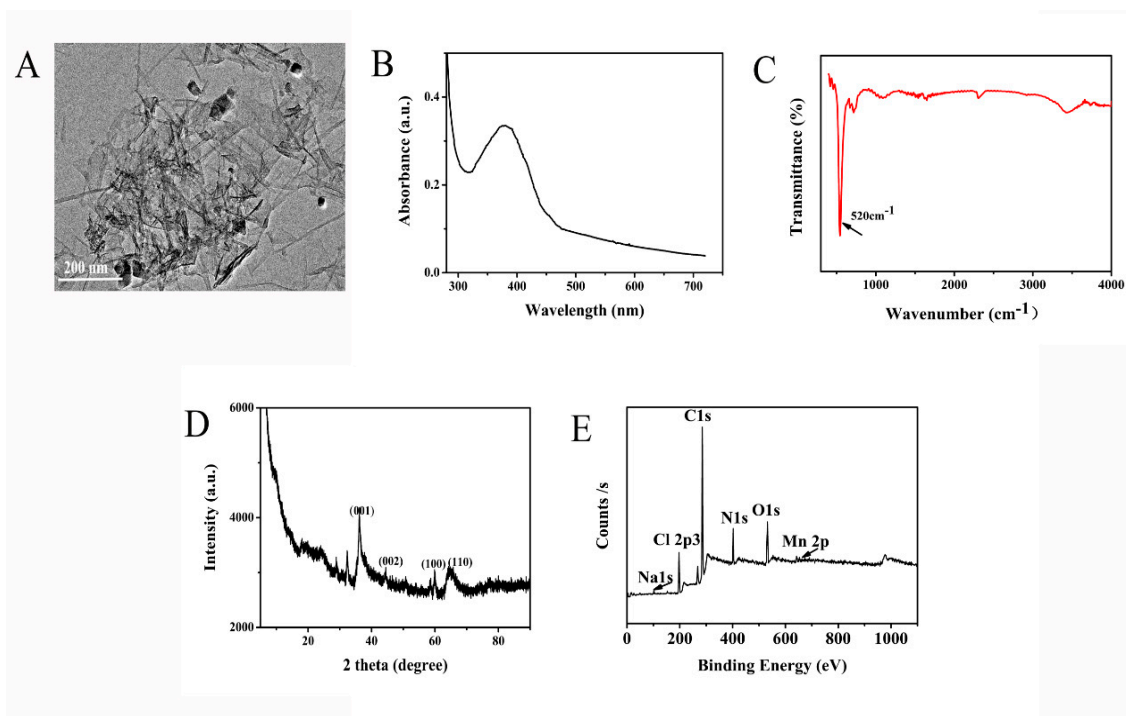


Figure.S2 (A) TEM image of manganese dioxide nanosheets (MnO₂), (B) UV-Vis absorption spectrum of MnO₂, (C) FT-IR spectrum of MnO₂, (D) XRD pattern of MnO₂ and (E) XPS survey spectrum of MnO₂.

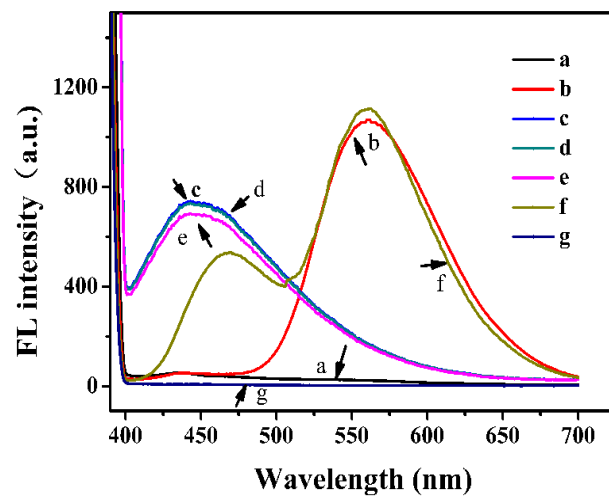


Figure.S3 Fluorescence spectra of various substances (a) MnO_2 , (b) OPD, (c) MnO_2 +OPD, (d) NCDs, (e) MnO_2 +NCDs, (f) MnO_2 +OPD+NCDs

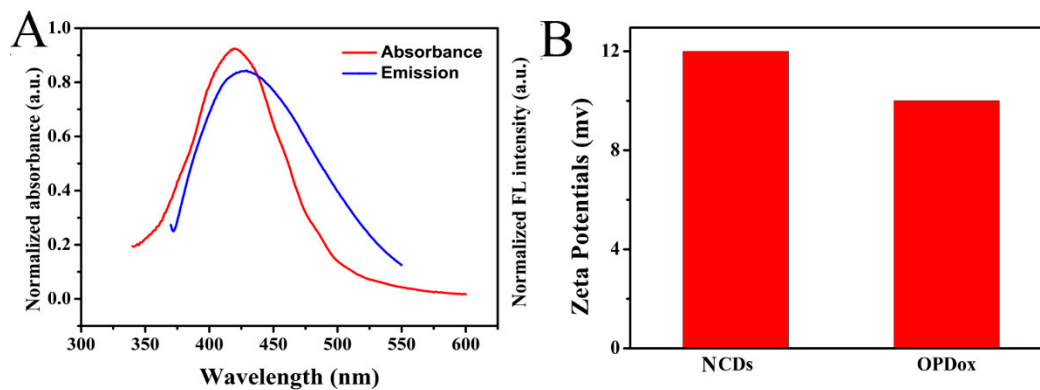


Figure.S4 (A) UV-vis absorption spectrum of OPDox and fluorescence emission spectrum of NCDs (EX=360nm), (B) the potential diagram of OPDox and NCDs

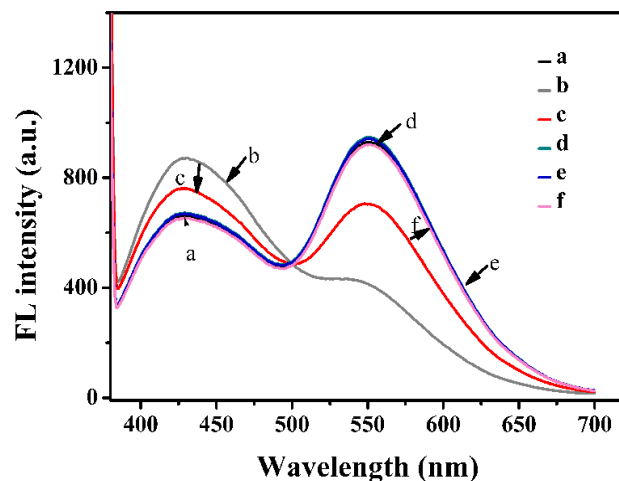


Figure.S5 Fluorescence spectra of ratio fluorescence system before and after adding ALP and 2, 4-D, before: (a) MnO_2 +OPD+NCDs, after: (b) MnO_2 +OPD+NCDs+ALP+AAP, (c) MnO_2 +OPD+NCDs+ALP+AAP+2,4-D, (d) MnO_2 +OPD+NCDs+2,4-D, (e) MnO_2 +OPD+NCDs+AAP, (f) MnO_2 +OPD+NCDs+ALP.

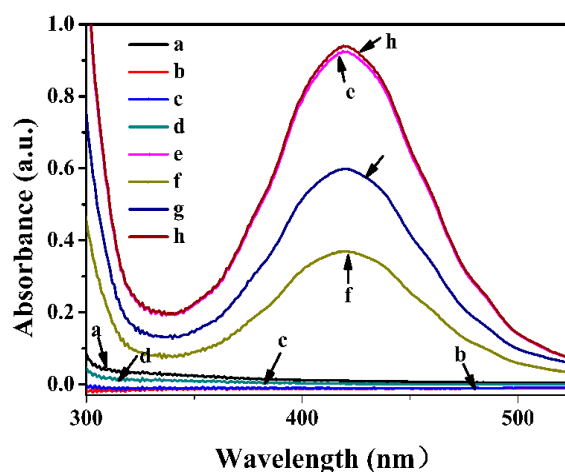


Figure. S6 Uv-vis absorption spectra of various substances (a) OPD, (b) ALP, (c) AAP, (d) ALP+AAP, (e) MnO_2 , (f) MnO_2 +OPD+AAP+ALP, (g) MnO_2 +OPD+AAP+ALP+2,4-D, (h) MnO_2 +OPD+2,4-D

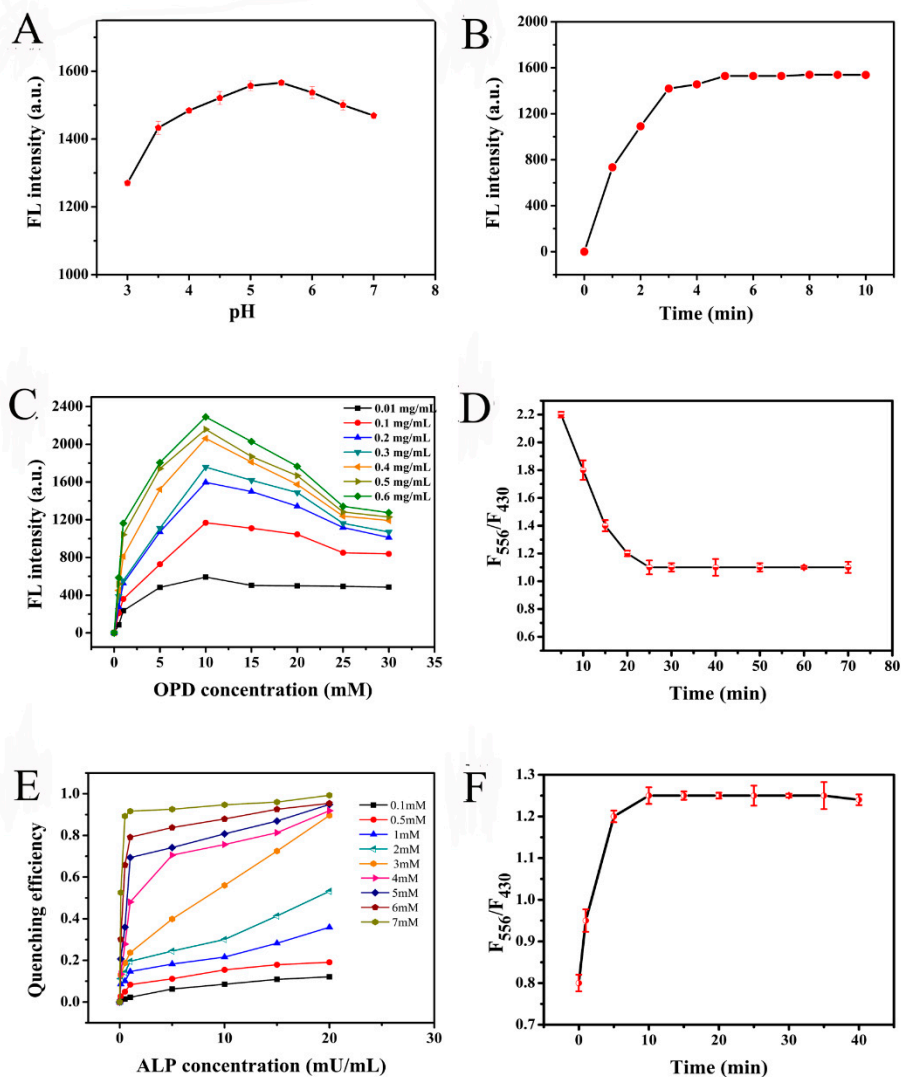


Figure.S7 Effects of (A) OPD-oxidation-related pH, (B) oxidation time, (C) concentrations of OPD and the MnO₂ nanosheets, (D) ALP hydrolysis time, (E) concentration of the hydrolyzed AAP substrate, and (F) time required for enzyme inhibition by 2,4-D

Table S1 Comparison of the 2,4-D sensor reported herein with similar previously reported sensors

Method	Linear range	Detection limit	Type	Reference
Electrochemistry	100–1000 nM	34 nM	Boron-doped diamond electrode	(Neto ,Oliveira,& Suarez,2022)

Fluorescence	0–5 $\mu\text{g L}^{-1}$	50 ng mL^{-1}	Gold nanobipyramids sensors	(Ye, Zhang,& Wang, 2022)
Electrochemistry	0.1–1 mg L^{-1}	50 $\mu\text{g L}^{-1}$	Prussian blue nanoparticles	(Arduini, Moscone, 2019)
Electrochemistry	1.4–2.7 μM	0.21 μM	Fe_3O_4 -polyaniline nanocomposite	(Goswami ,Mahanta, 2021)
Ratiometric fluorescence	0.05–30 $\mu\text{g/mL}$	0.013 $\mu\text{g/mL}$	MnO_2 -NCDs-OPD	Present study

References

Neto J C D, Dos Santos V B, de Oliveira S C B, Suarez W T, de Oliveira J L (2022) In situ voltammetric analysis of 2, 4-dichlorophenoxyacetic acid in environmental water using a boron doped diamond electrode and an adapted unmanned air vehicle sampling platform. *Anal. Methods* 14(13):1311-1319

Ye X, F, Yang L, Yang W, Zhang L, Wang Z (2022) Paper-based multicolor sensor for on-site quantitative detection of 2,4-dichlorophenoxyacetic acid based on alkaline phosphatase-mediated gold nanobipyramids growth and colorimeter-assisted method for quantifying color. *Talanta* 245:123489

Arduini F, Cinti S, Caratelli V, Amendola L, Palleschi G, Moscone D (2019) Origami multiple paper-based electrochemical biosensors for pesticide detection. *Biosens. Bioelectron.* 126:346-354

Goswami B, Mahanta D (2021) Fe_3O_4 -Polyaniline Nanocomposite for Non-enzymatic Electrochemical Detection of 2,4-Dichlorophenoxyacetic Acid. *ACS Omega* 6(27):17239-17246