

CoMnO_x Nanoflower-Based Smartphone Sensing Platform and Virtual Reality Display for Colorimetric Detection of Ziram and Cu²⁺

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Reagents and Chemicals

Cobaltous nitrate (Co(NO₃)₂), hydrogen peroxide (H₂O₂, 30%), tert-butanol (TBA), sodium azide (NaN₃), potassium permanganate (KMnO₄), dimethyl sulfoxide (DMSO), ethylenediamine tetraacetic acid disodium salt (EDTA), riboflavin, *p*-benzoquinone (PBQ), and ethanol (CH₃CH₂OH) were supplied by KeLong (Chengdu, China, <http://www.cdkelong.com/>). Ziram was purchased from Macklin Biochemical Co. Ltd. (Shanghai, China, <http://www.macklin.cn/>). 3,3',5,5'-Tetramethylbenzidine (TMB) was obtained from Shanghai Ryon Biological Technology Co. Ltd. (Shanghai, China, <http://ruien.company.lookchem.cn/>). All the chemicals used in experiments were analytically pure. All experiments used deionized water (18.25 MΩ·cm⁻¹).

Characterization Equipment

The crystal structure, morphological structure, elemental composition and surface functional groups were analyzed by X-ray diffraction (XRD, DX-2700, Dandong, China), transmission electron microscope (TEM, JSM4800F, JEOL, Japan), scanning electron microscope (SEM, JEOL-2100F, Japan), X-ray photoelectron spectroscope

(XPS, ESCALAB-250Xi, China), and Fourier transform infrared spectrometer (FTIR, Bruker Vertex 70, China), respectively.

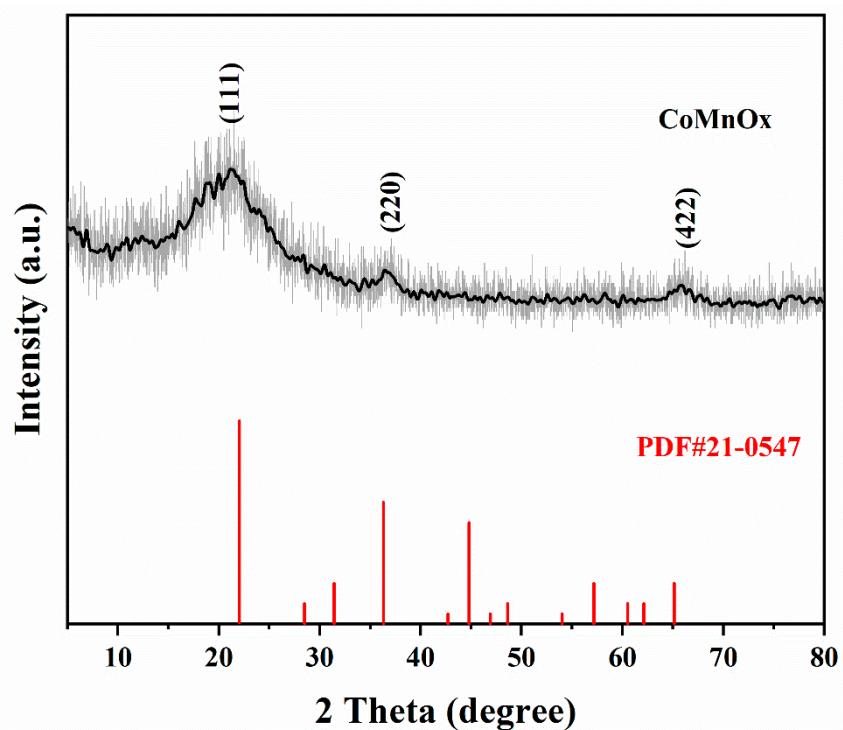


Figure S1. XRD image of CoMnO_x.

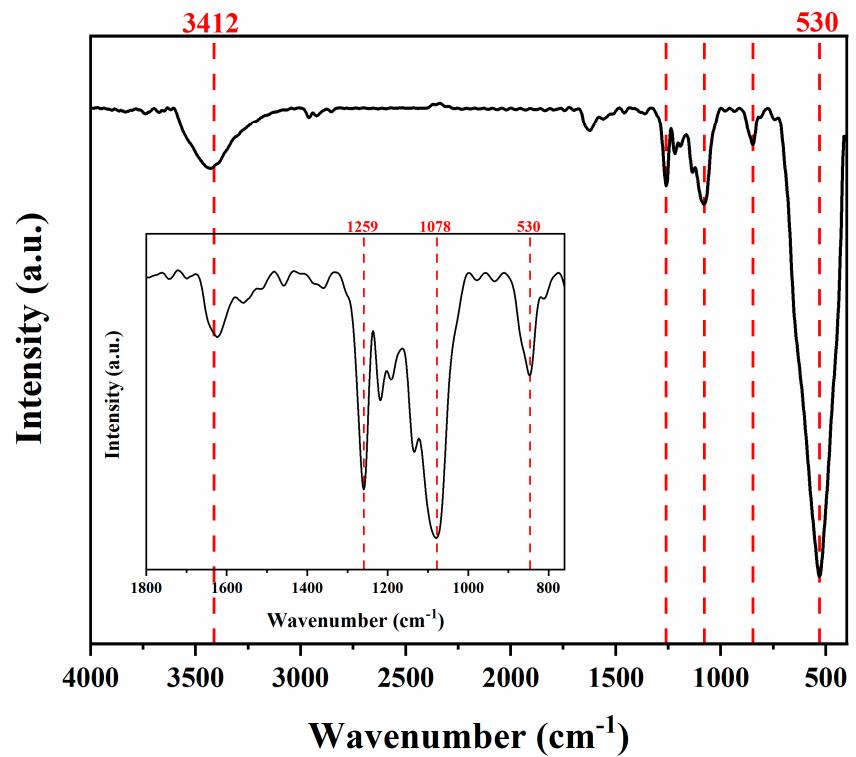


Figure S2. FTIR spectrum of CoMnO_x.

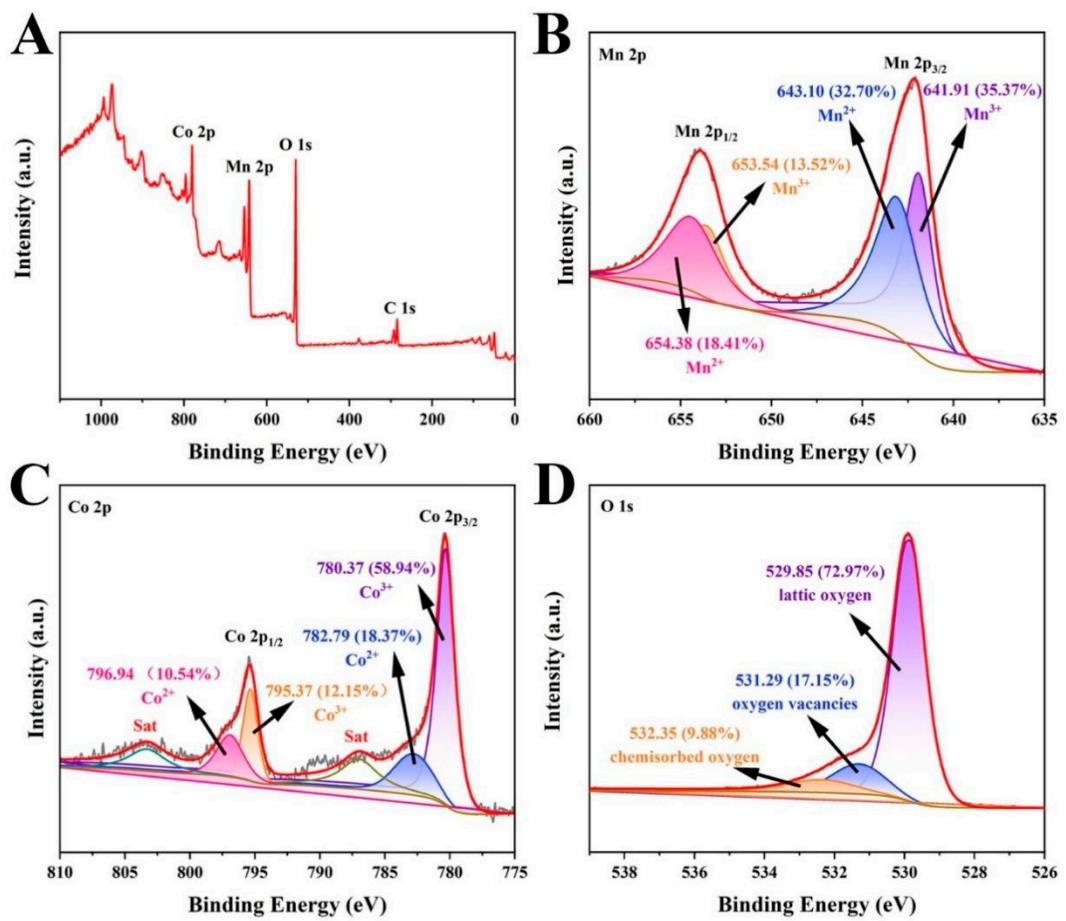


Figure S3. (A) XPS full spectrum; (B) Mn 2p, (C) Co 2p and (D) O 1s spectra of CoMnO_x .

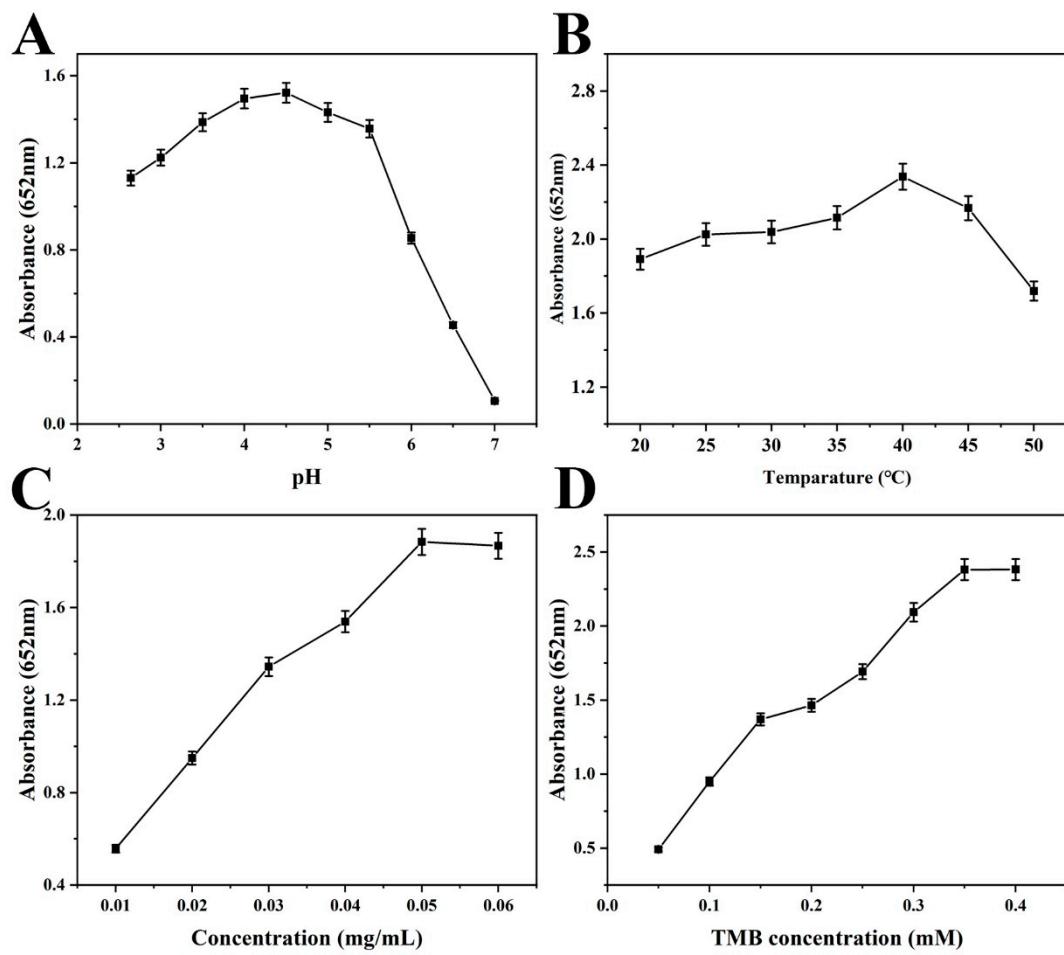


Figure S4. Optimization of reaction conditions for CoMnO_x oxidase-like activity. (A) pH; (B) Temperature; (C) Material concentration; (D) TMB concentration ($n=3$).

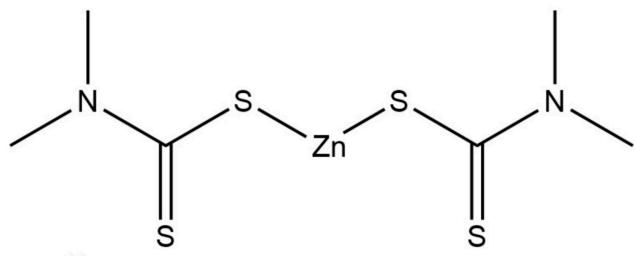


Figure S5. Chemical structure of Ziram

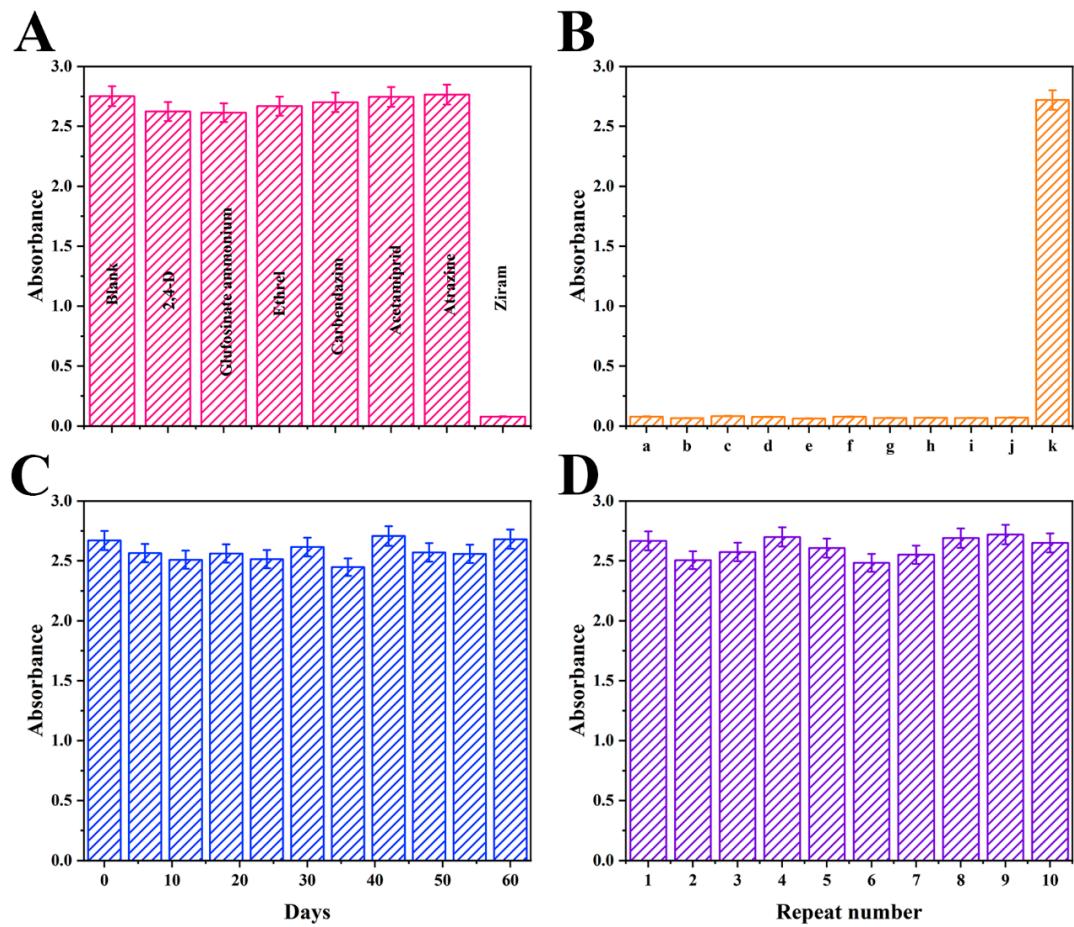


Figure S6. Potential interferences of other substances for the detection of (A) Ziram and (B) Cu^{2+} (a: control, b: Na^+ , c: Mn^{2+} , d: Mg^{2+} , e: Zn^{2+} , f: Ca^{2+} , g: K^+ , h: Al^{3+} , i: Pb^{2+} , j: Cd^{2+} , k: Cu^{2+}); (C) Sensor stability; (D) Reproducibility of our method ($n=3$).

Table S1. Comparison of steady-state kinetic parameters for the oxidase-like activity of CoMnO_x and other nanozymes.

Nanozyme	<i>K_m</i> (mM)	<i>V_{max}</i> (μM·s ⁻¹)	Reference
Co ₃ O ₄	0.0469	0.0459	Error! Reference source not found. Error!
Co ₃ O ₄ NPs	0.051	0.033	Reference source not found. Error!
Co-Fe LDH	0.218	0.263	Reference source not found. Error!
Co ₃ O ₄ /rGO	0.19	0.1071	Reference source not found. Error!
MnO ₂ nanosticks	0.147	1.34	Reference source not found. Error!
FO@ZMFO@FM-MOG	0.291	0.149	Reference source not found. Error!
MOF(Co/2Fe)	0.199	0.39	Reference source not found. Error!
Fe ₃ O ₄ @MnO ₂	0.12	—	Reference source not found. Error!
FeCo NPs@PNC	0.091	0.0939	Reference source not found. Error!
CNF/MnCo ₂ O _{4.5}	0.04	0.0645	Reference source not found.

CoMnO _x	0.0022	0.1809	This work
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Table S2. XPS analysis results of Mn 2p and Co 2p.

Sample	Mn ²⁺ /Mn ³⁺	Co ²⁺ /Co ³⁺
CoMnO _x	1.054	0.407
CoMnO _x +TMB	1.001	0.726

Table S3. XPS analysis results of O 1s.

Sample	Chemisorbed oxygen	Oxygen vacancy	Lattice oxygen
CoMnO _x	9.88%	17.15%	72.97%
CoMnO _x +TMB	9.26%	26.57%	64.17%

Table S4. Comparison of different methods for the detection of Ziram.

Material	Detection method	Linear range (μM)	LOD (μM)	Reference
p-AgSAE	Voltammetry	0.5-5	0.24	Error! Reference source not found. Error!
Carbon dots	Fluorometry	0-16	1.80	Reference source not found.
CsPbBr ₃	Fluorometry	0.32-163.49	0.28	Error!

				Reference source not found. Error!
Oct/Au@AgNPs	SERS ^a	0.33-32.70	0.050	Reference source not found. Error!
Silver nanoparticles	Colorimetry	0.64-2.4	0.49	Reference source not found. Error!
Pt/Co ₃ O ₄	Colorimetry	5-50	3.36	Reference source not found. Error!
Copper foam	SEIRAS ^b	60-1000	58.87	Reference source not found. Error!
AuNPs	Fluorometry	0.052-0.50	0.052	Reference source not found.
CoMnO _x	Colorimetry	5-280	1.475	This work

a: surface enhanced Raman spectroscopy

b: surface enhanced infrared absorption spectroscopy

Table S5. Comparison of different methods for the detection of Cu²⁺.

Material	Detection method	Linear range (μM)	LOD (μM)	Reference
Chol-RN	Fluorometry	5-40	1.12	Error! Reference source not found. Error!
Rhodamine-based derivative	Fluorometry	10-300	3.42	Reference source not found. Error!
Furfuraldehyde fluorescein hydrazone	Colorimetry	6.6-330	6.6	Reference source not

				found.
	Fluorometry			Error!
Rhodamine B hydrazone		2.63-200	2.63	Reference source not found.
	Fluorometry			Error!
Cu ₂ (BDC) ₂ (BPY)		50-250	0.005	Reference source not found.
	Fluorometry			Error!
Ferrocene benzobisimidazole		0-10	1.6	Reference source not found.
Rhodamine-based fluorescent probe	Fluorometry	2-20	0.1	Reference source not found.
Flu@Mea-CdS	Fluorometry	4-14	0.17	Reference source not found.
Cu ²⁺ -triethylamine	Colorimetry	0.0625-8.0	0.0625	Reference source not found.
CoMnO _x	Colorimetry	80-360	3.906	This work

Table S6. Assay results of Ziram in soil and water samples (*n*=3).

Sample	Add (μM)	Determined (μM)	RSD (%)	Recovery (%)
Soil	0	0.87	1.04	—
	60	59.28	2.23	98.81
	90	91.03	2.70	101.14
	120	122.61	3.46	102.18

	0	0.87	1.04	—
Water	50	48.72	3.67	97.44
	100	100.84	1.60	100.84
	150	144.92	2.85	96.60

Table S7. Assay results of Cu²⁺ in soil and water samples (*n*=3).

Sample	Add (μM)	Determined (μM)	RSD (%)	Recovery (%)
Soil	0	0.94	1.21	—
	60	59.61	0.57	99.35
	90	89.33	2.16	99.25
	120	118.48	4.69	98.73
Water	0	0.94	1.21	—
	50	49.95	0.52	99.90
	100	100.42	1.42	100.42
	150	150.26	2.11	100.18